

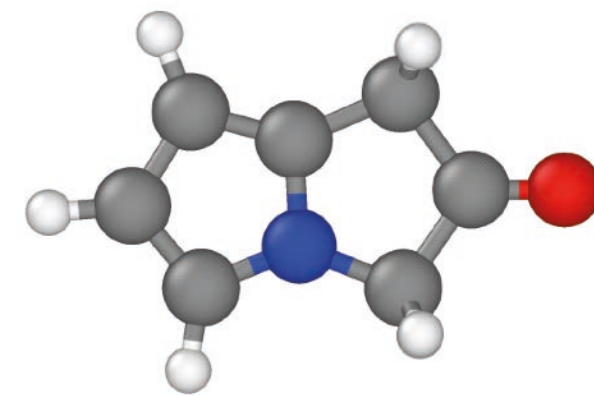
Data-driven approaches to atomistic modeling beyond interatomic potentials

Chiheb Ben Mahmoud

NHR4CES Community Workshop 2025 — Online — 07.05.2025

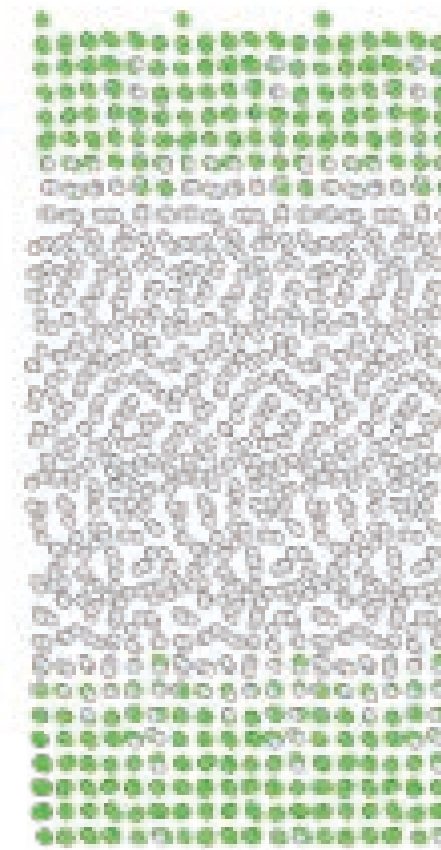
QM results at a fraction of the cost

1/accuracy



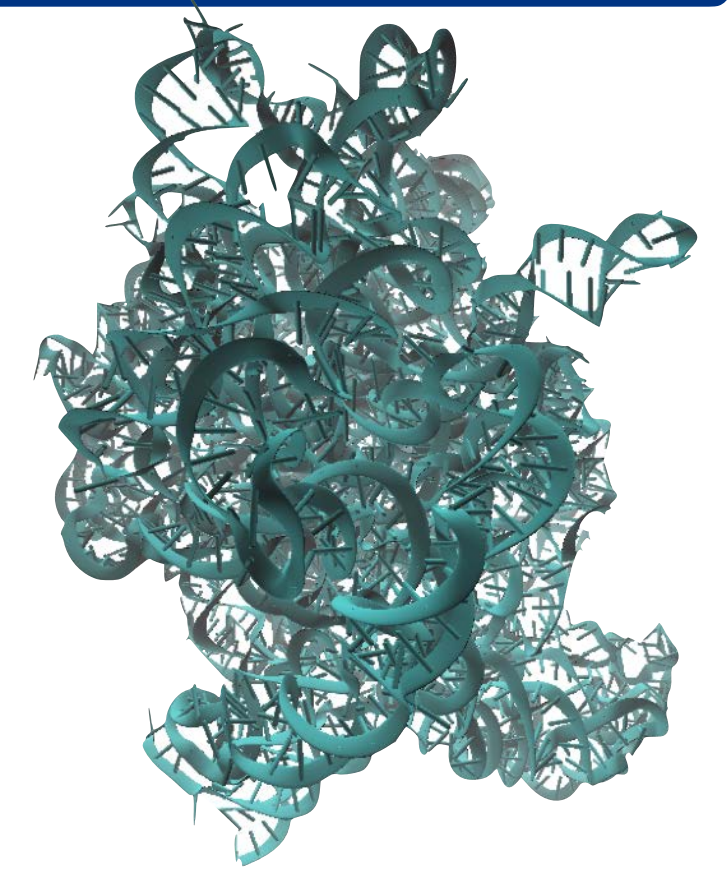
DFT

Quantum
chemistry



Empirical
potentials

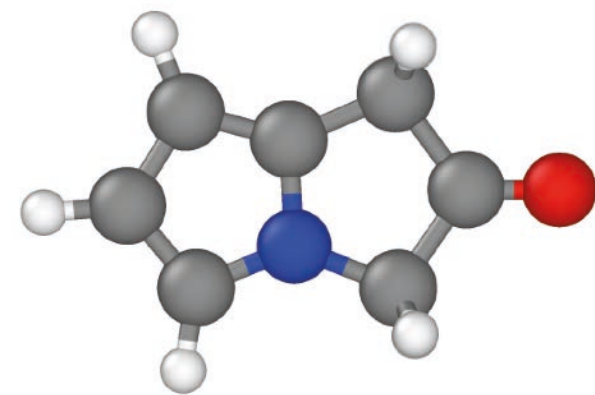
Coarse
grained
models



length scale

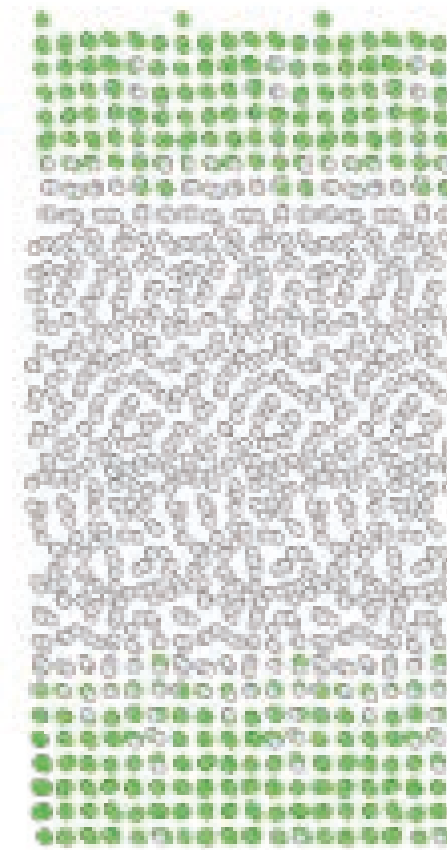
QM results at a fraction of the cost

1/accuracy



DFT

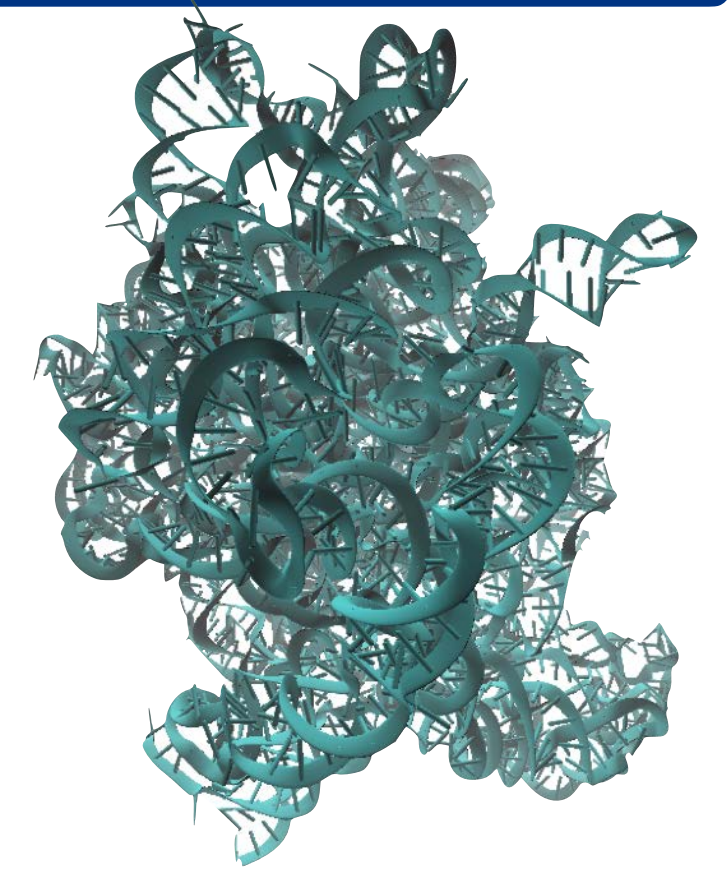
Quantum
chemistry



Empirical
potentials

Machine learning
interatomic
potentials

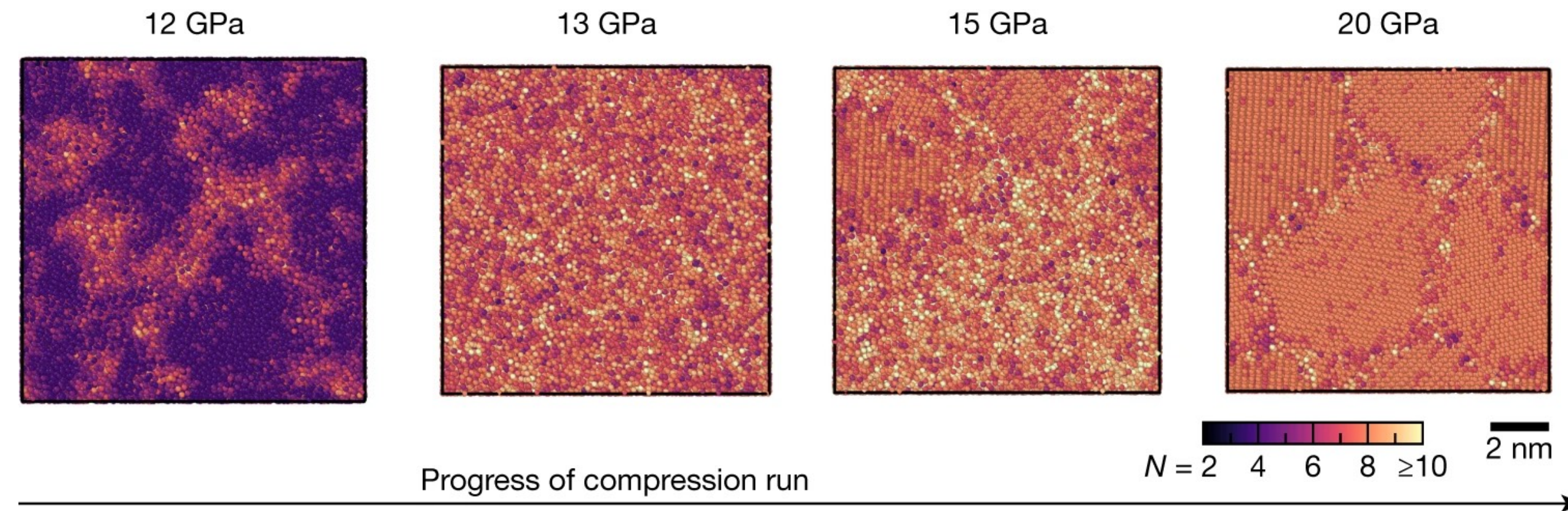
Coarse
grained
models



length scale

QM results at a fraction of the cost

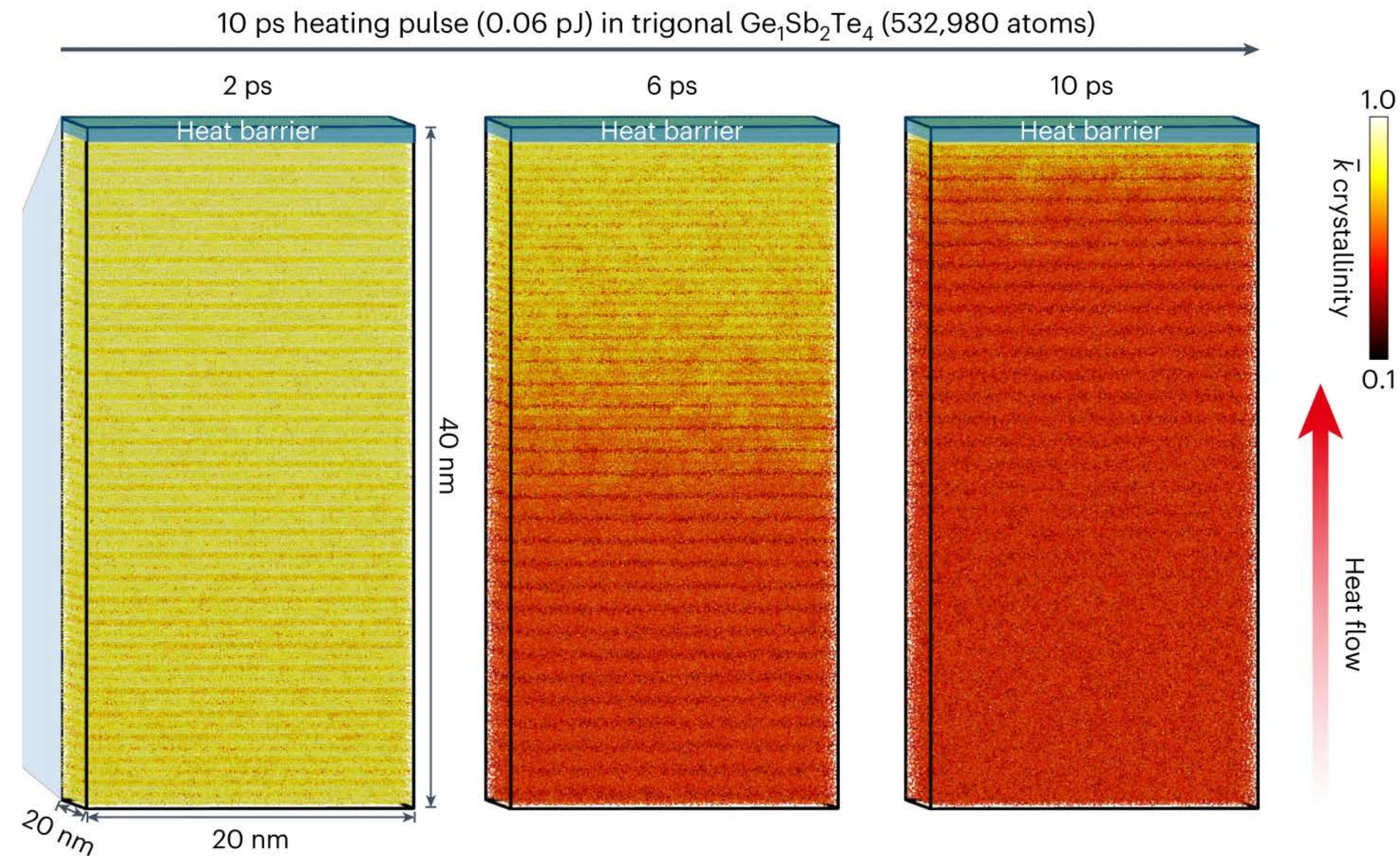
Amorphous Si compression -
100,000 atoms



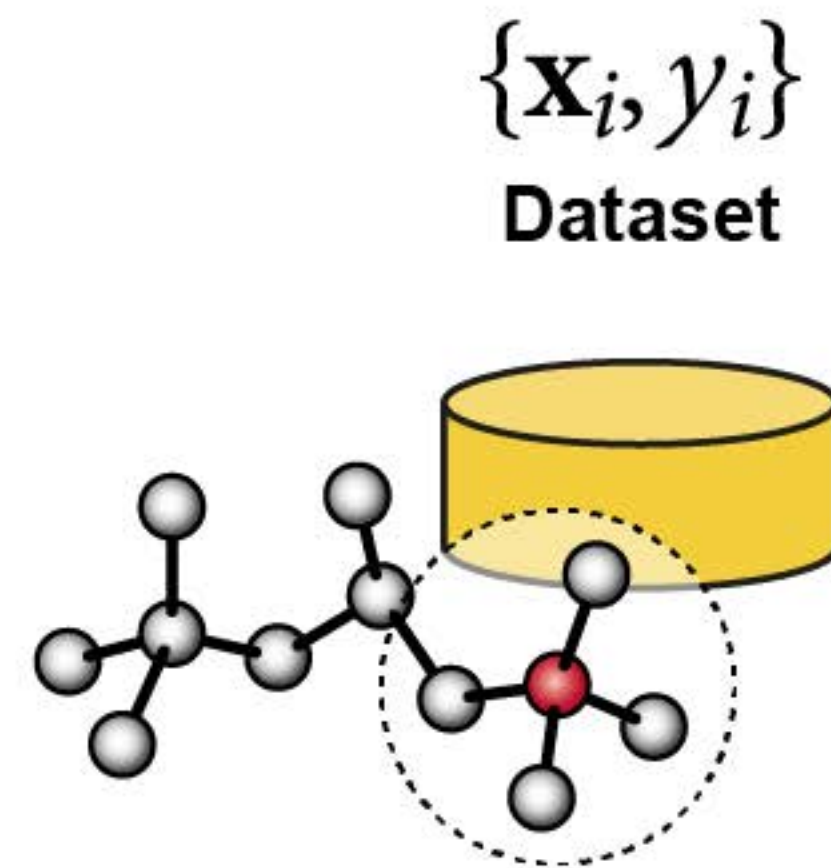
Deringer et al, 2021

QM results at a fraction of the cost

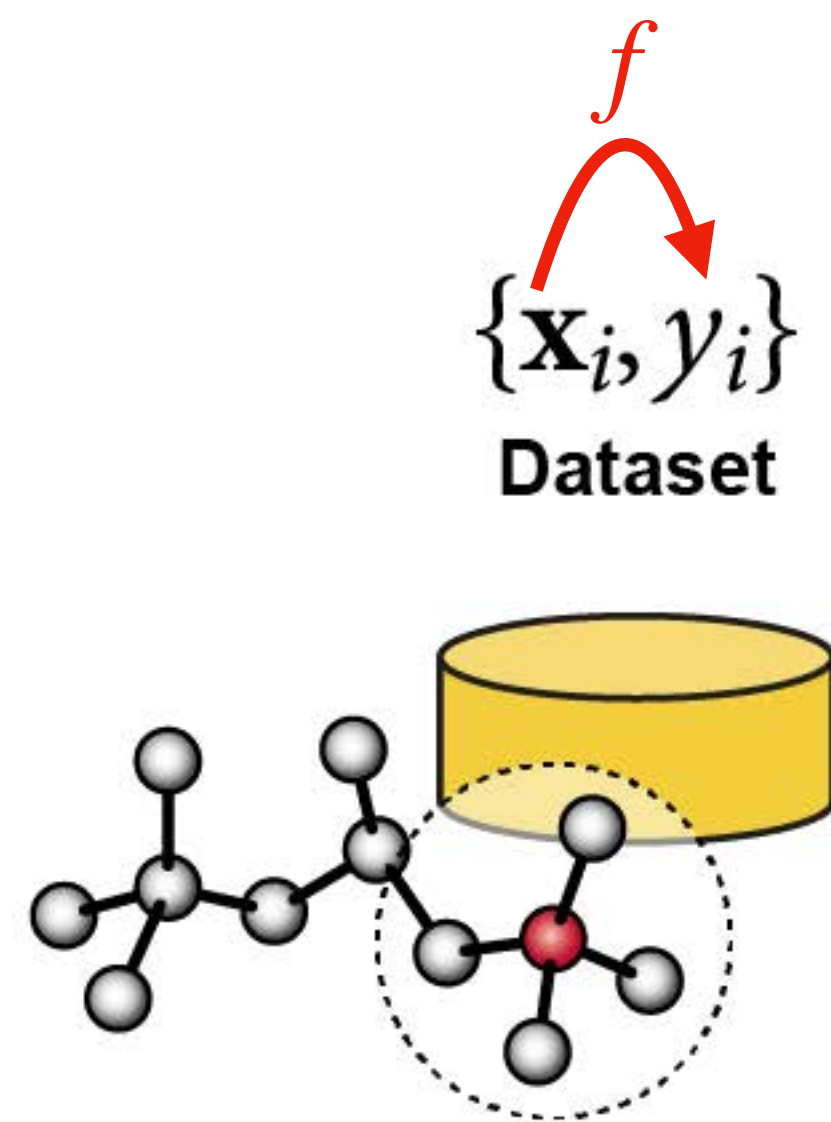
Heat flow in phase change materials - ~500,000 atoms



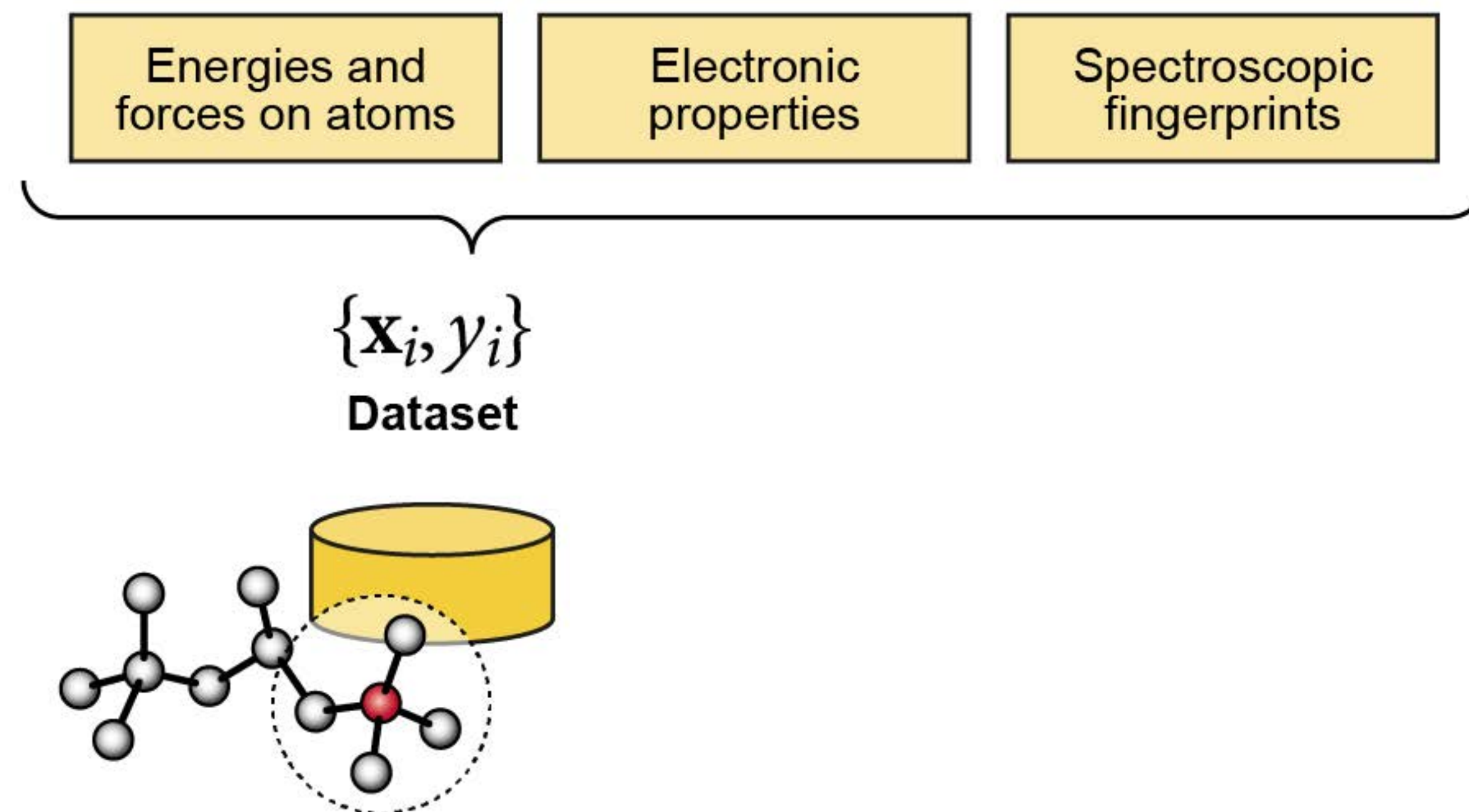
Important atomistic ML ingredients



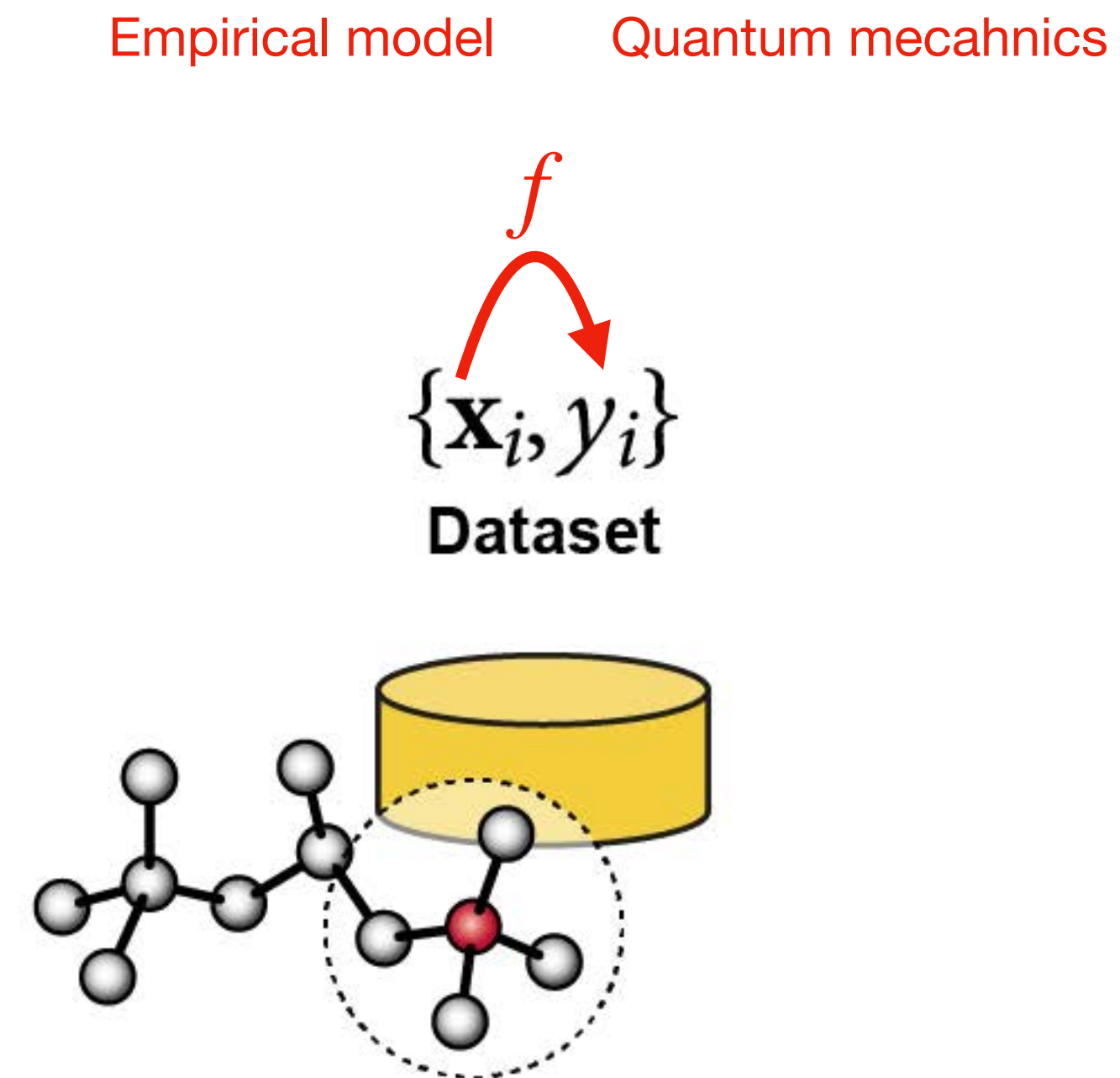
Important atomistic ML ingredients



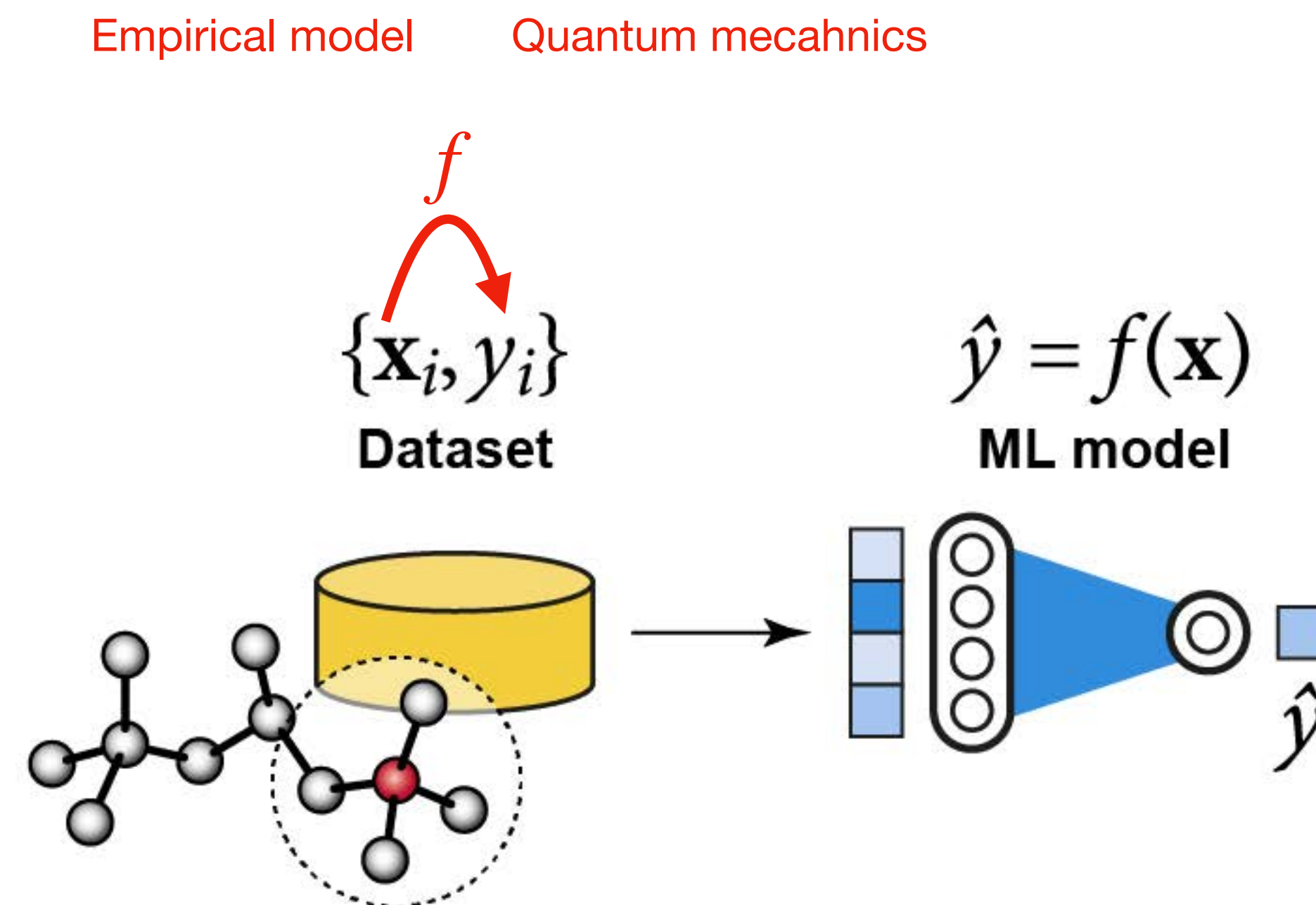
Important atomistic ML ingredients



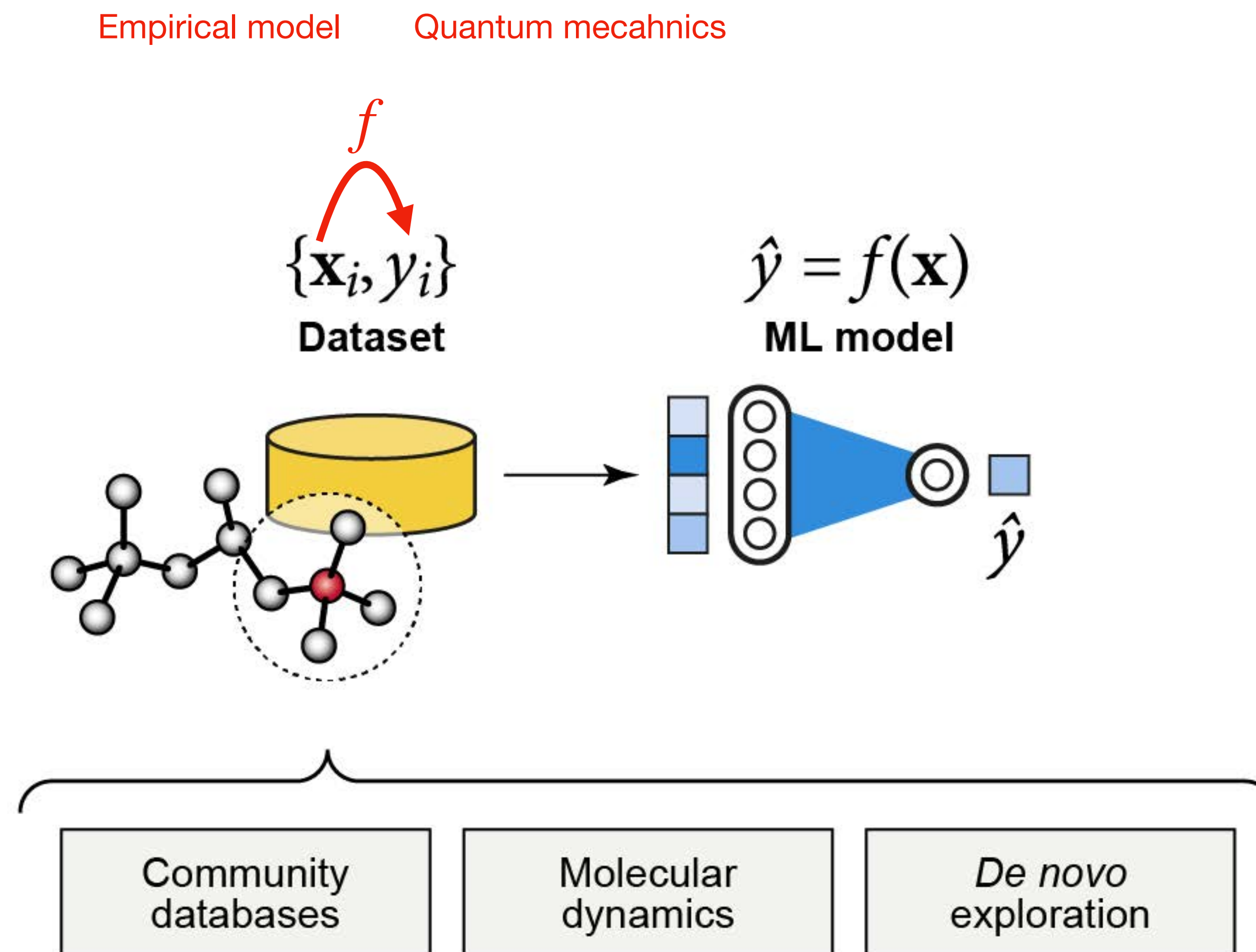
Important atomistic ML ingredients



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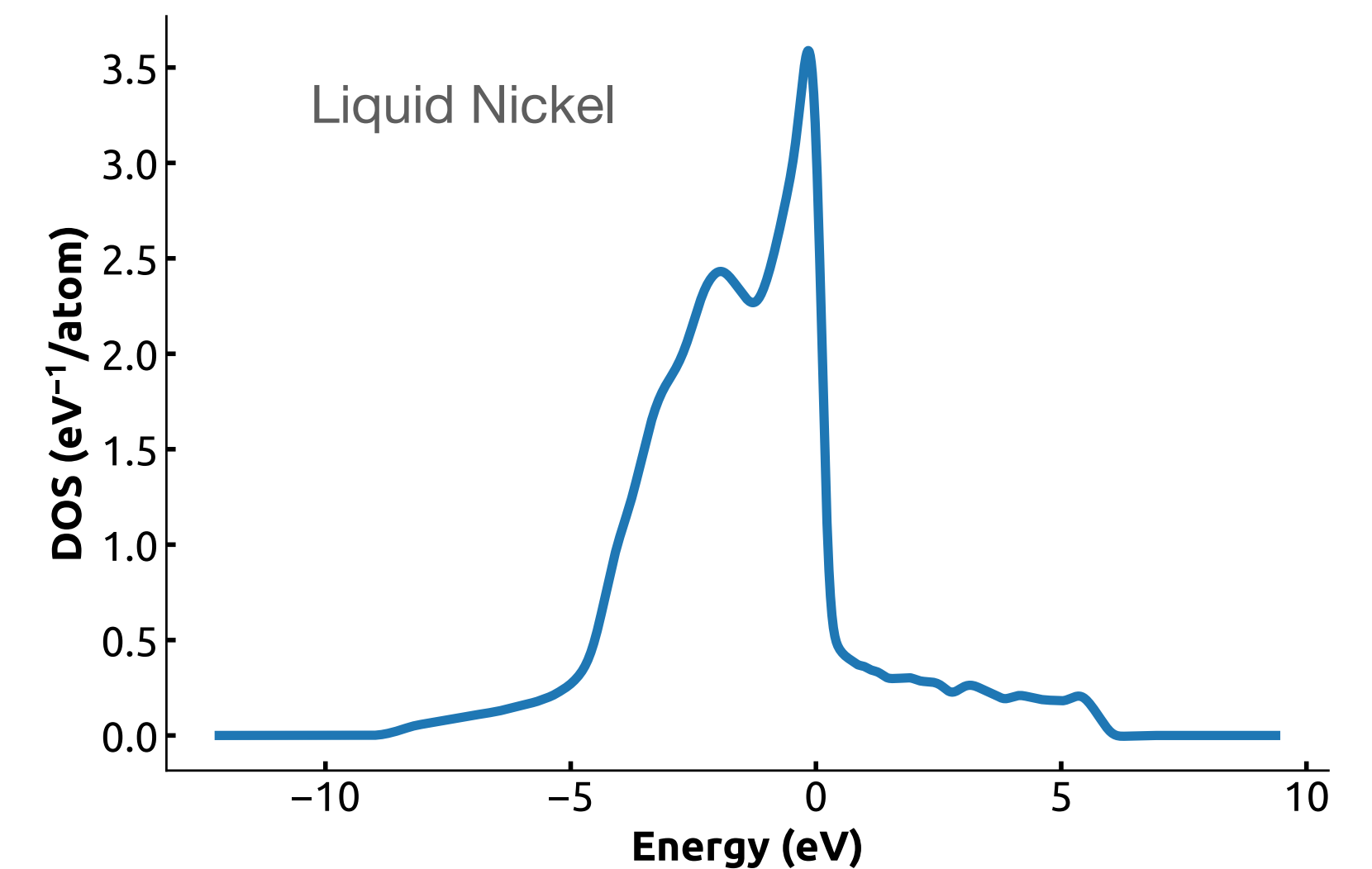
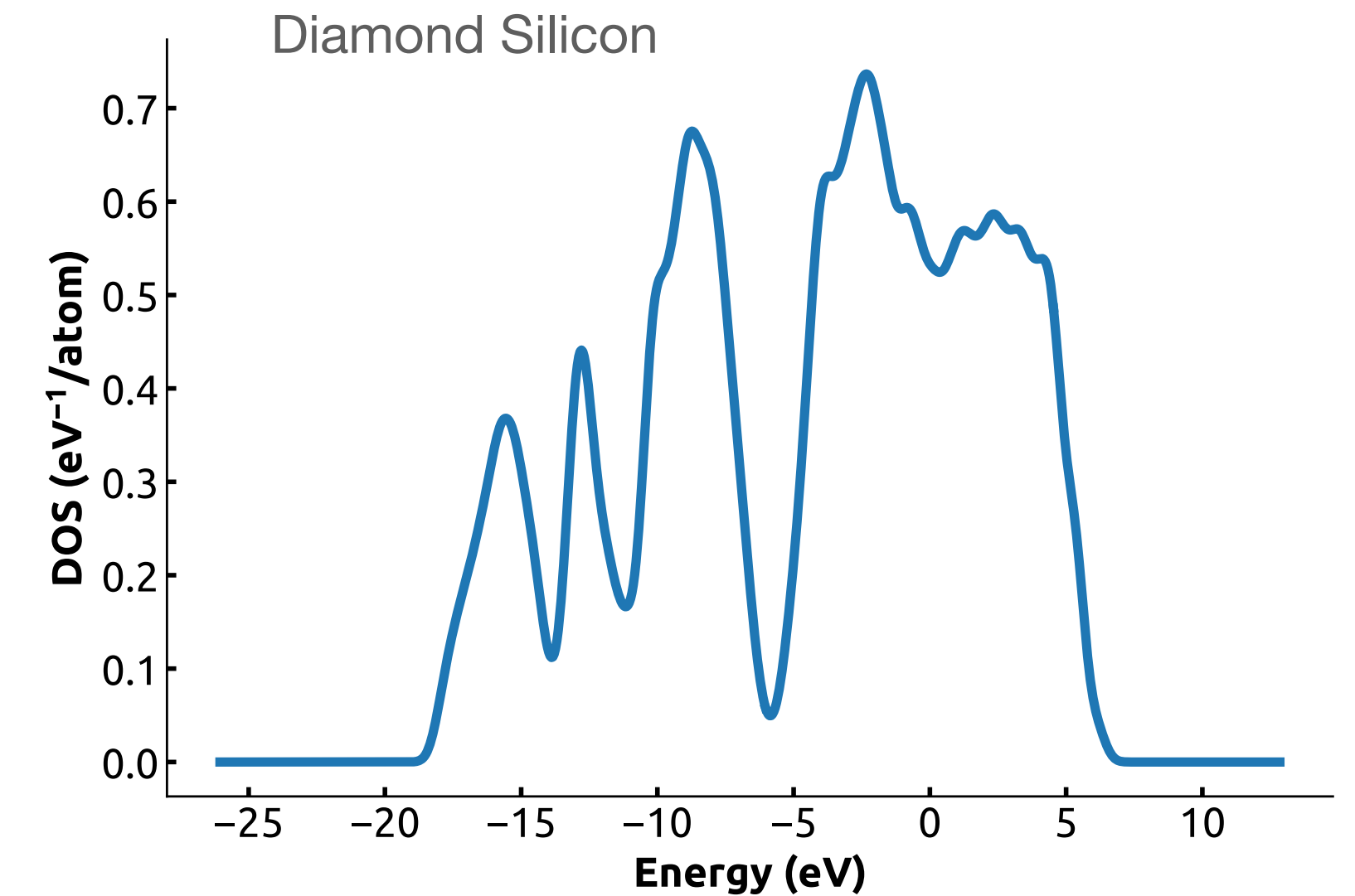
Important atomistic ML ingredients



Property 1:
electronic density of states

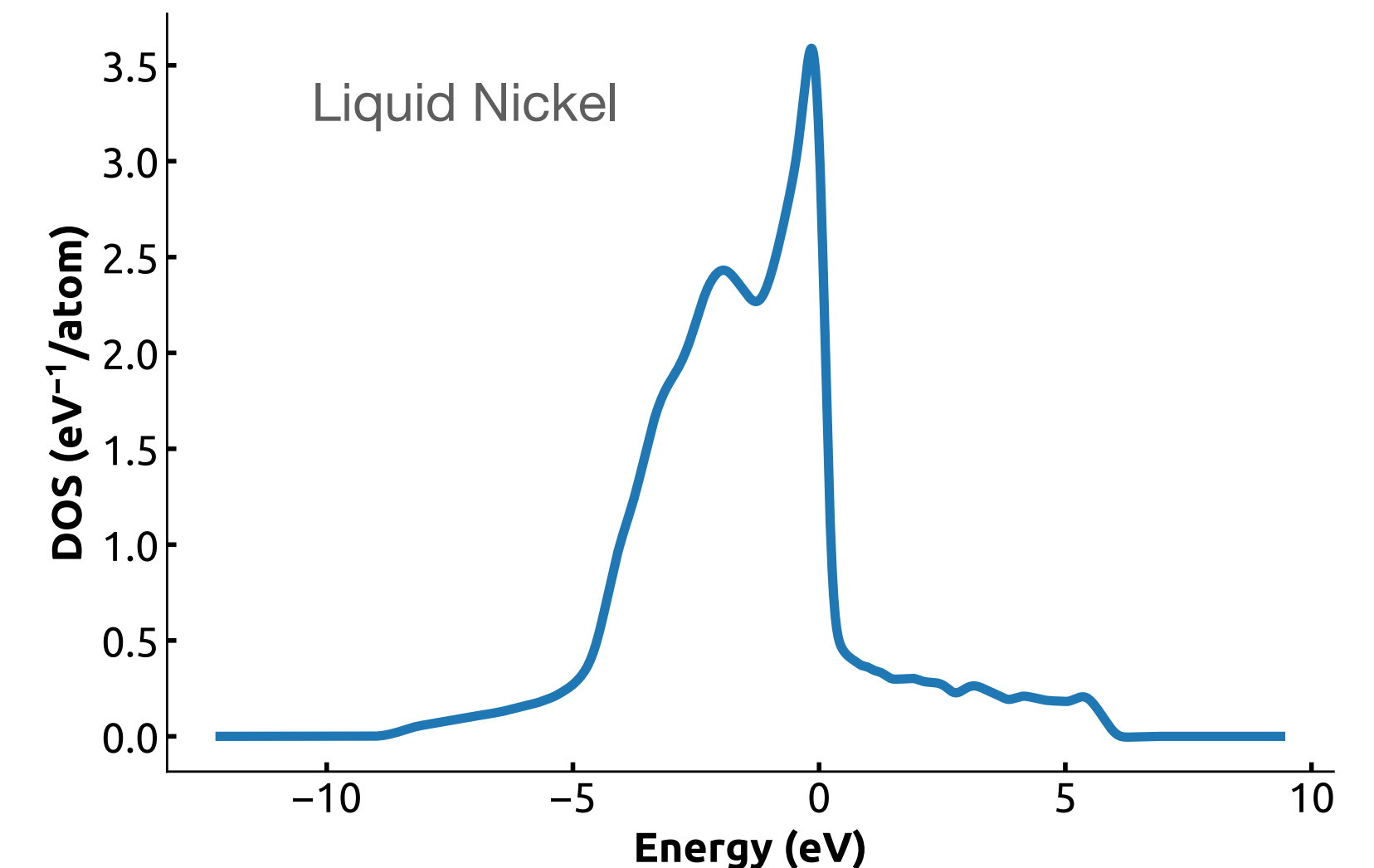
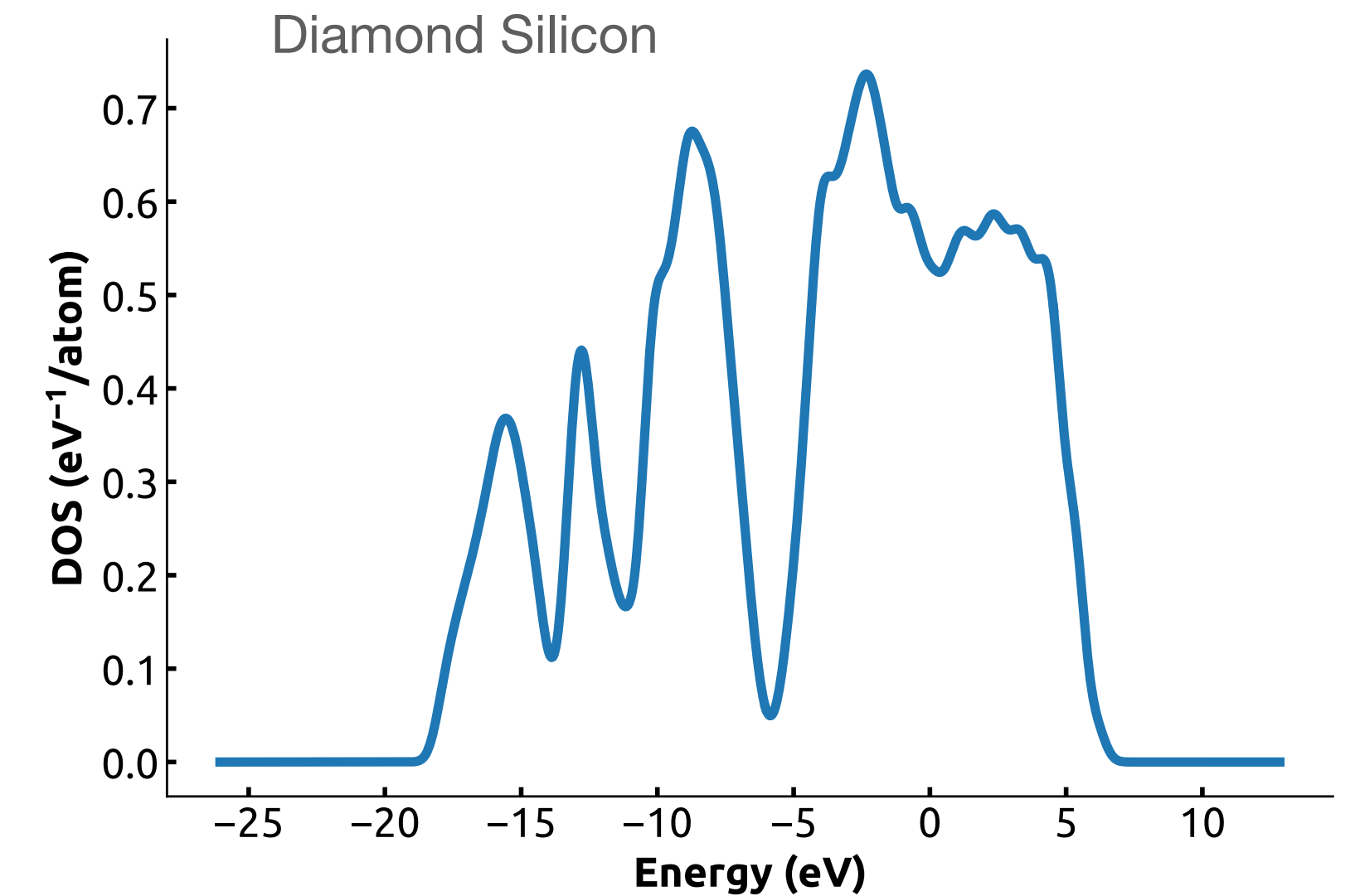
Property 1: Electronic density of states

- The electronic density of states (DOS) is useful for:



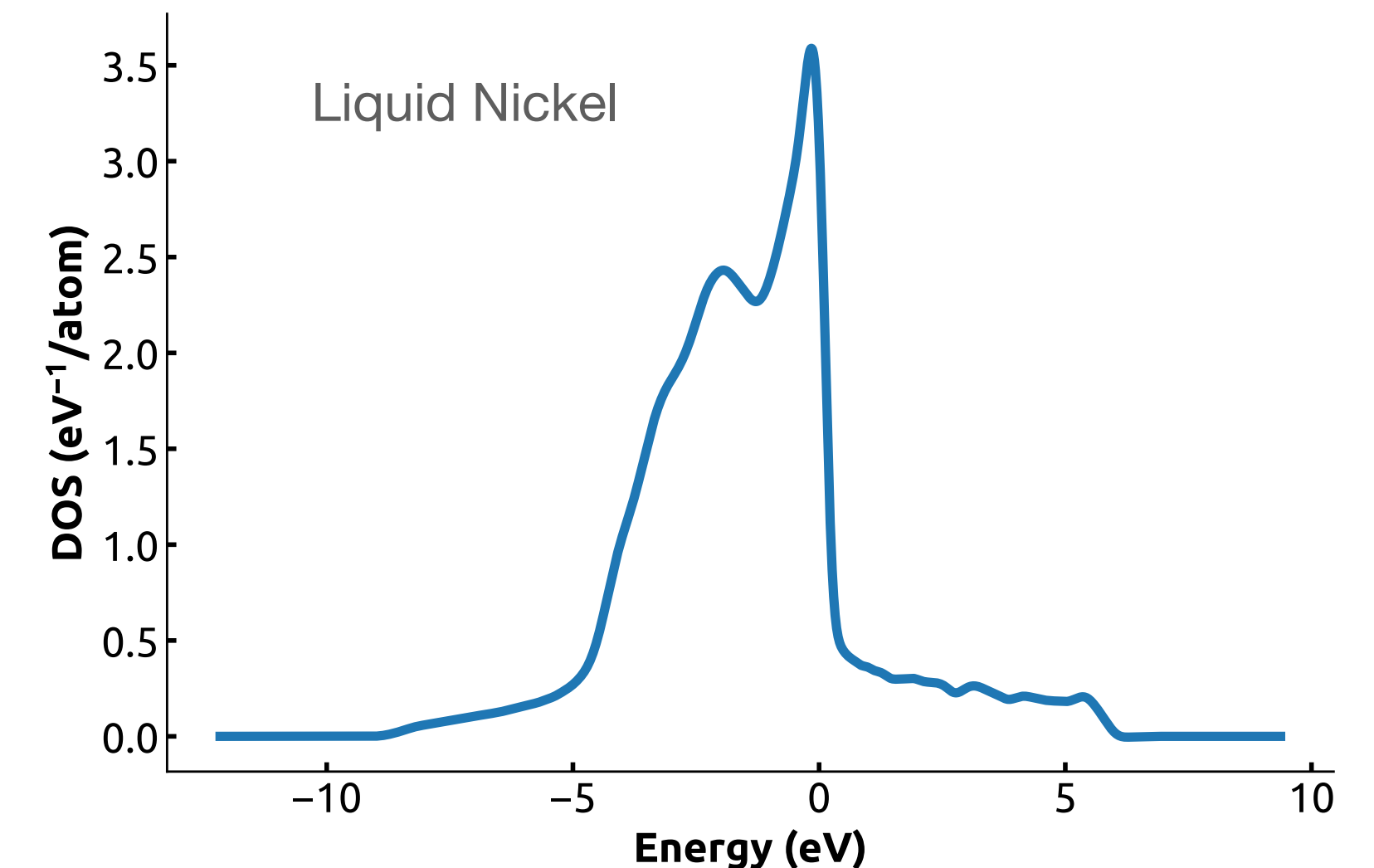
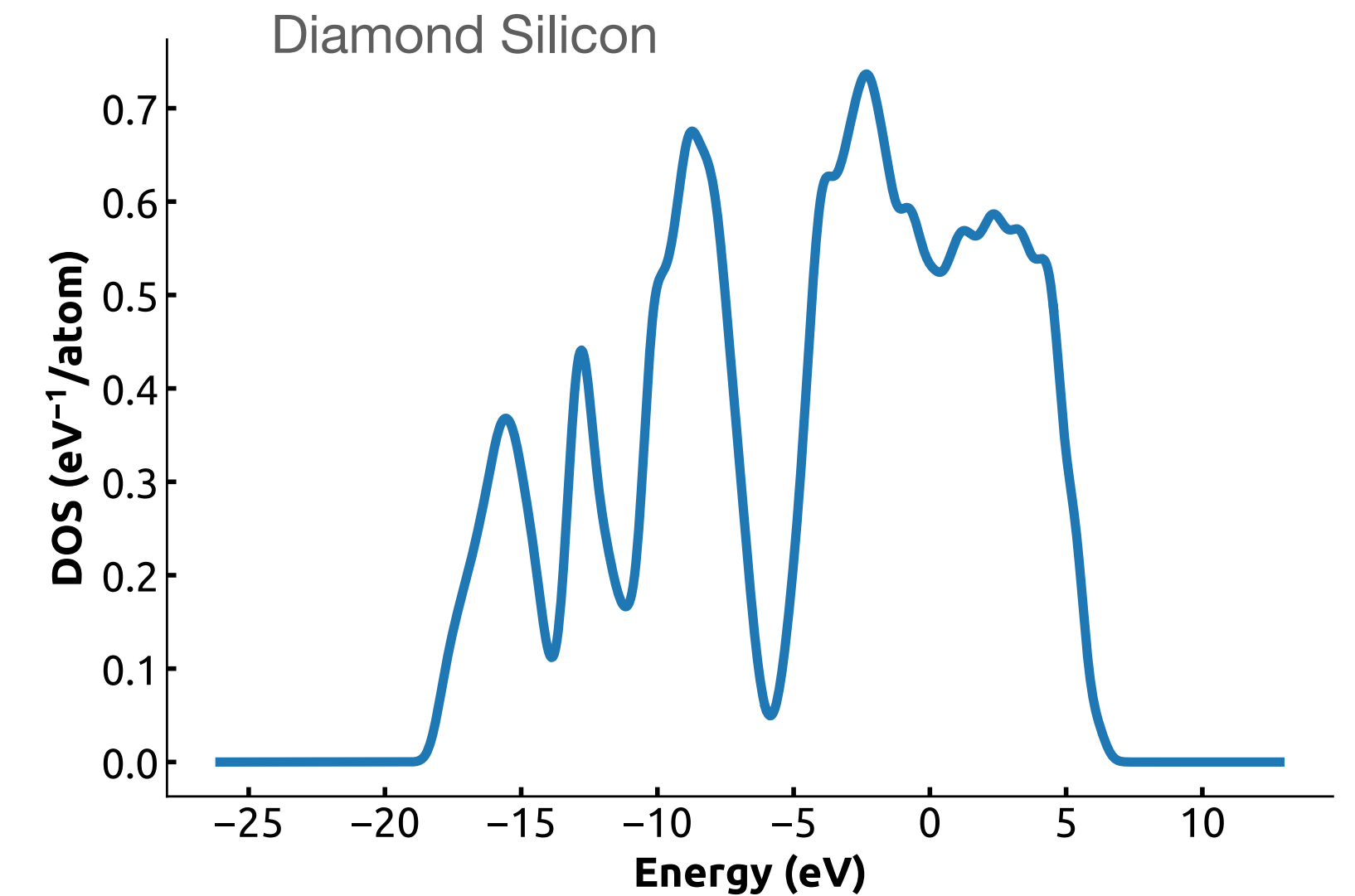
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 - **optical properties:** band gap and joint density of states



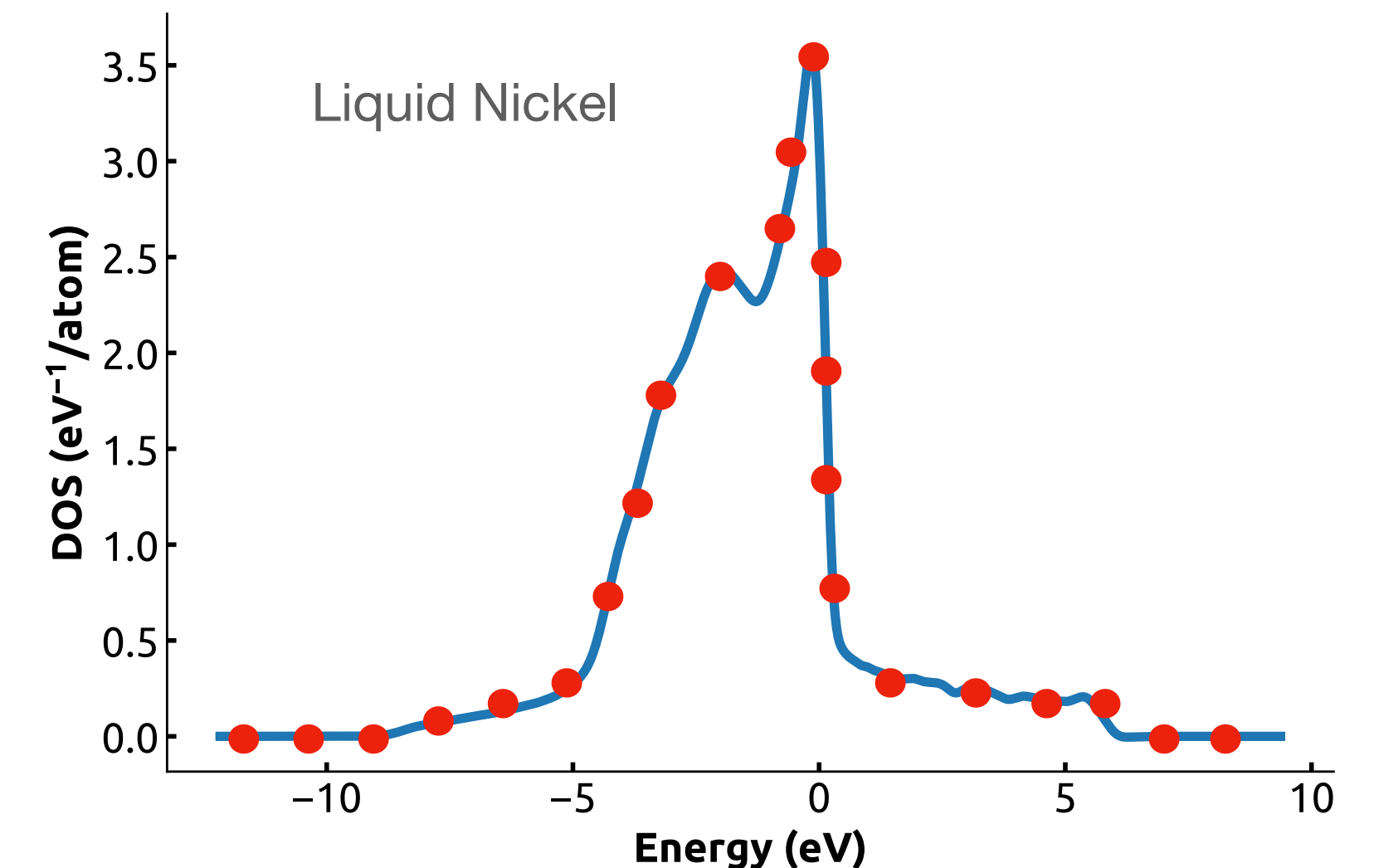
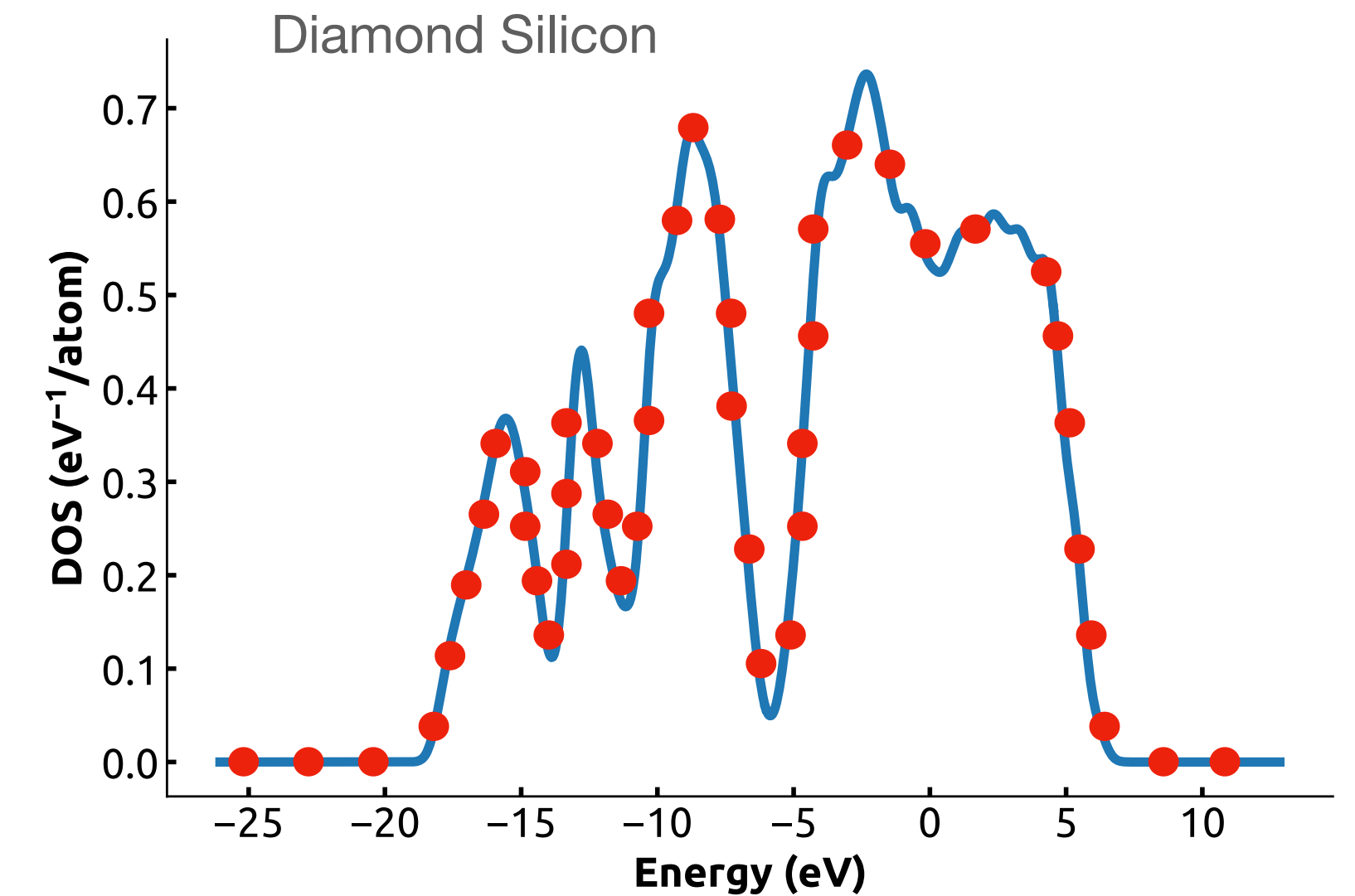
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Objectives

- Build a **transferable ML model** for the DOS
- Identify the **challenges** in modelling and interpreting the DOS
- Apply the ML DOS models to study :
 - the **feature/property** relations
 - thermal properties of materials in physics-based approximations

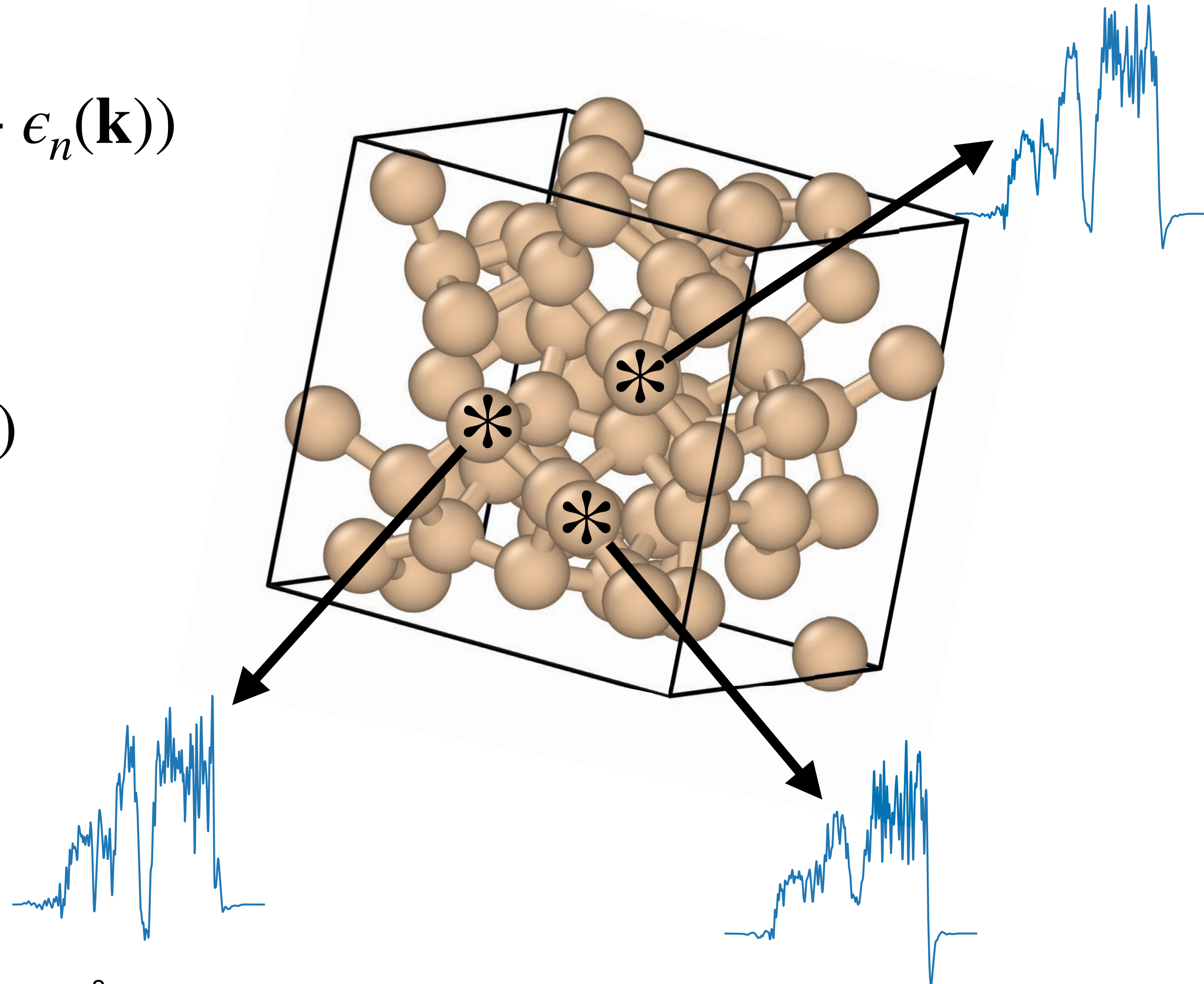
Atom-centred model for the DOS

- $$\text{DOS}(\mathcal{A}, E) = \frac{2}{N_k} \sum_n^{\text{bands}} \sum_{\mathbf{k}} \tilde{\delta}(E - \epsilon_n(\mathbf{k}))$$

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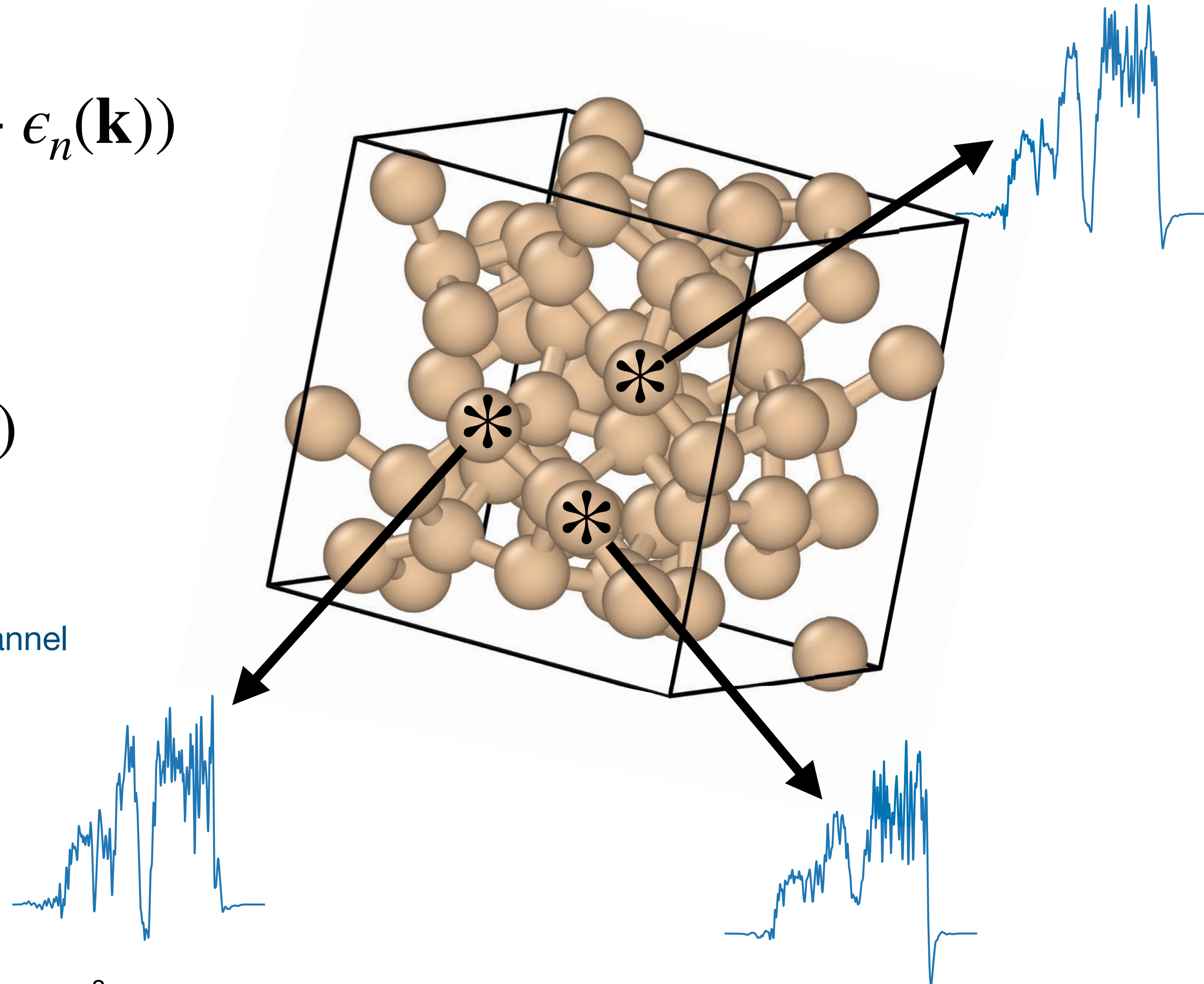


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Build a machine learning model for each energy channel



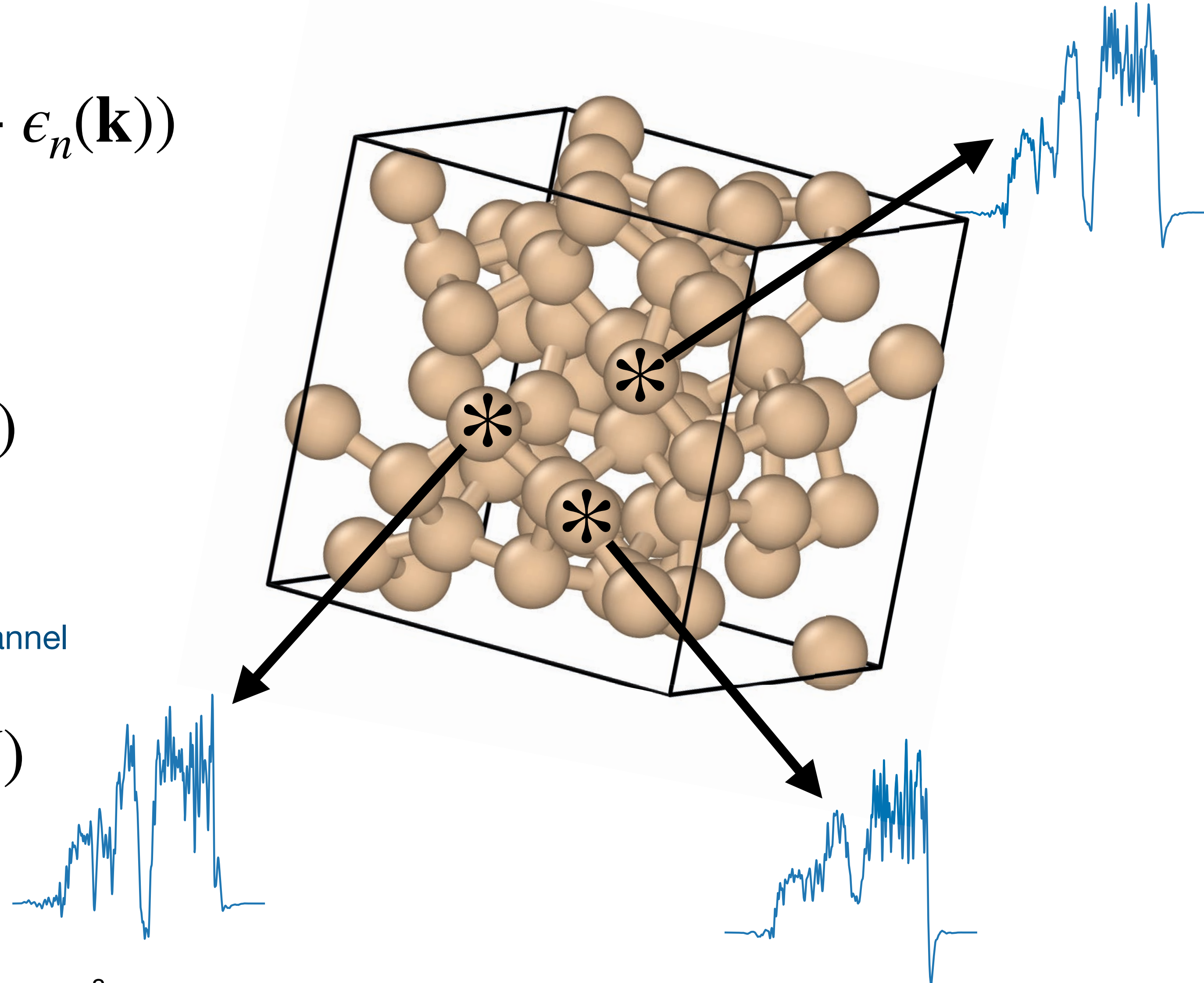
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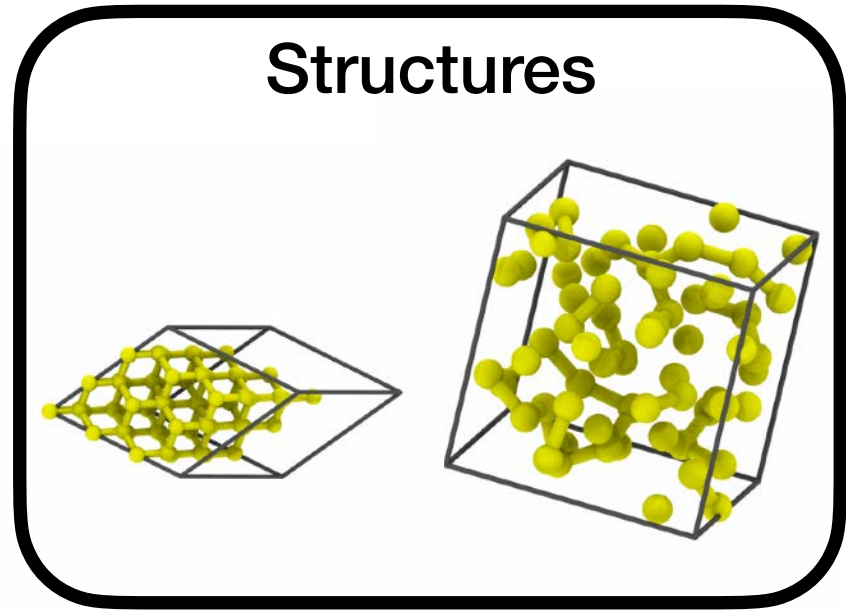
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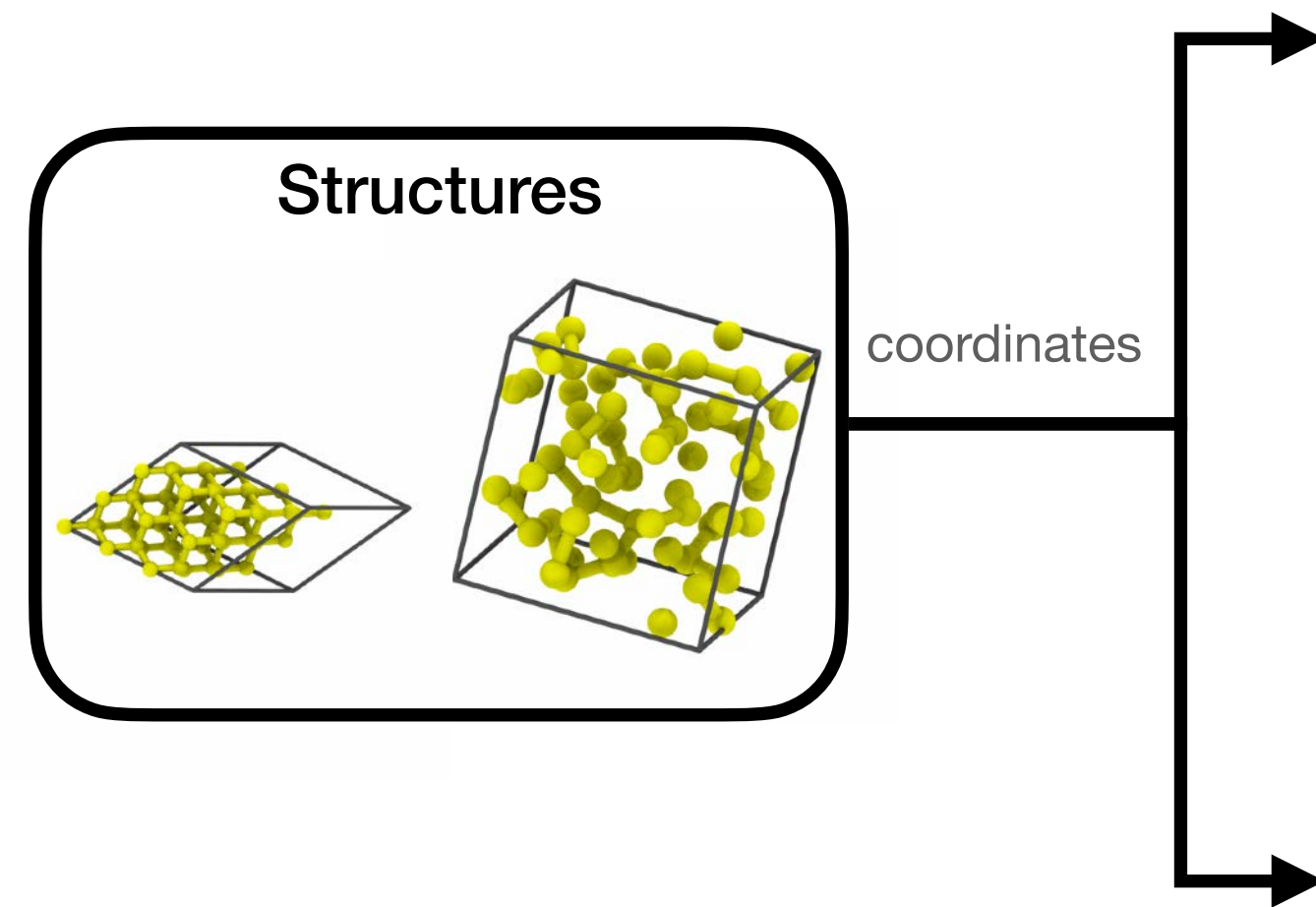
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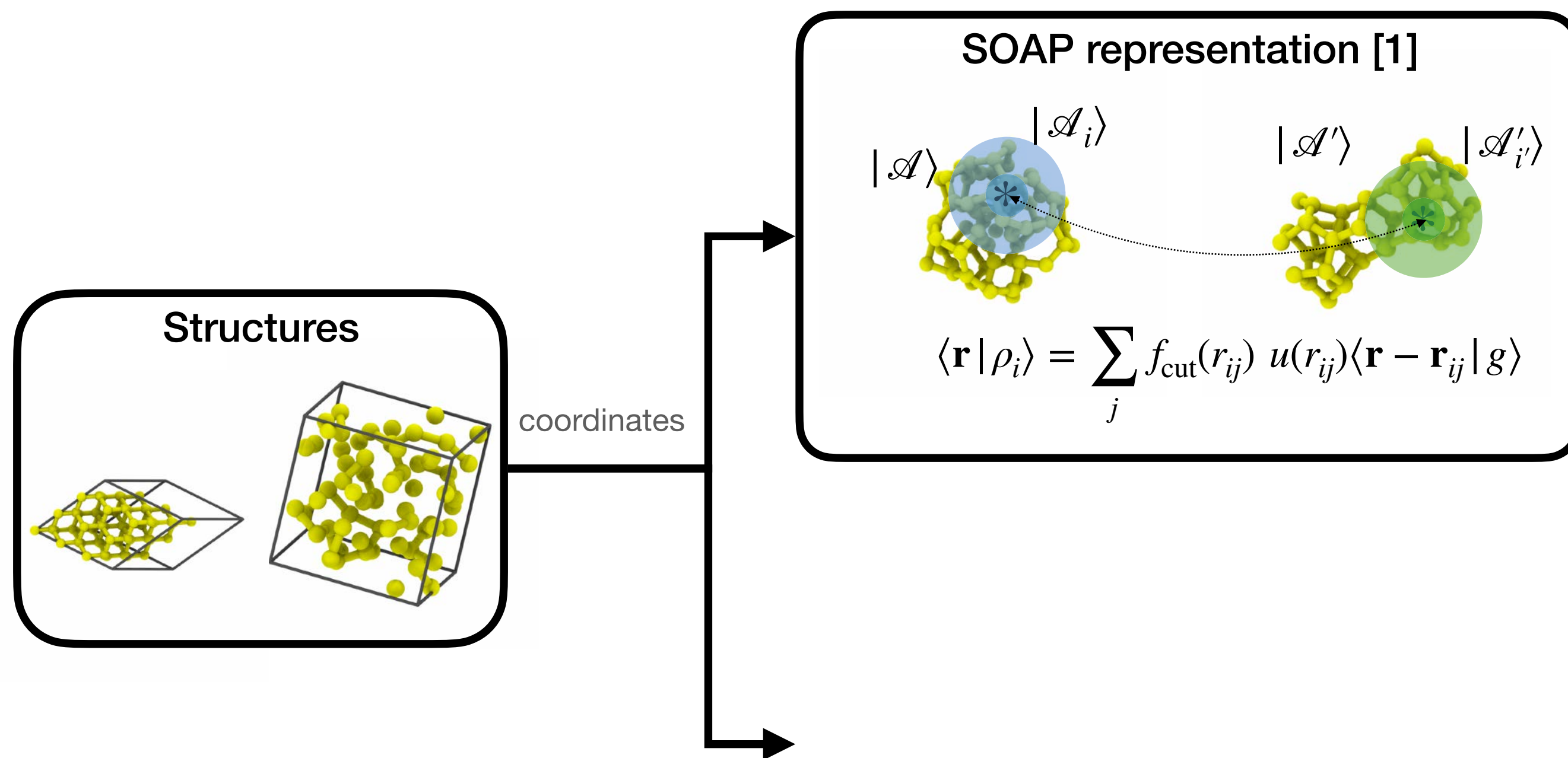
Build a machine learning model for each energy channel

- $$\text{DOS}(\mathcal{A}, \epsilon) = \sum_{\mathcal{A}_j \in \mathcal{A}} \mathbf{x}_j(\epsilon) k(\mathcal{A}_j, M)$$



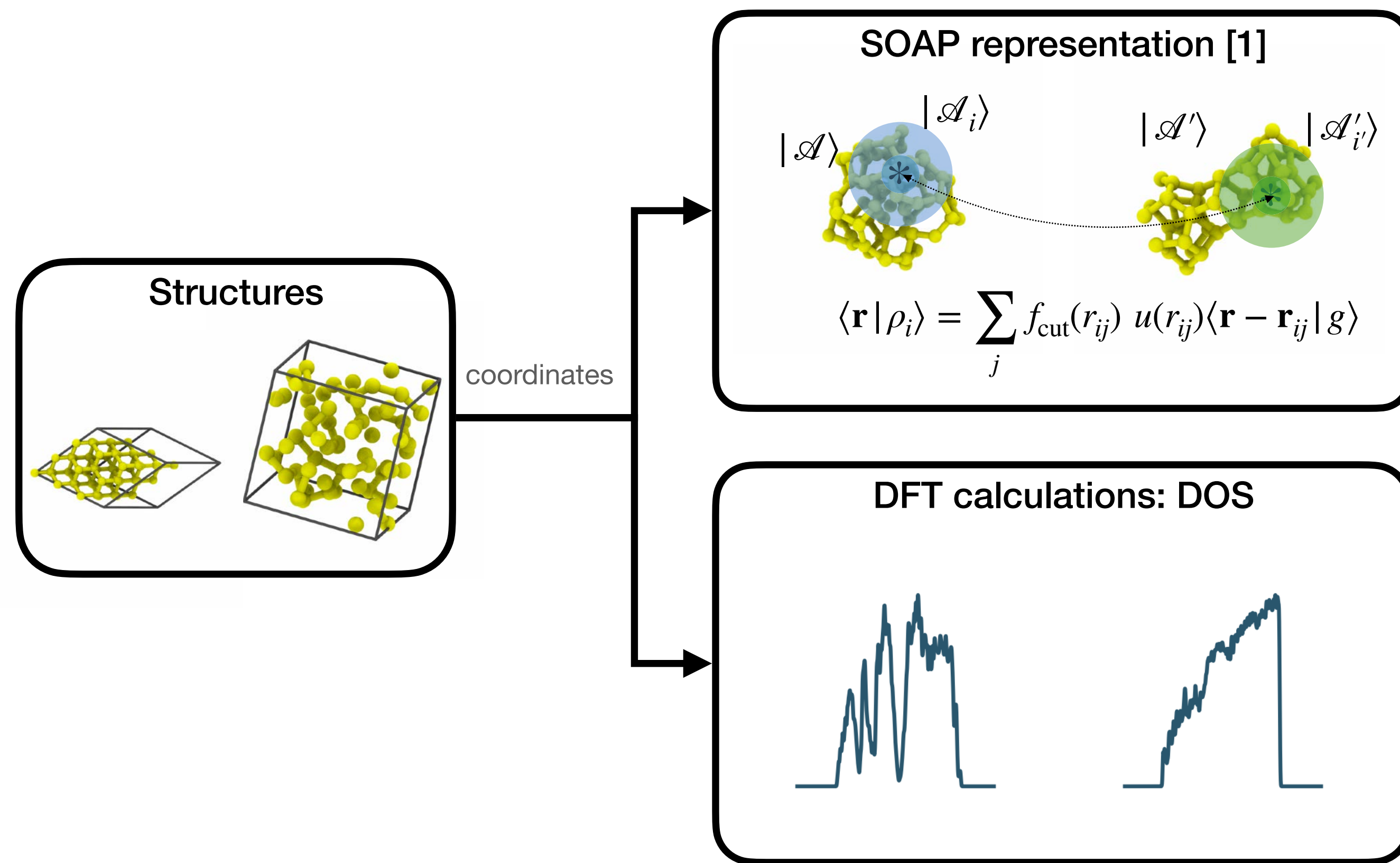






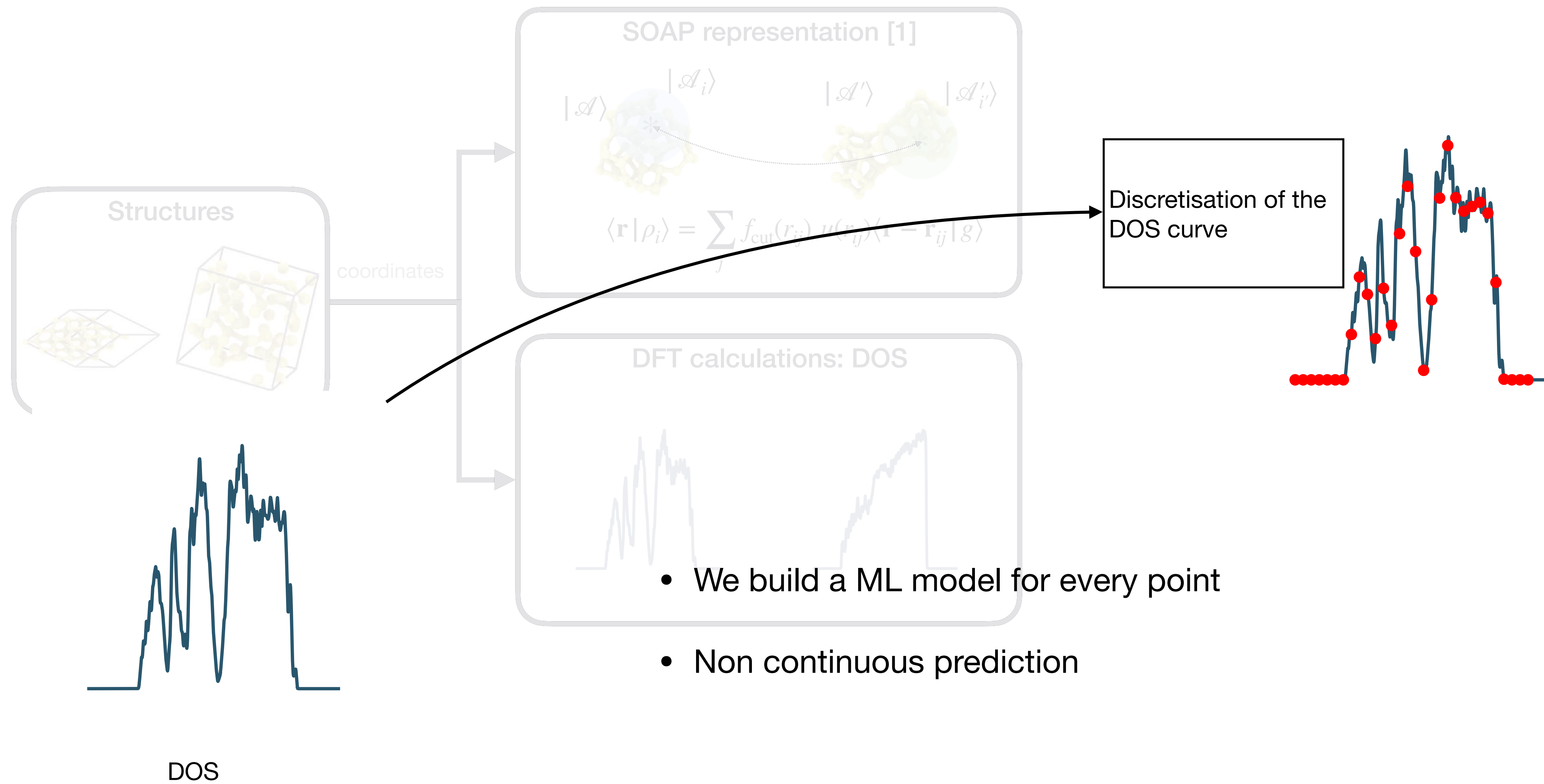
[1] Bartók, A.P., Kondor, R., Csányi, G., (2013). On representing chemical environments. Phys. Rev. B 87, 184115.

CBM, Anelli, A., Csányi, G., Ceriotti, M., 2020.
Learning the electronic density of states in condensed matter.
Phys. Rev. B 102, 235130

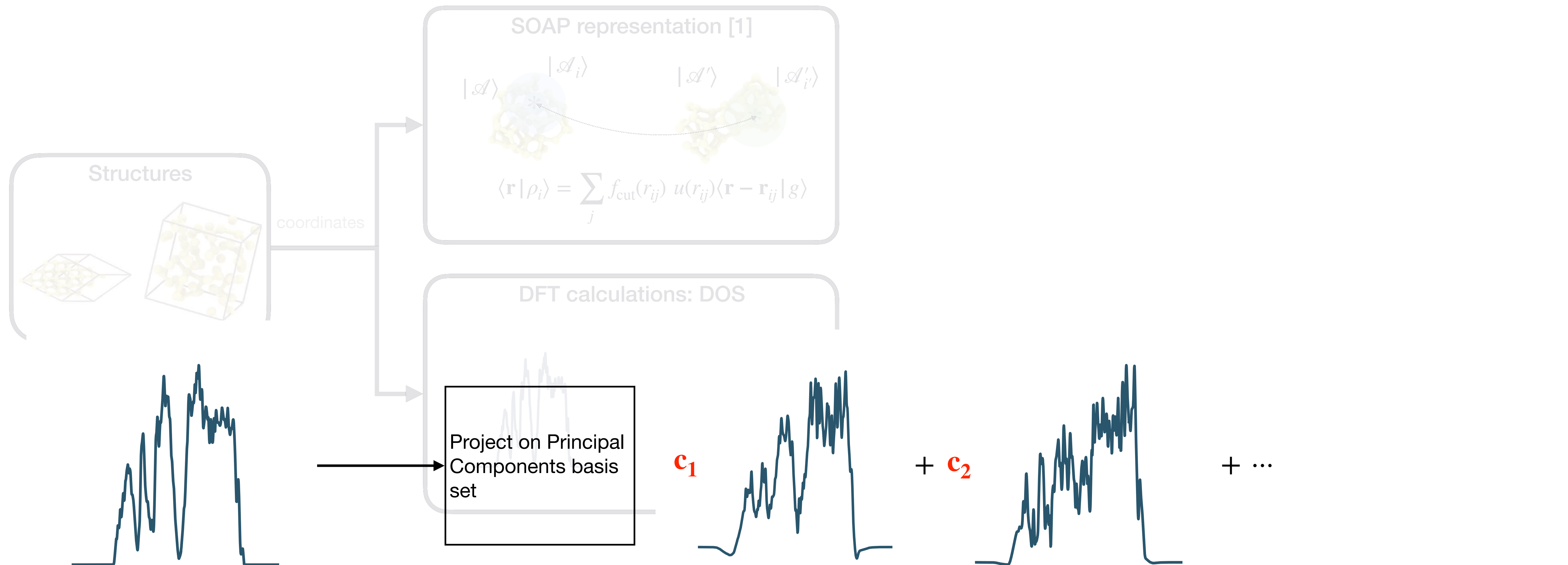


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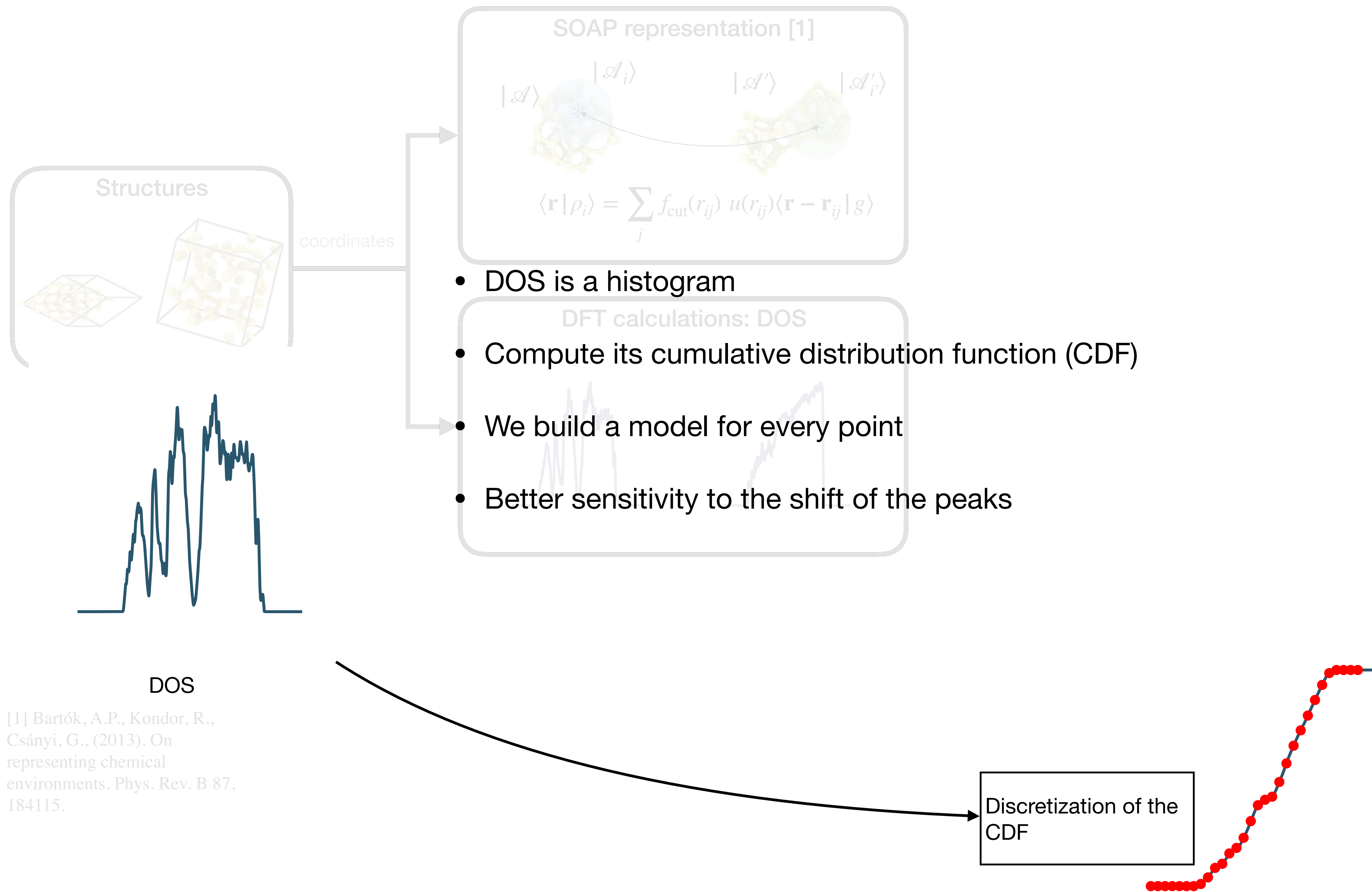
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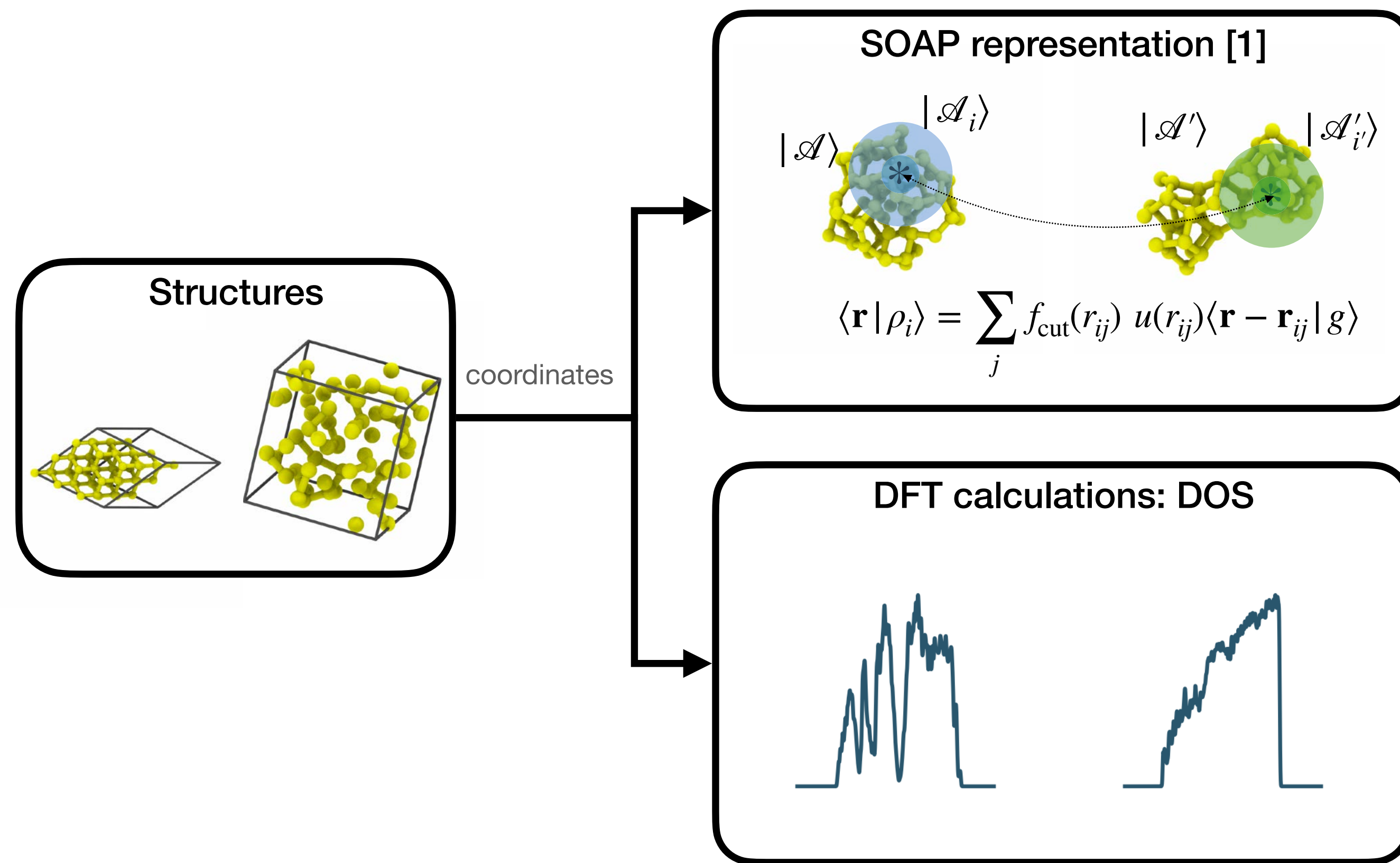
- We build a ML model for every coefficient
- Continuous prediction
- Extra smoothening

[1] Bartók, A.P., Kondor, R., Csányi, G., (2013). On representing chemical environments. Phys. Rev. B 87, 184115.

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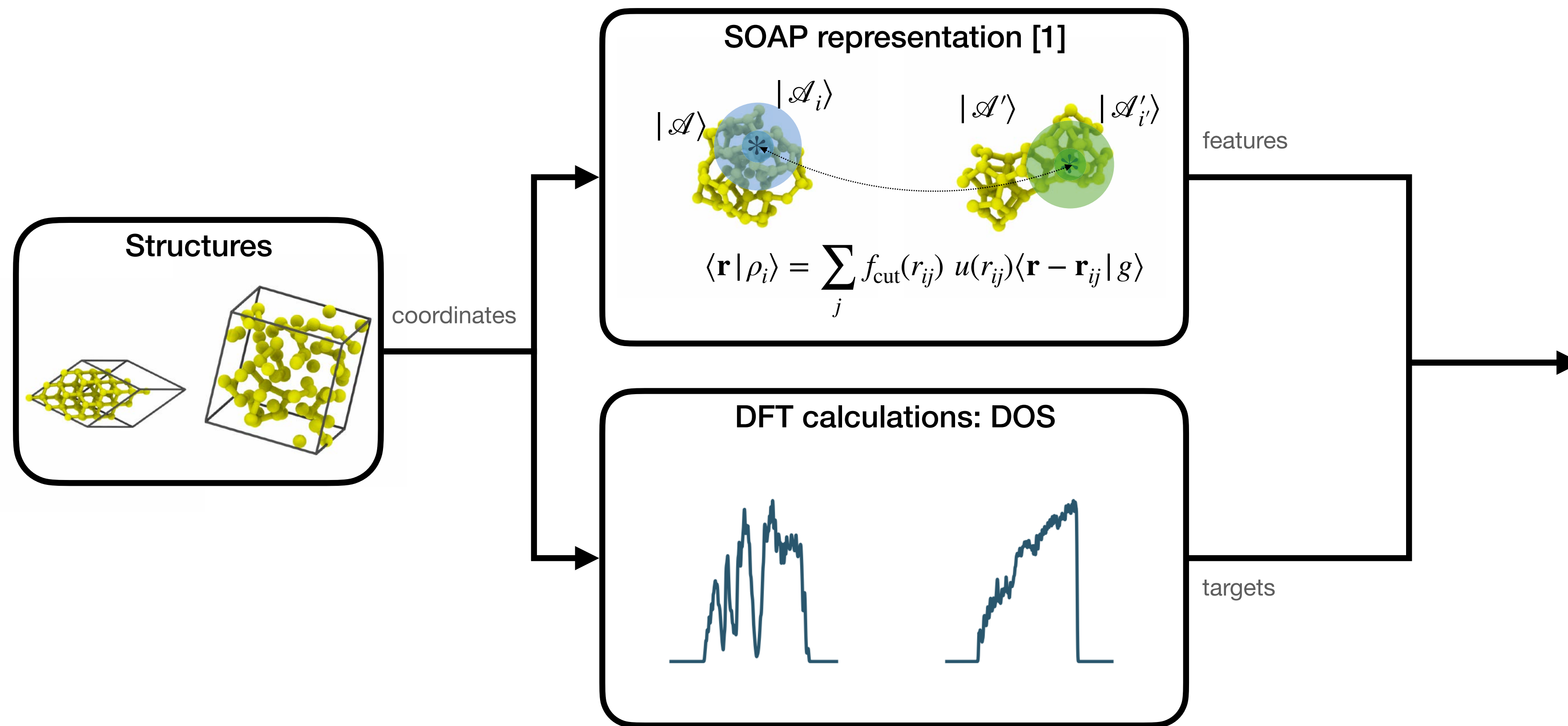


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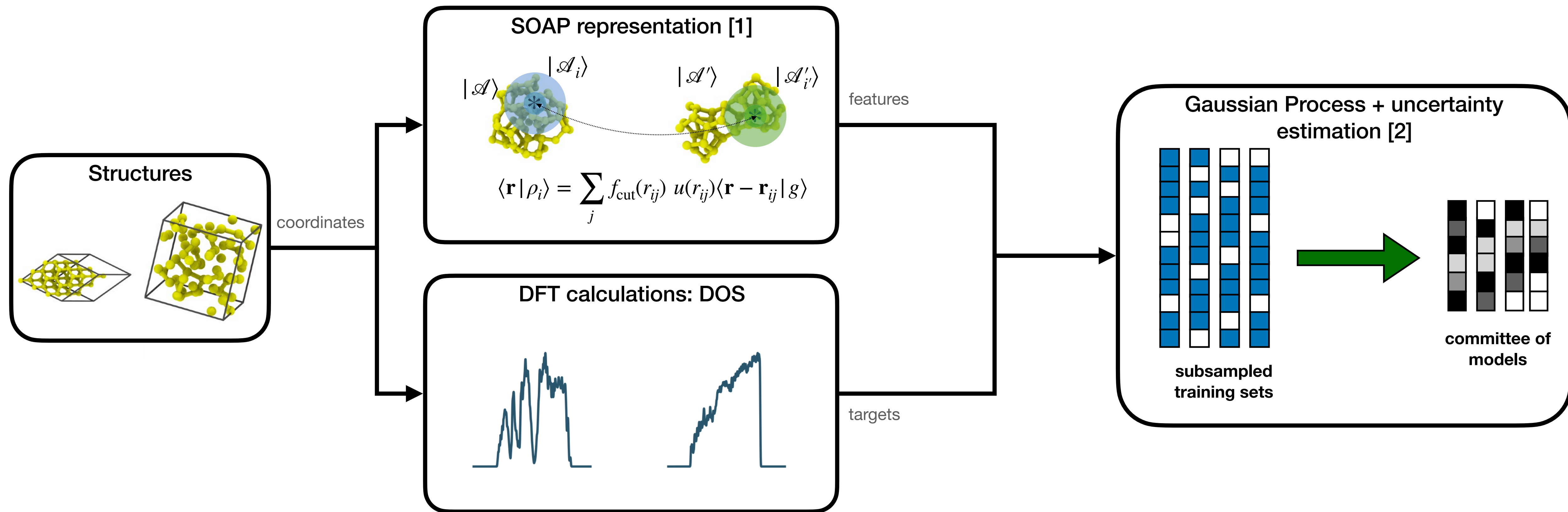
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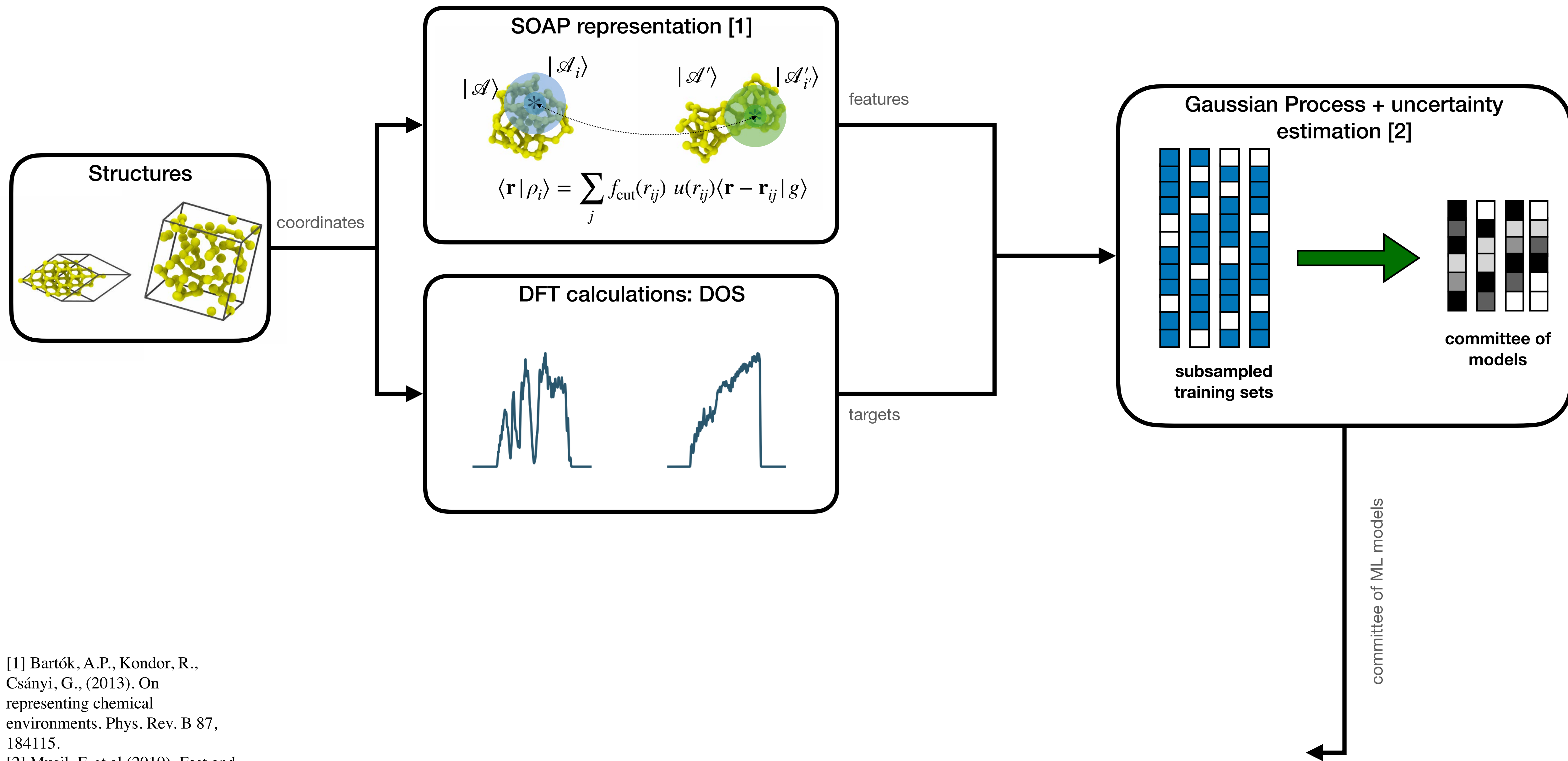
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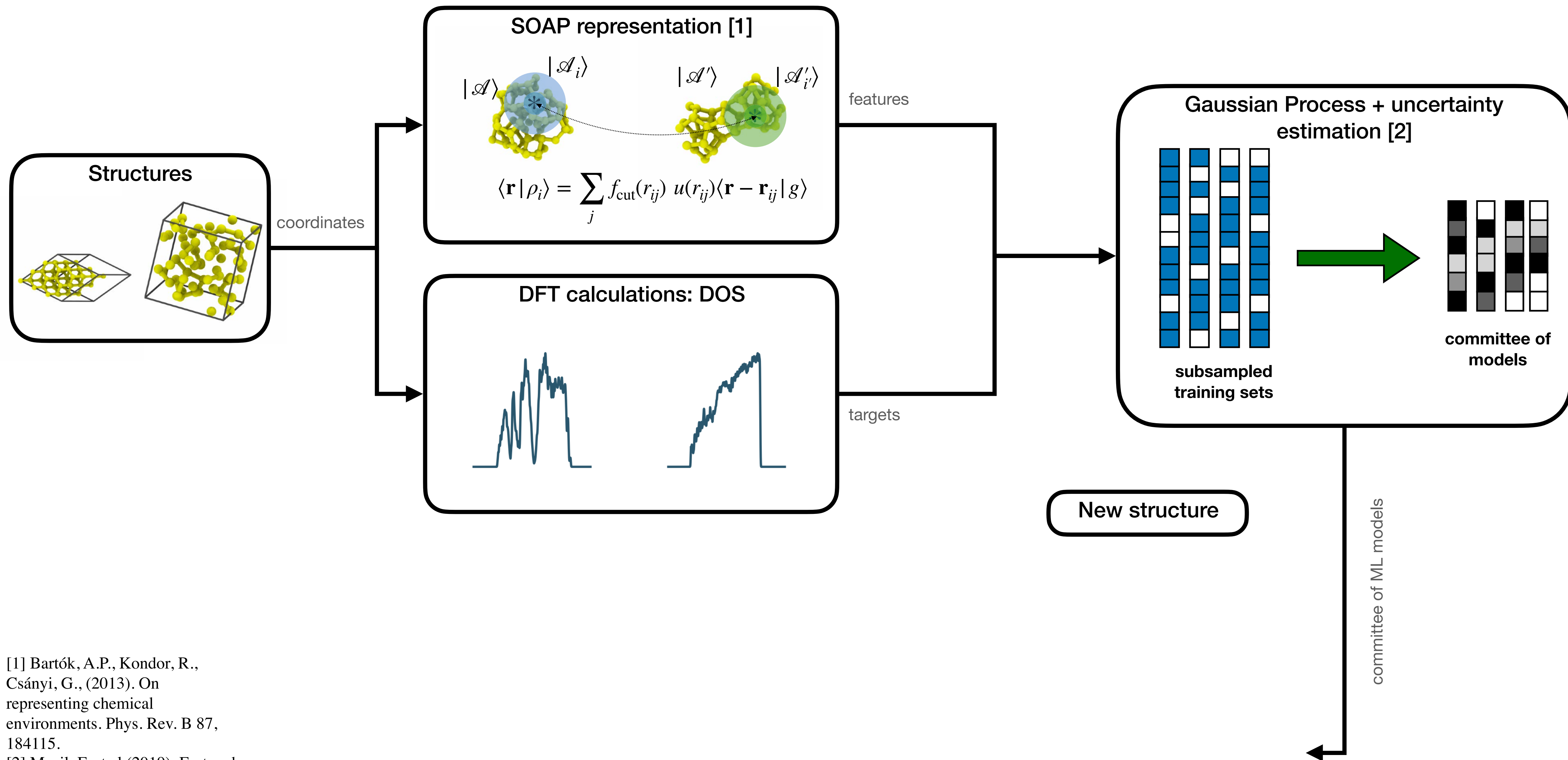
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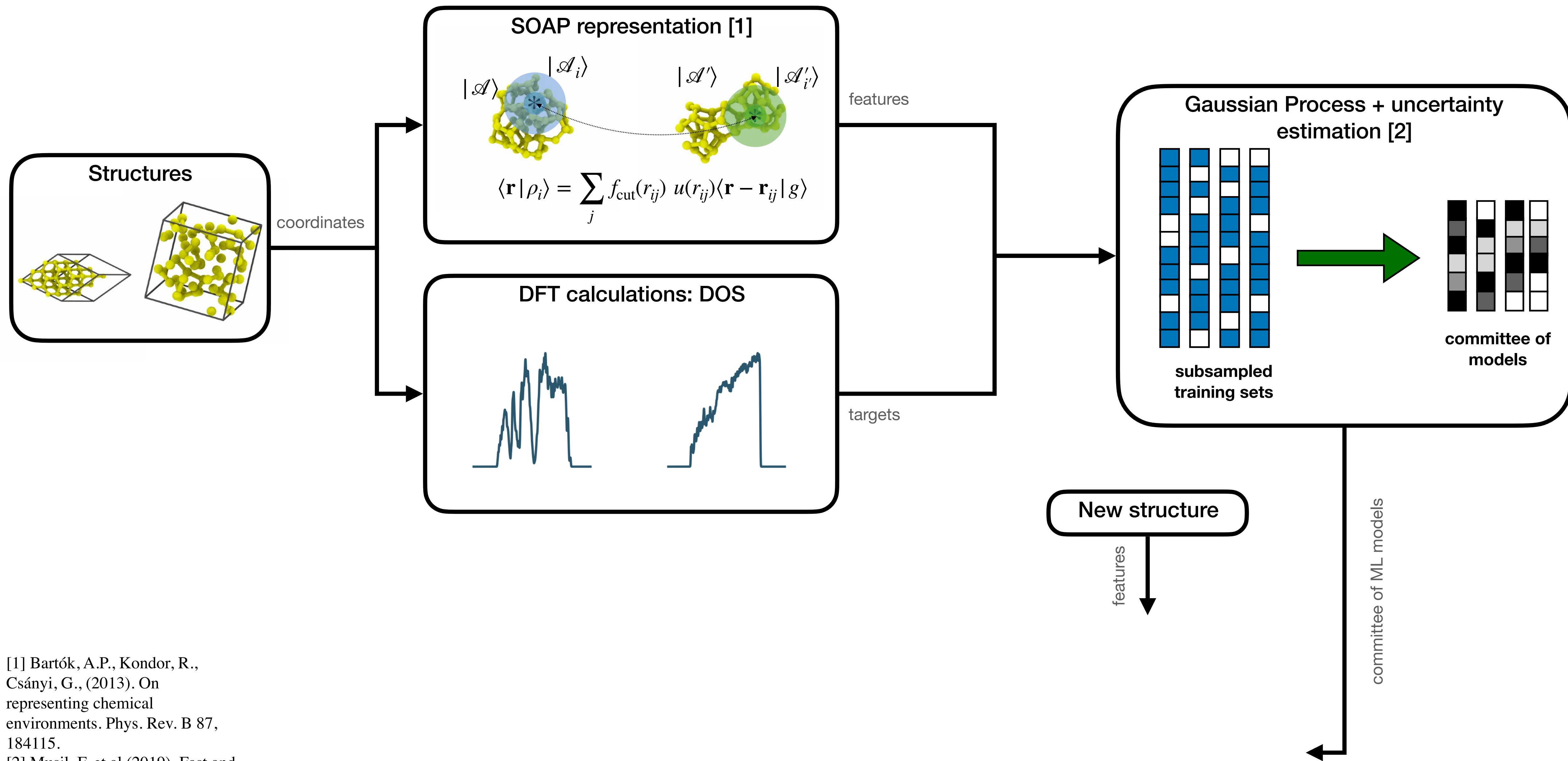
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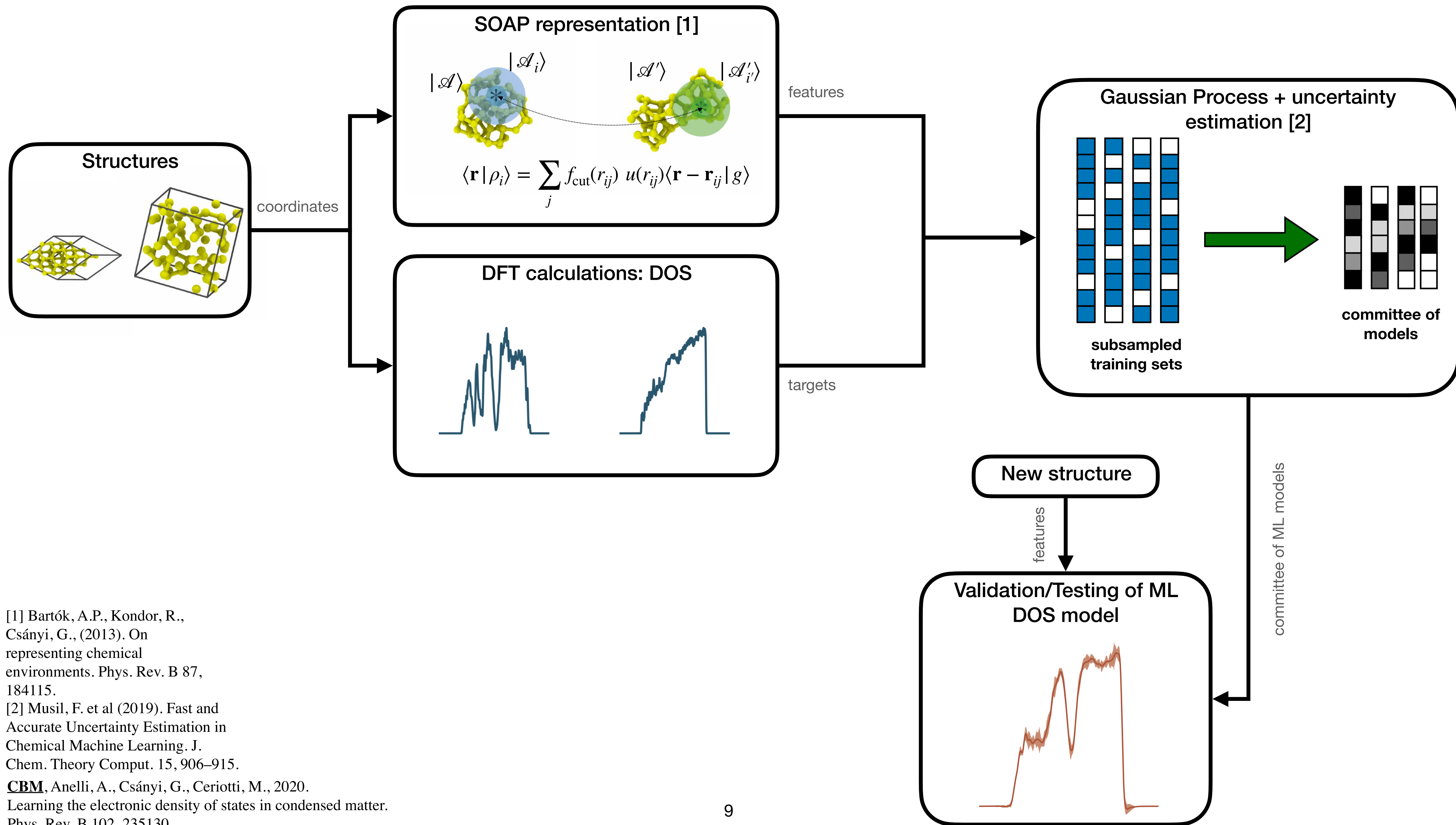
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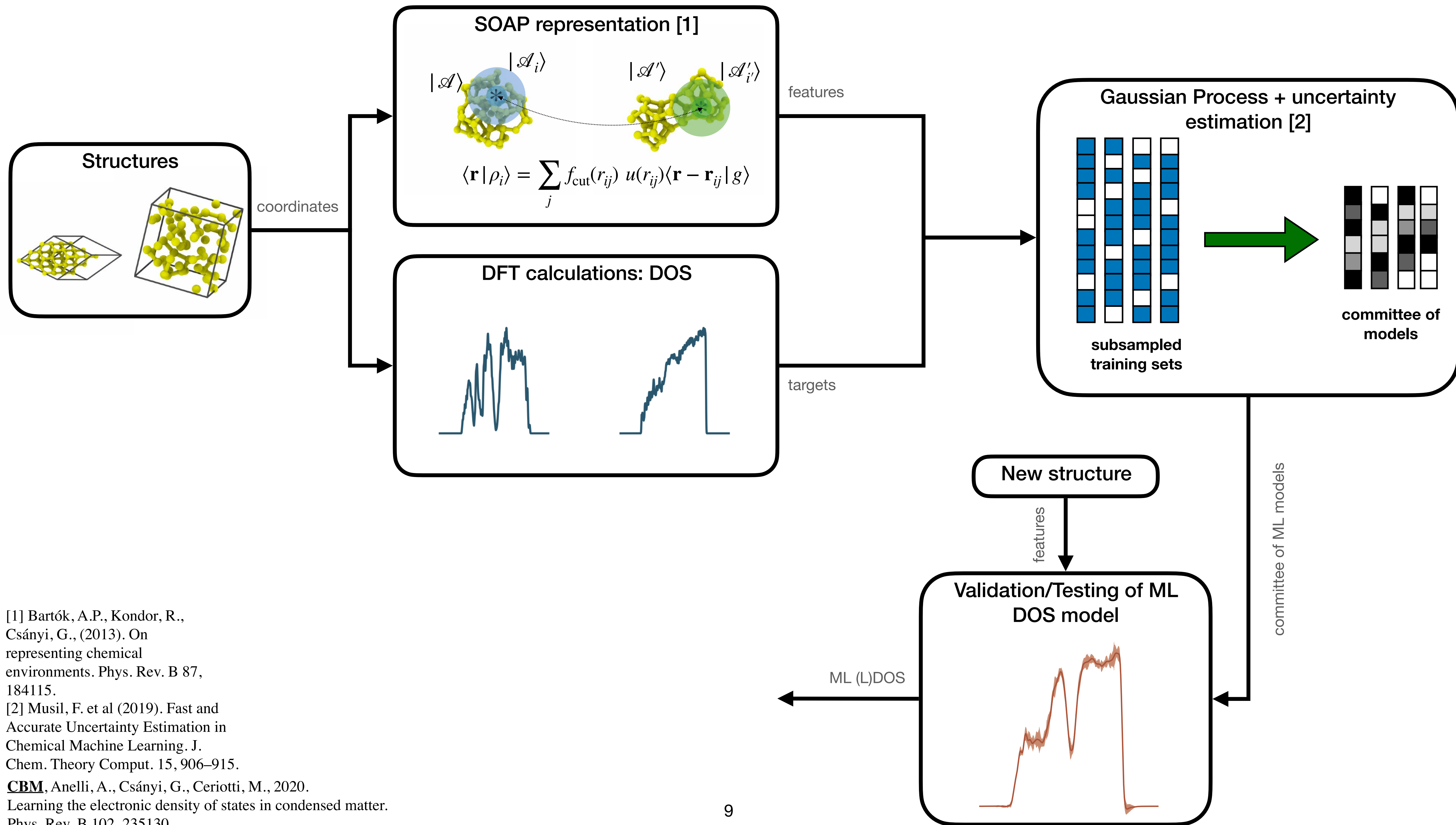
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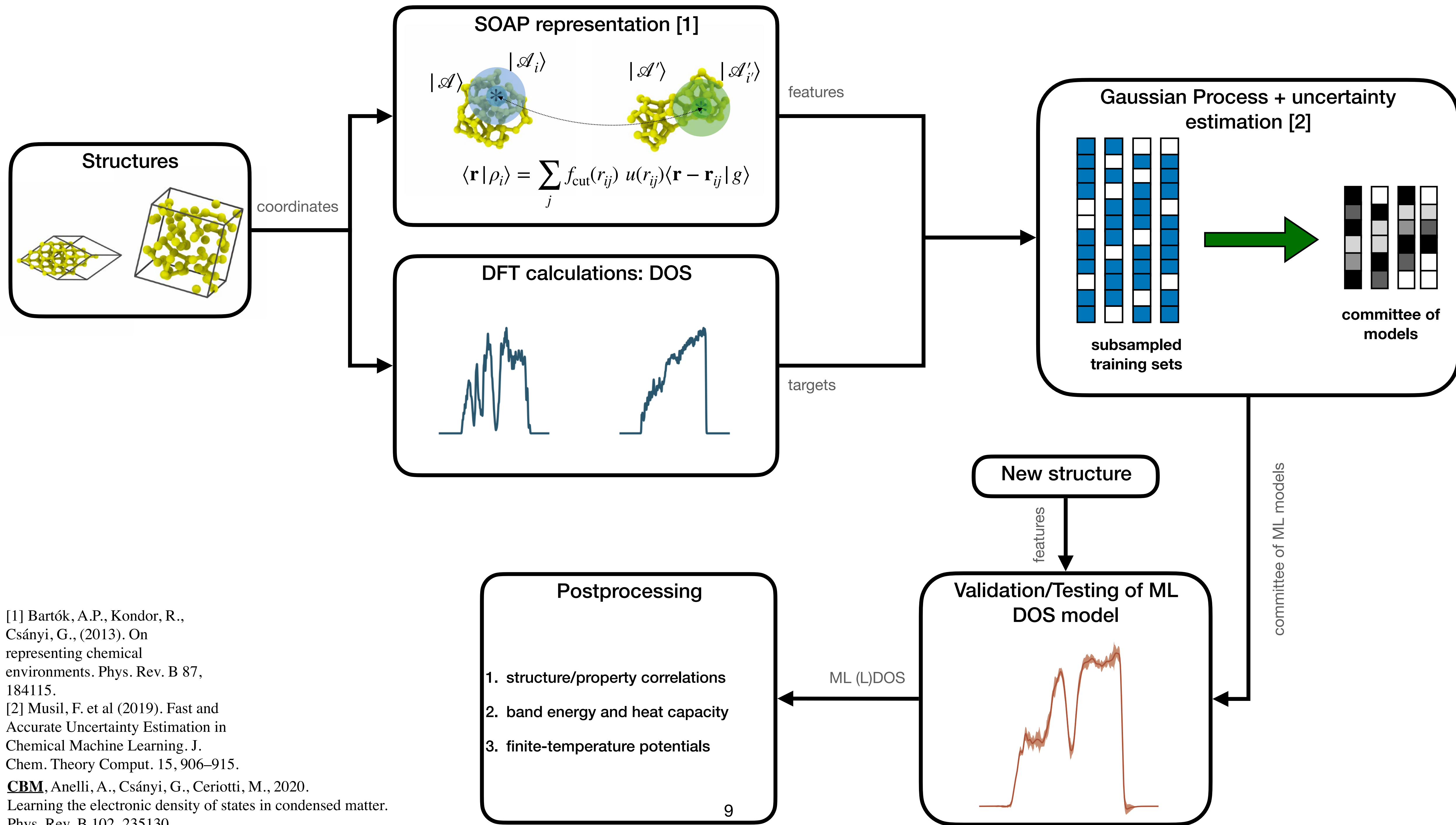
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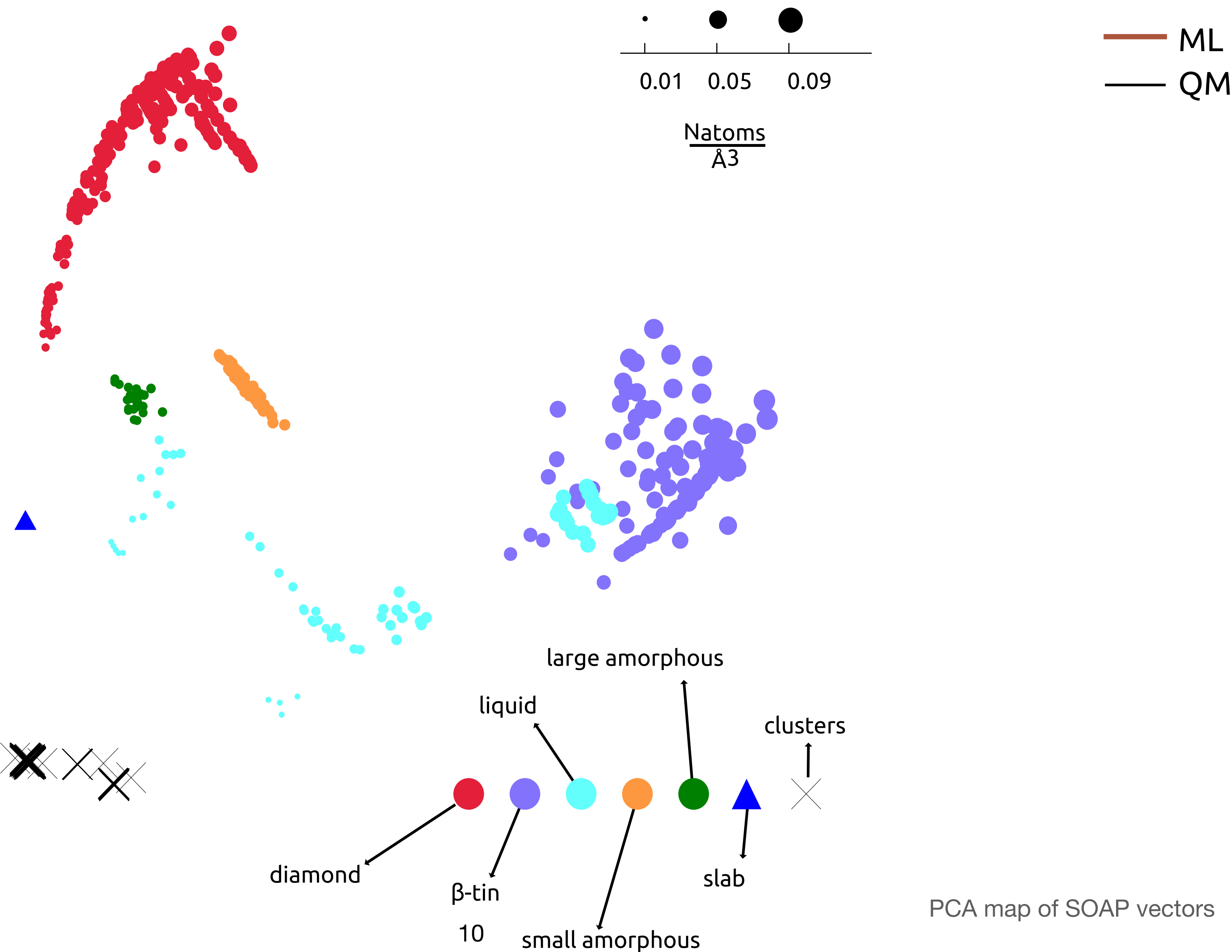
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Benchmark on diverse silicon data set

1039 Silicon structures

PCA map of SOAP vectors

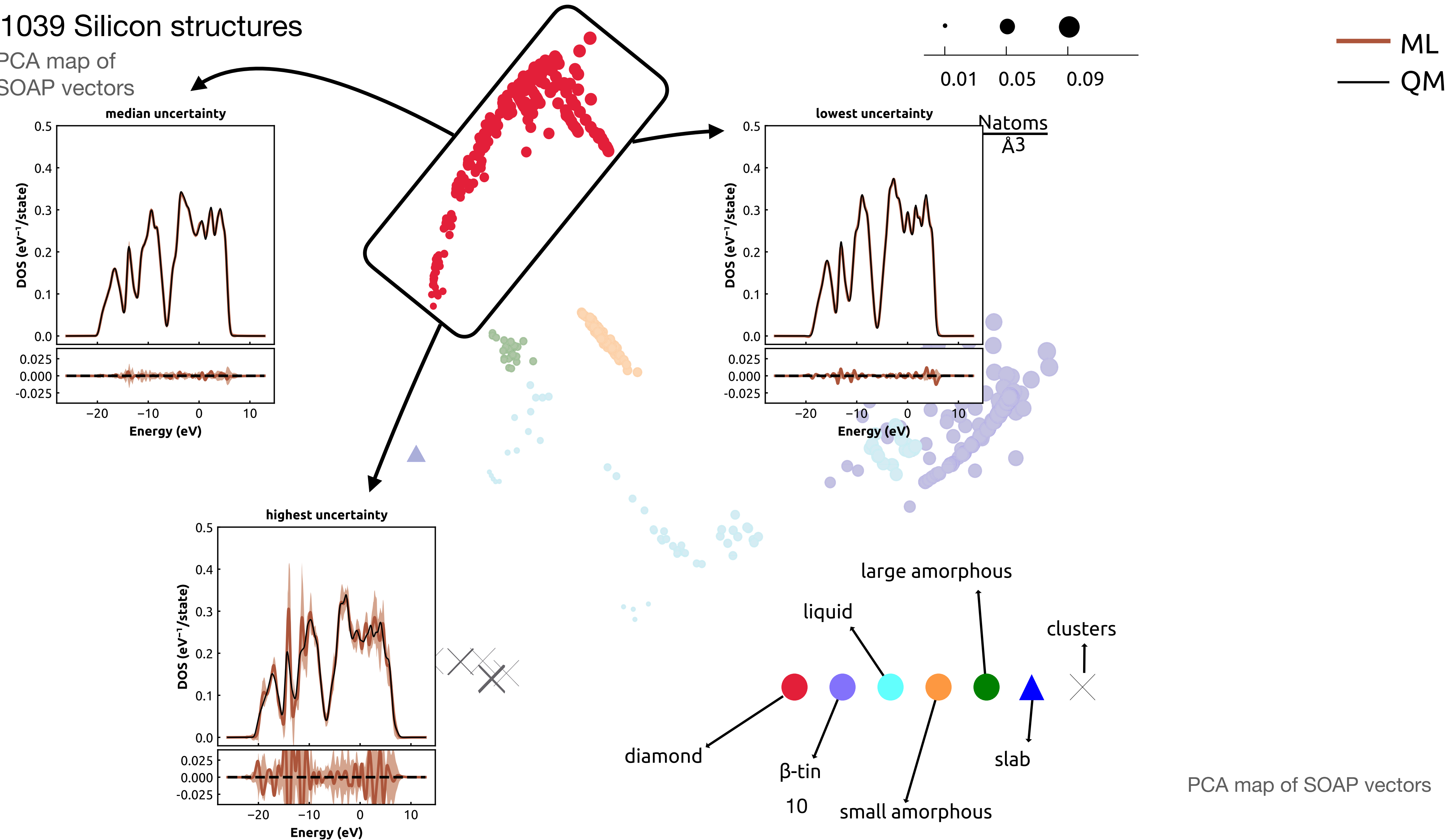


PCA map of SOAP vectors

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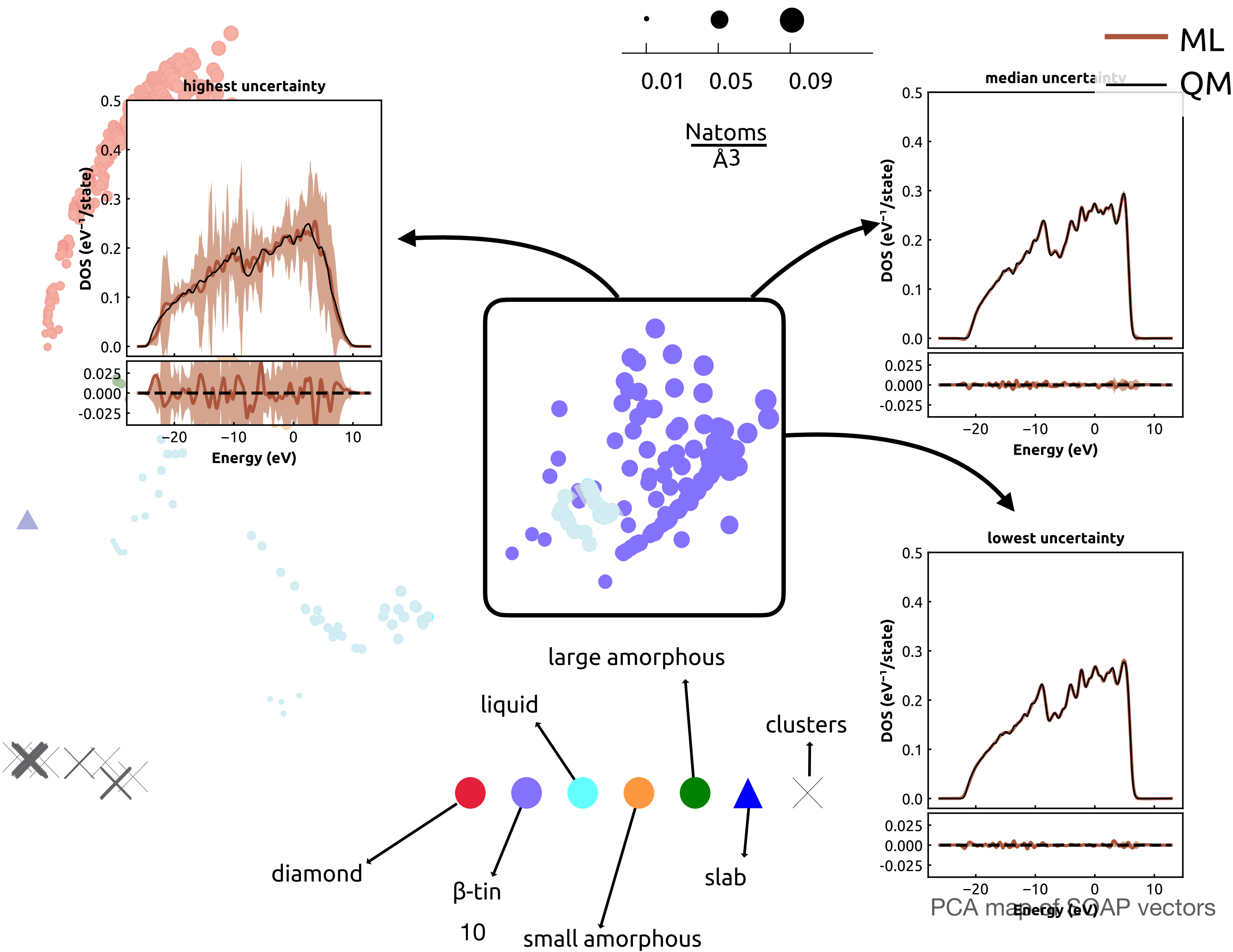
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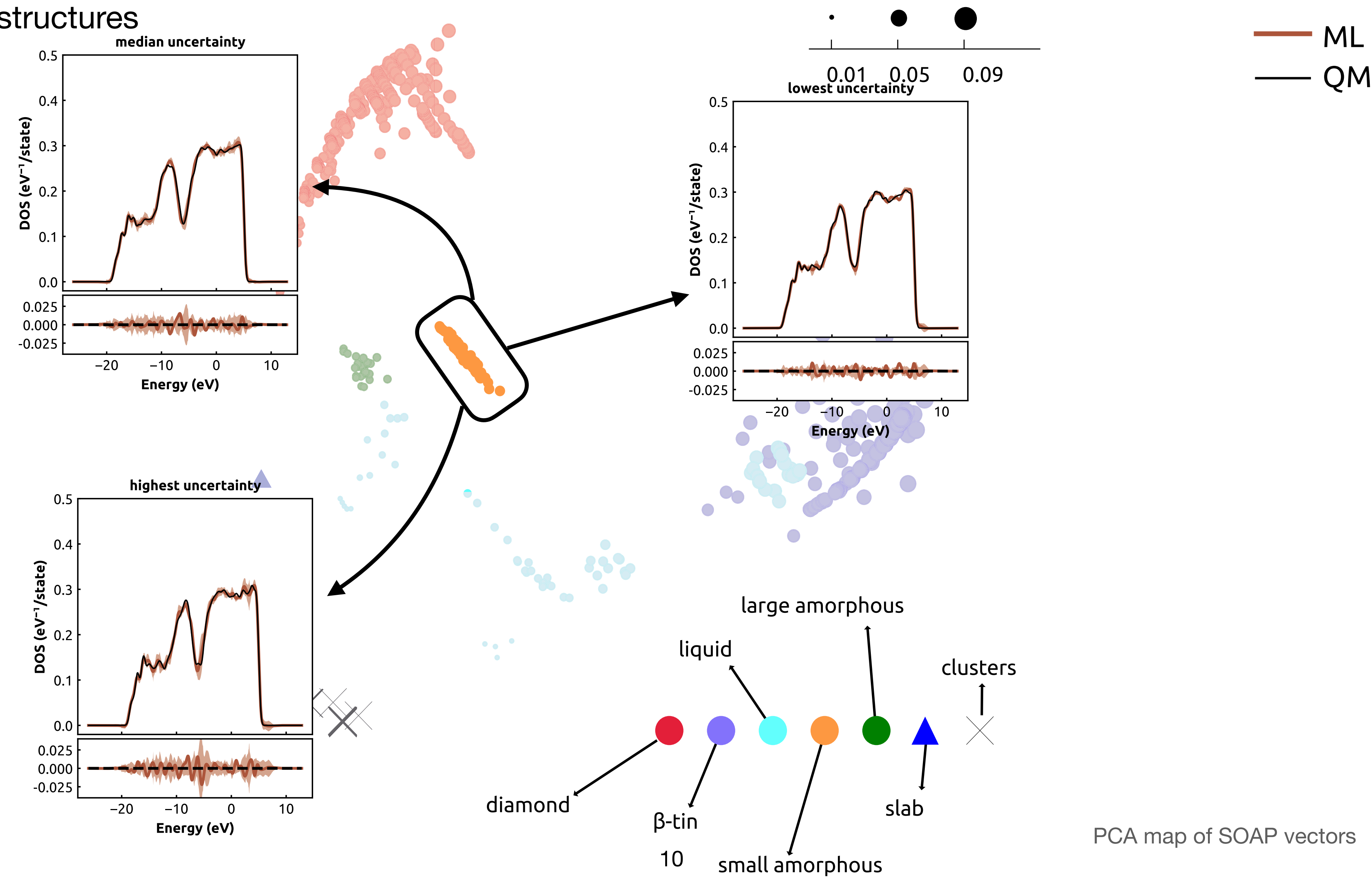
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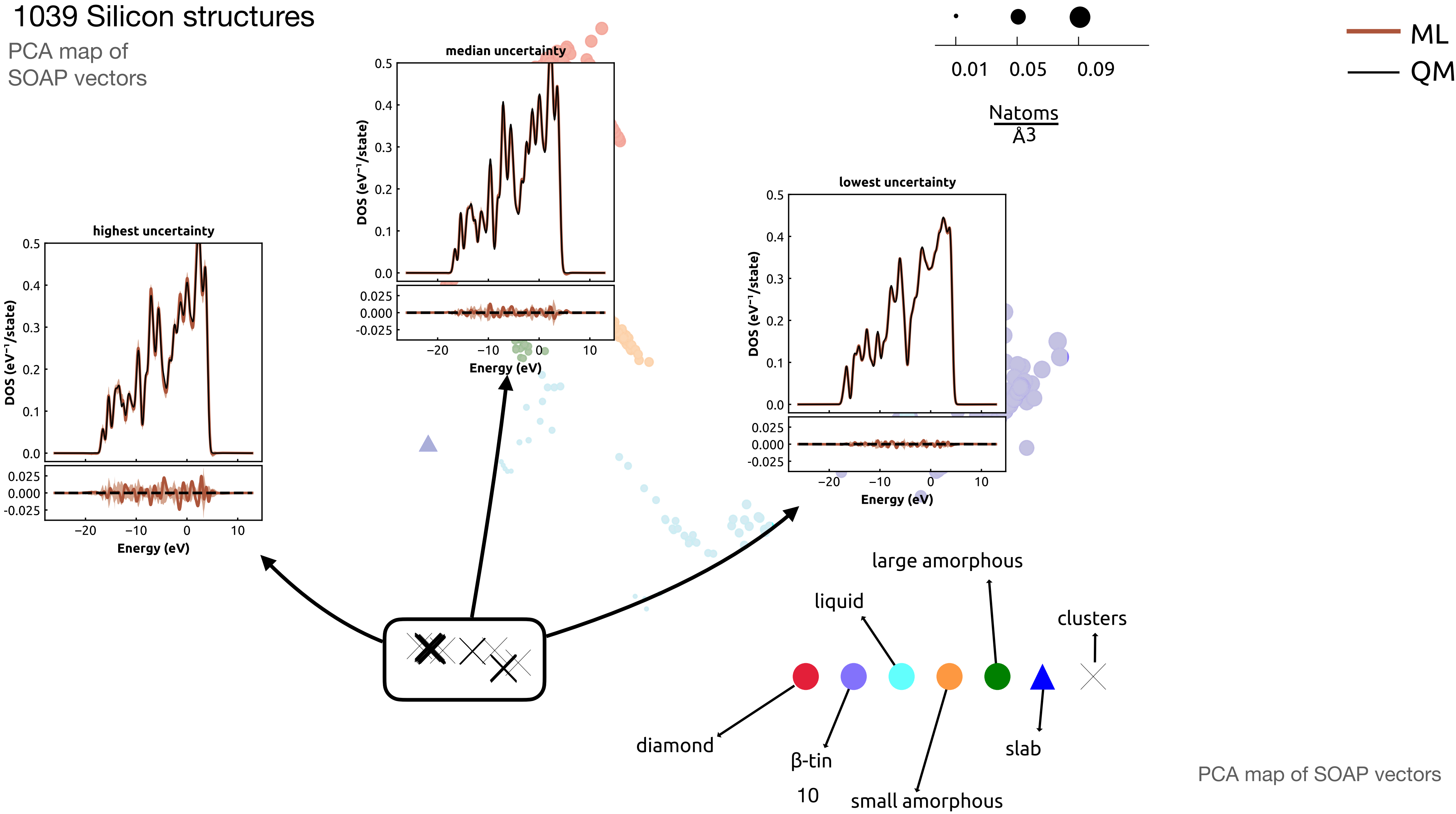


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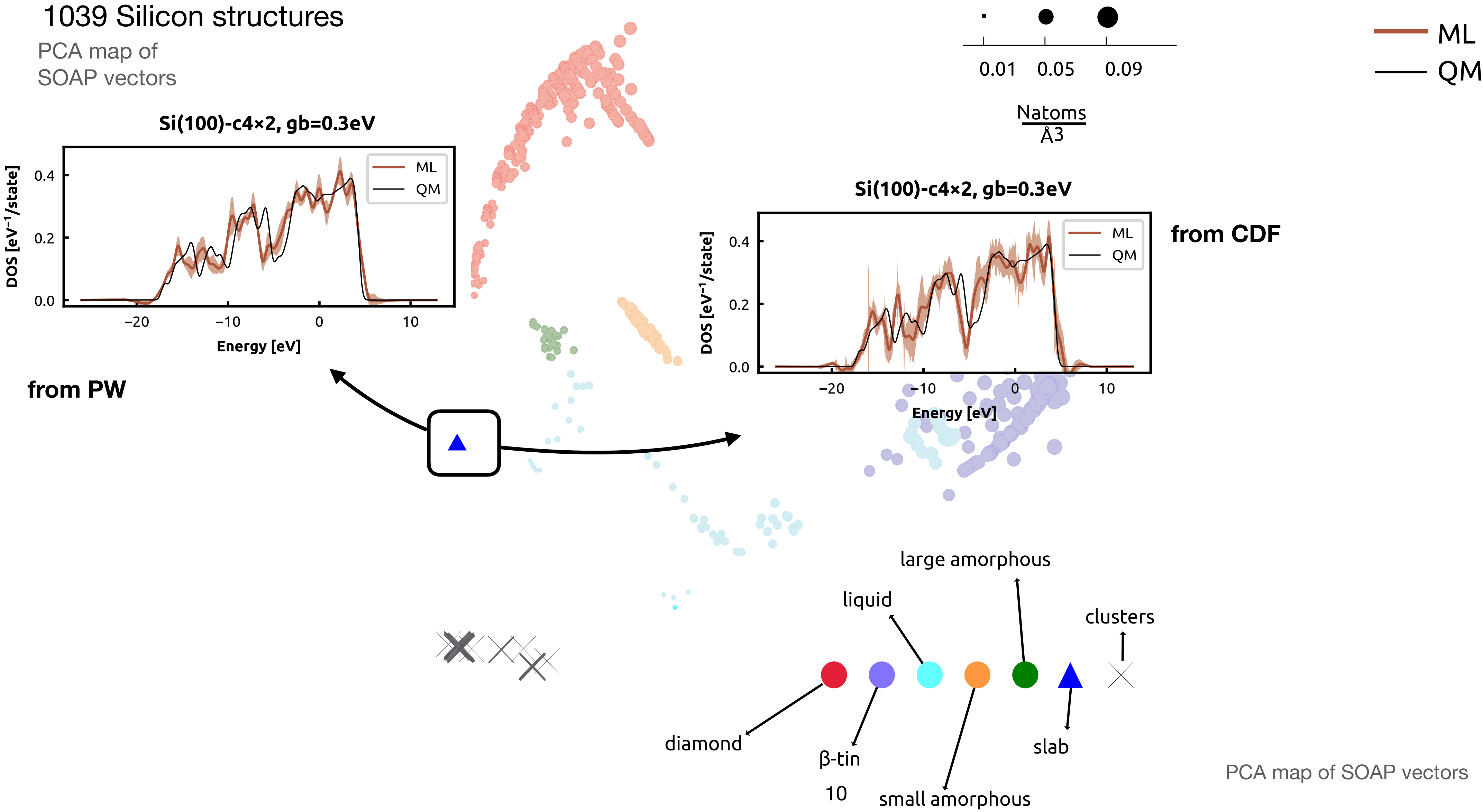


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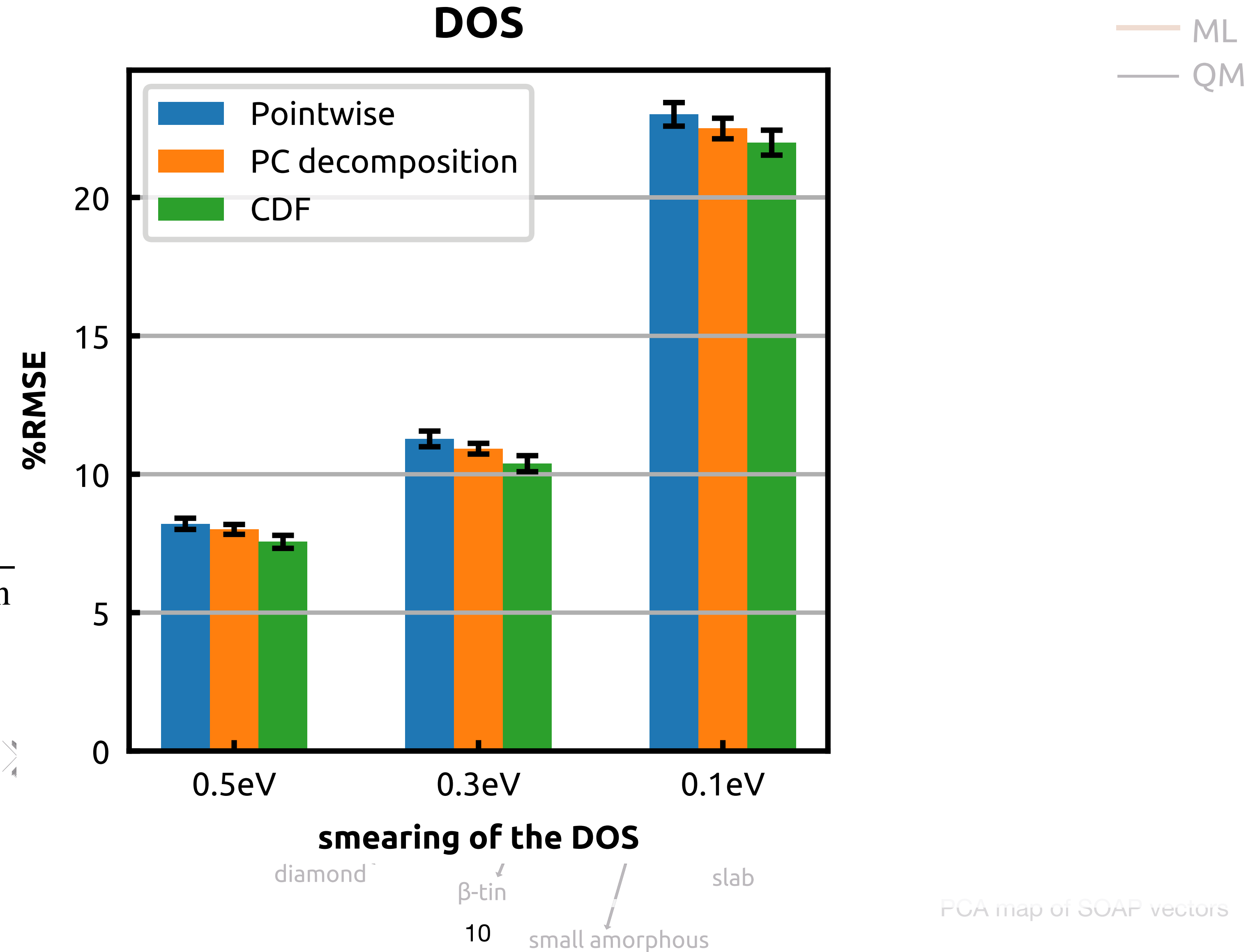


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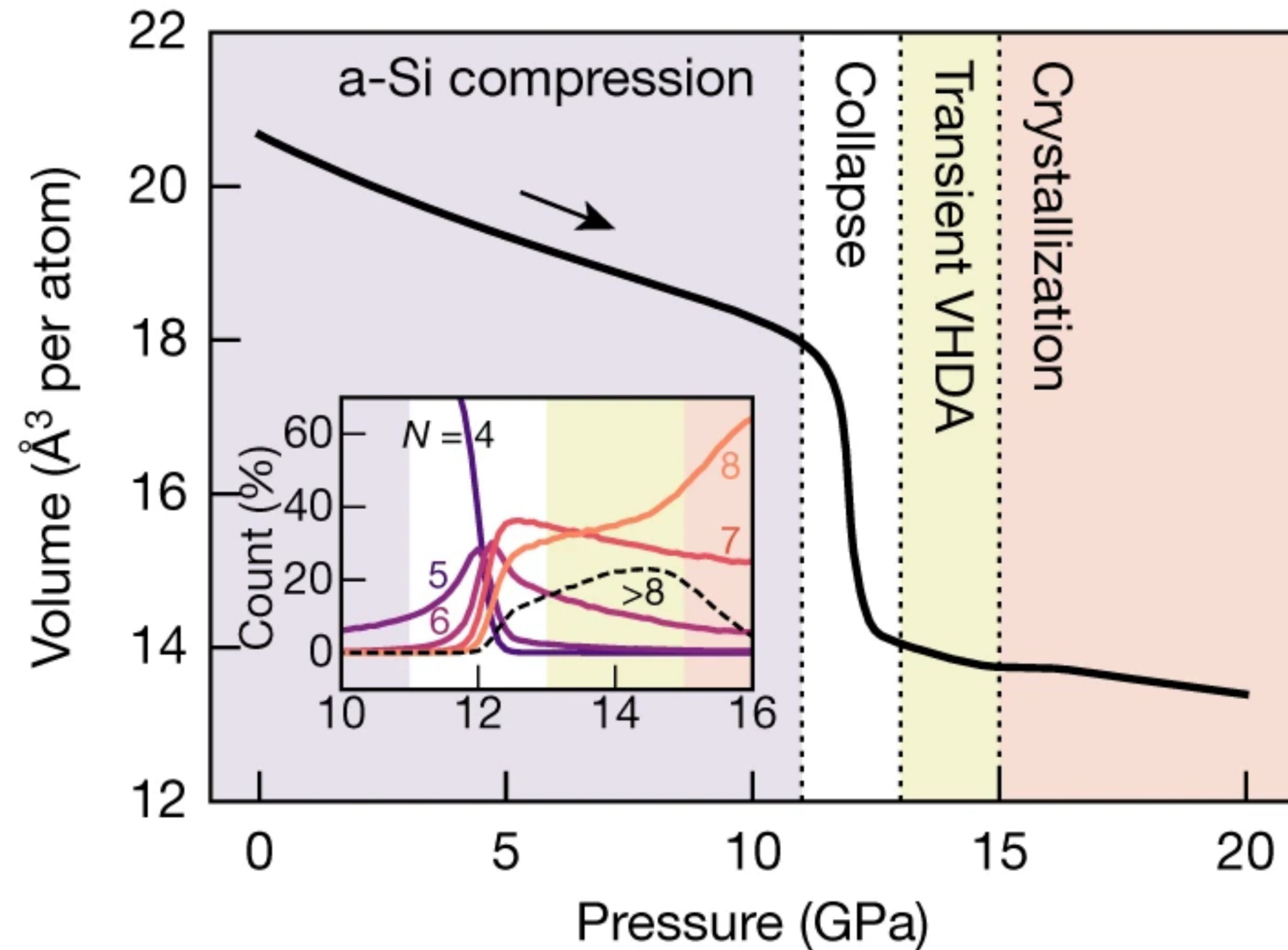
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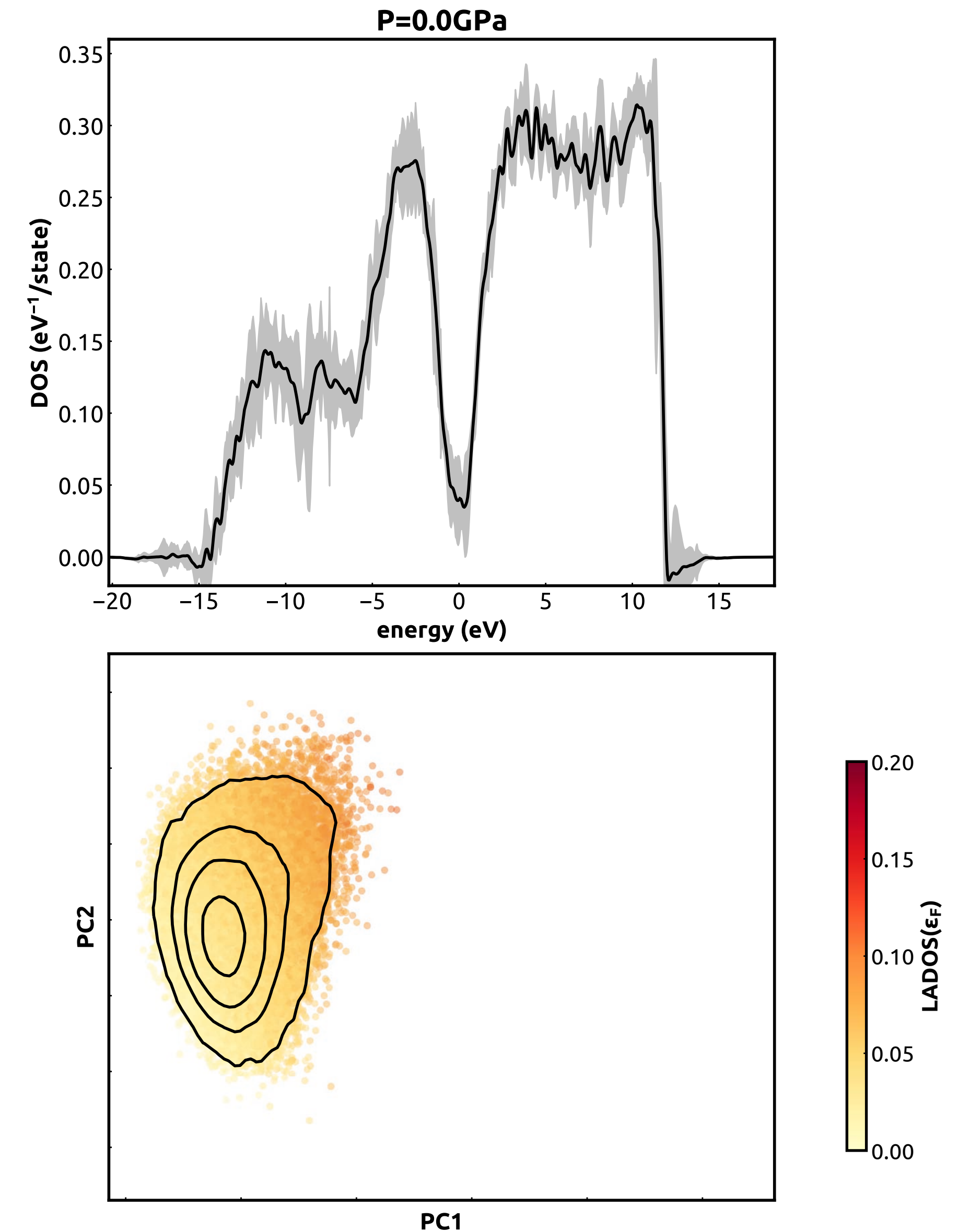
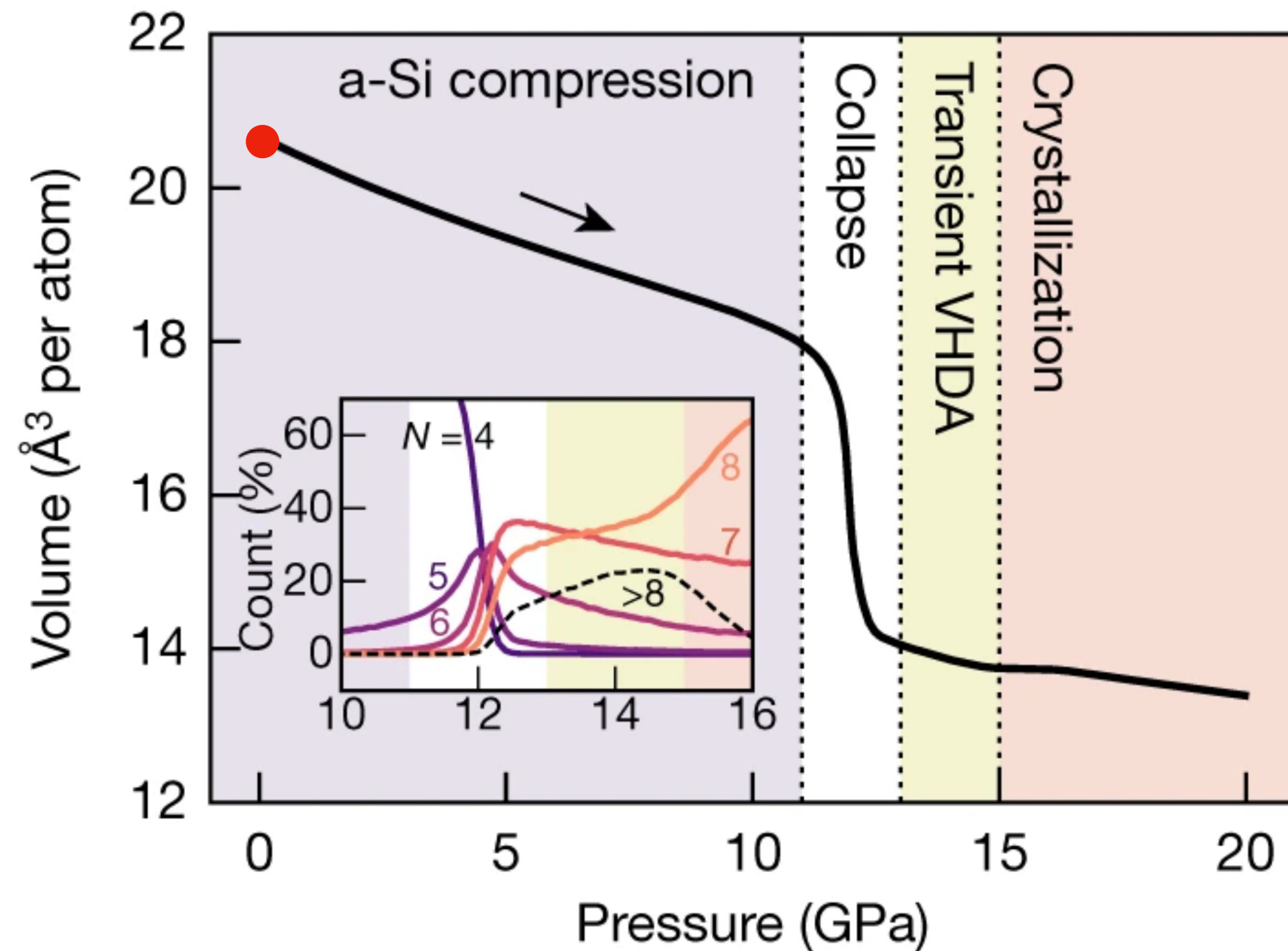
$$\% \text{RMSE} = 100 \times \frac{\text{RMSE}}{\text{standard deviation}}$$



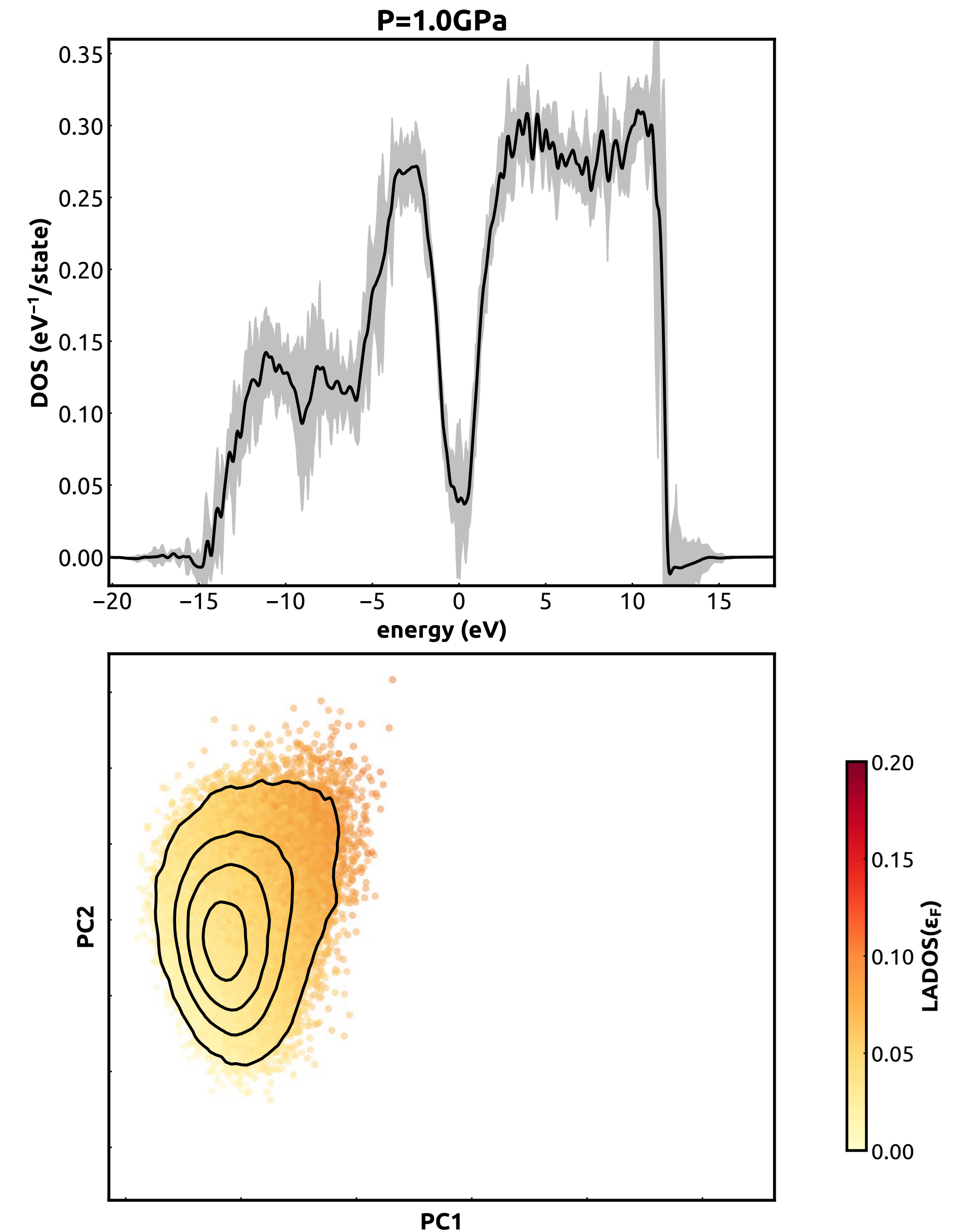
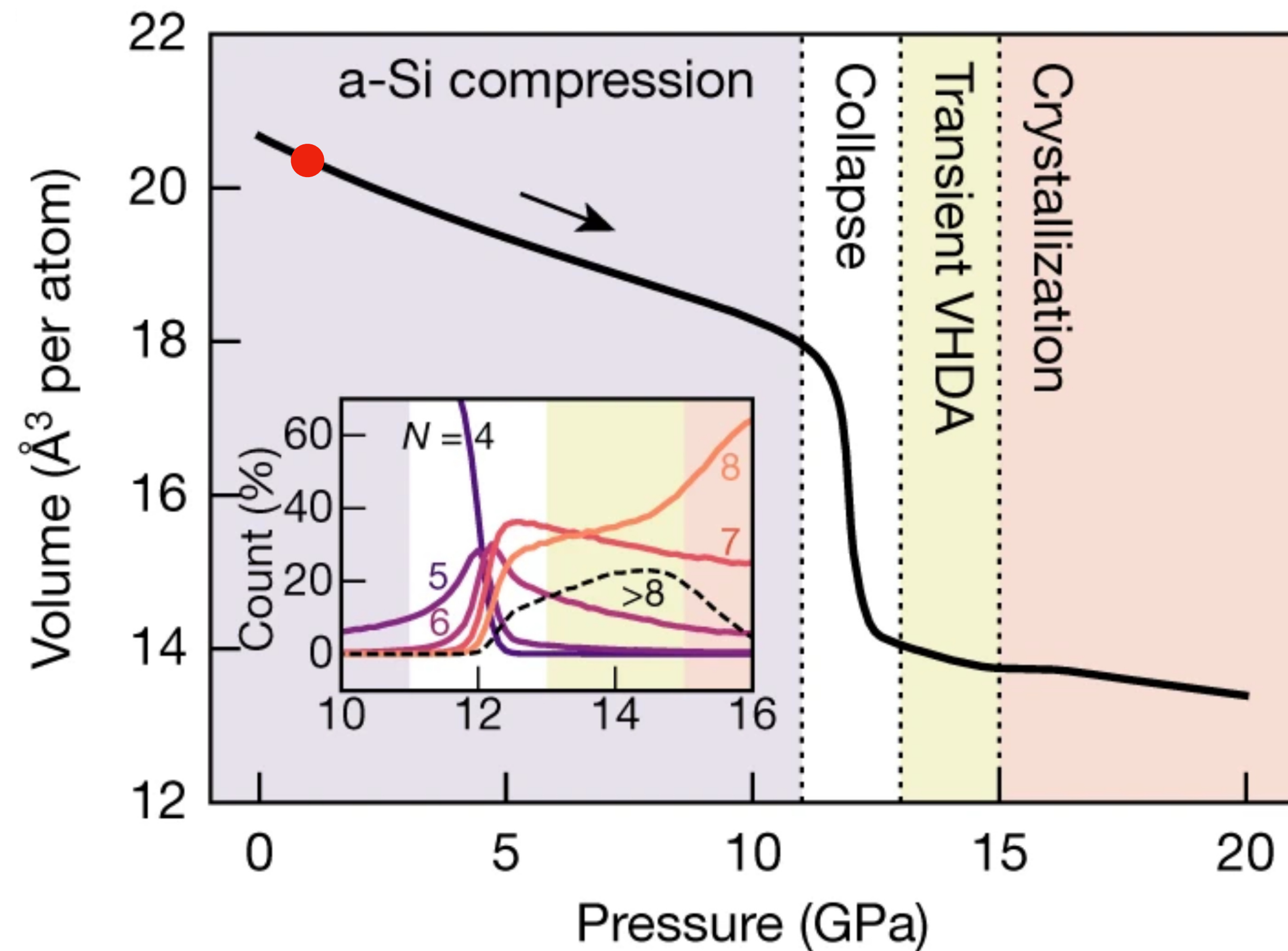
Combining structural and electronic fingerprints



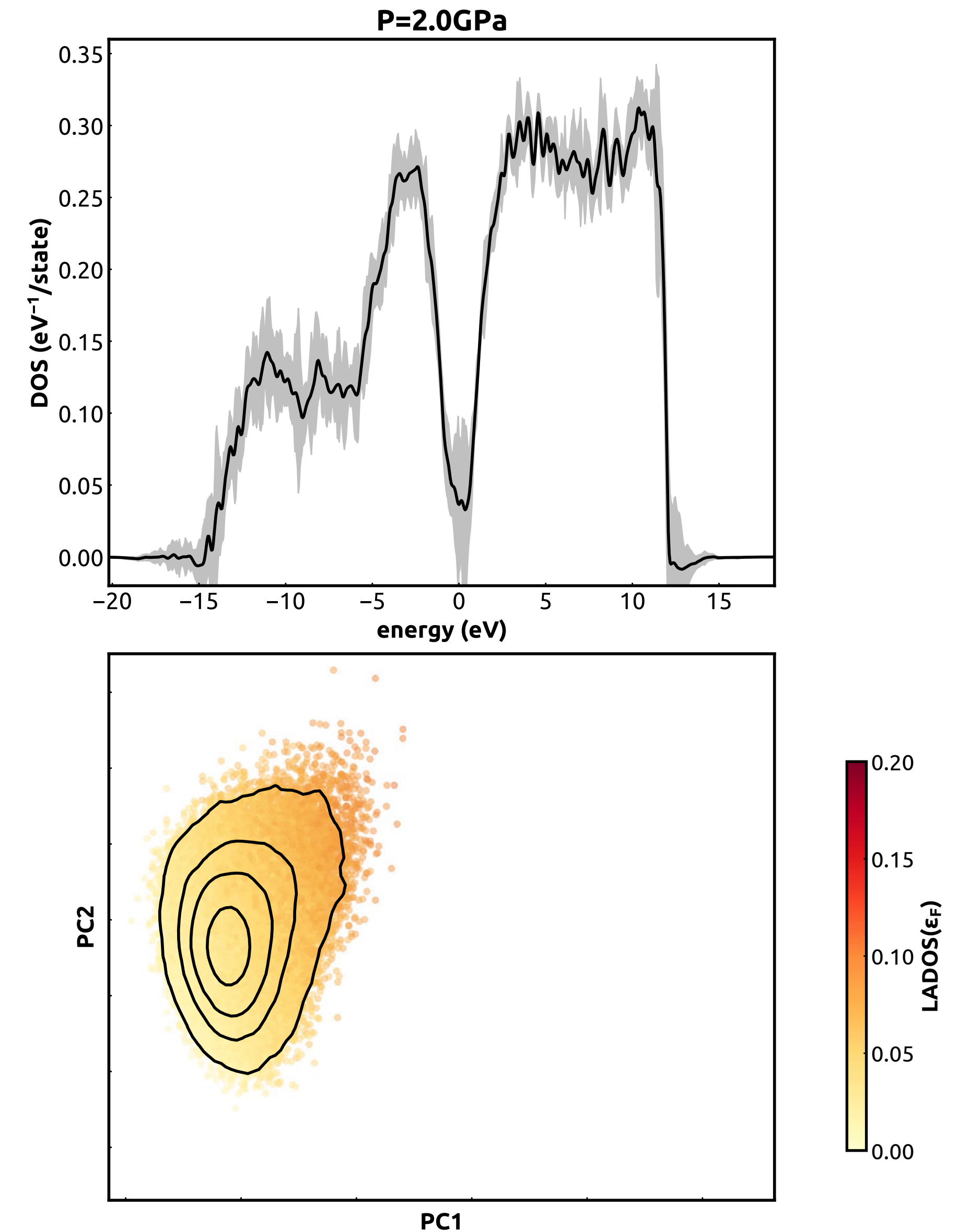
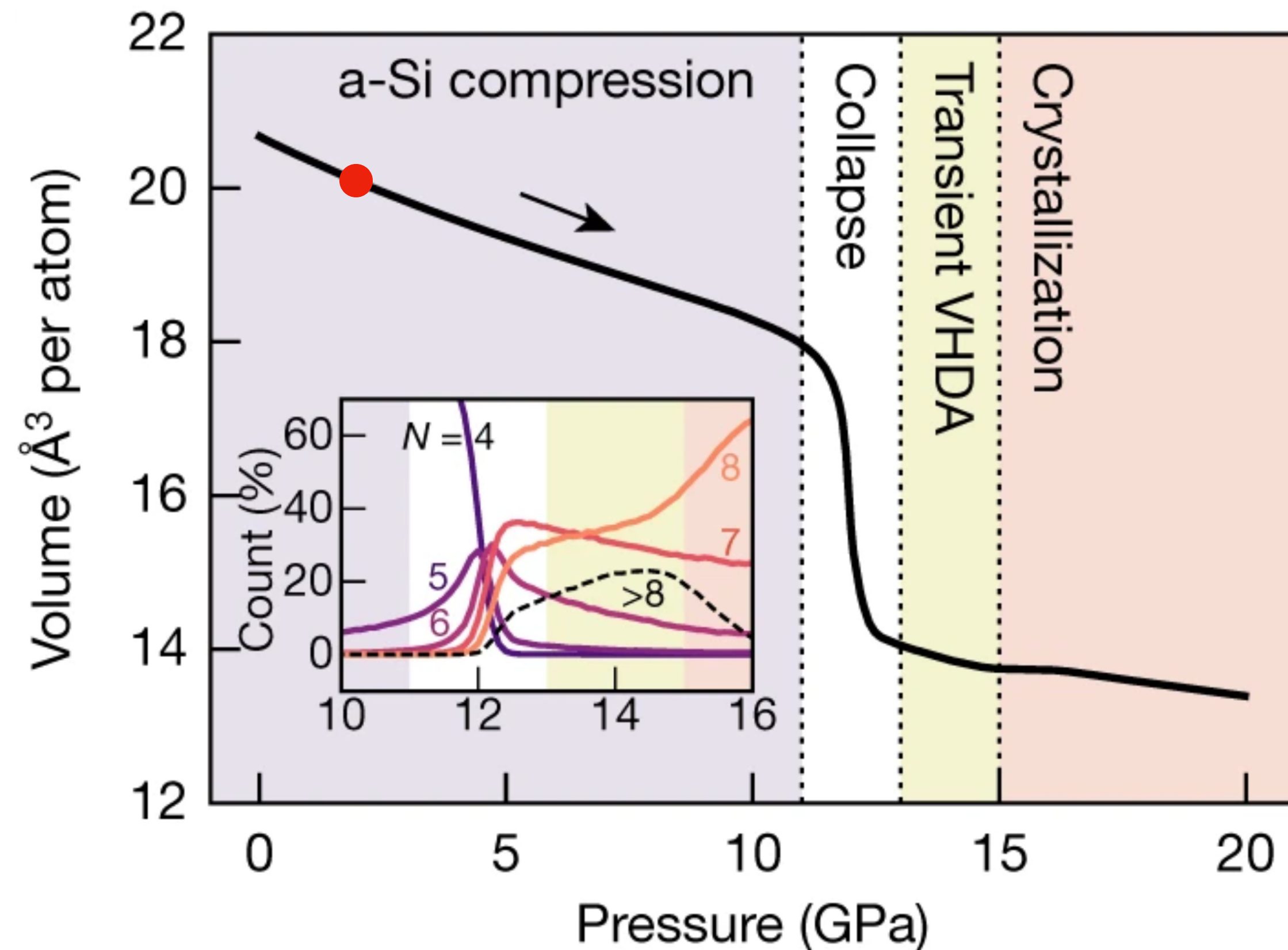
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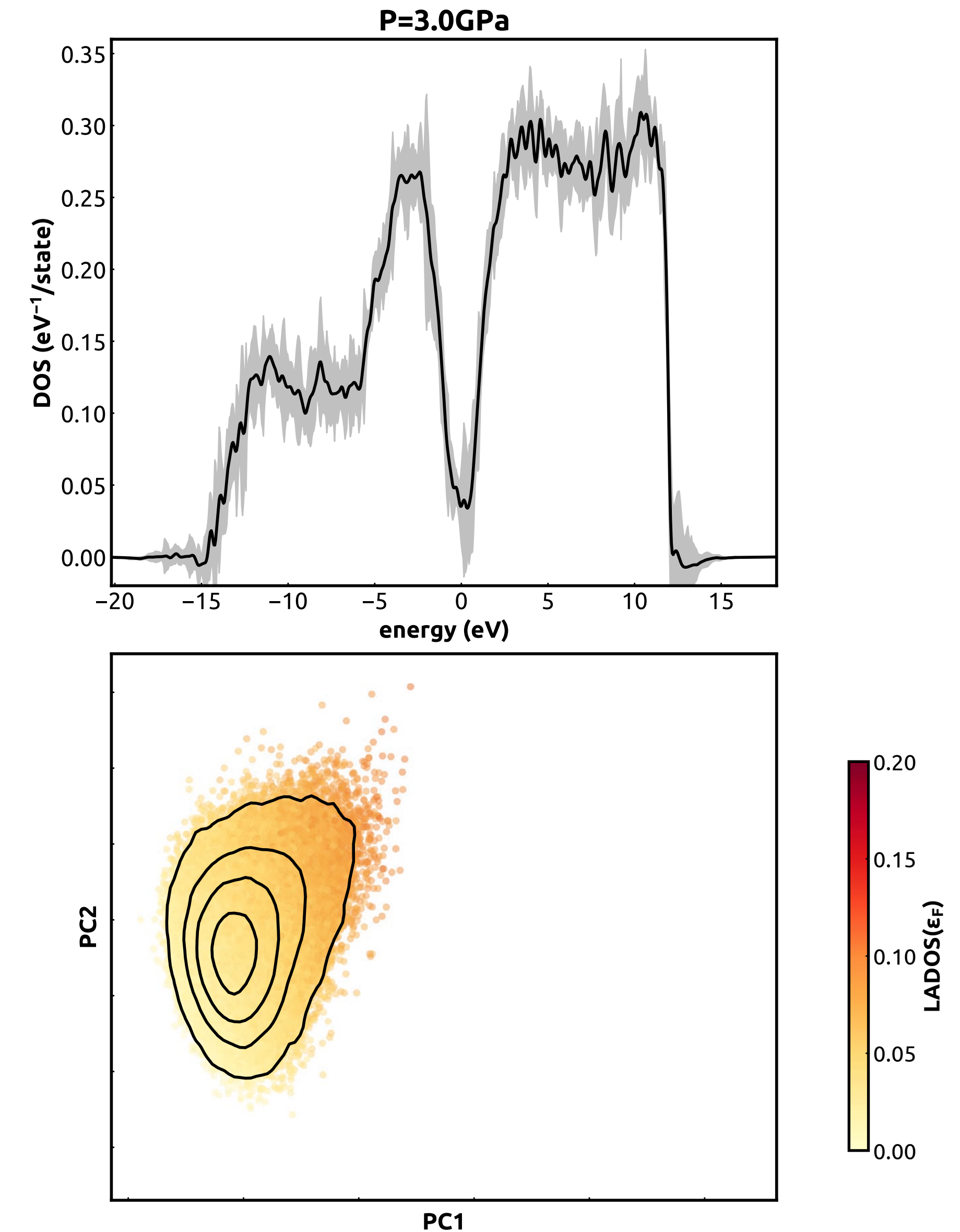
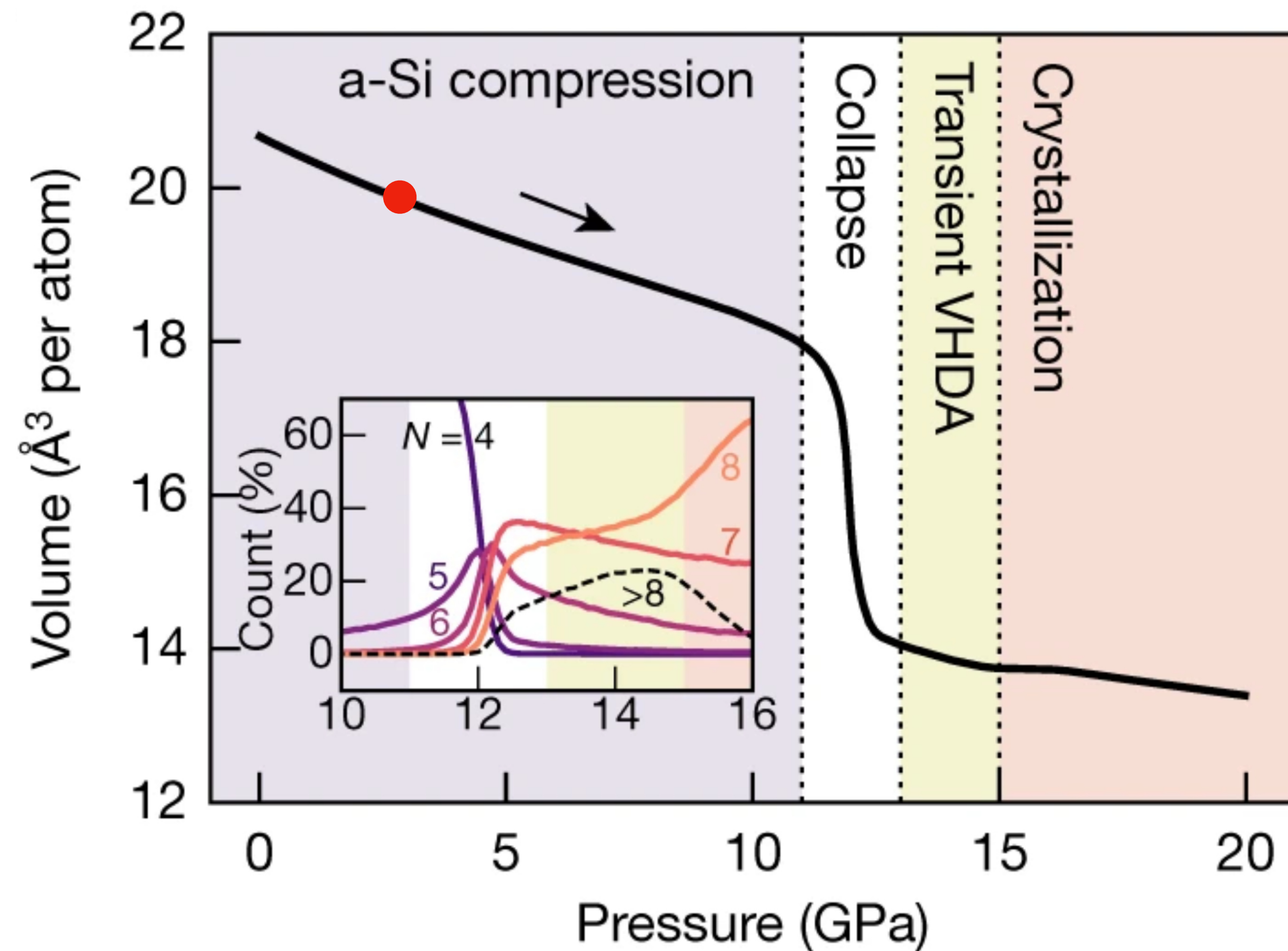
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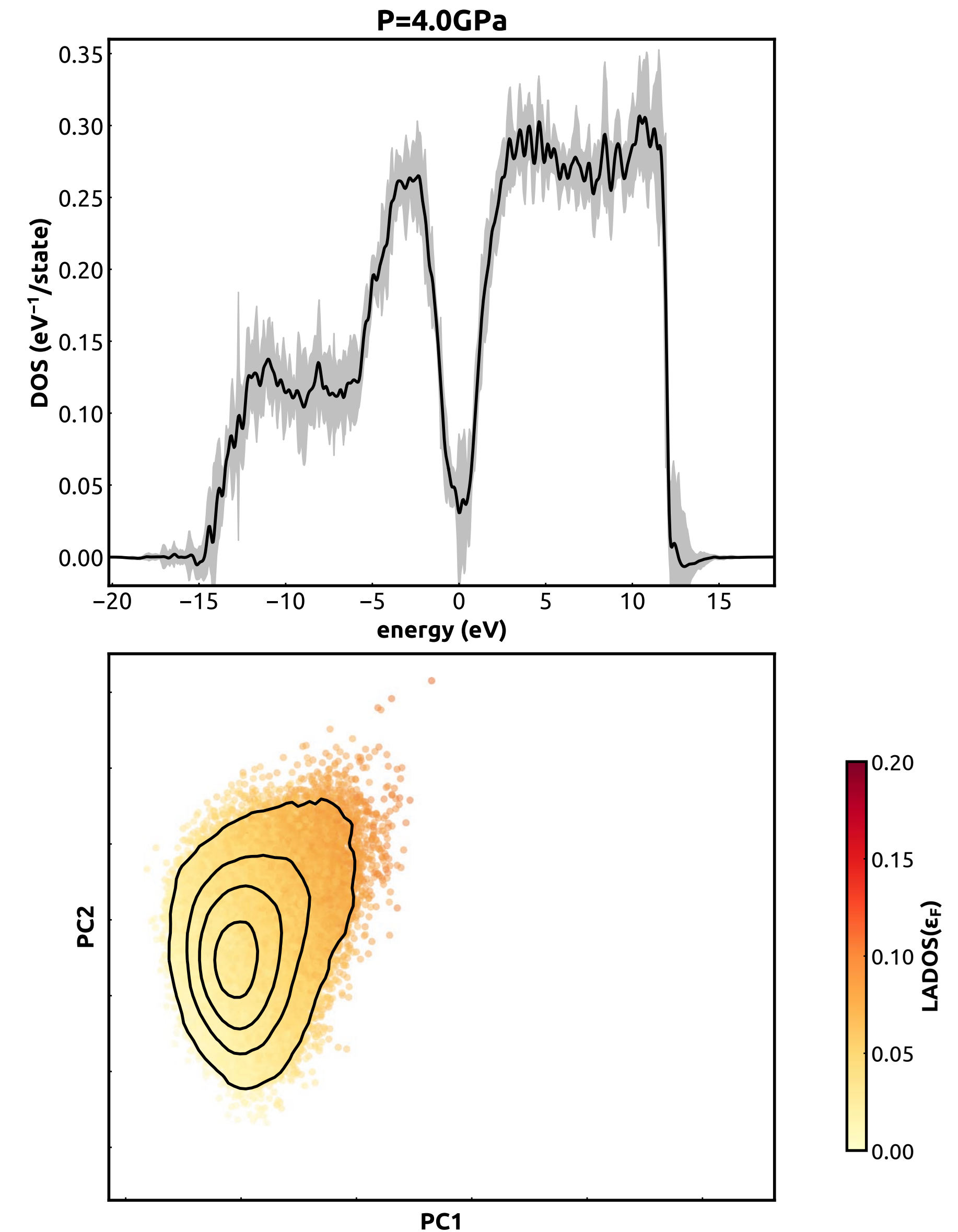
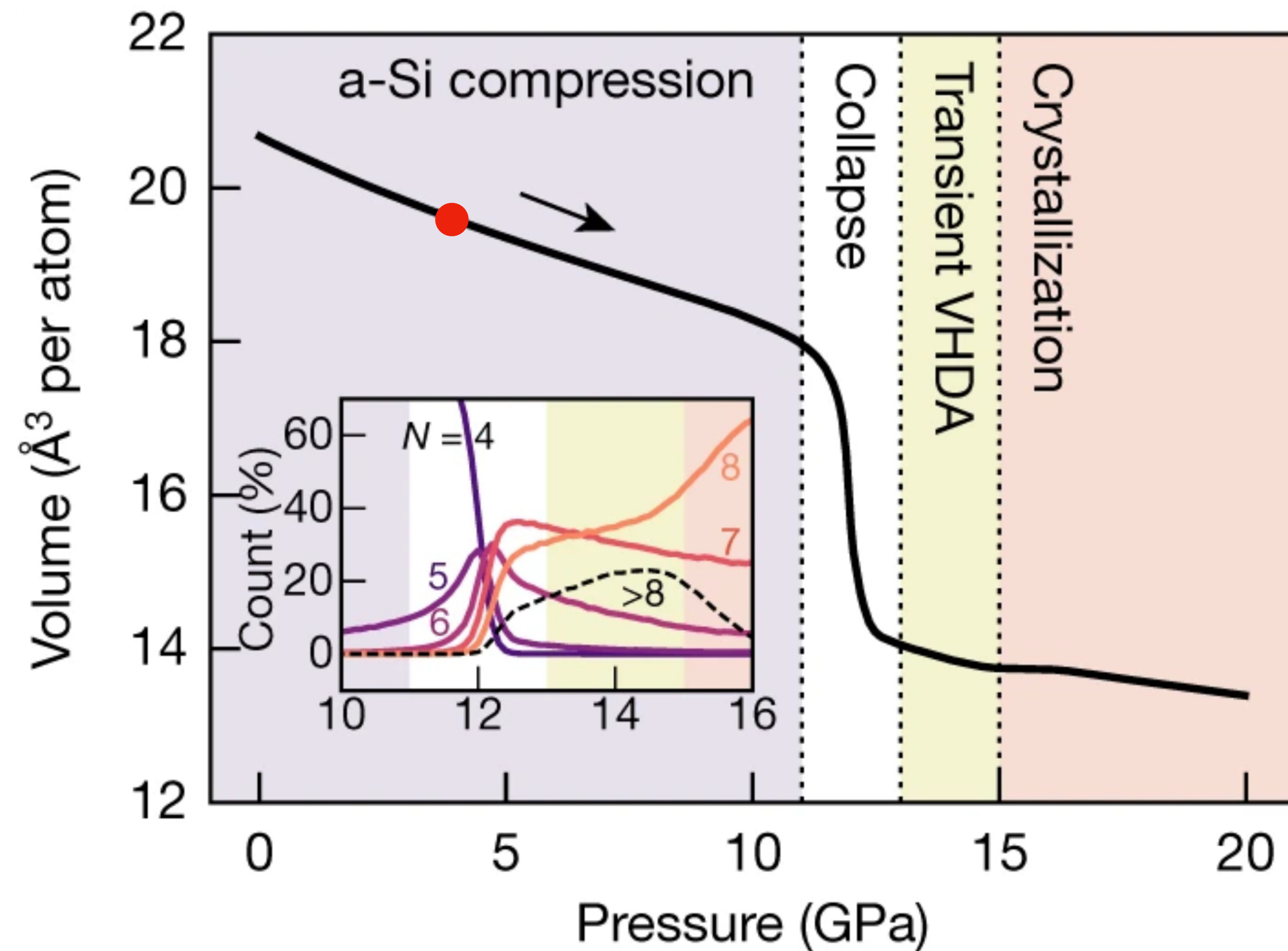
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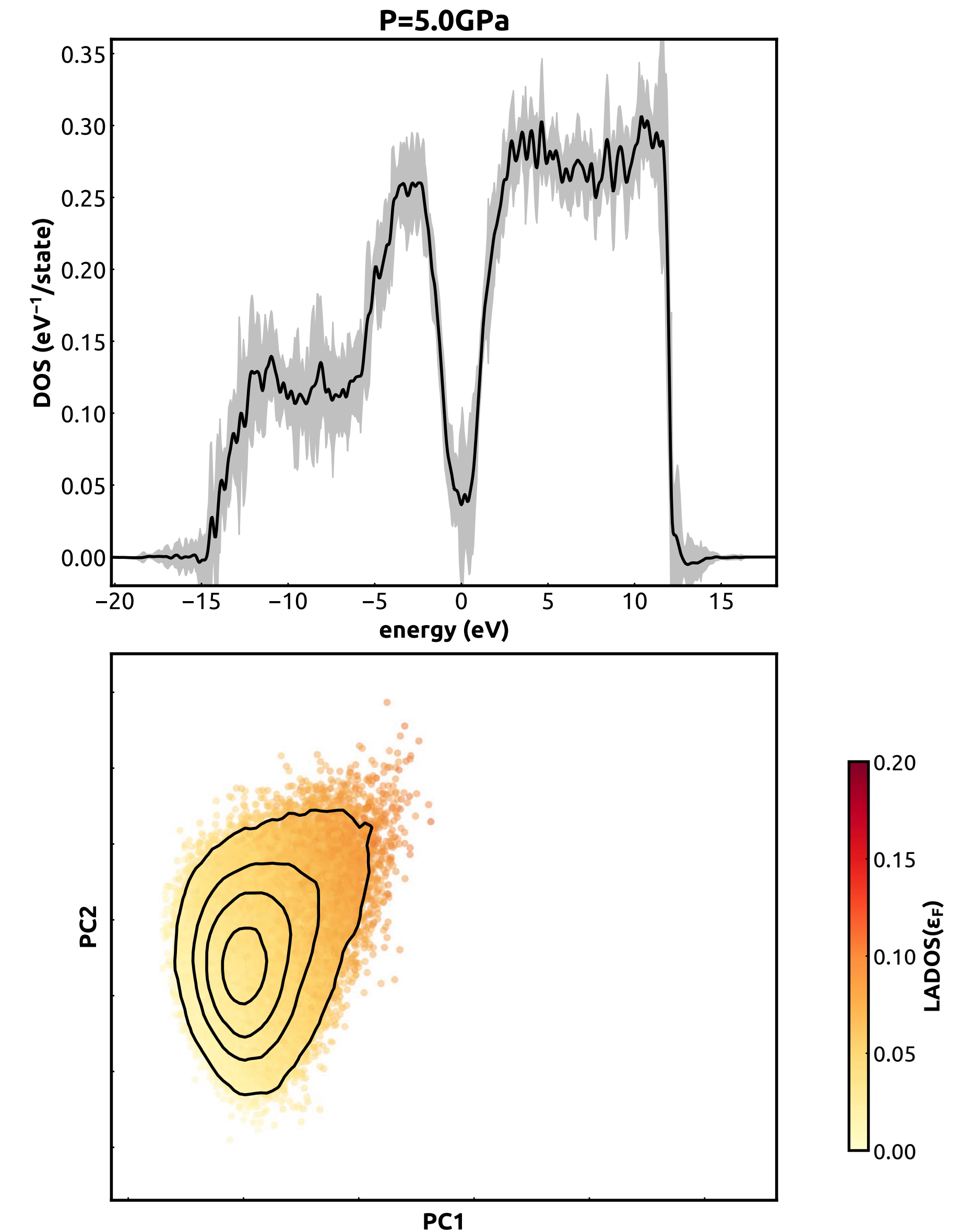
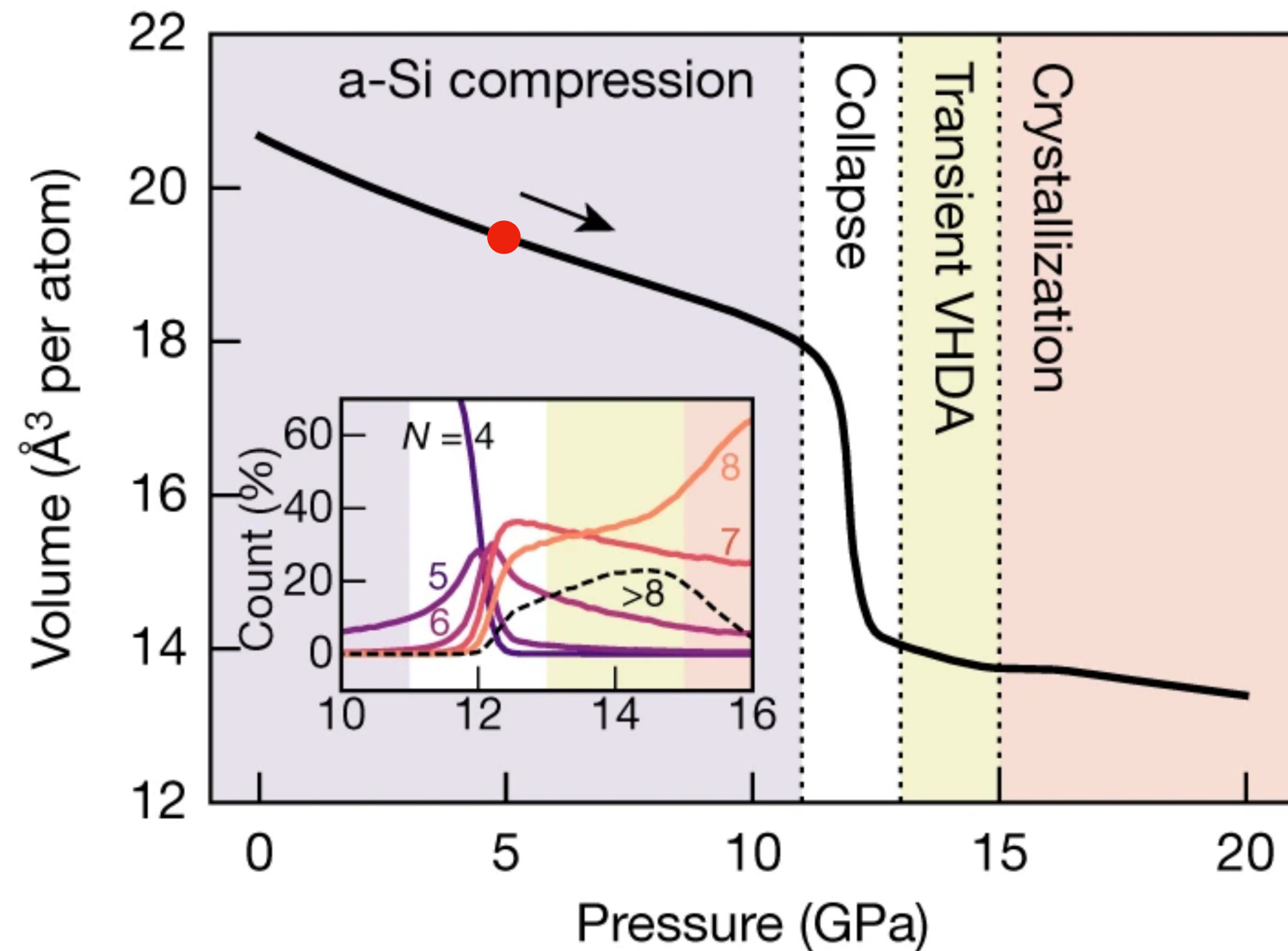
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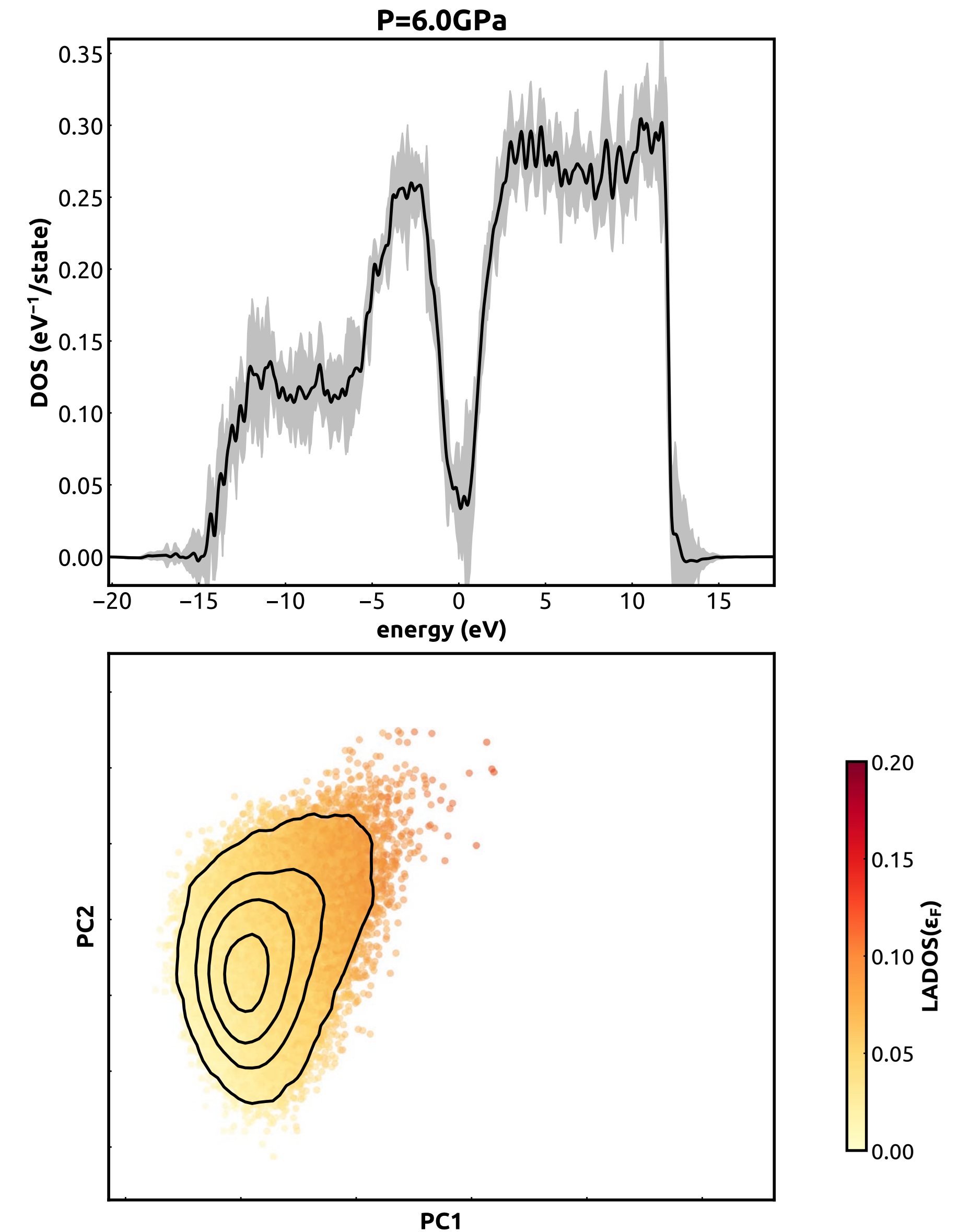
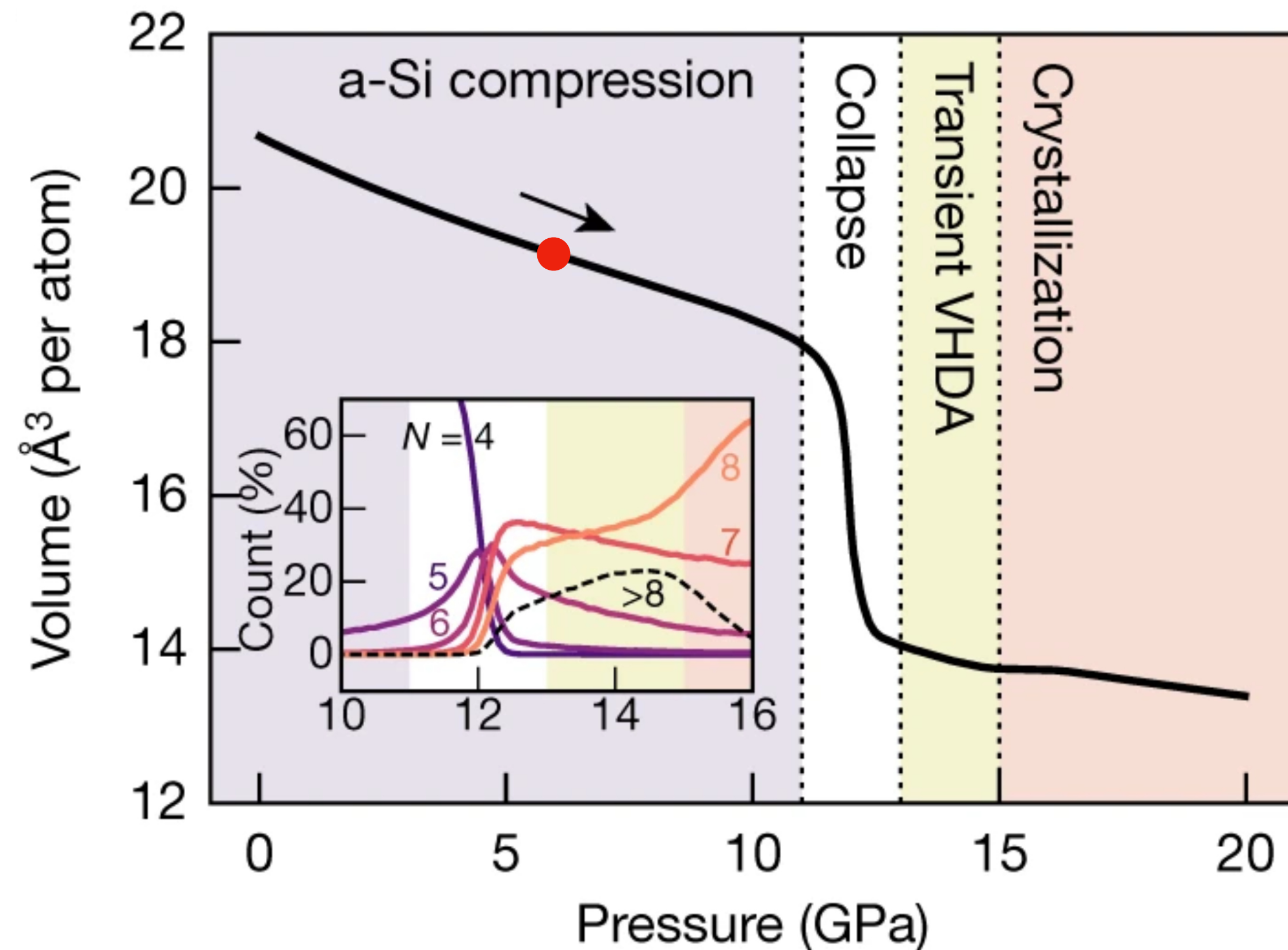
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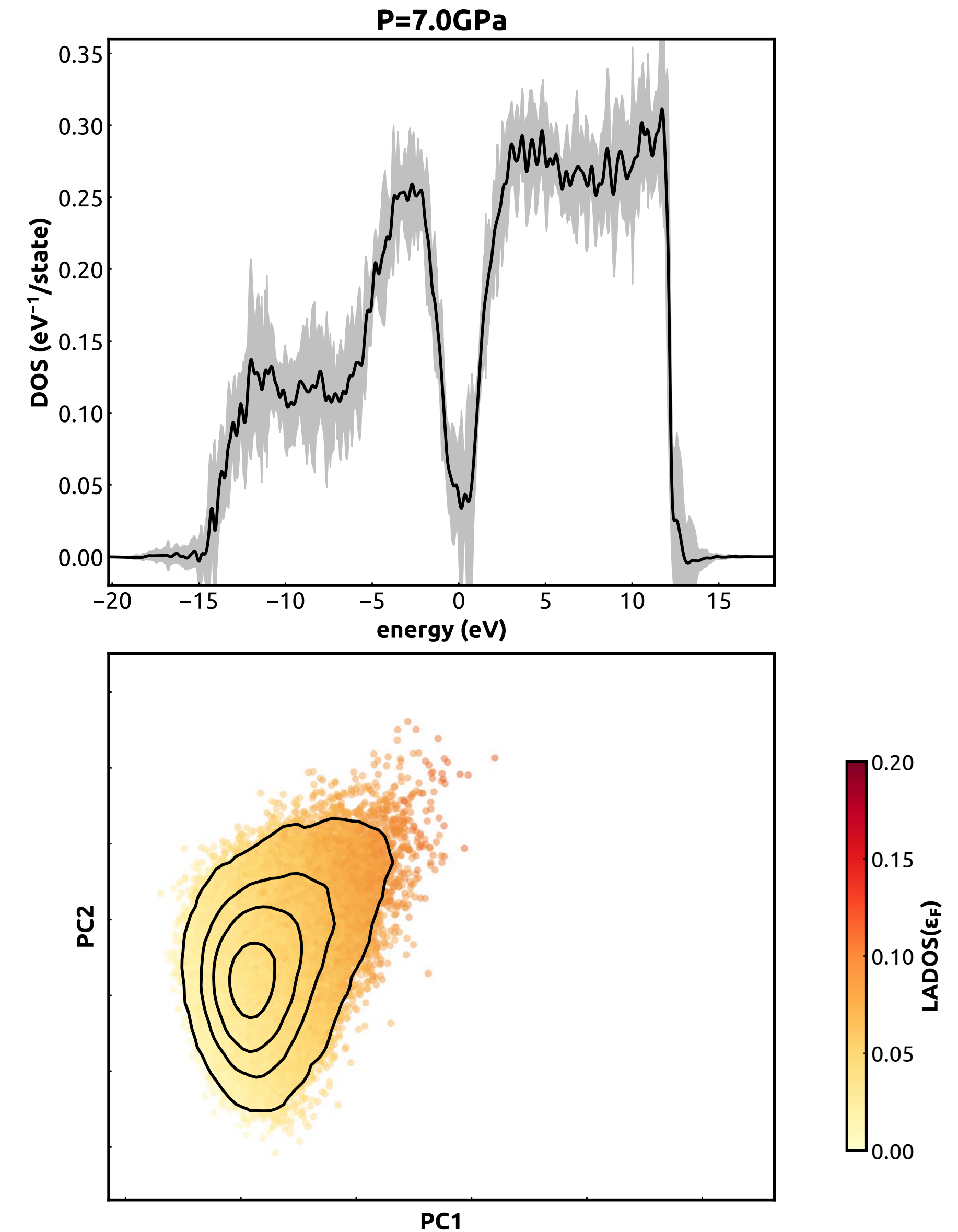
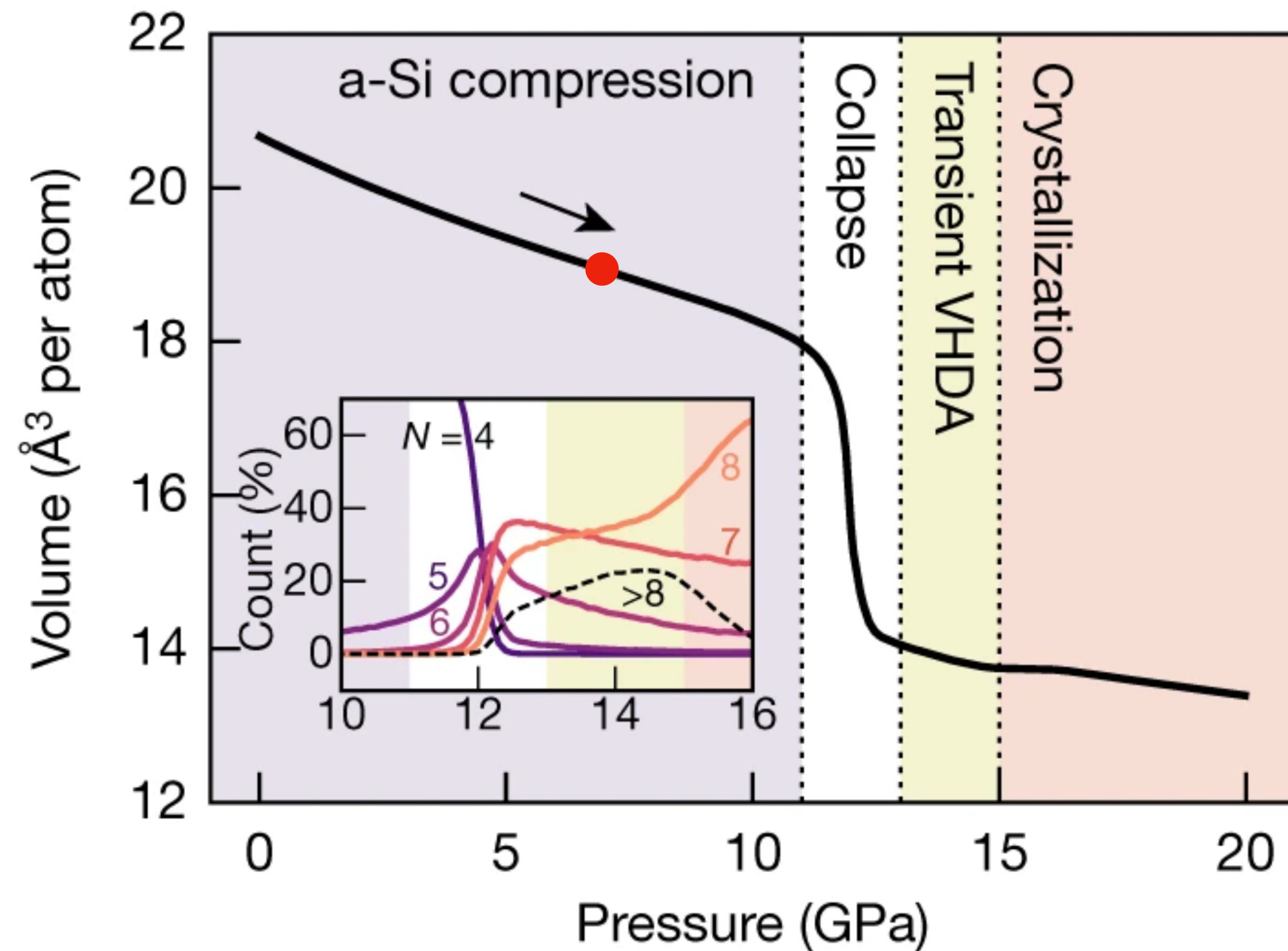
Combining structural and electronic fingerprints



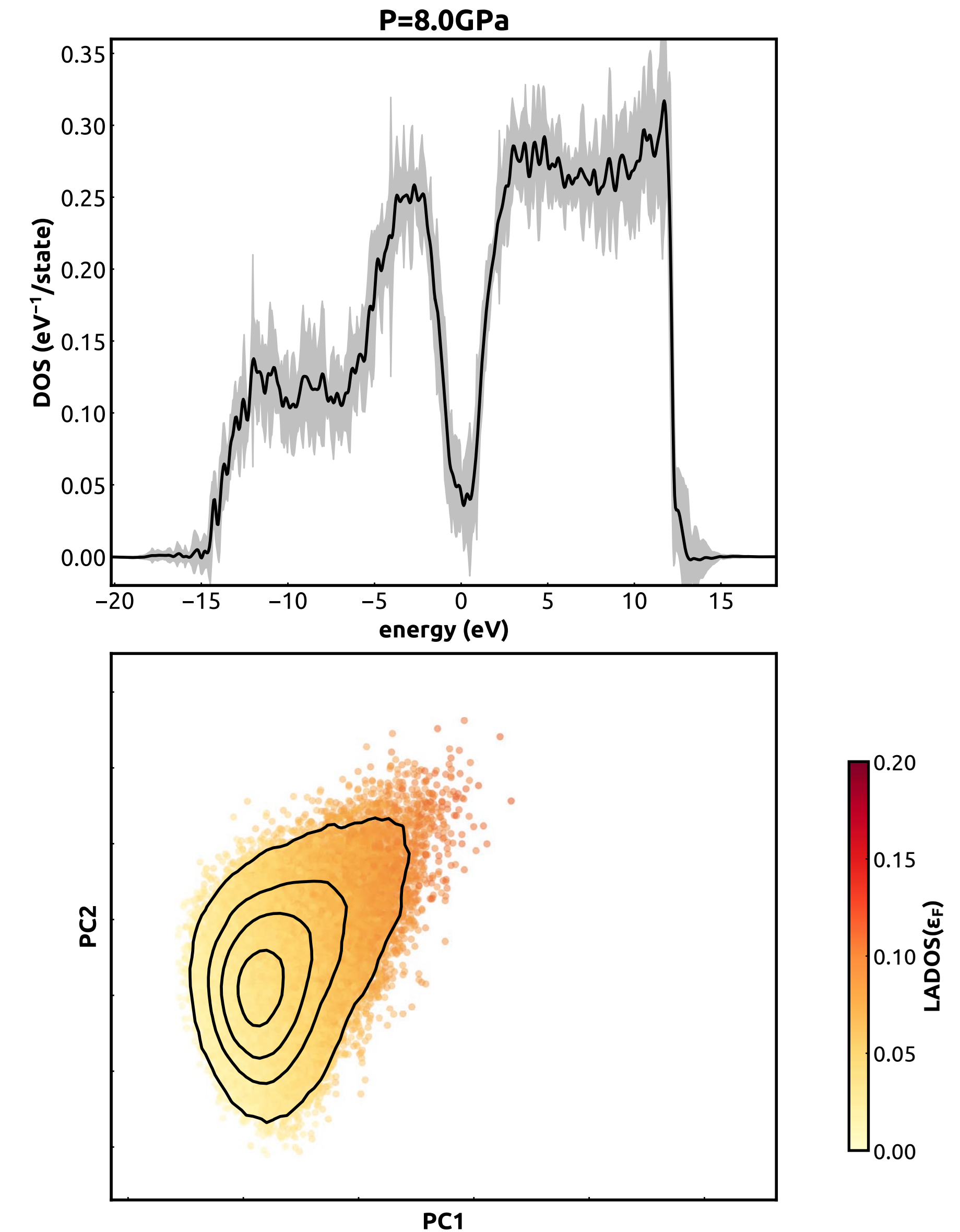
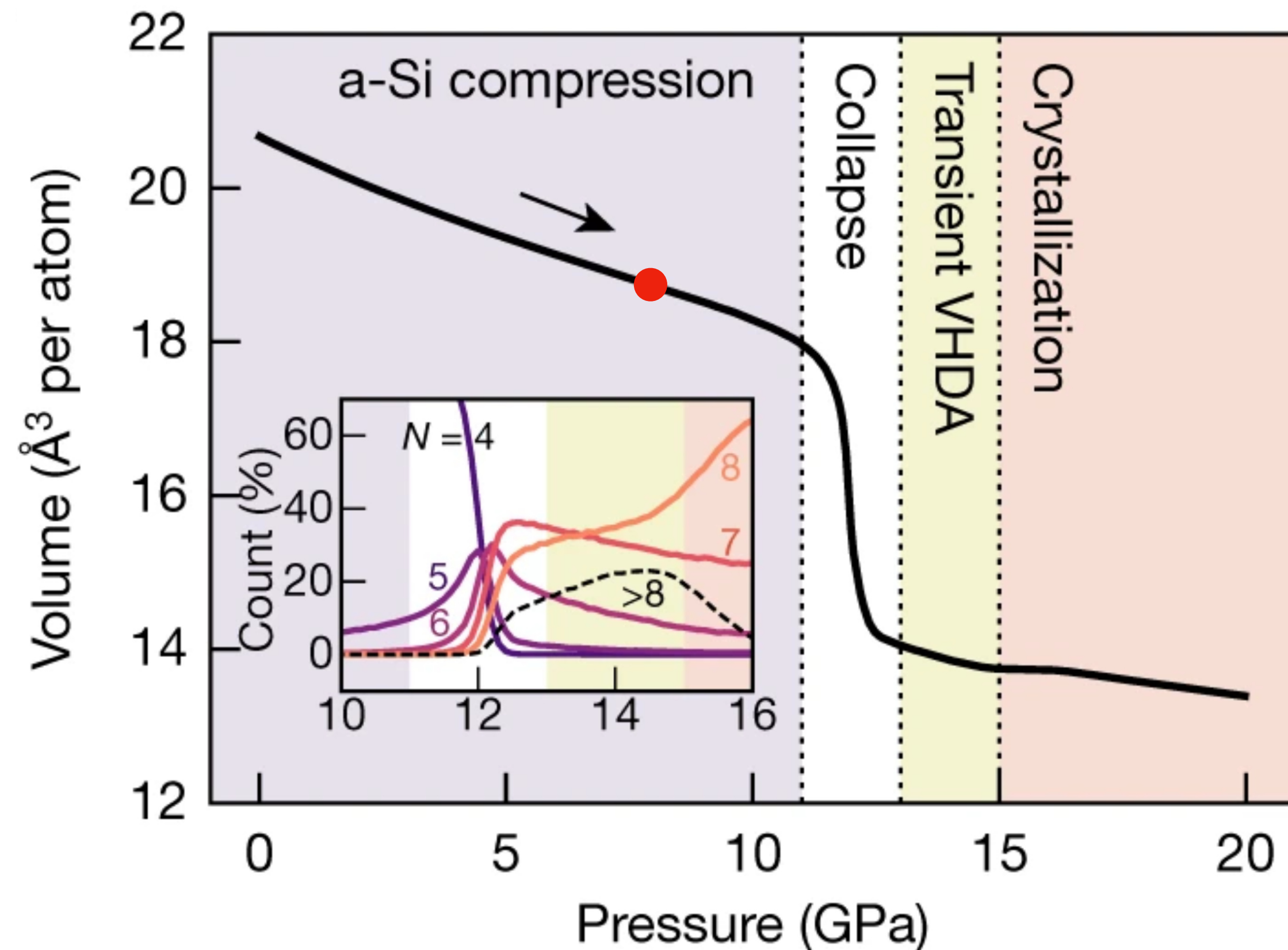
Combining structural and electronic fingerprints



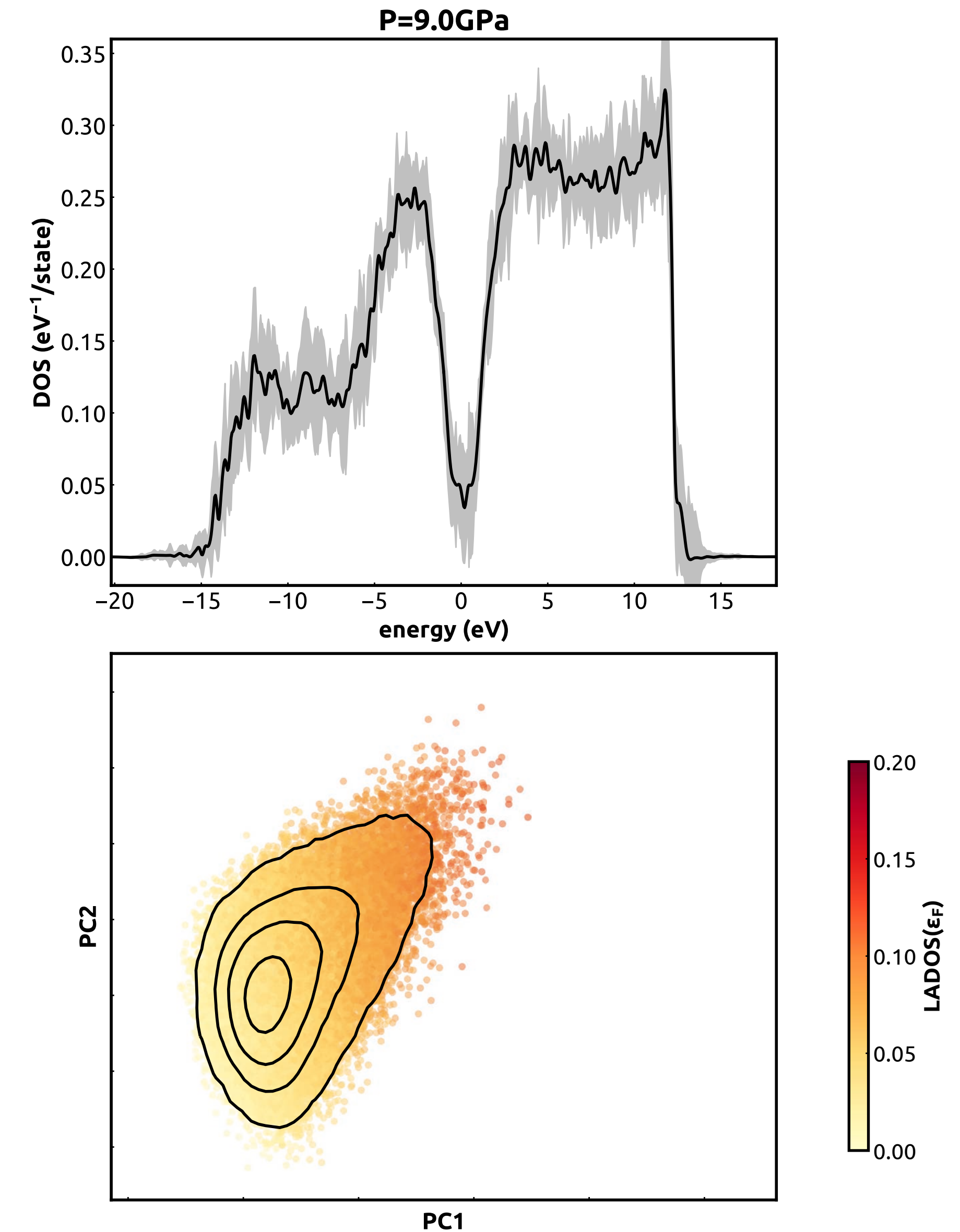
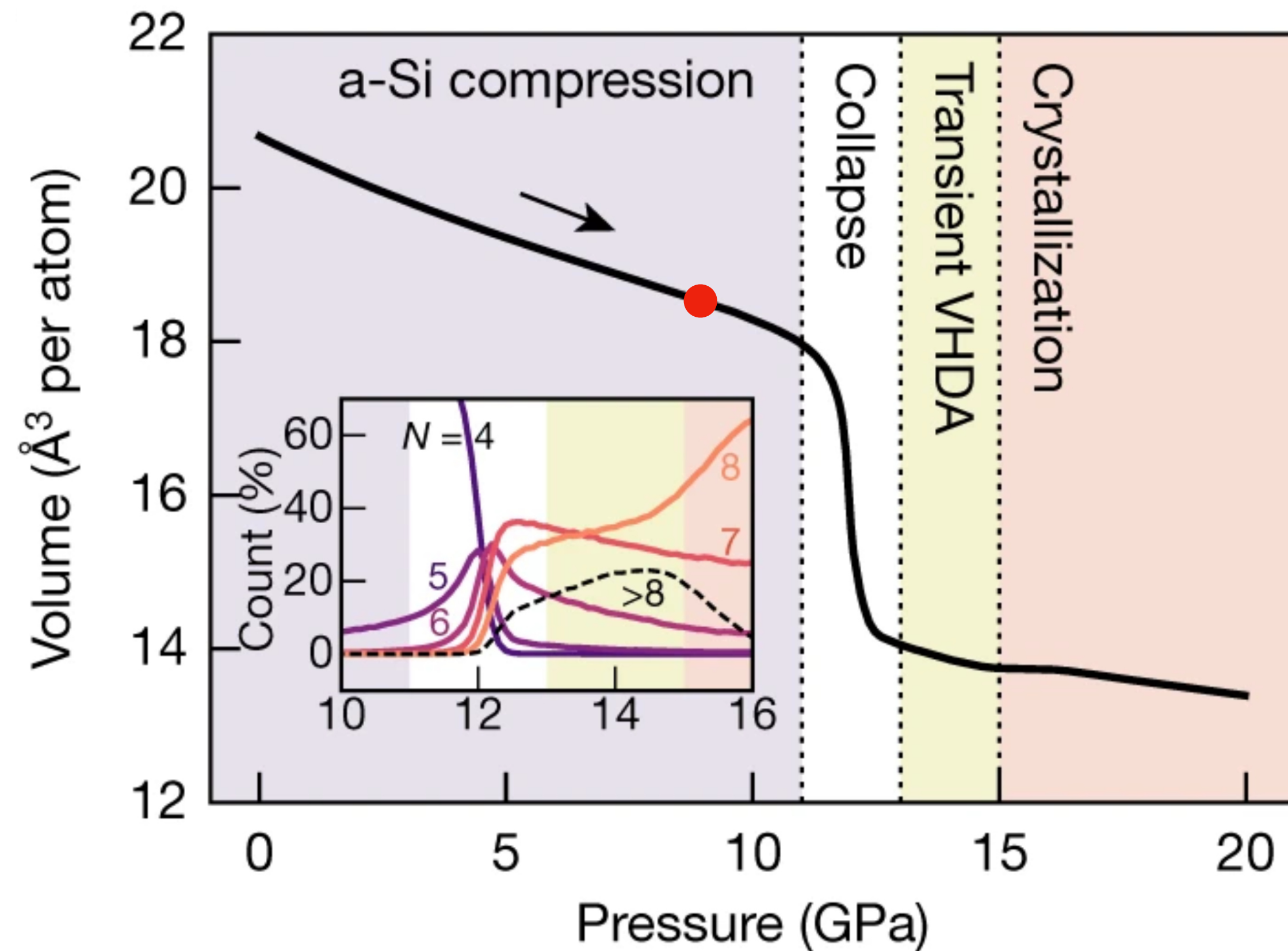
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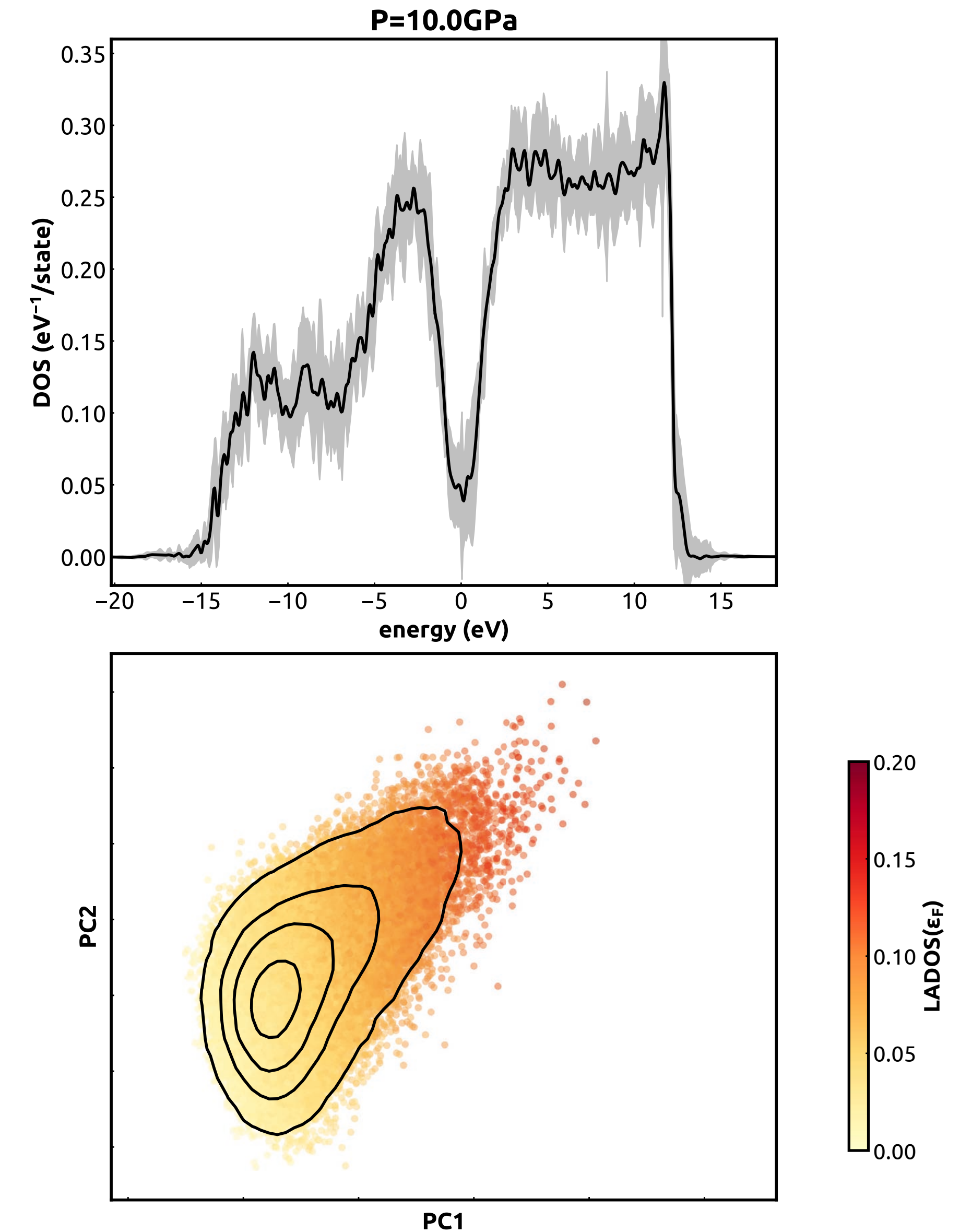
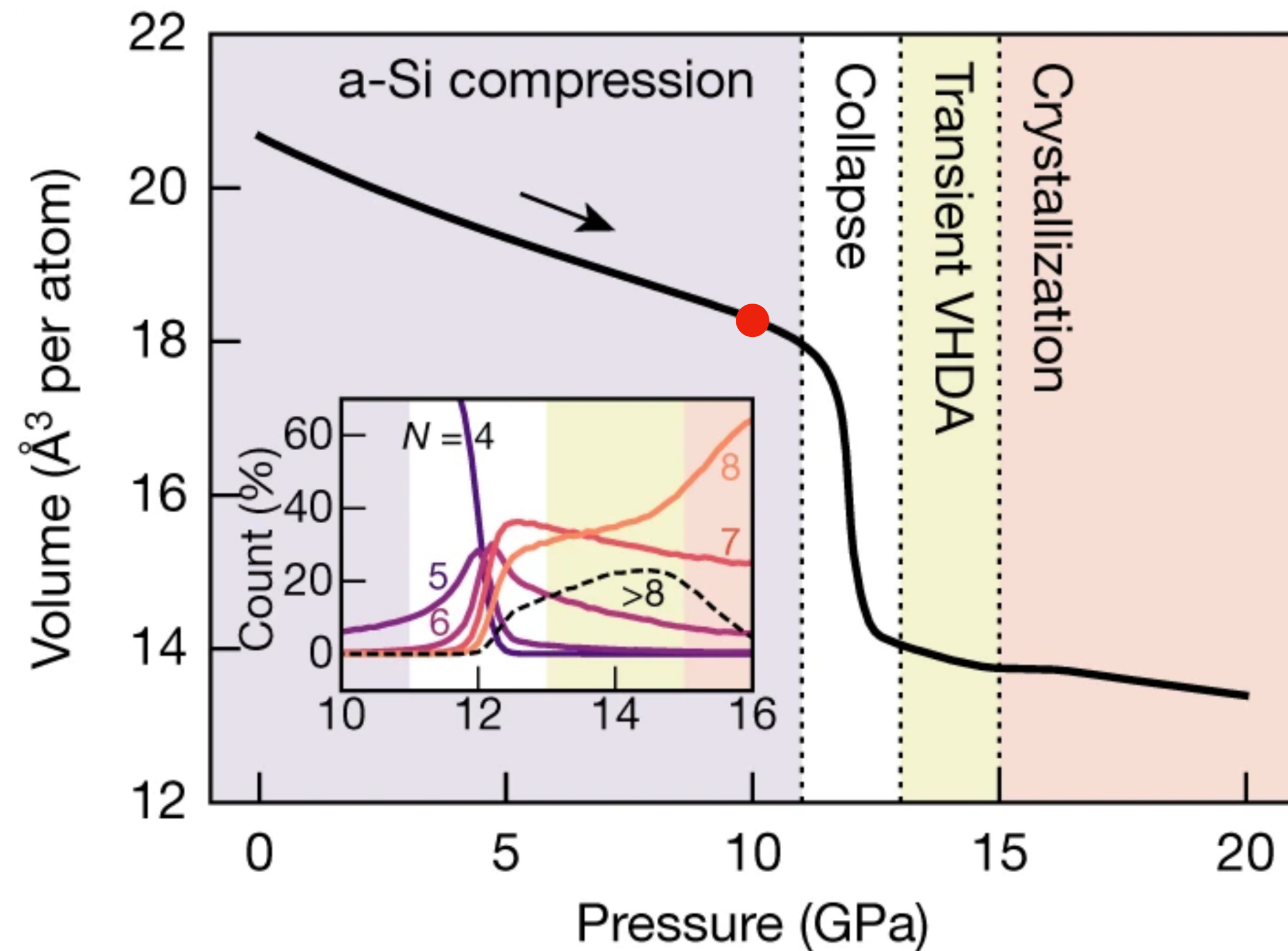
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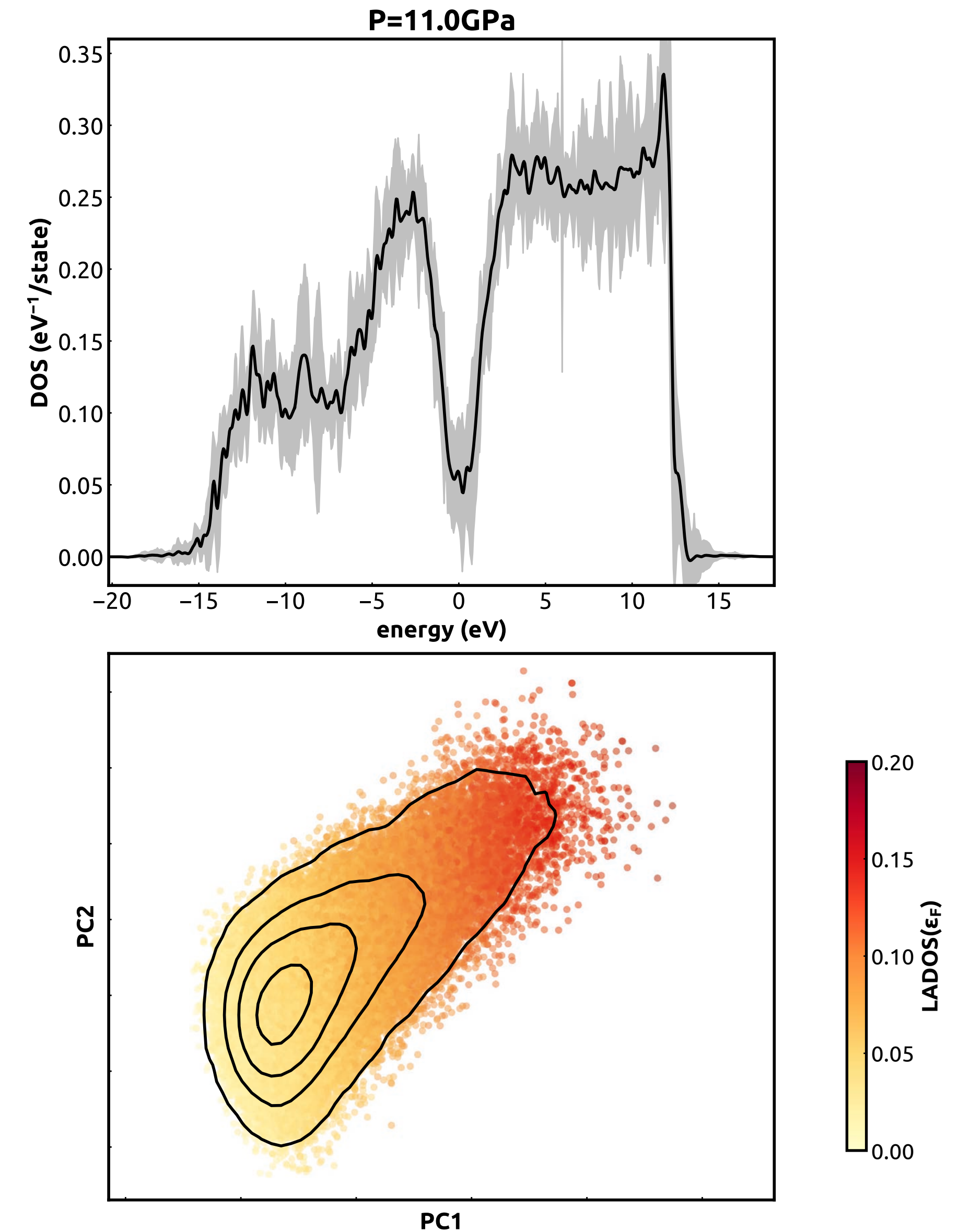
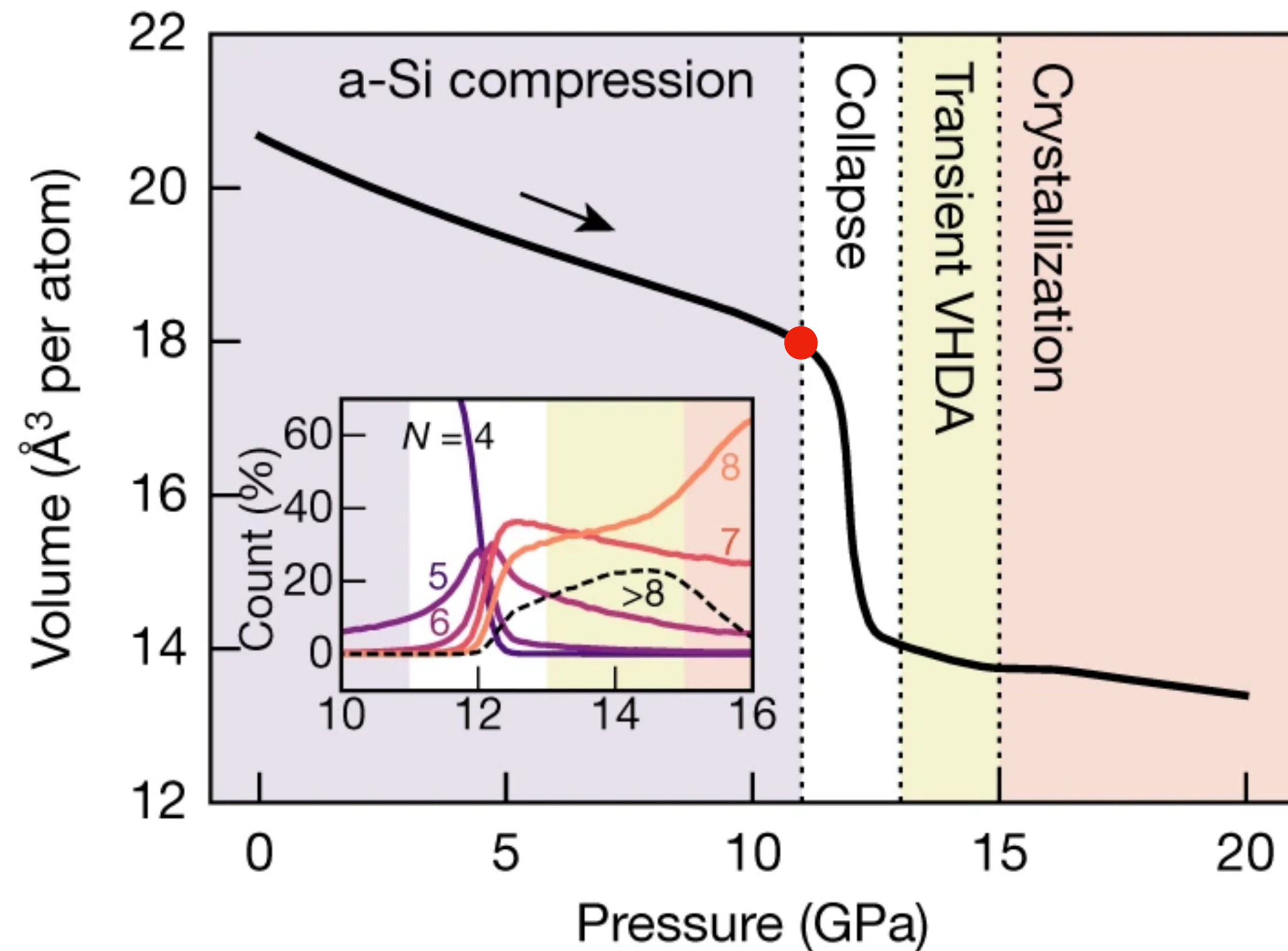
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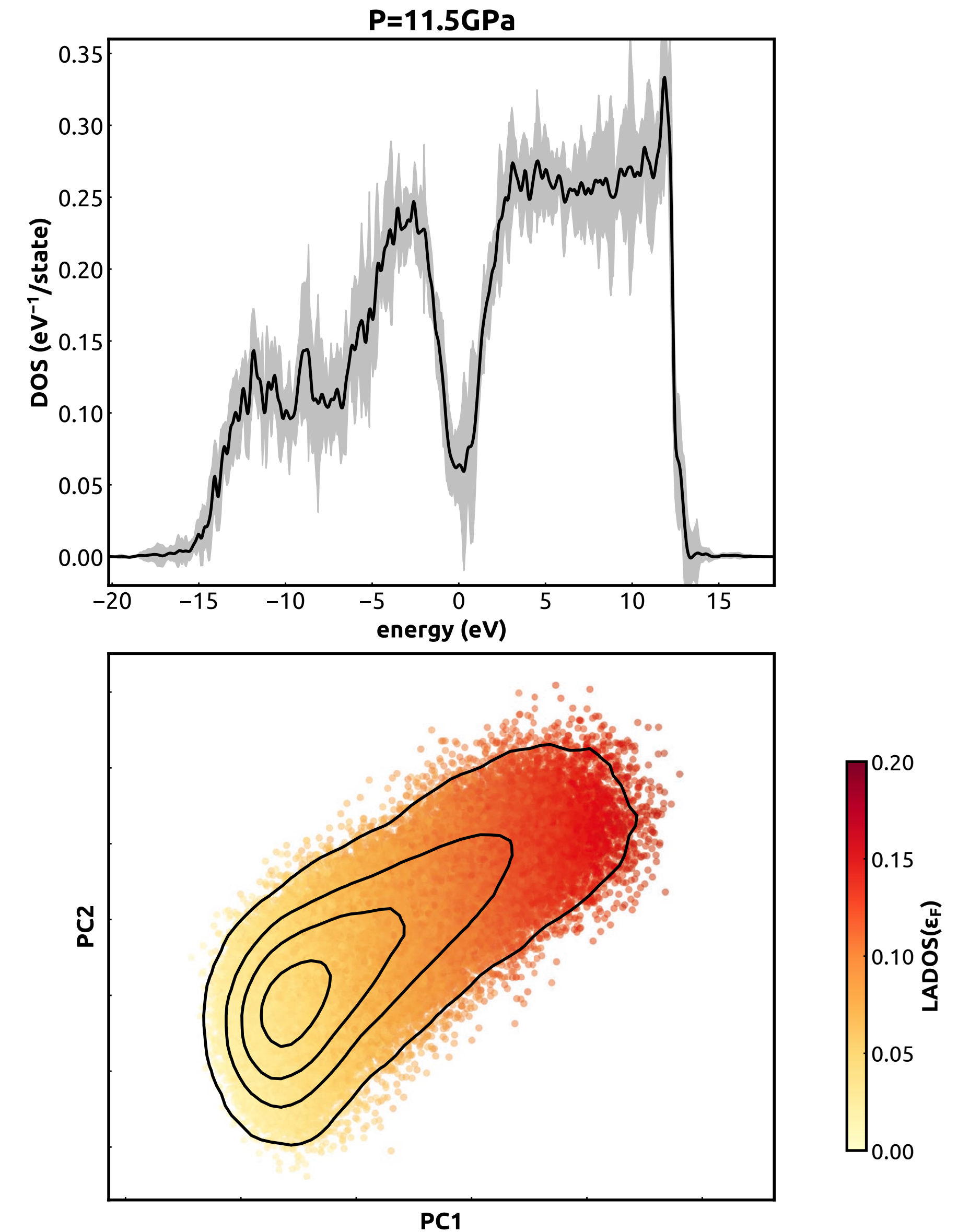
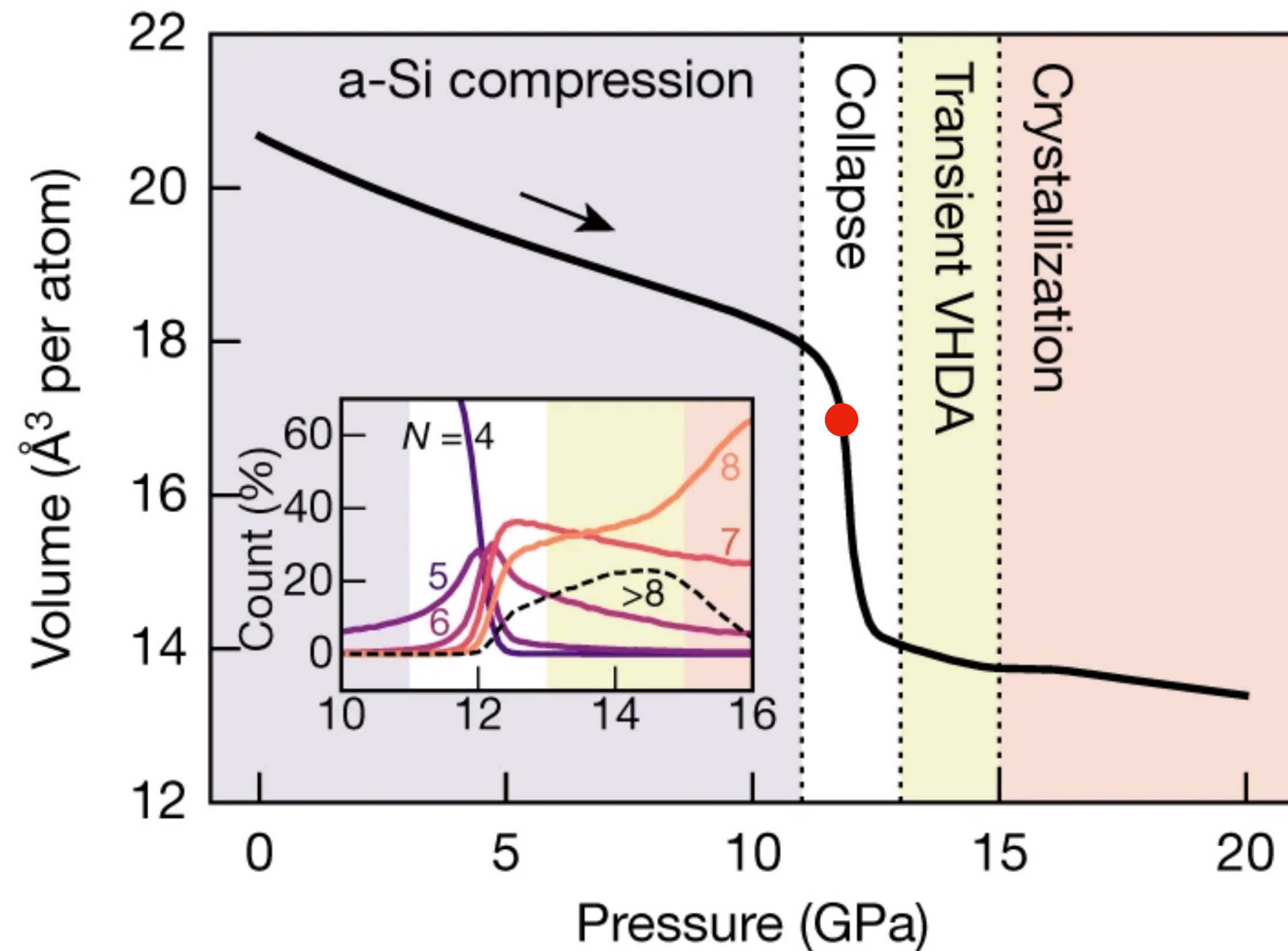
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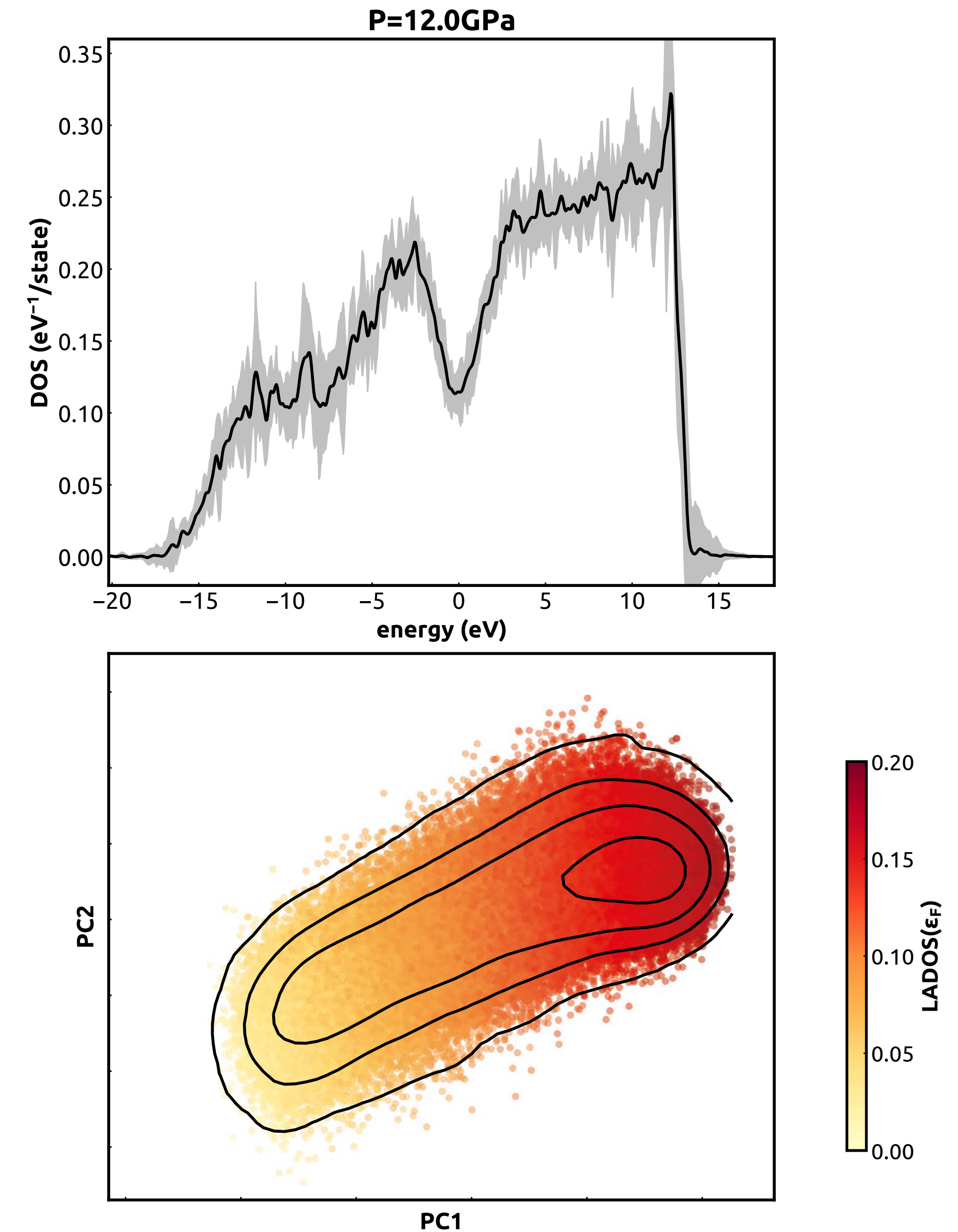
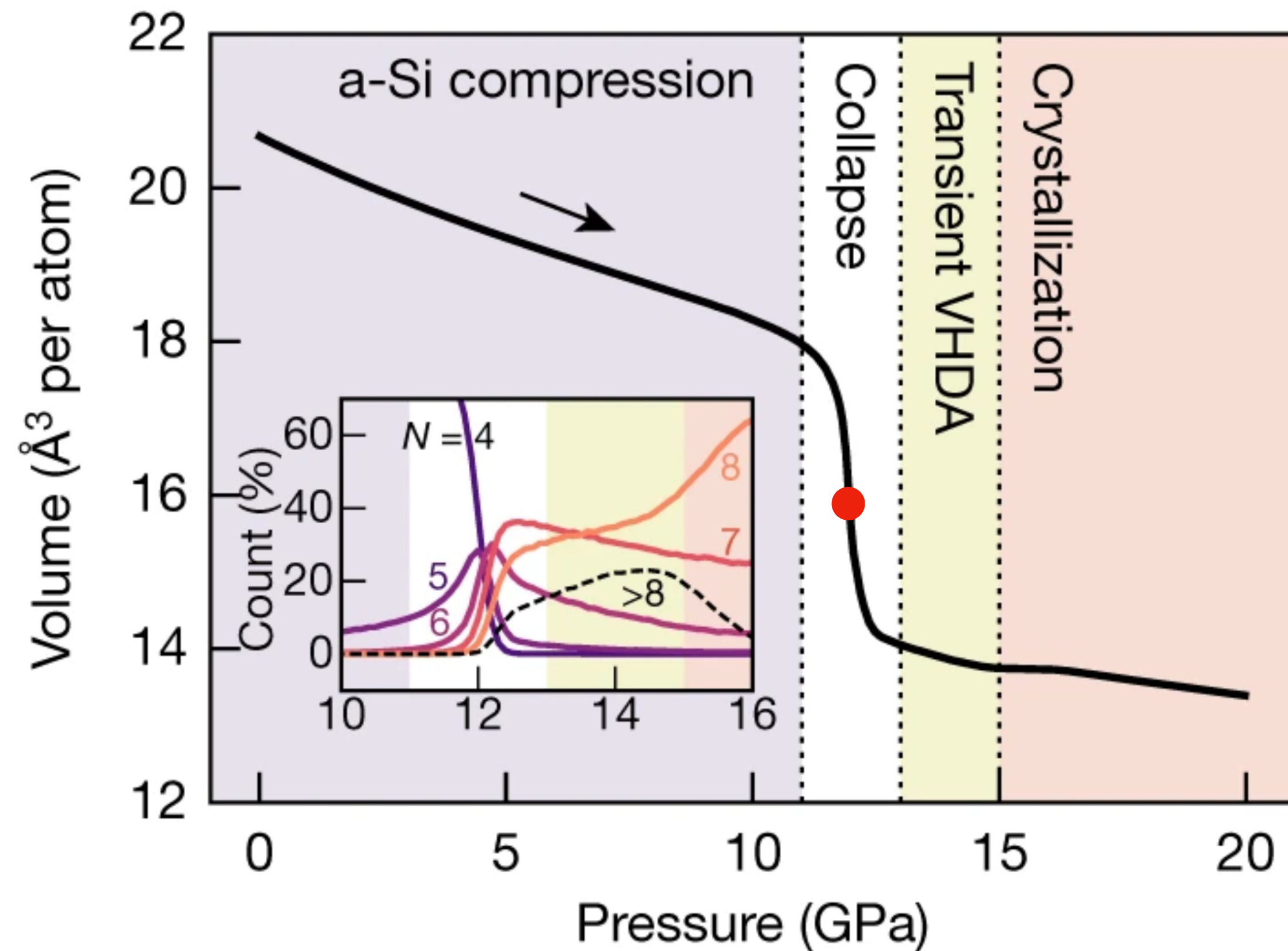
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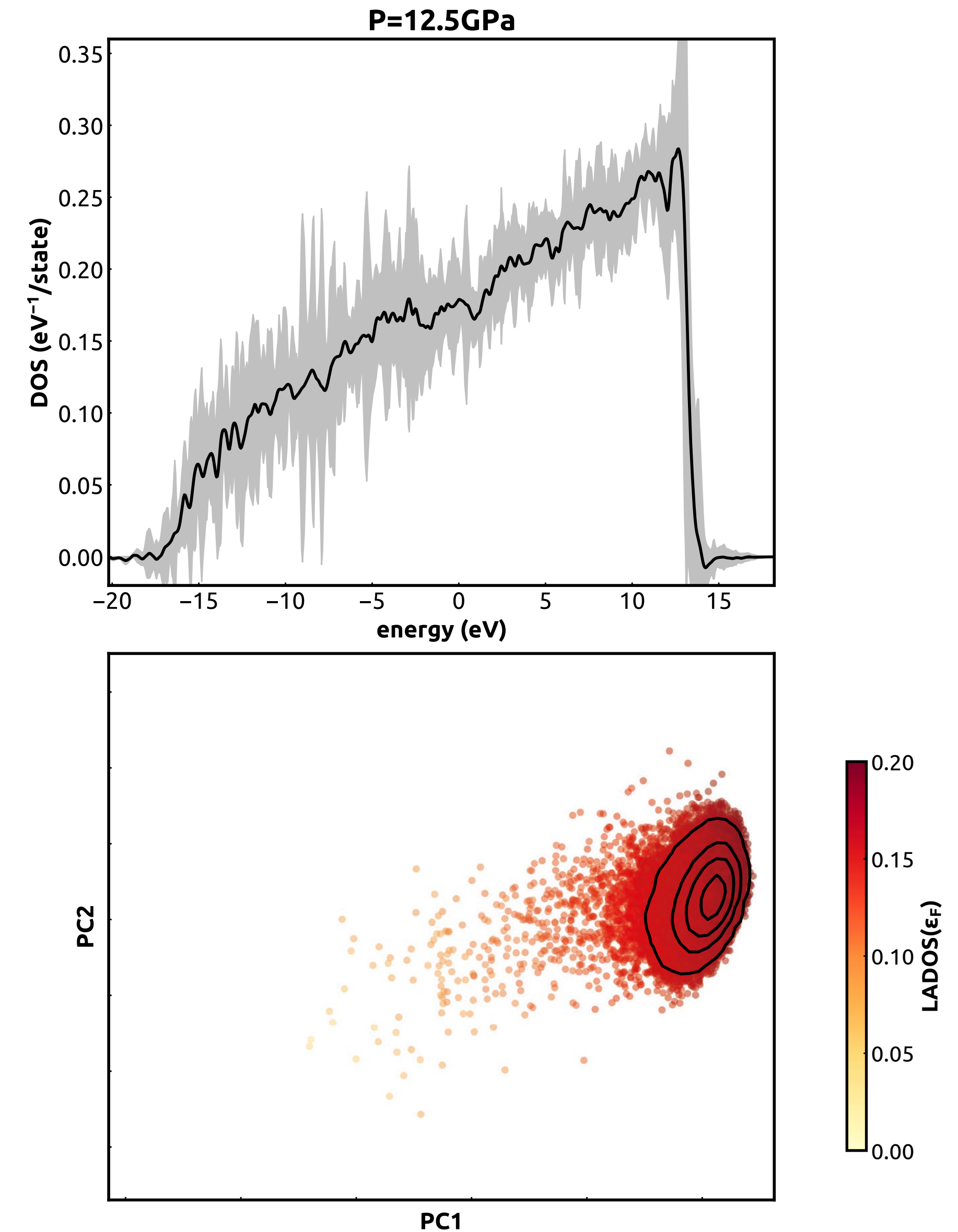
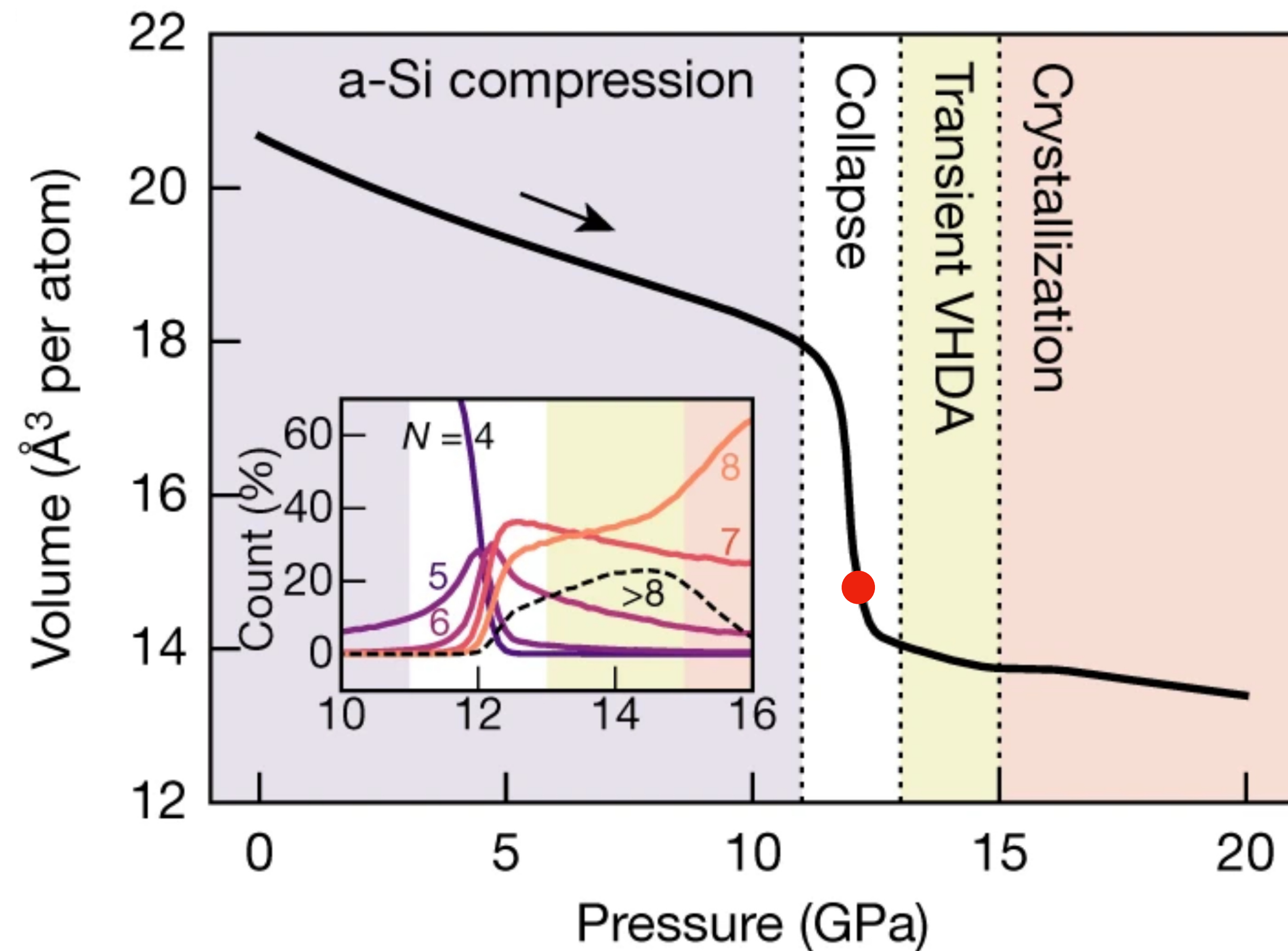
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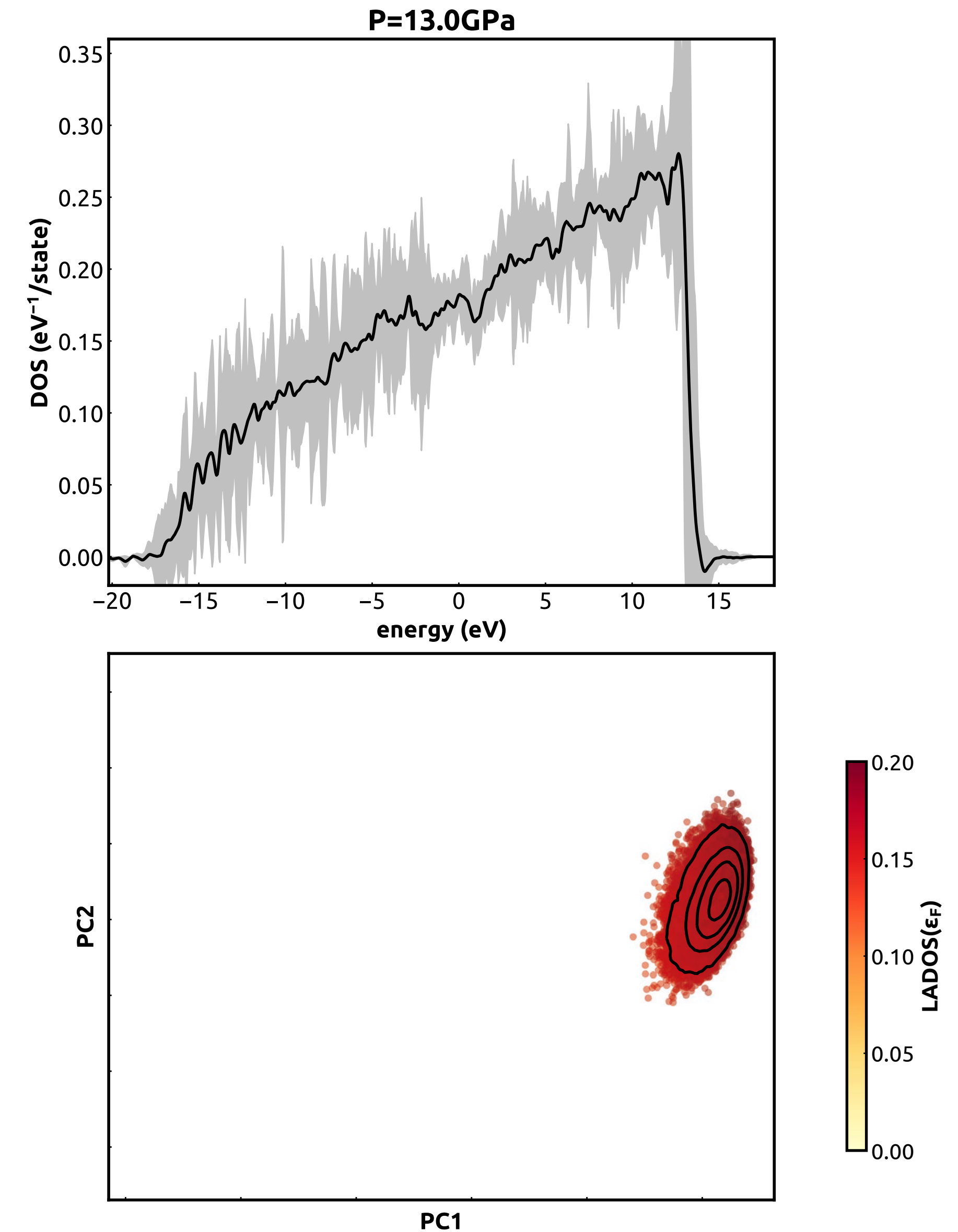
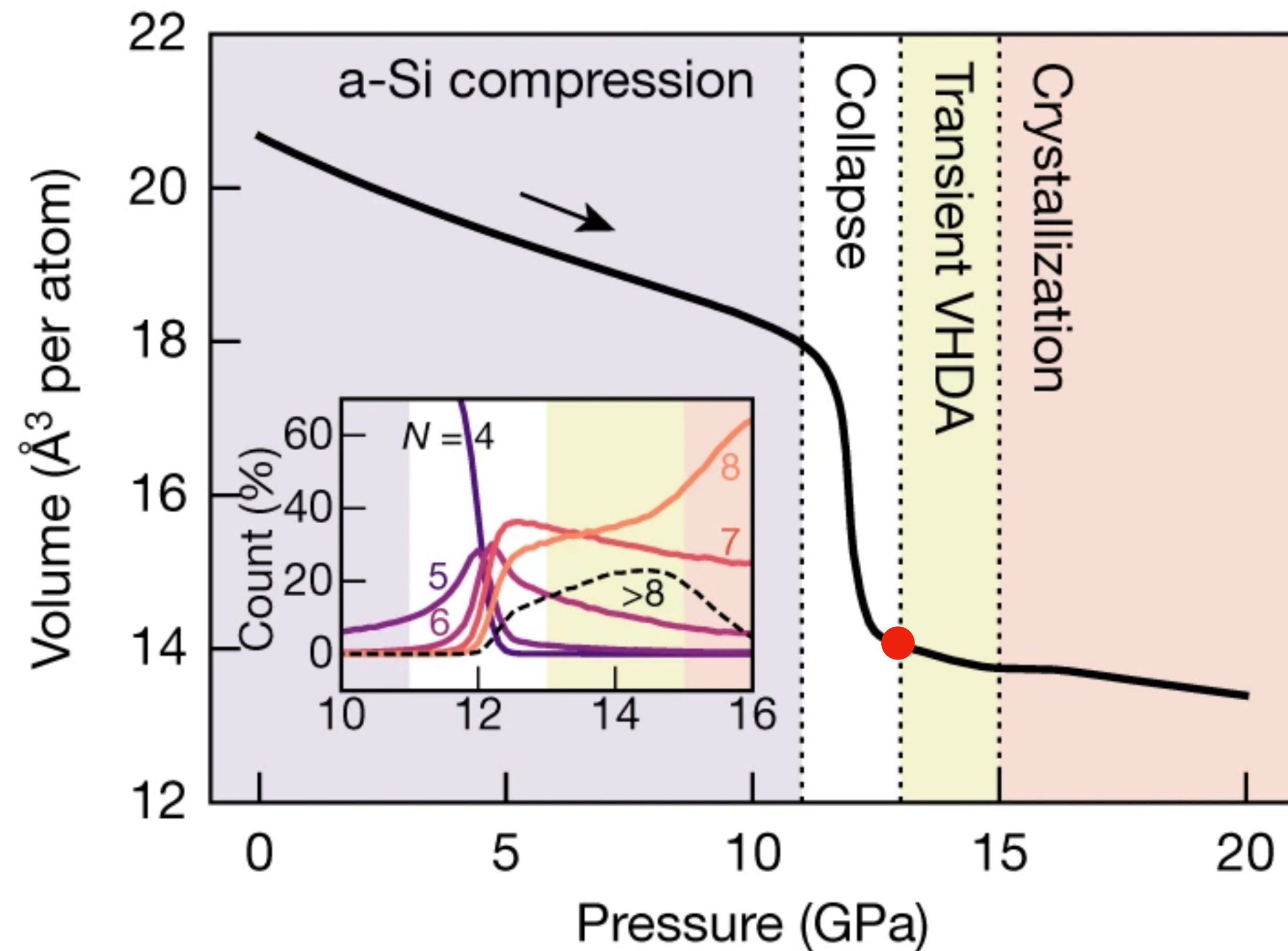
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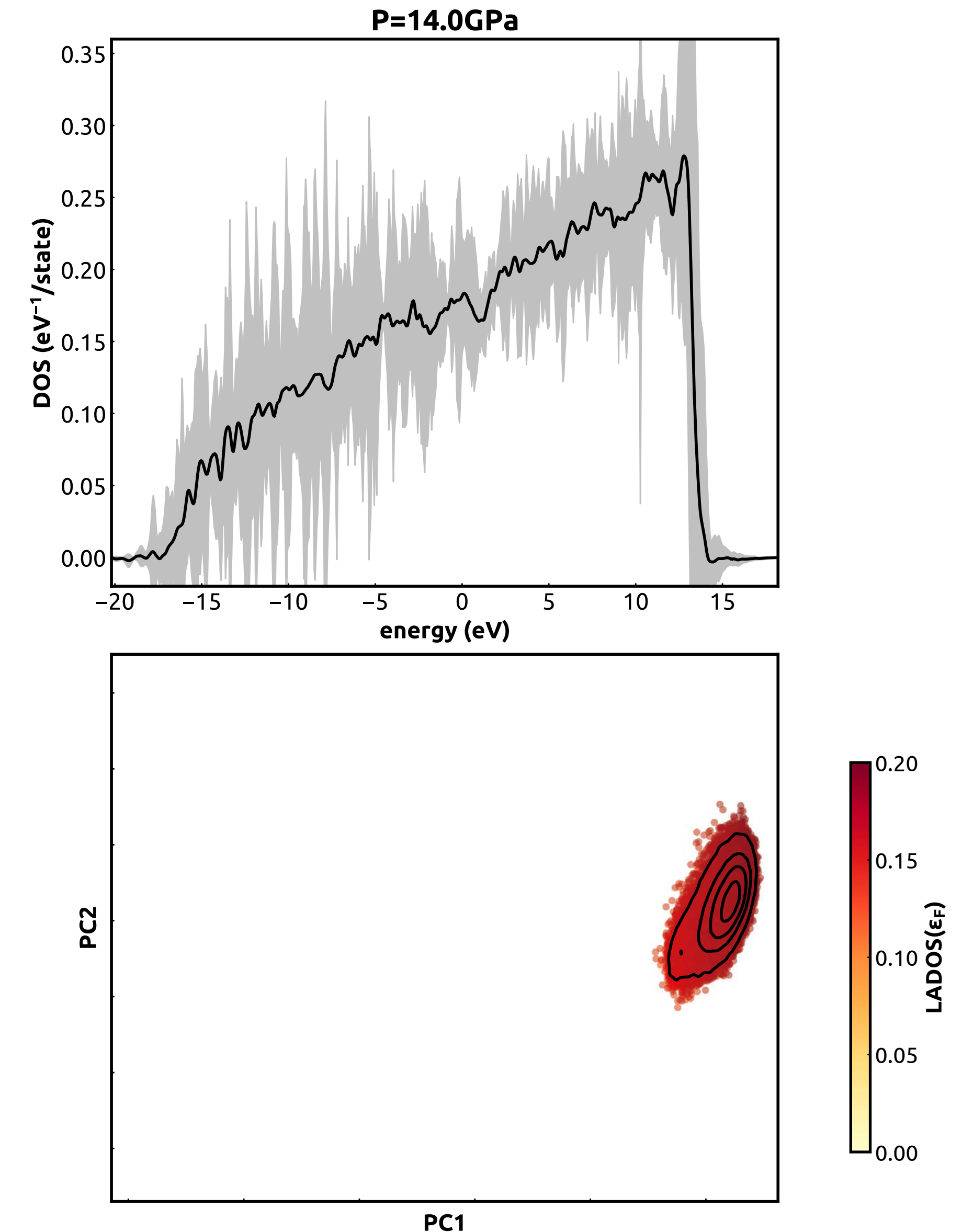
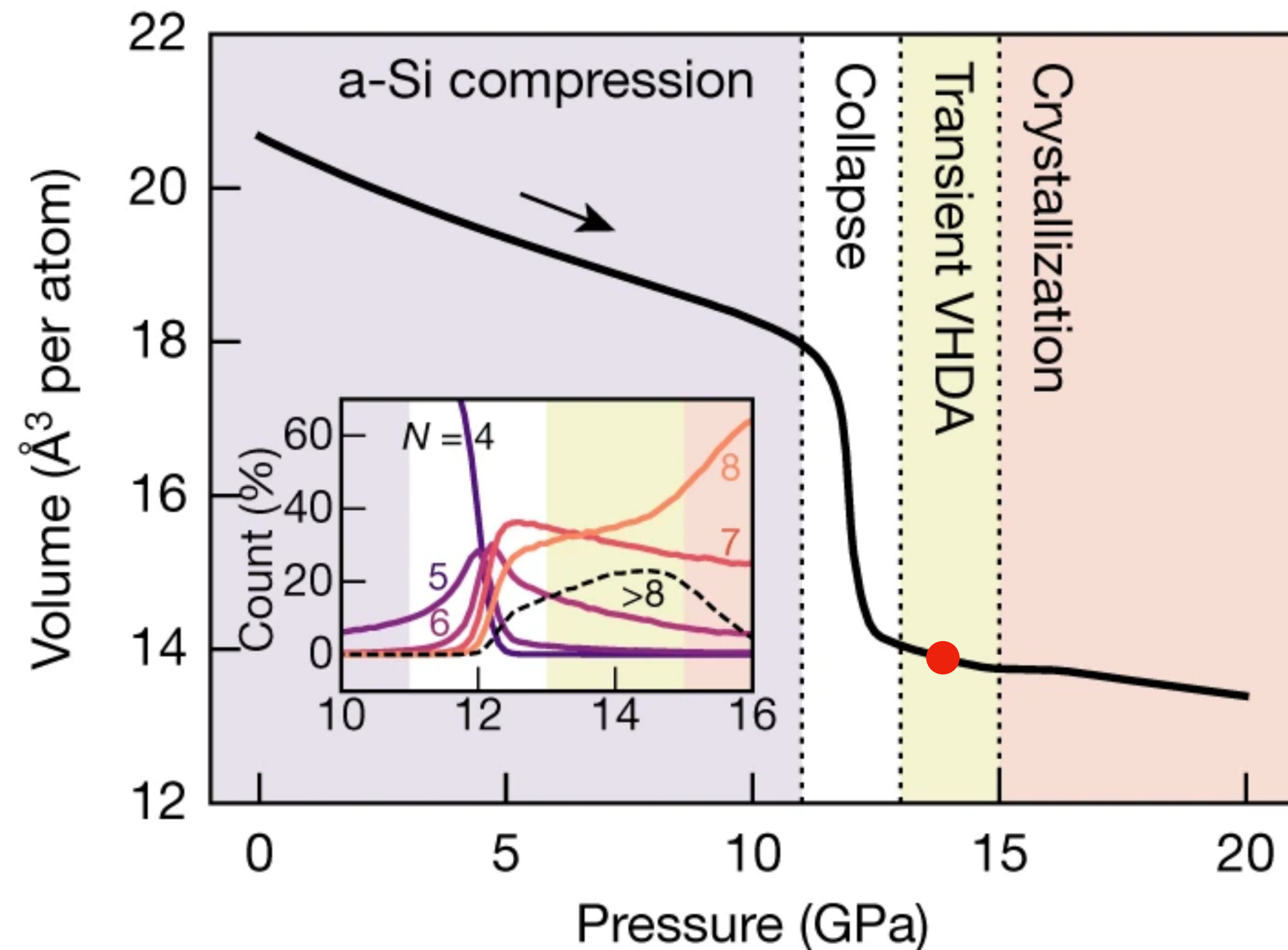
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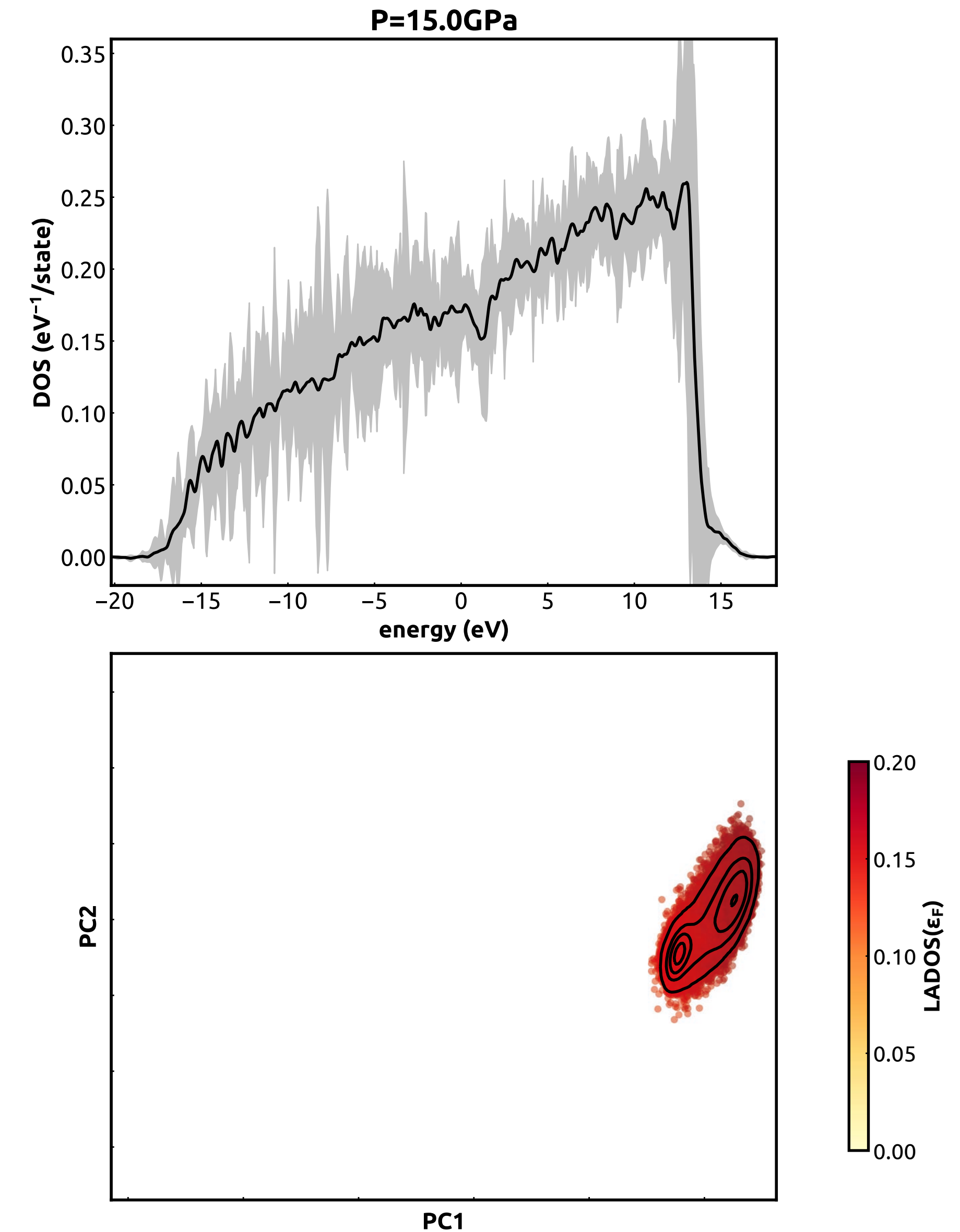
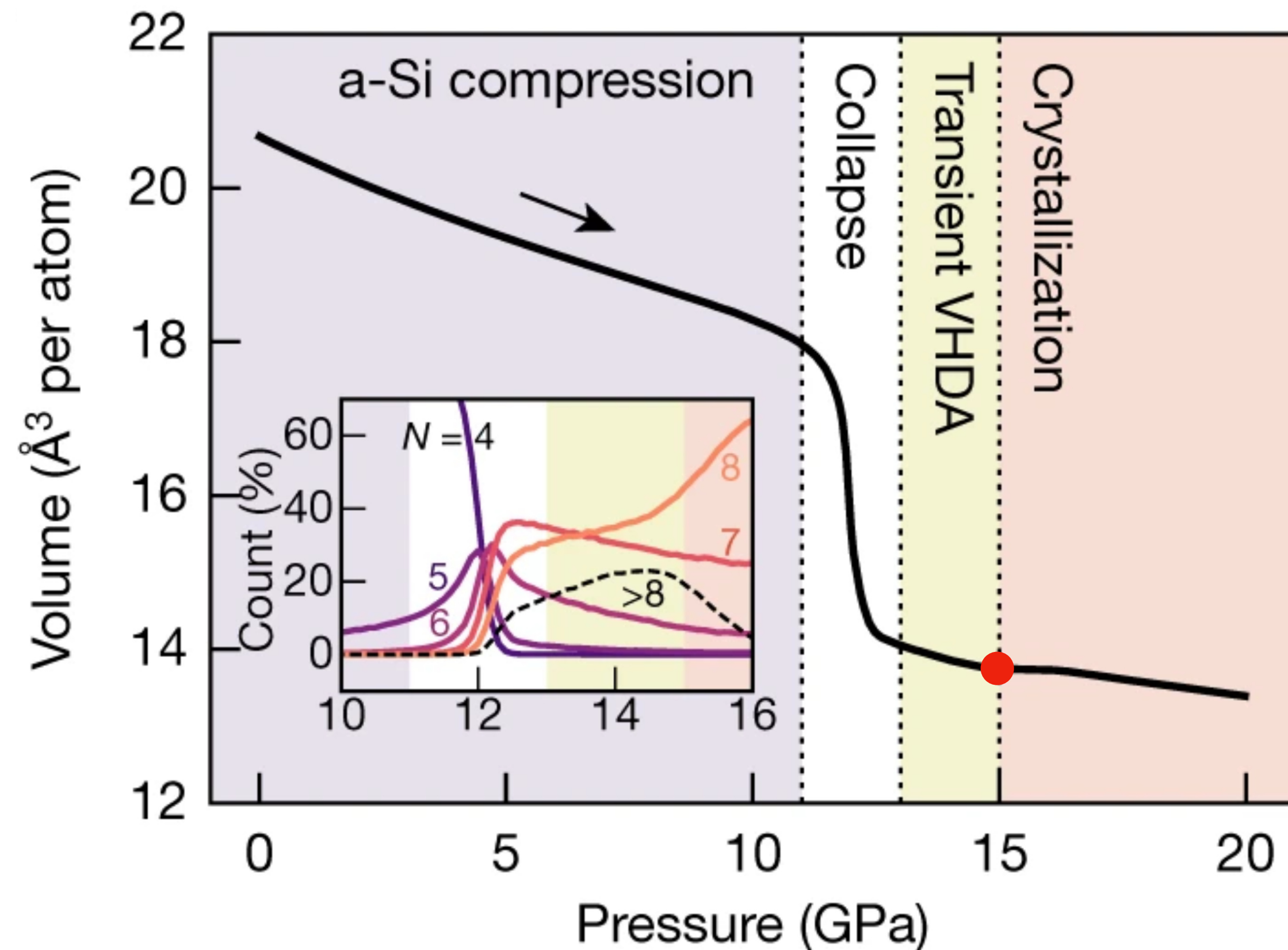
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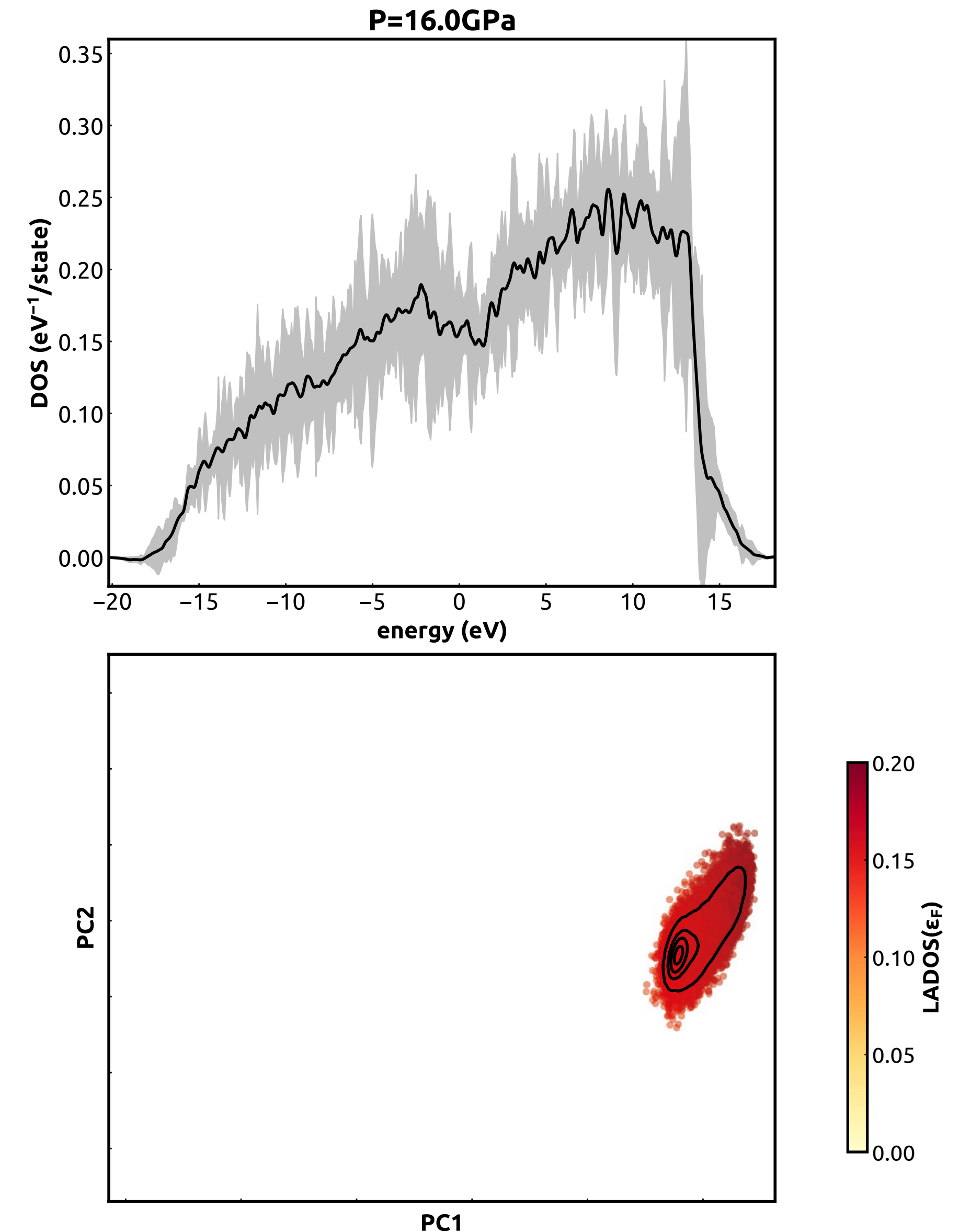
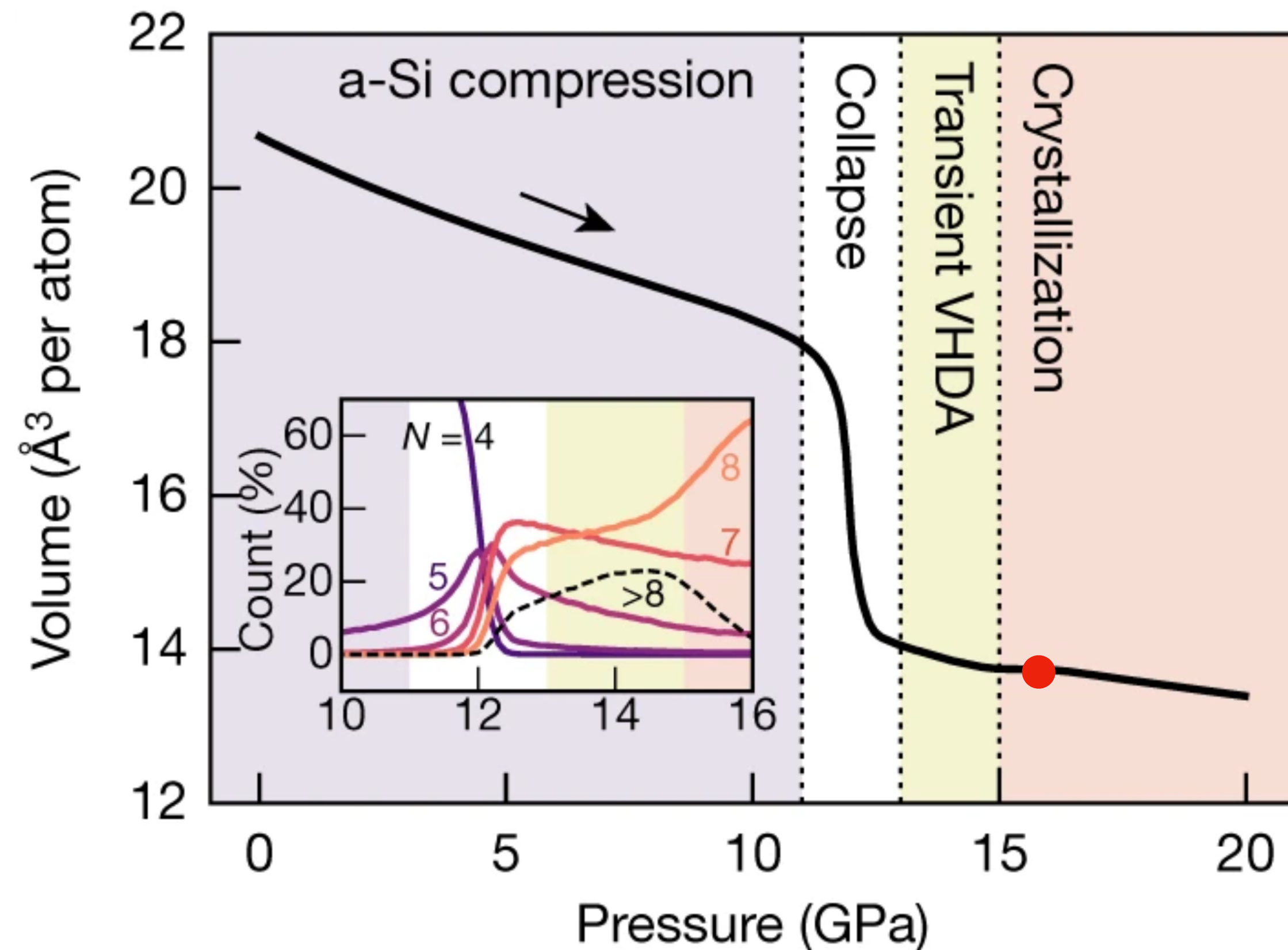
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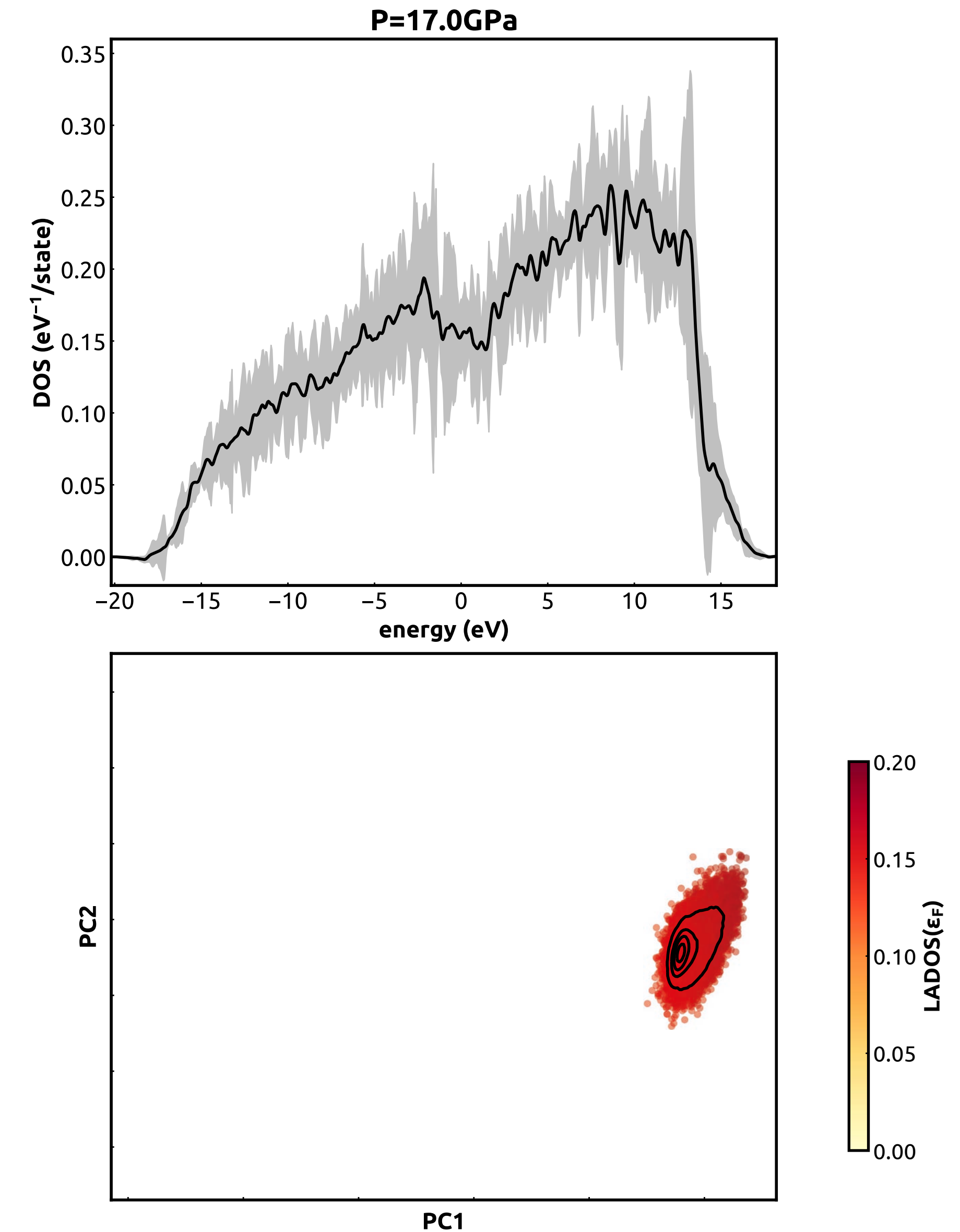
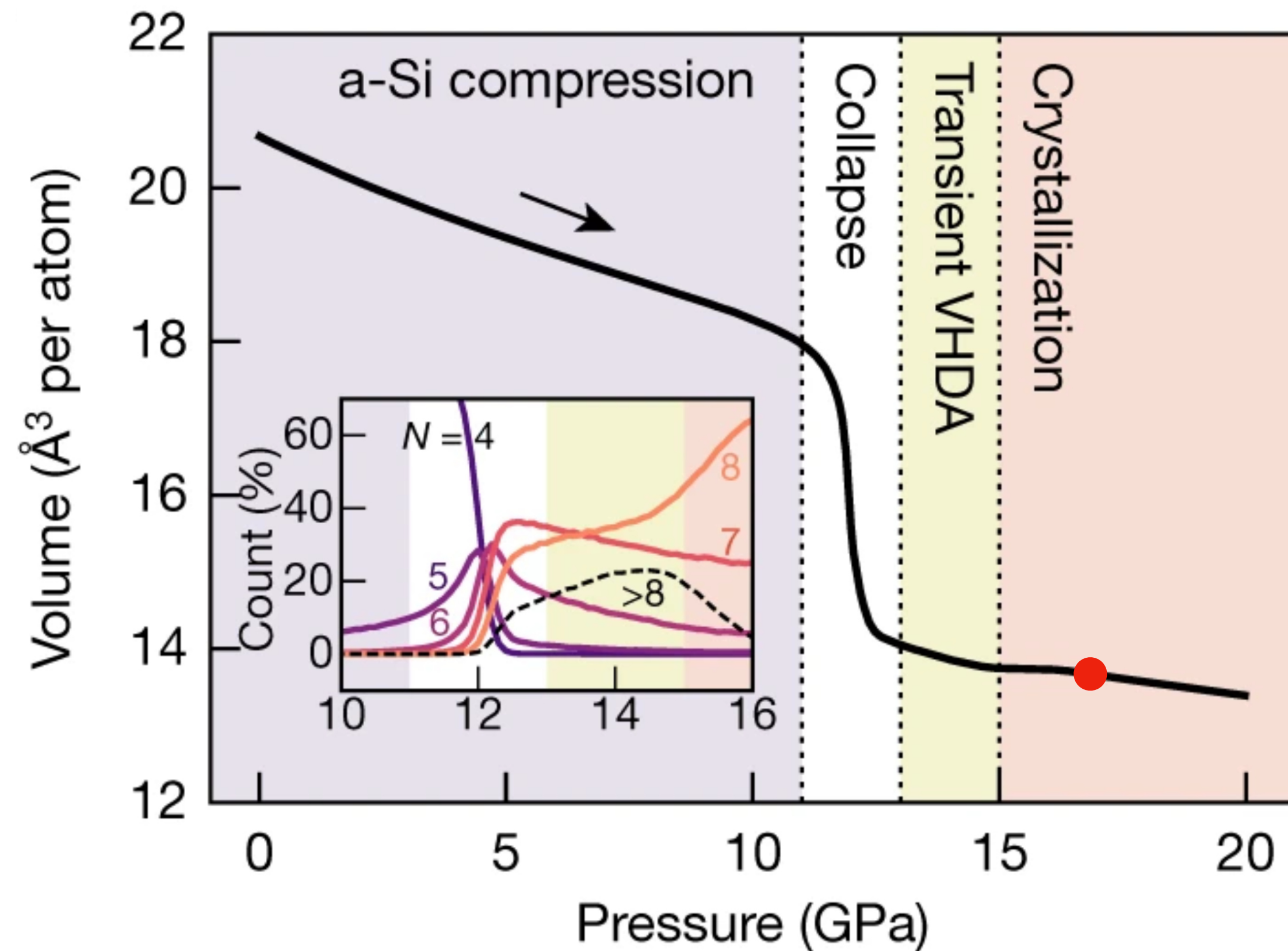
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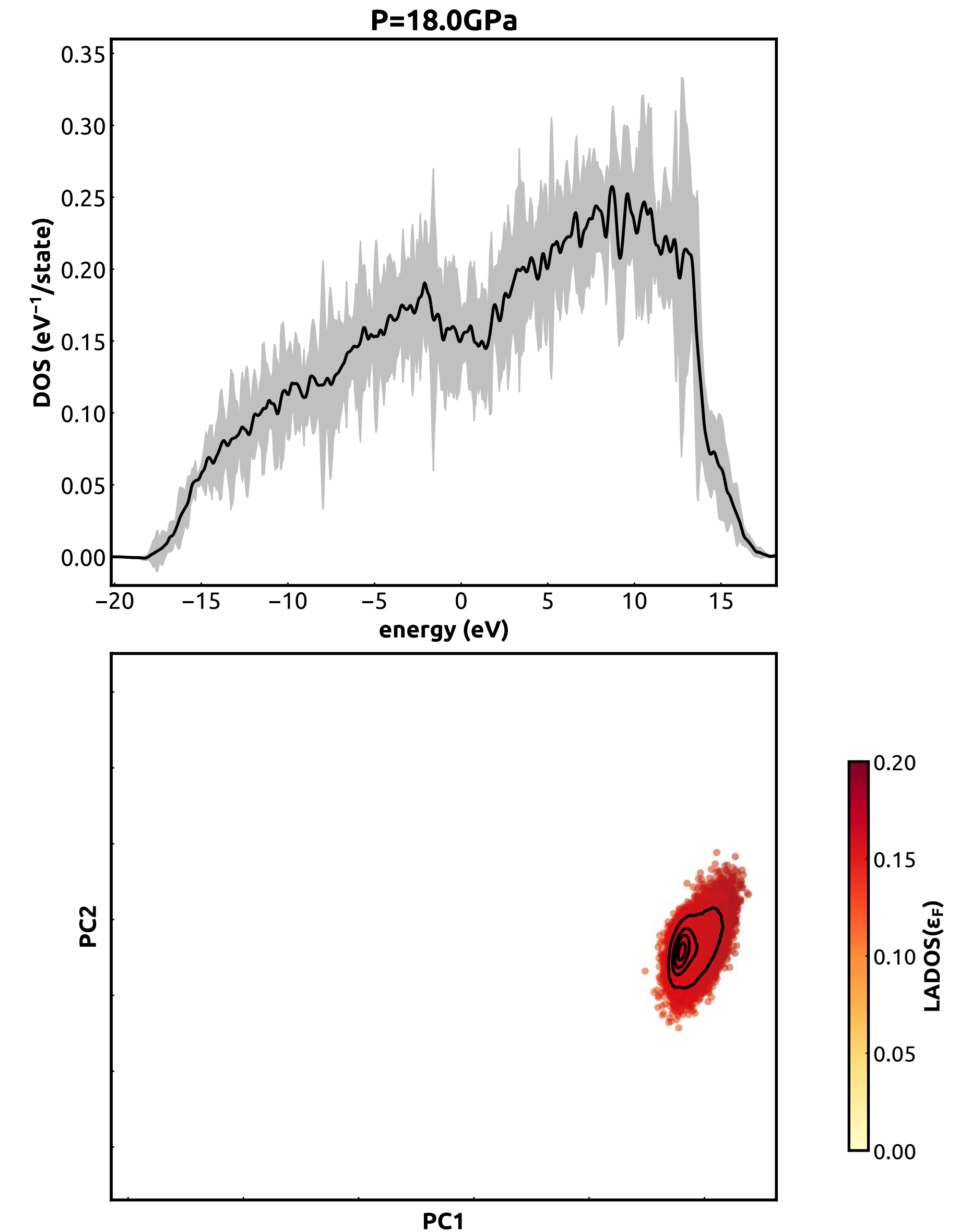
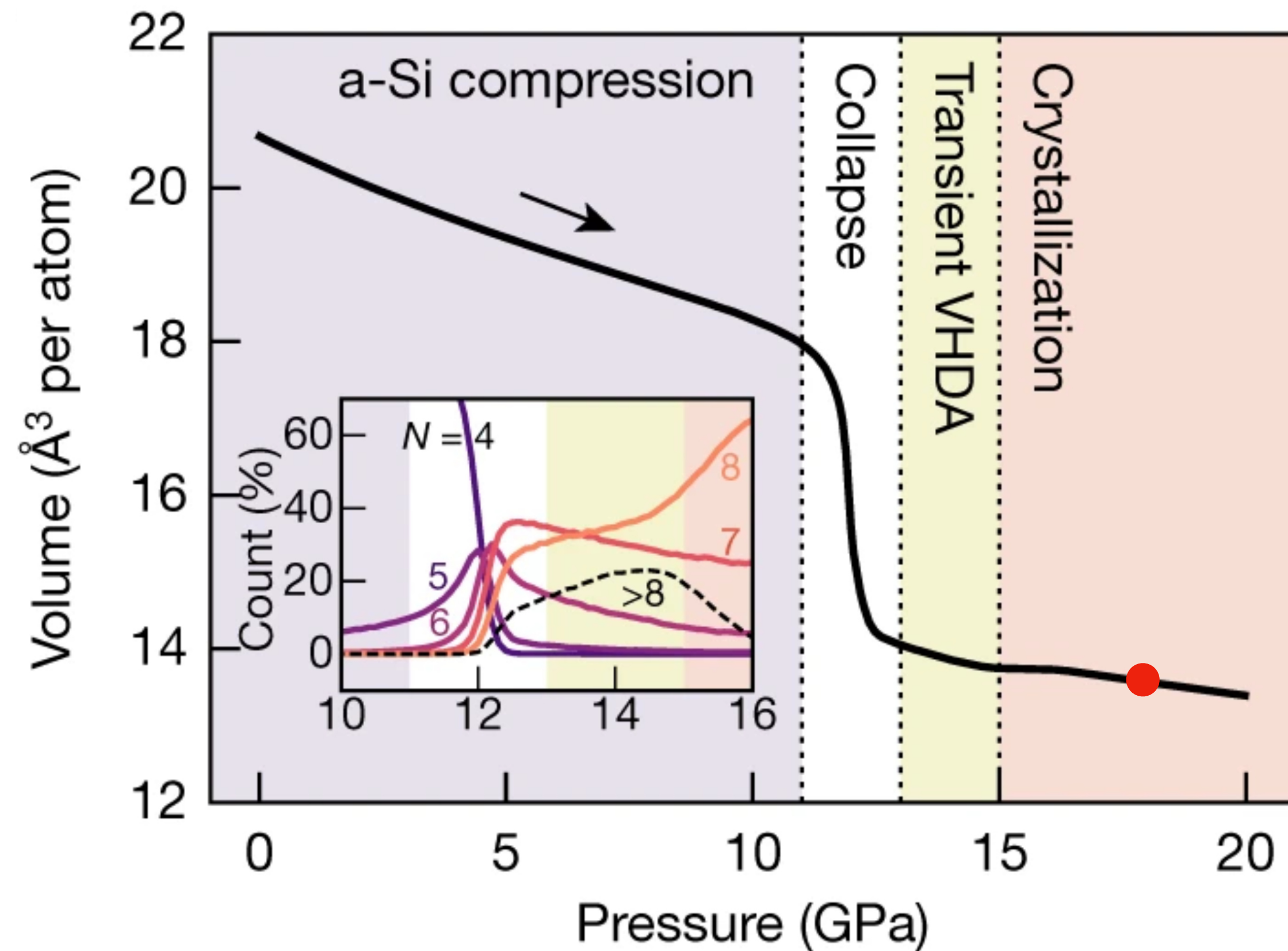
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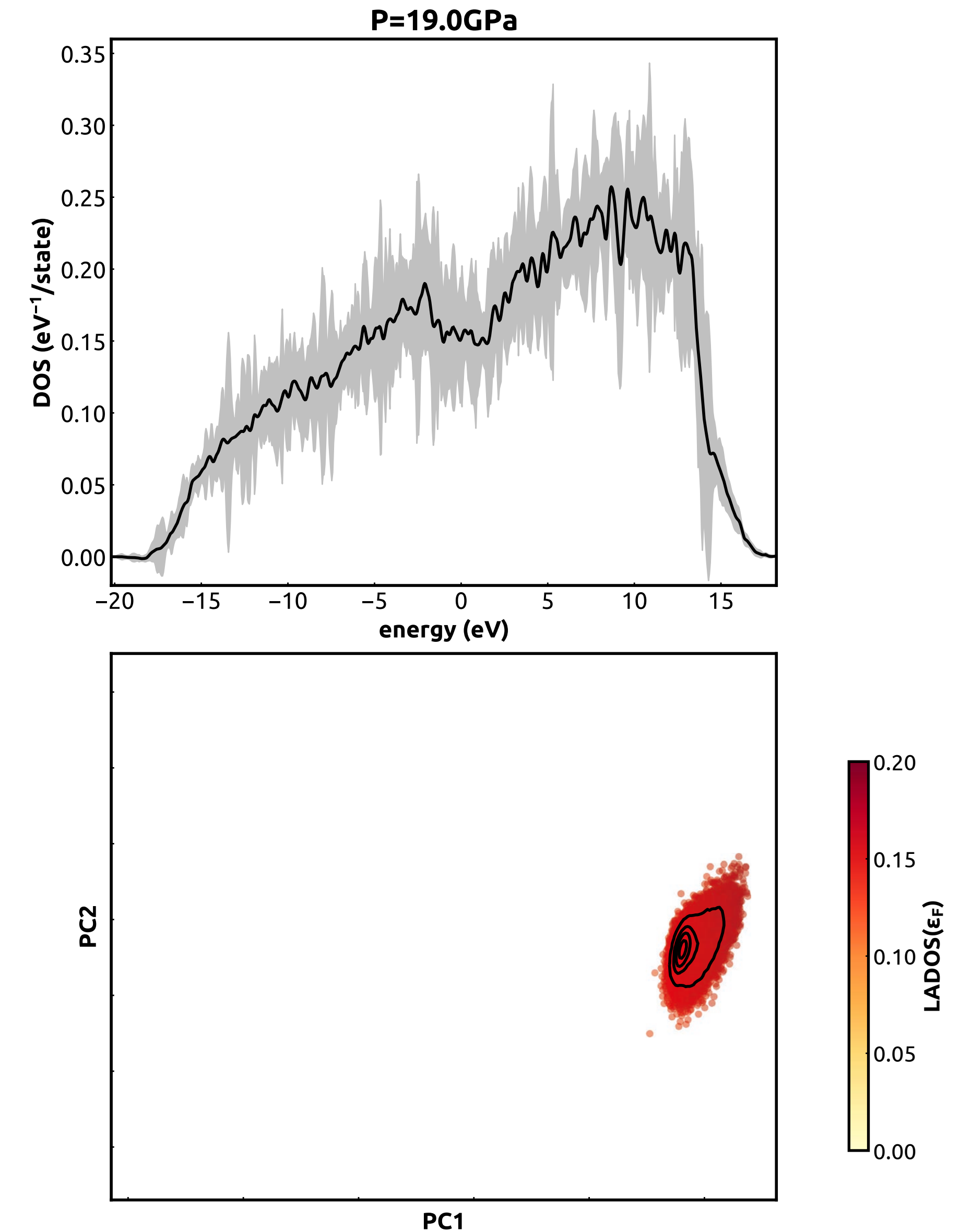
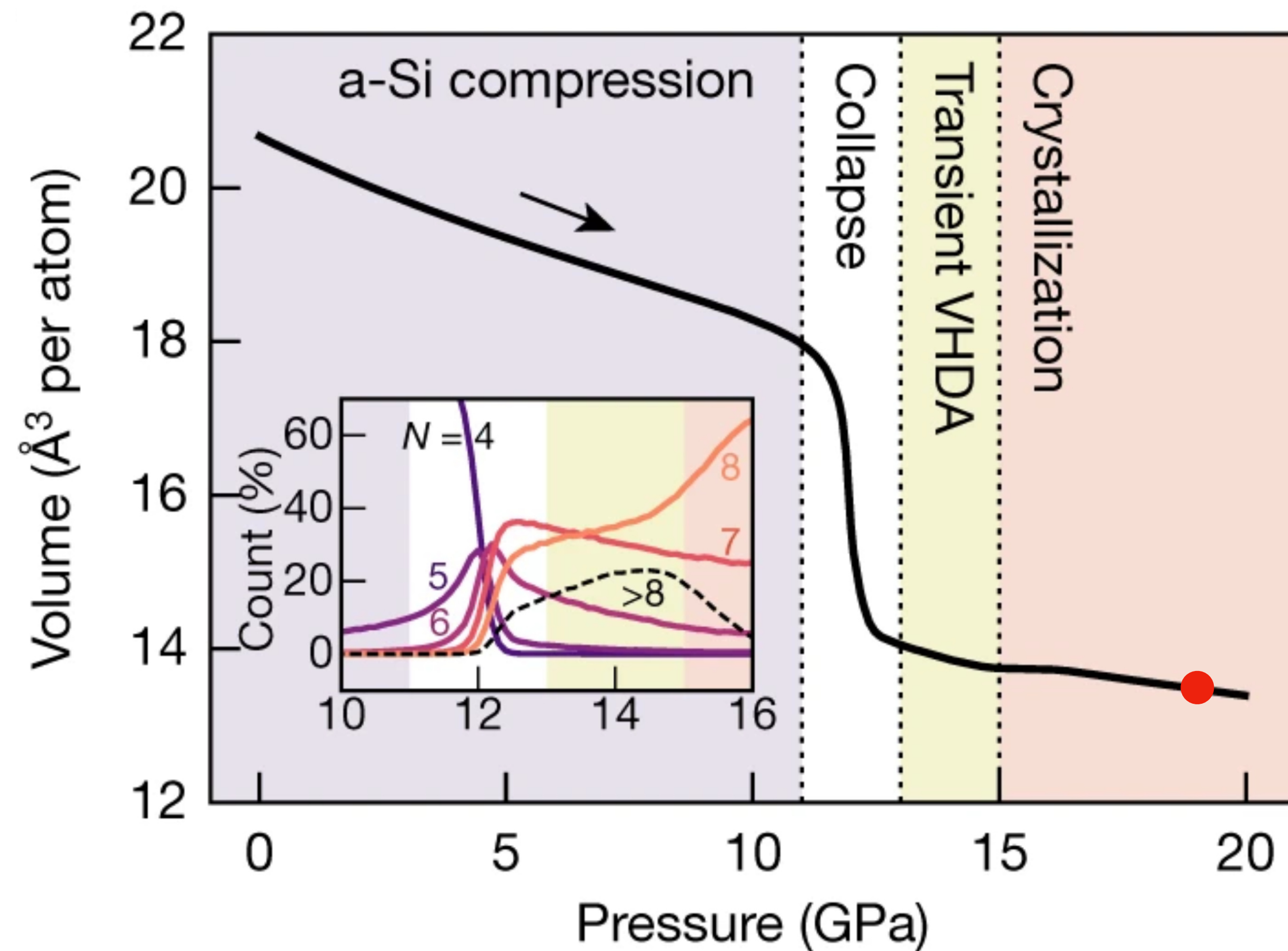
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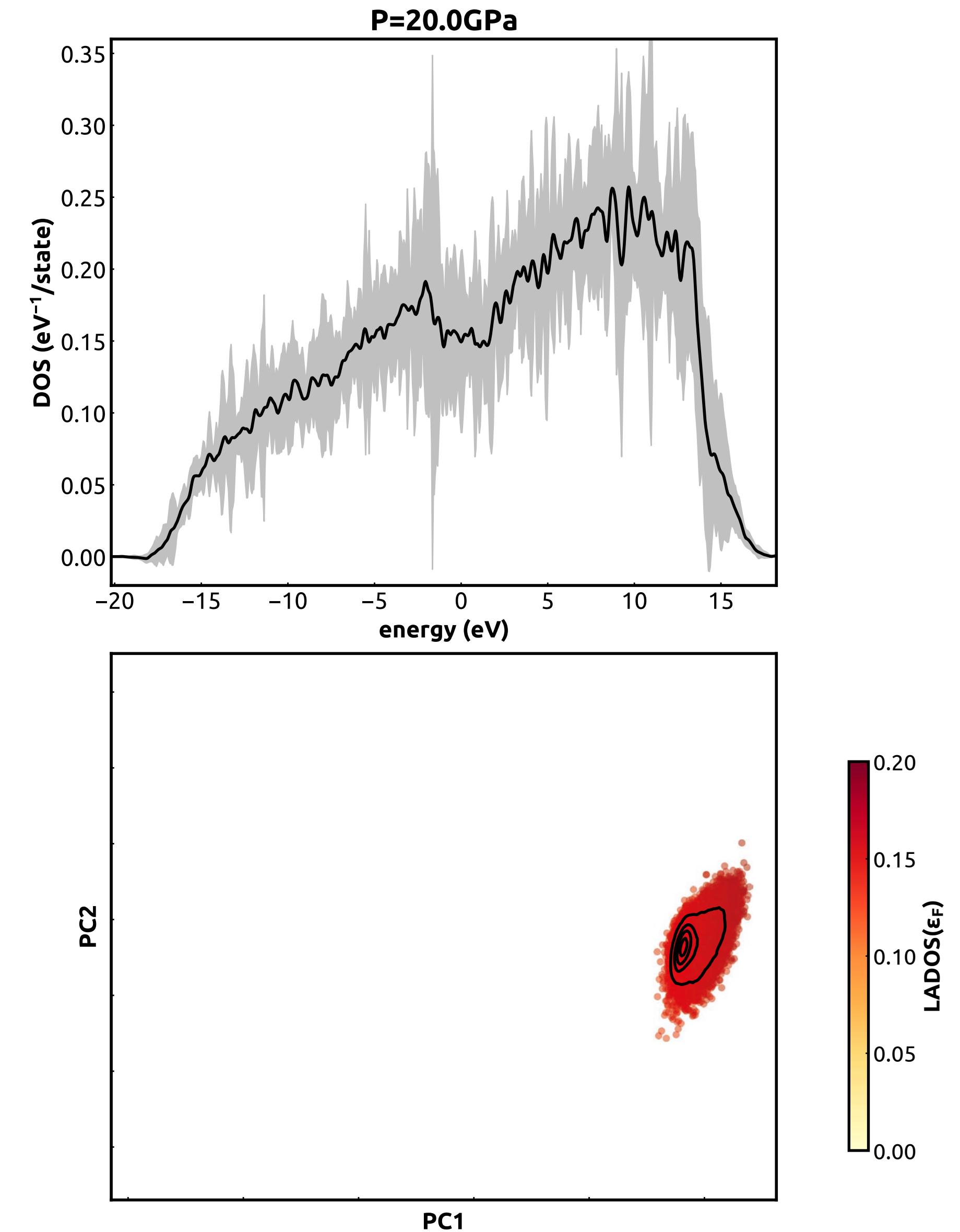
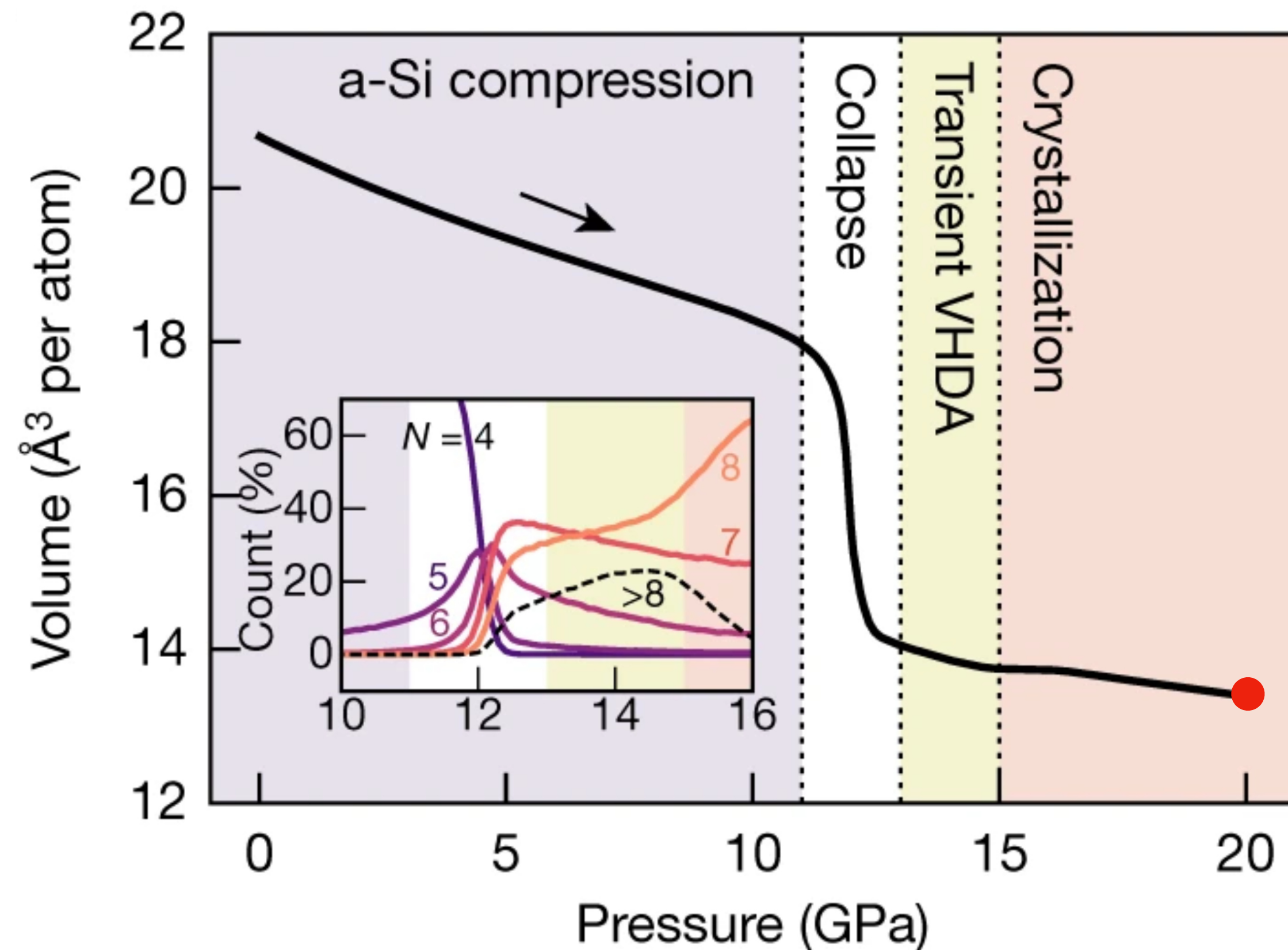
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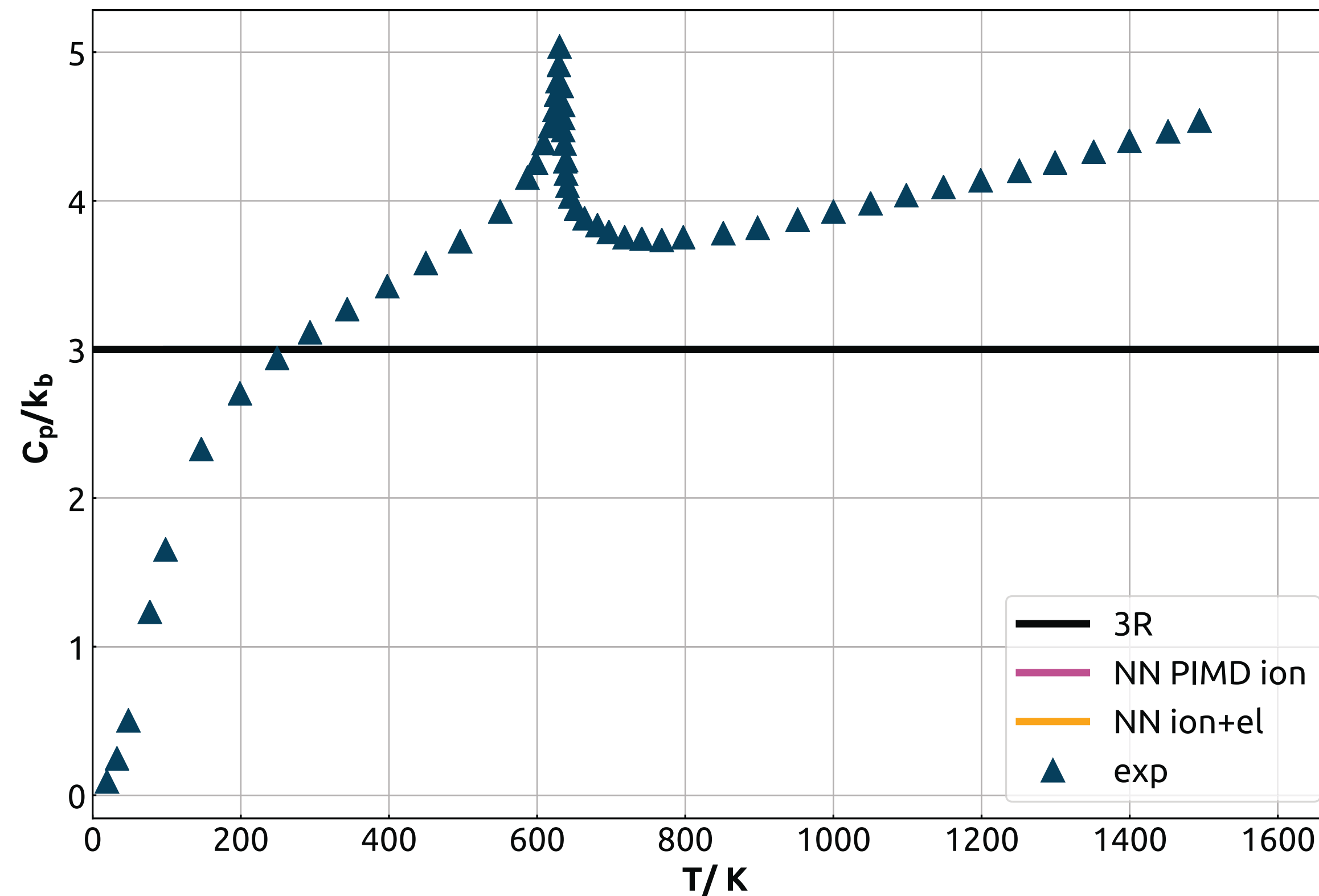
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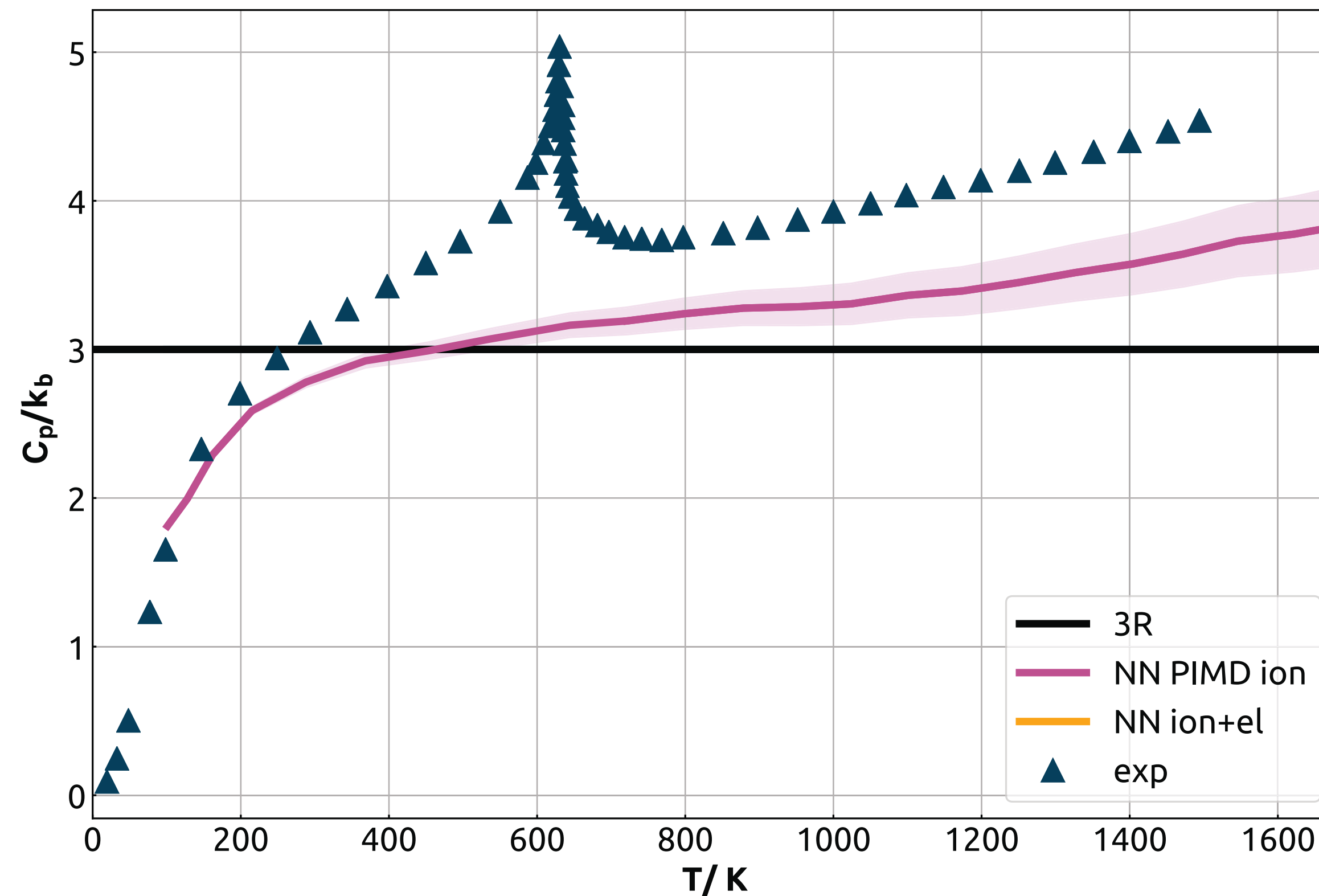


Thermodynamic properties from ground-state



Constant pressure heat capacity of nickel

Thermodynamic properties from ground-state

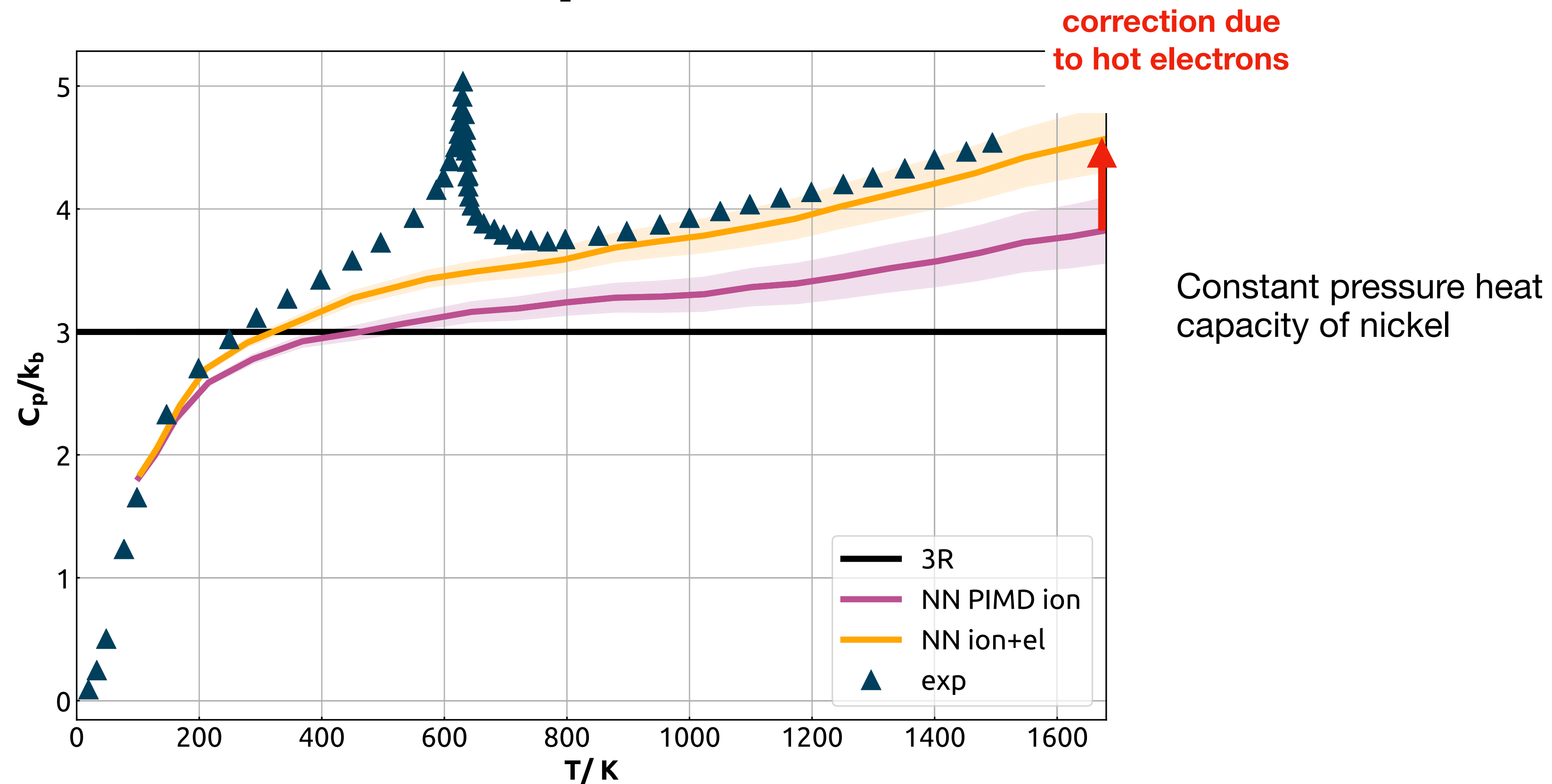


Constant pressure heat capacity of nickel

Thermodynamic properties from ground-state

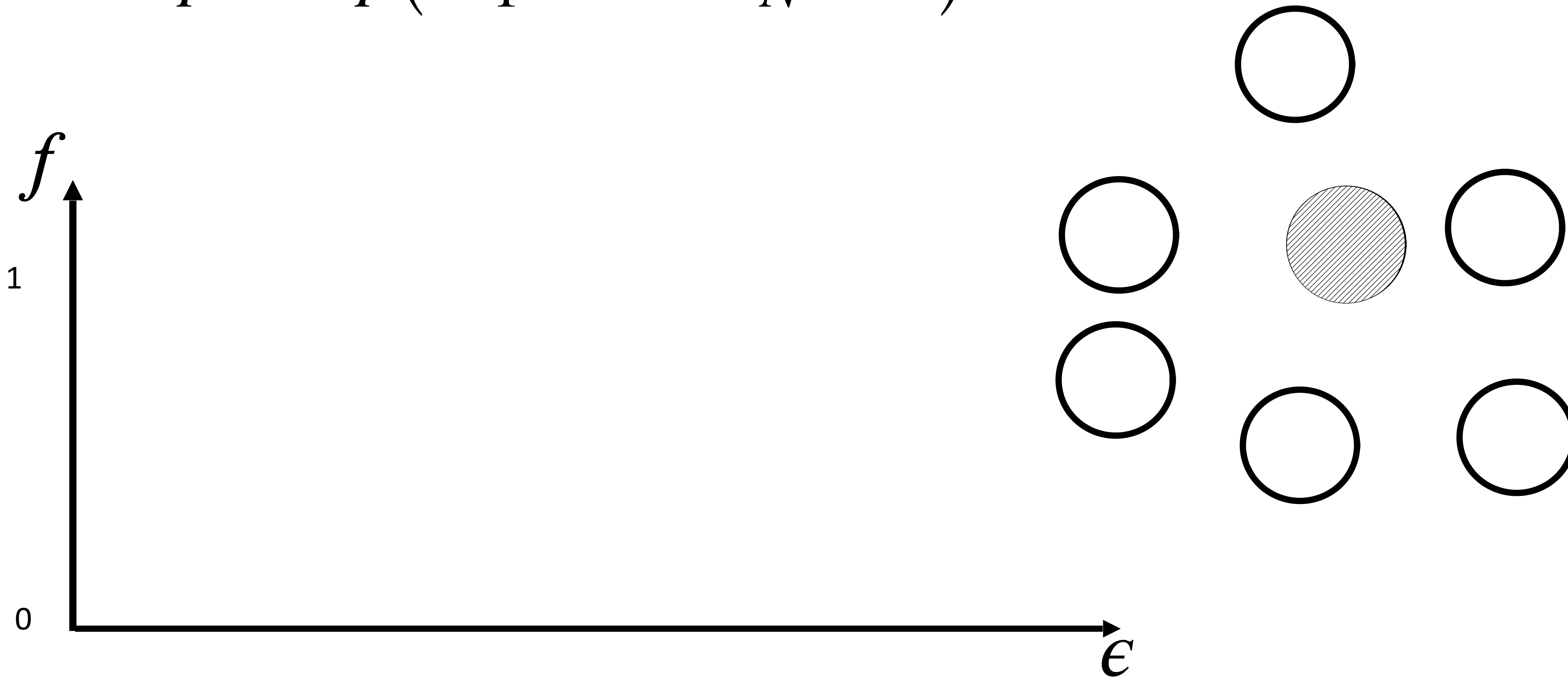
$$U_{\text{DOS}}^{\text{el}}(T) = \int_{-\infty}^{\infty} \varepsilon \text{DOS}(\varepsilon) [f_{\text{FD}}(\varepsilon - \mu(T), T) - f_{\text{FD}}(\varepsilon - \varepsilon_F, T = 0)] d\varepsilon$$

$$C_P^{\text{el}}(T) = \frac{\partial \langle U_{\text{DOS}}^{\text{el}}(T) \rangle}{\partial T}$$



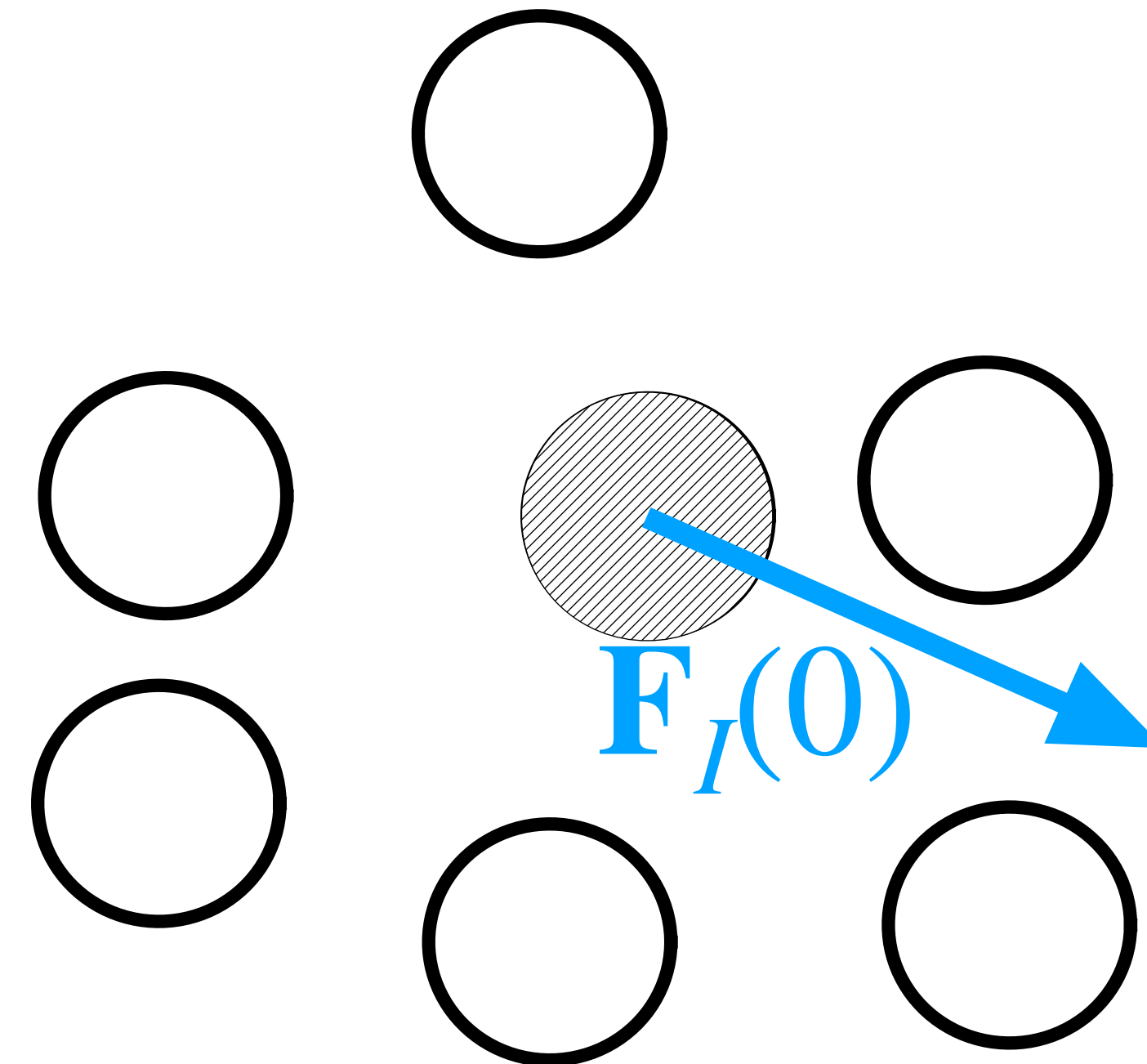
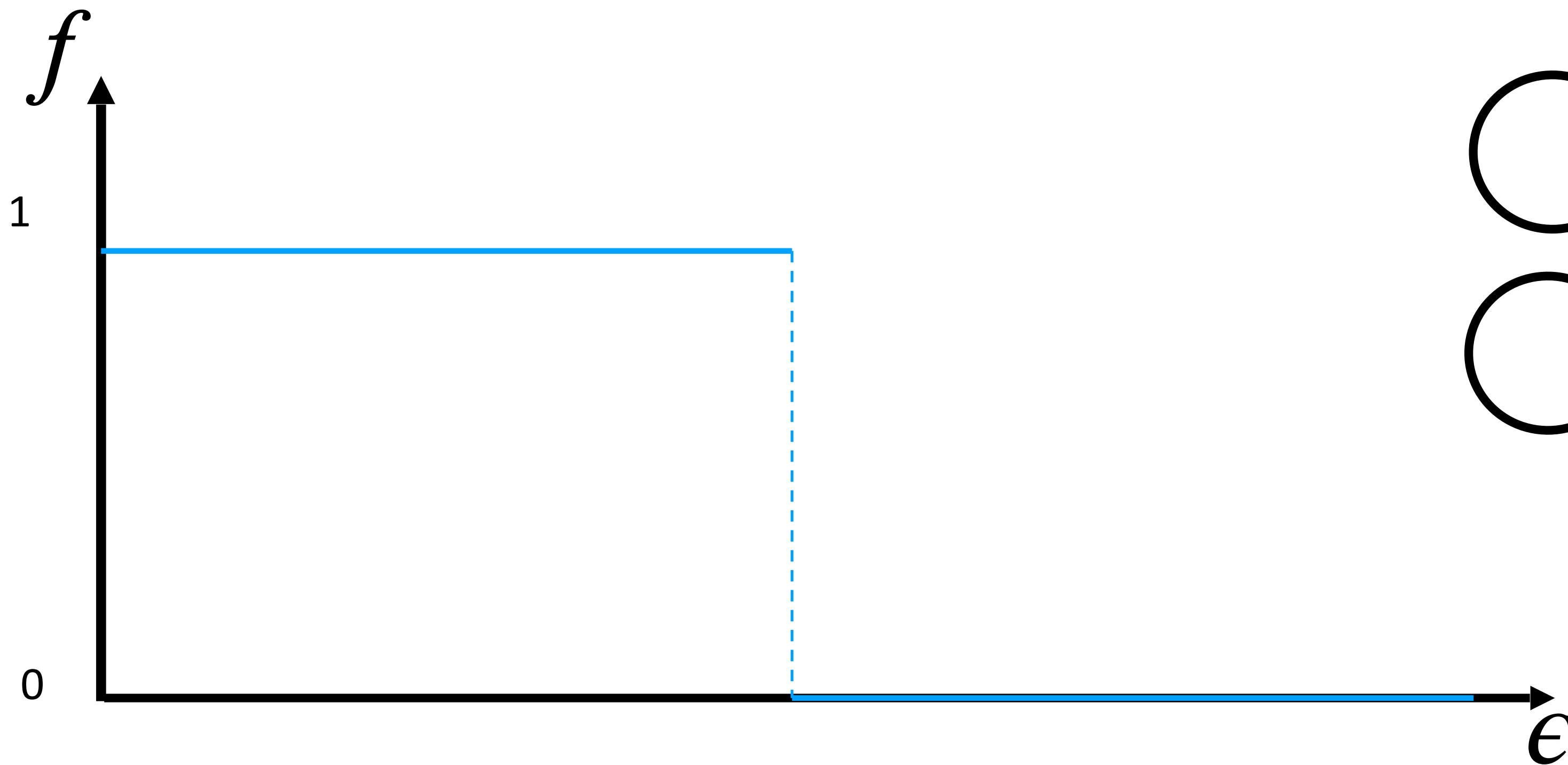
The finite-temperature problem

$$\mathbf{F}_I = \mathbf{F}_I(\mathbf{R}_1, \dots, \mathbf{R}_N; T^{el})$$



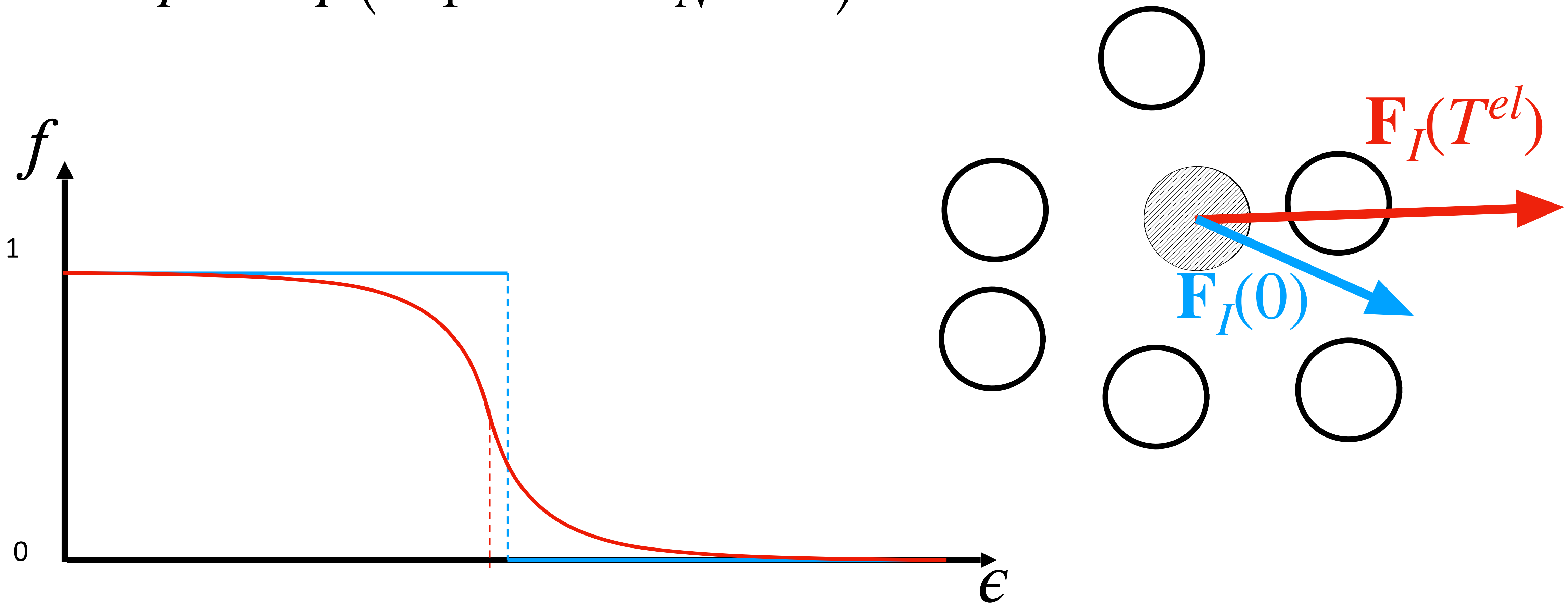
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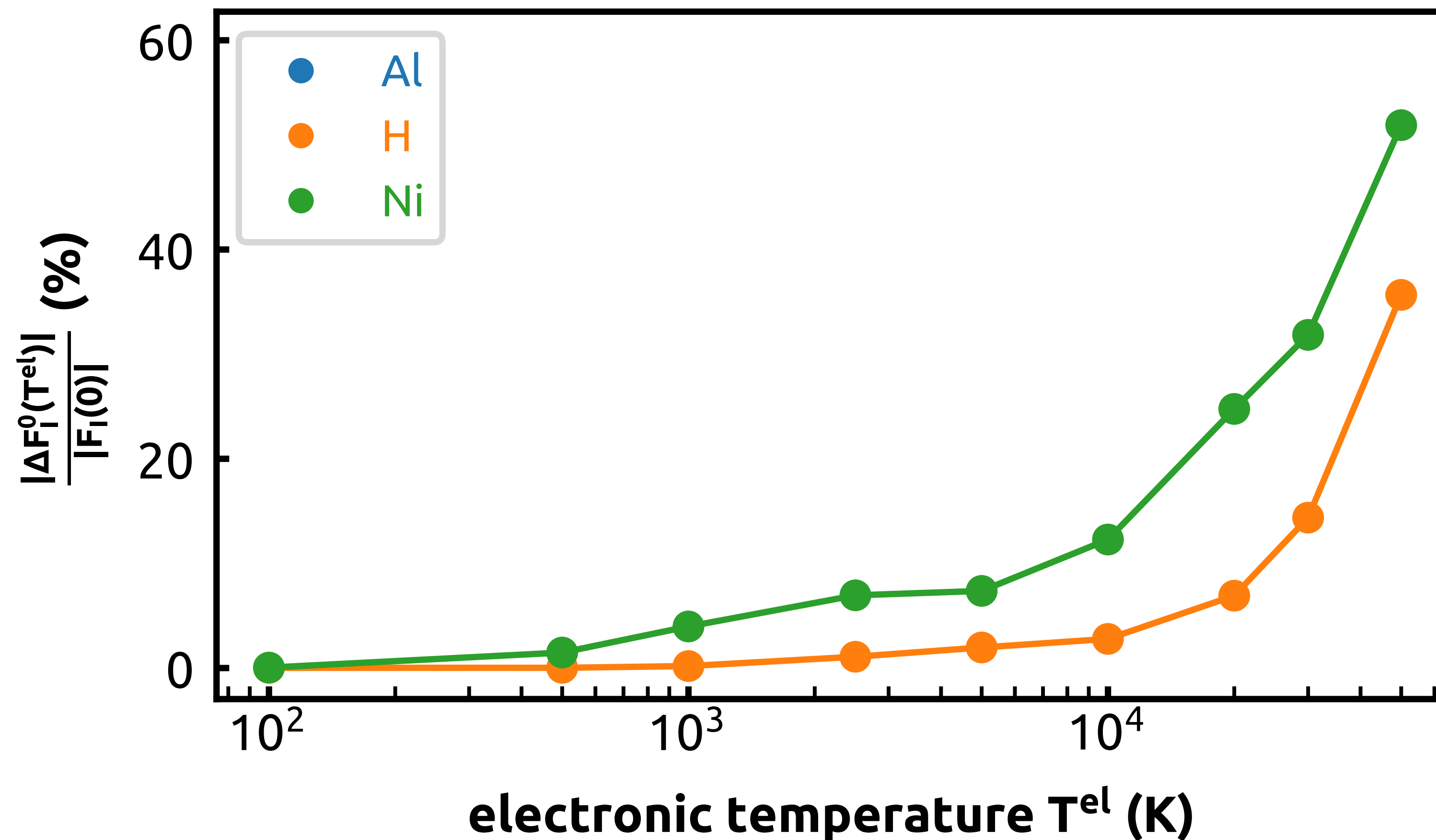
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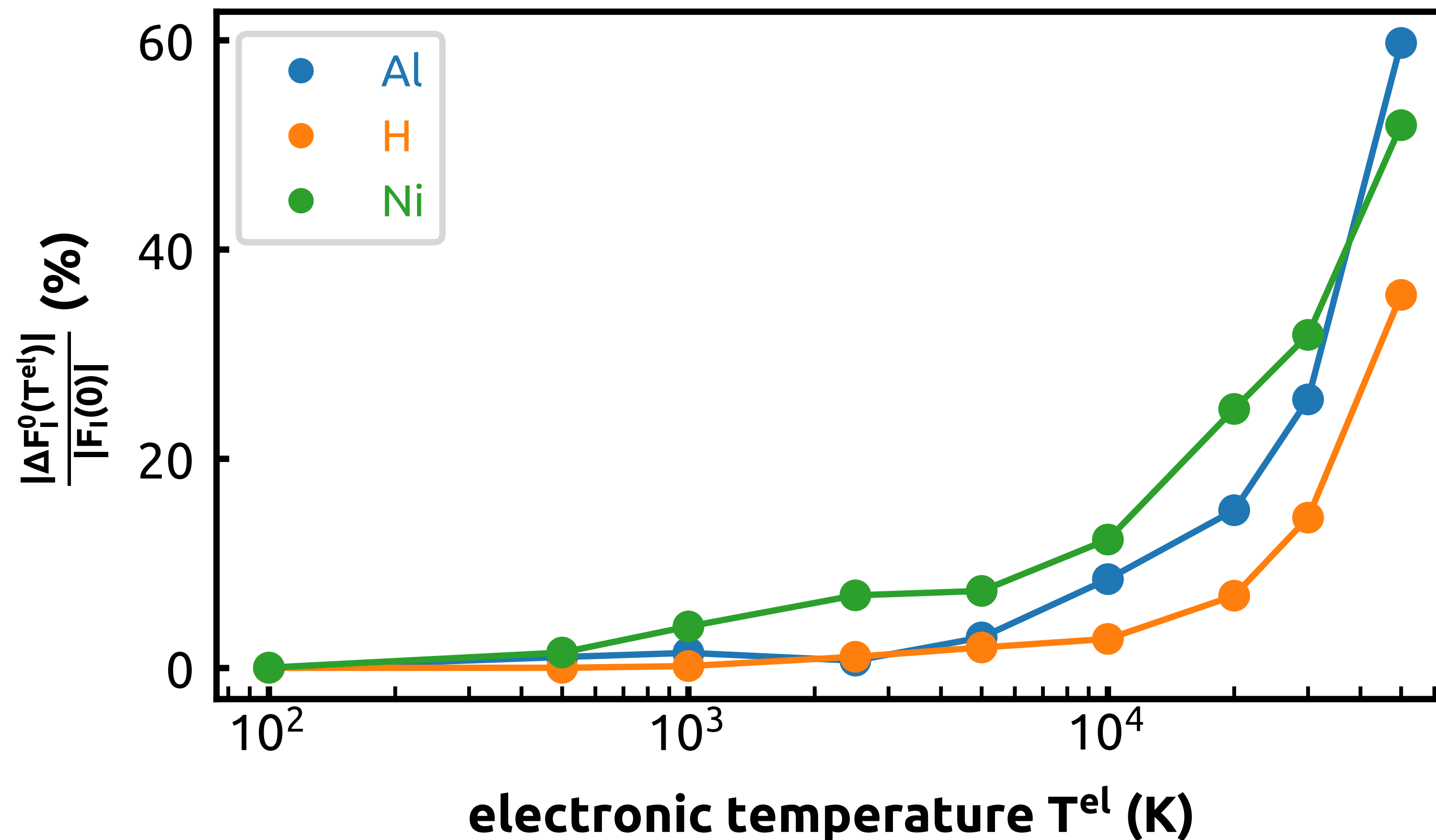
The finite-temperature problem

- High errors with respect to $T=0$ forces

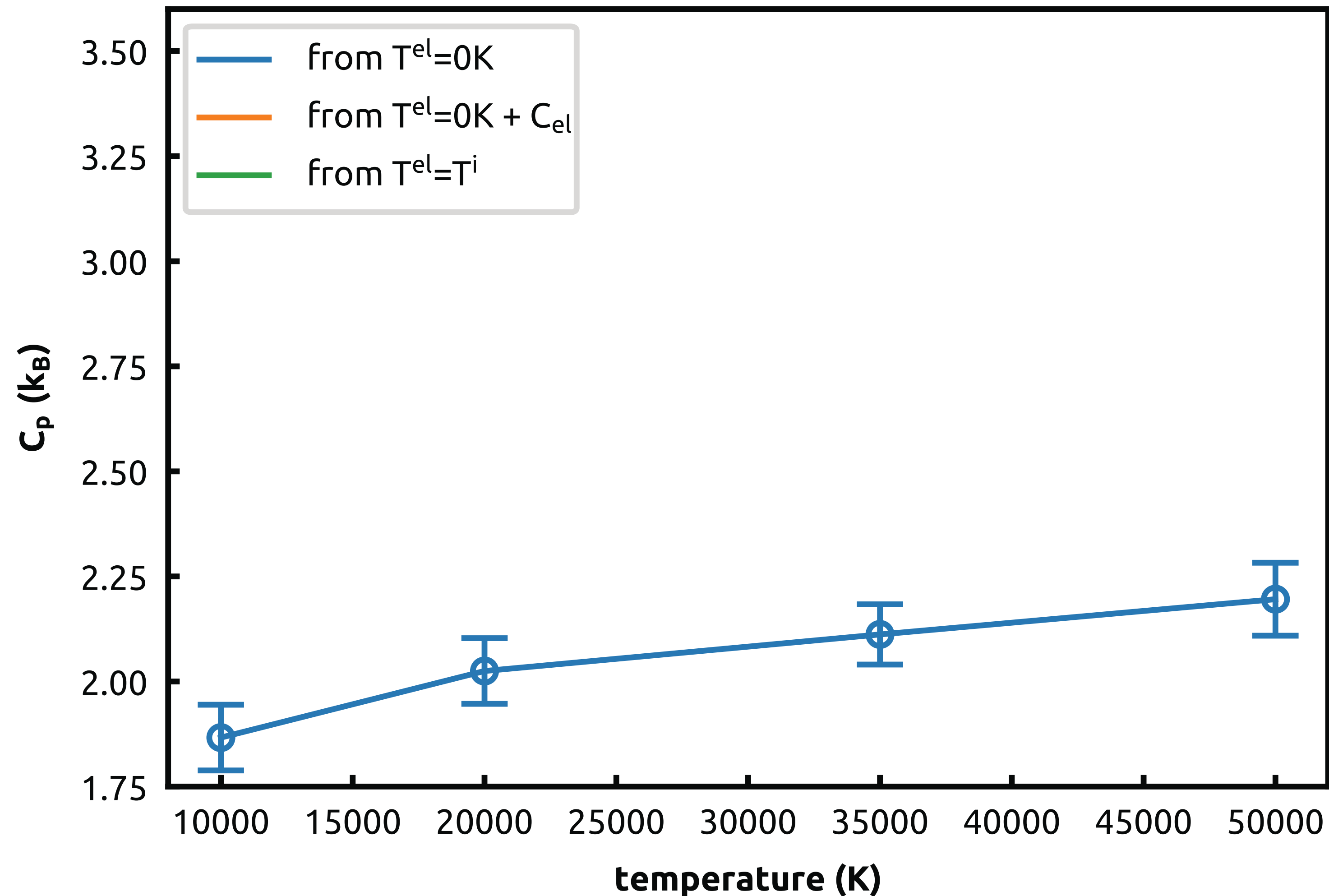


The finite-temperature problem

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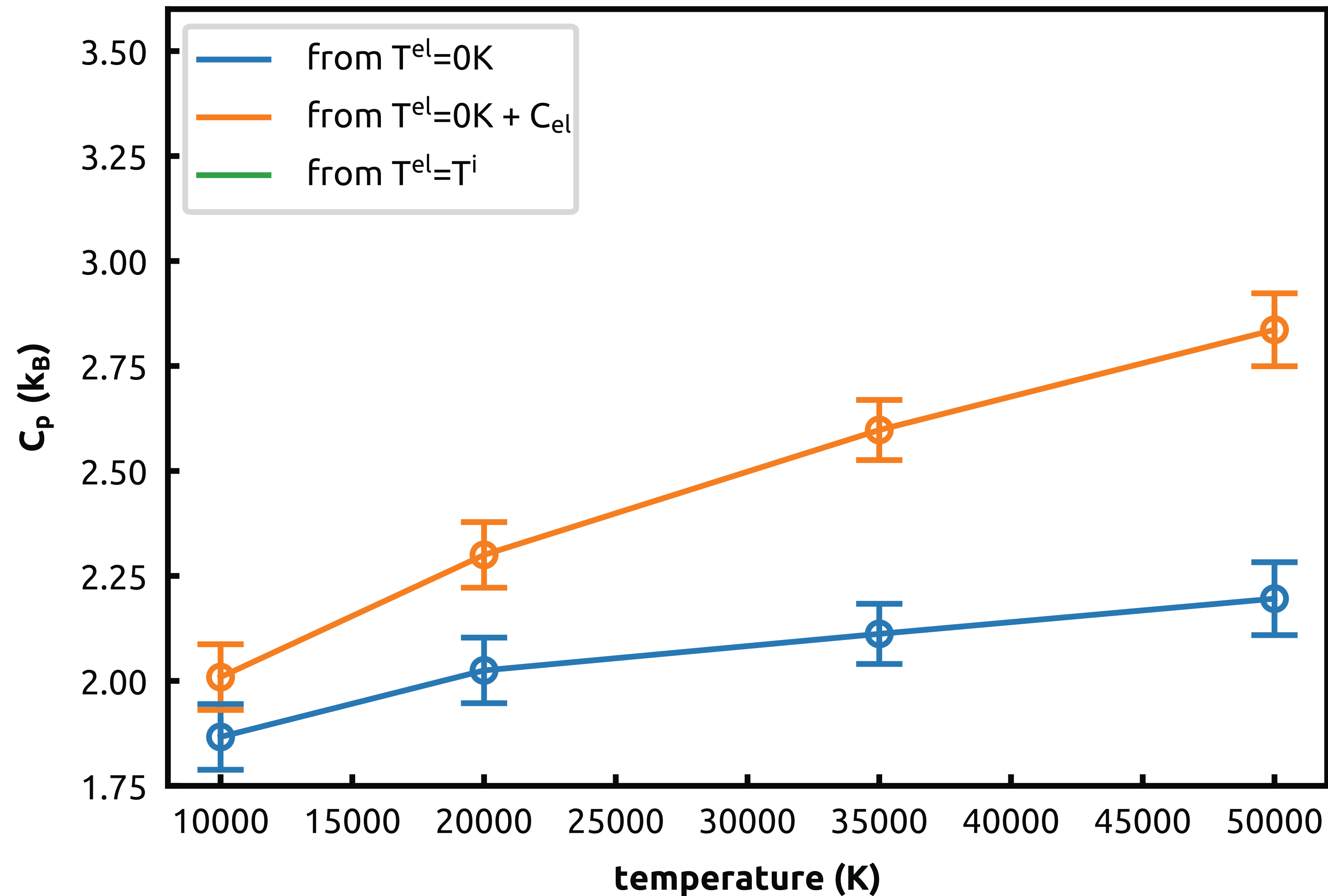


The finite-temperature problem



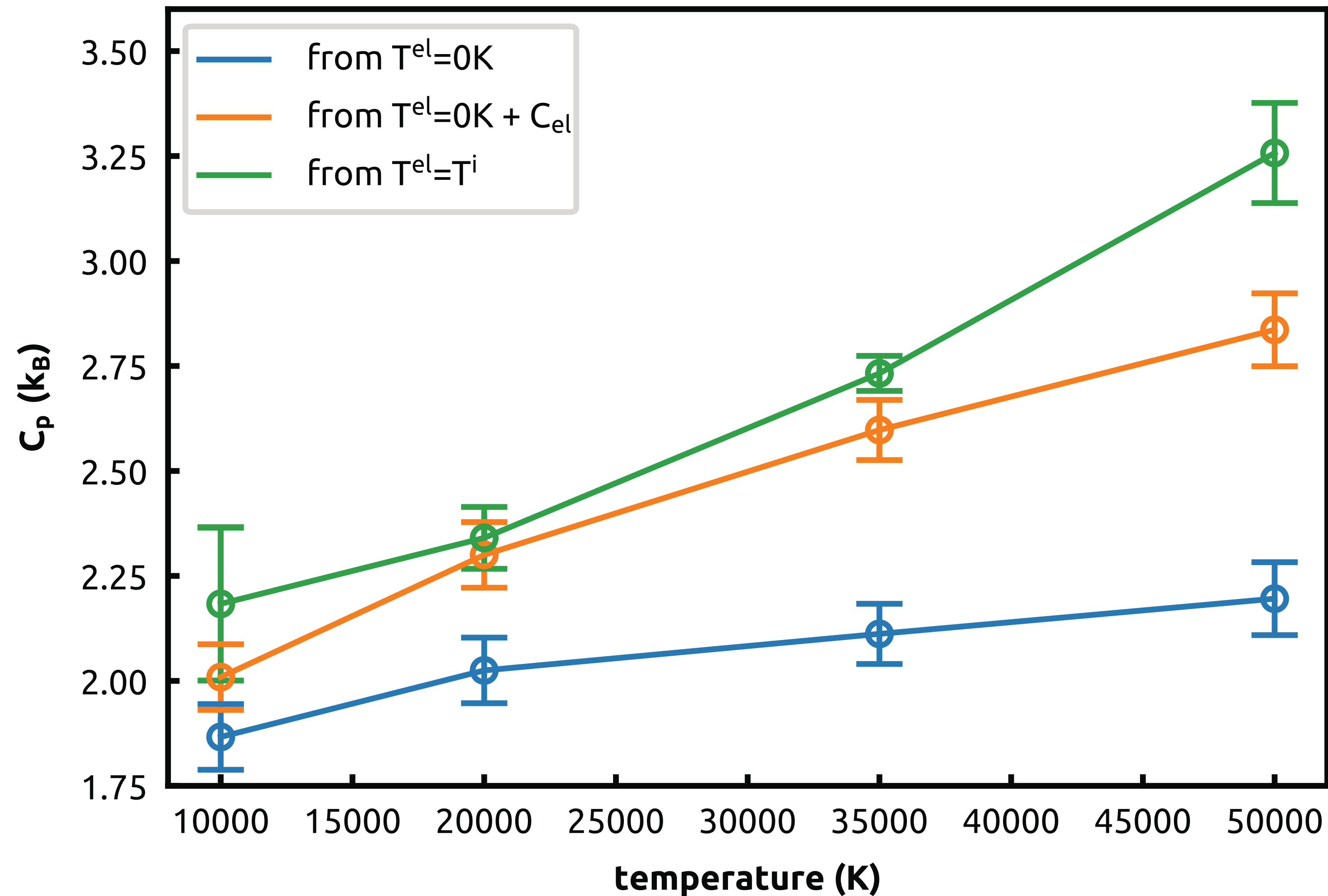
Constant pressure
heat capacity of
hydrogen at 400GPa
from DFT simulations

The finite-temperature problem



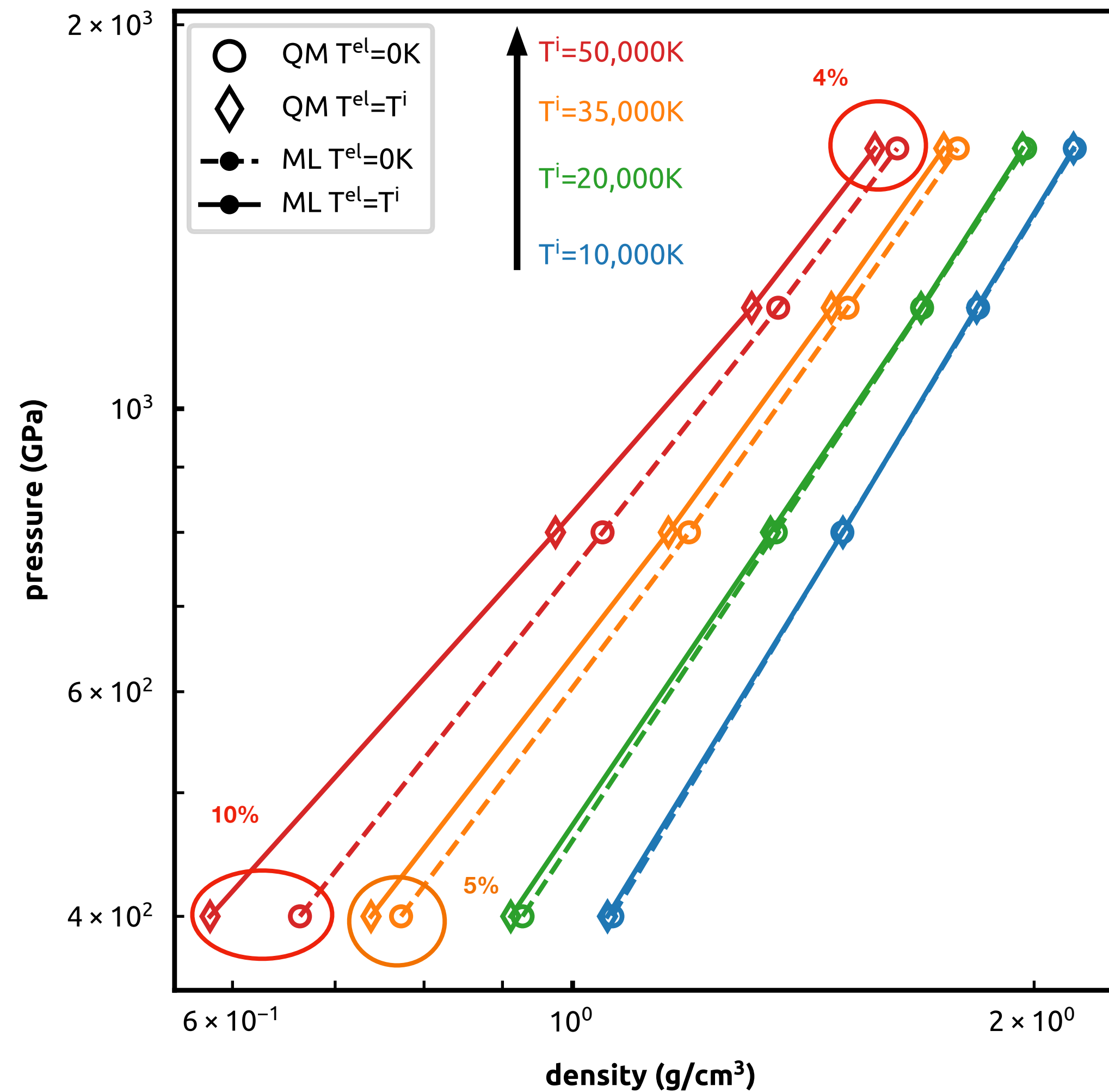
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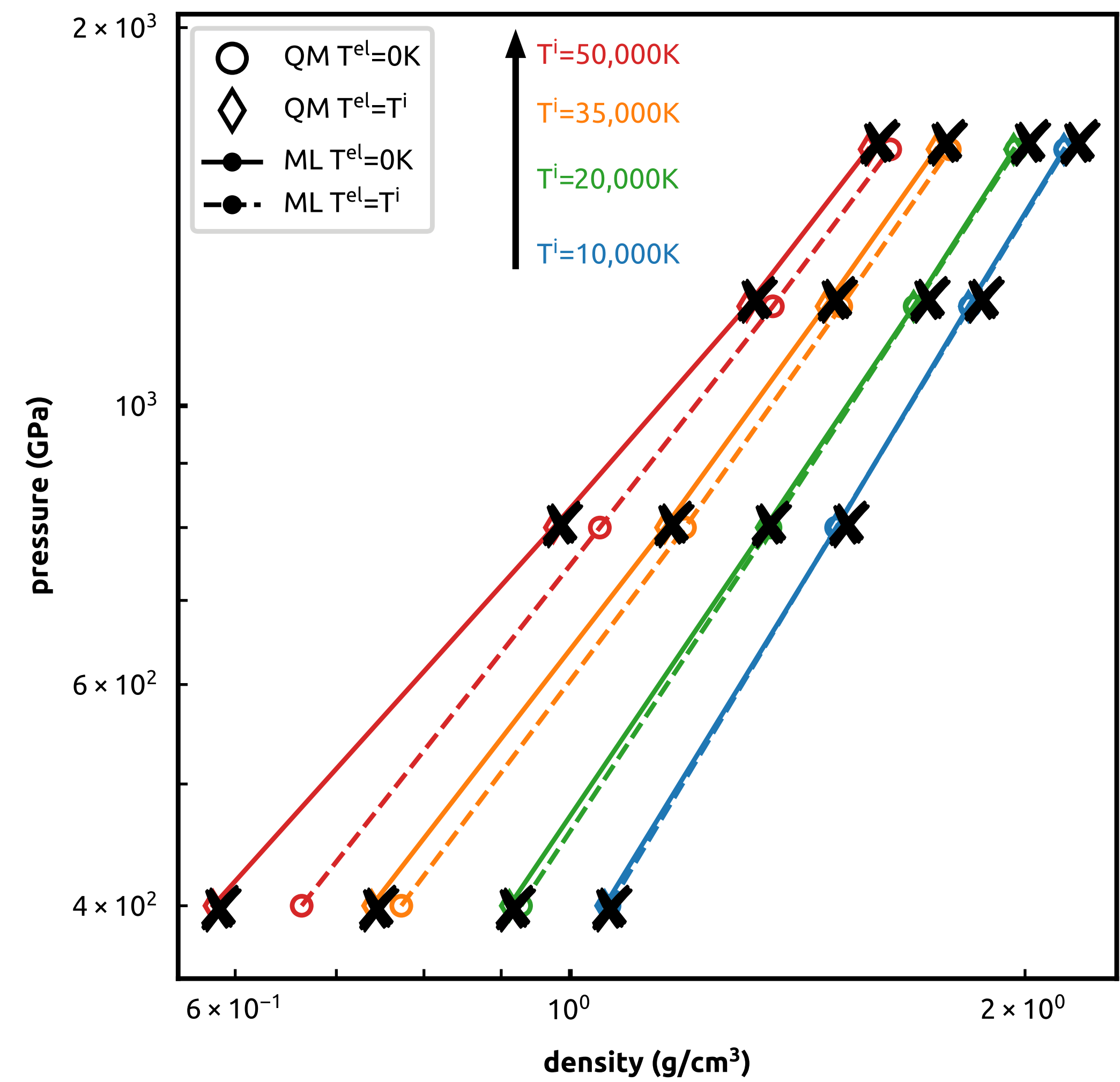
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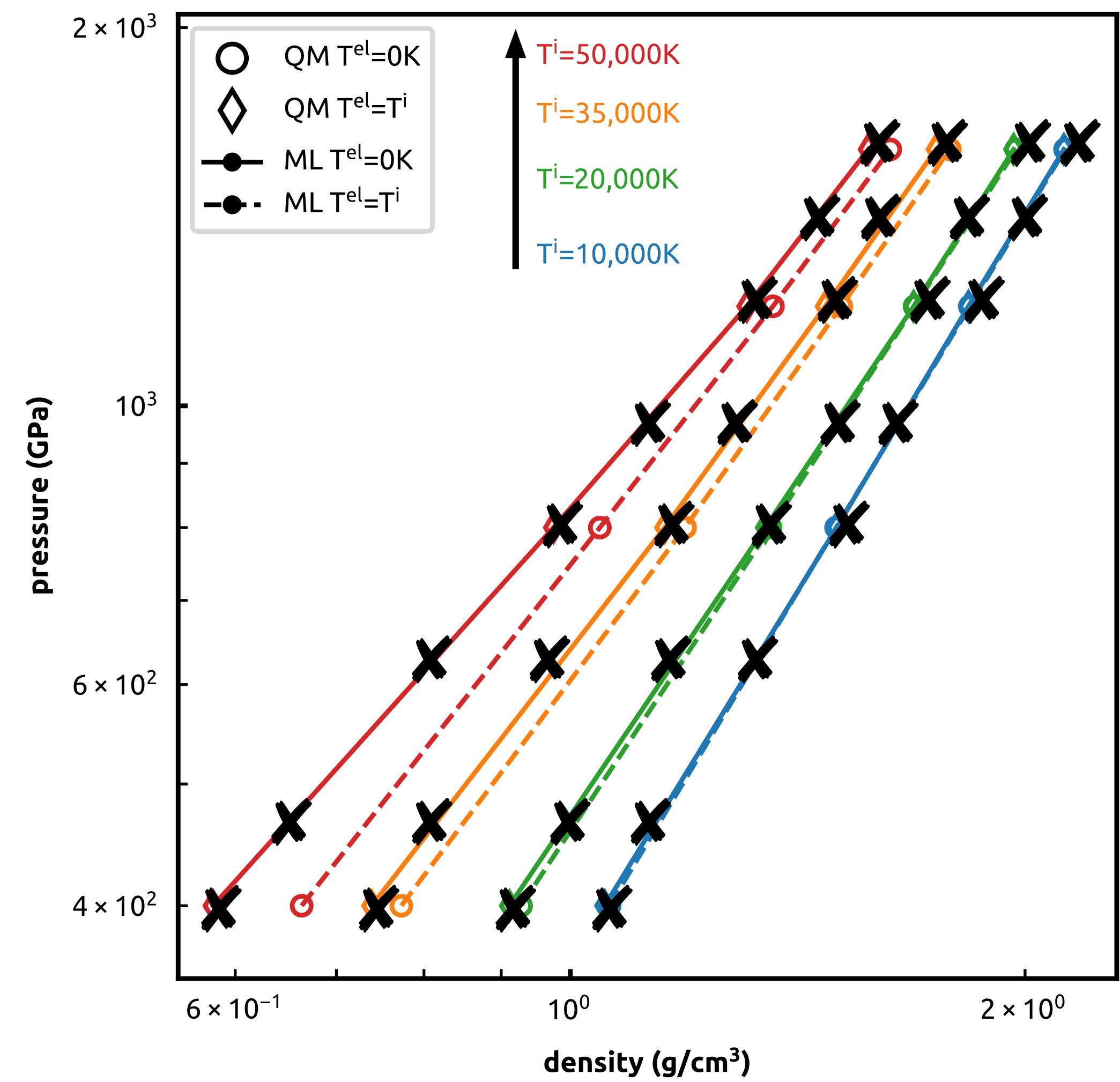


Phase diagram of hydrogen computed with DFT

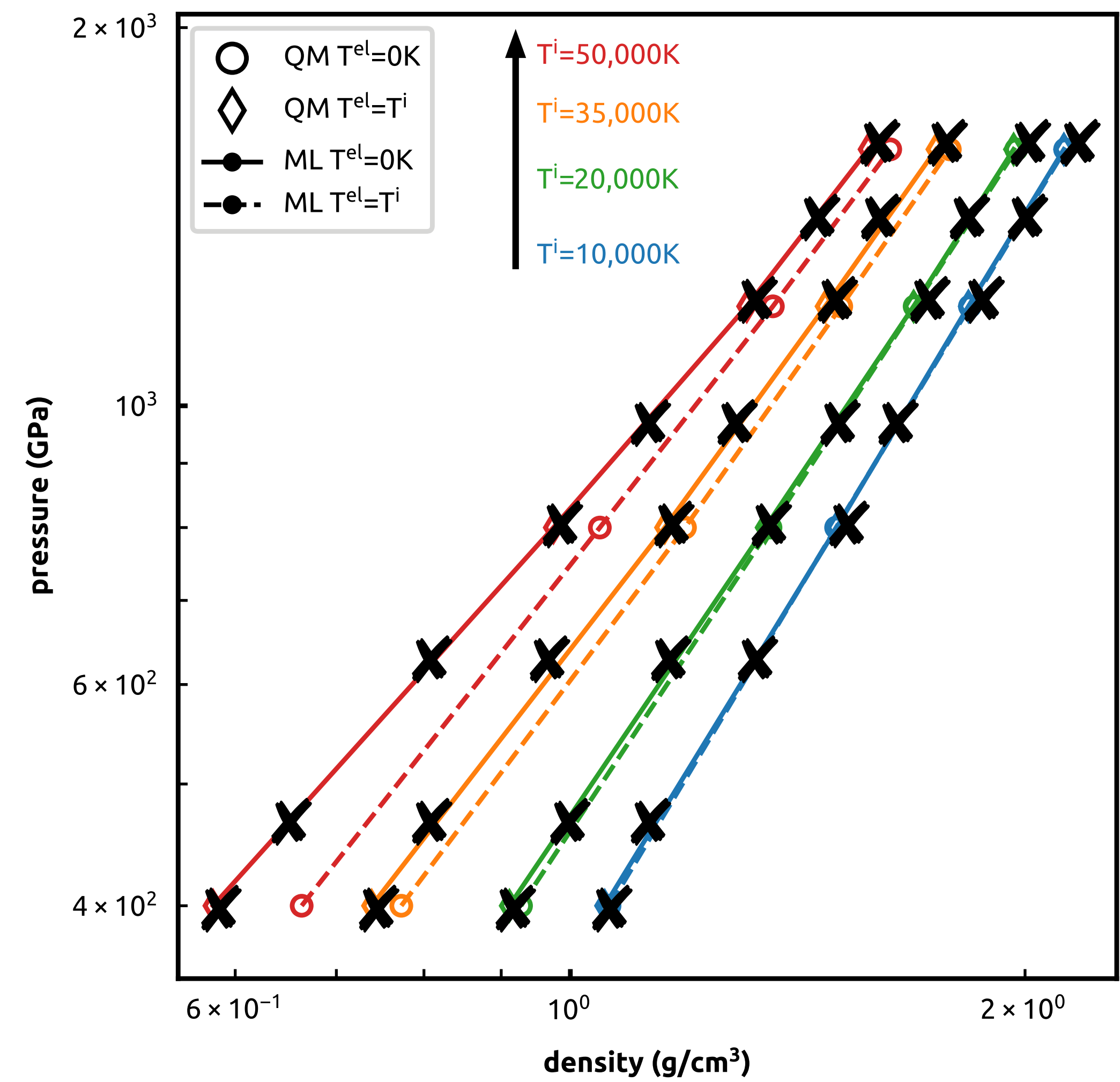
Naive solution



Naive solution

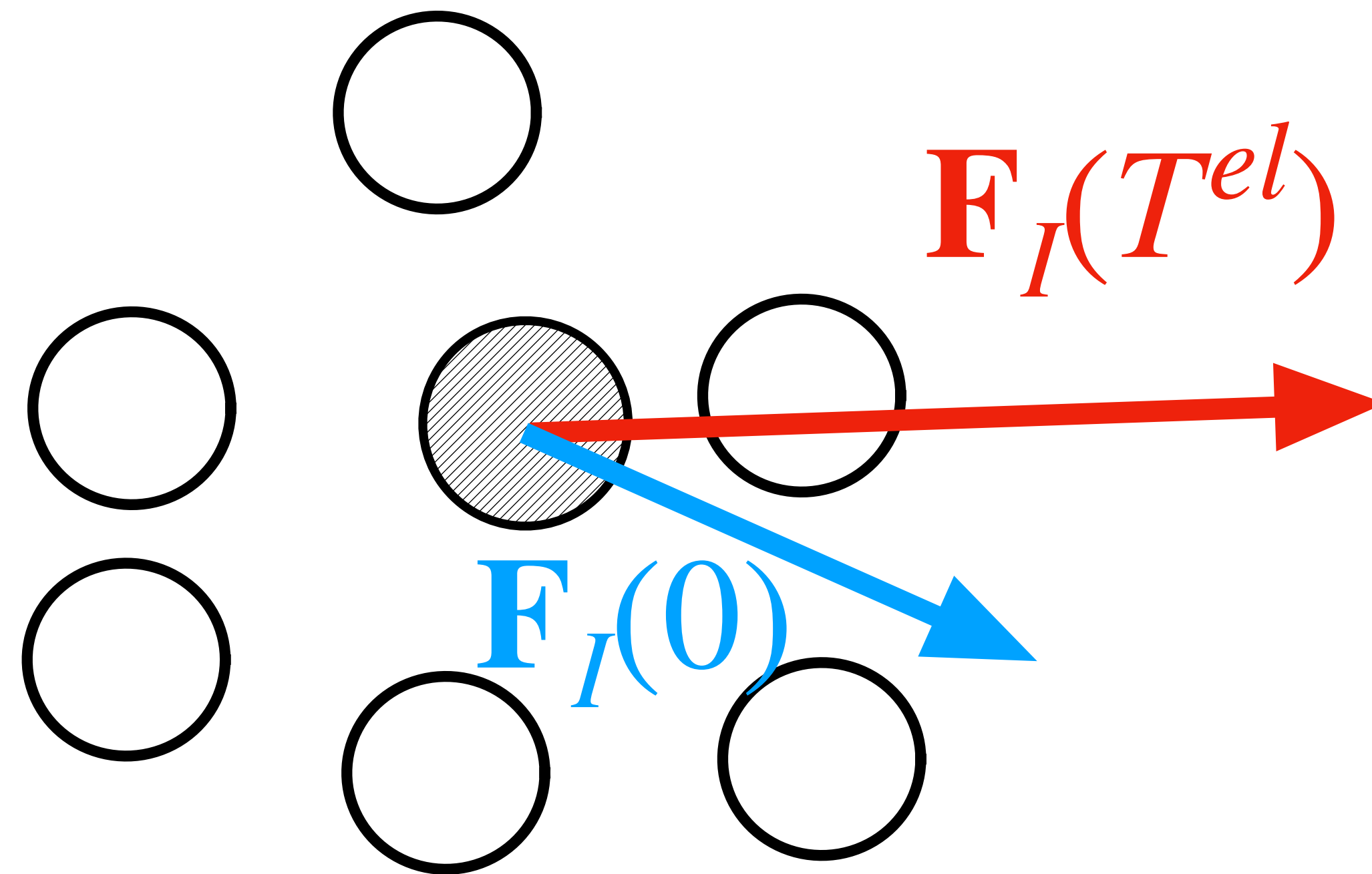


Naive solution

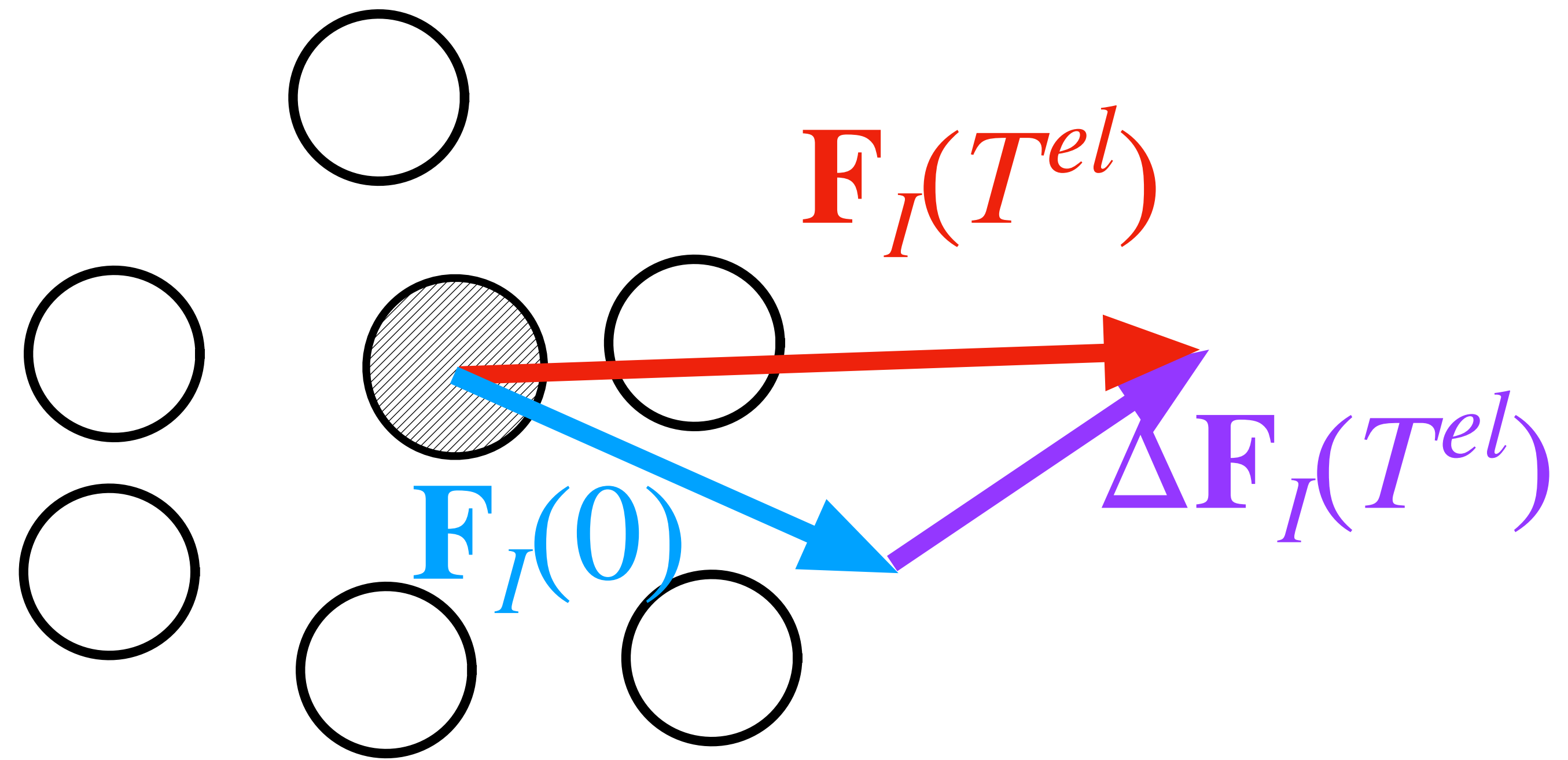


New ML potential for every temperature (4 in total)

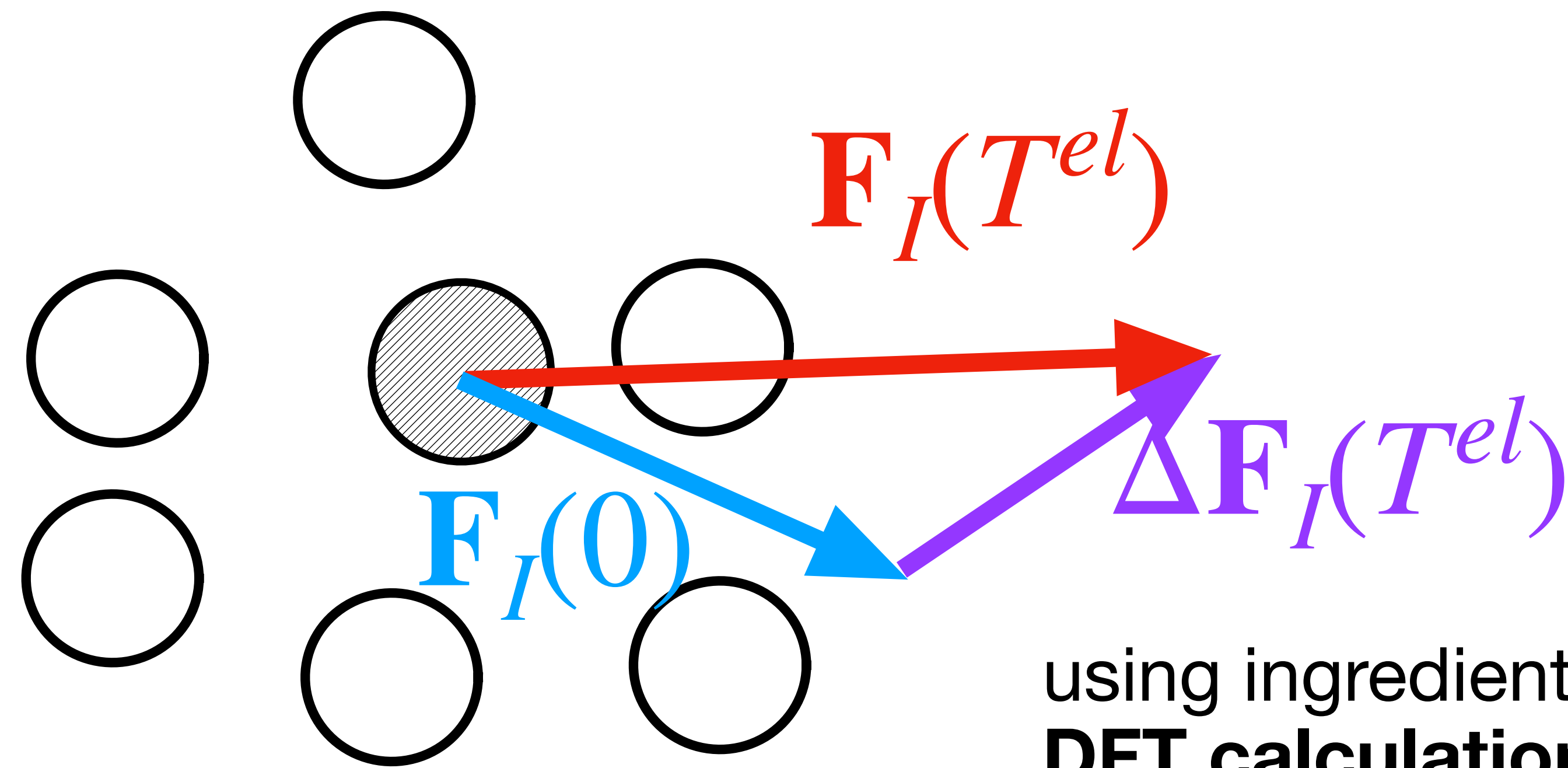
Sketch of the idea



Sketch of the idea



Sketch of the idea



using ingredients from **the same DFT calculation** and extrapolating to **ANY** electronic temperature

Physics-motivated framework

Physics-motivated framework

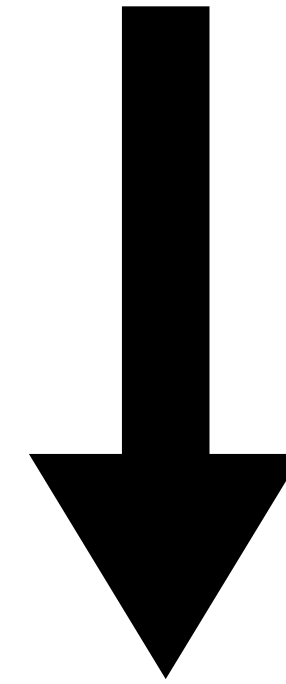
- Ensemble DFT[1, 2]: $A(T^{\text{el}}) = E(T^{\text{el}}) - T^{\text{el}} S_{KS}(T^{\text{el}})$

Physics-motivated framework

- Ensemble DFT[1, 2]: $A(T^{\text{el}}) = E(0) + \Delta E(T^{\text{el}}) - T^{\text{el}} S_{KS}(T^{\text{el}})$

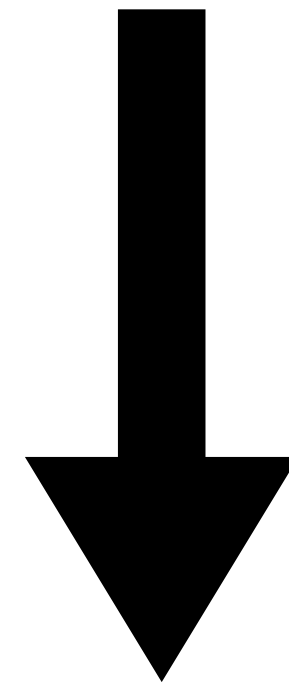
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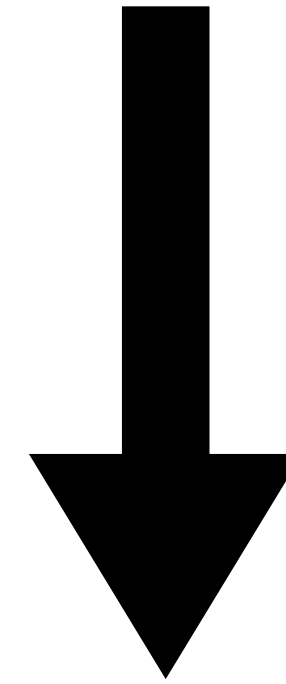
- Approximation: $A(T^{\text{el}}) \approx E(0) + \Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}}) + \mathcal{O}(f^2)$

[1]: Mermin, 1965

[2]: Marzari and Vanderbilt, 1997

Physics-motivated framework

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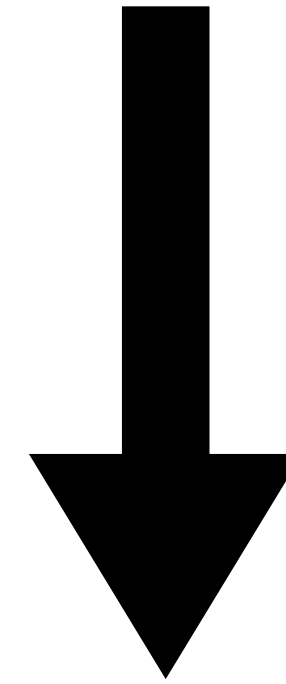
$$\Delta E_{\text{band}}^0(T^{\text{el}}) = \int d\epsilon \text{DOS}^0(\epsilon) \epsilon [f(\mu, T^{\text{el}}) - f(\mu_0, T = 0)] \quad S^0(T^{\text{el}}) = -k_B \int d\epsilon \text{DOS}^0(\epsilon) [f \log(f) - (1 - f) \log(1 - f)]$$

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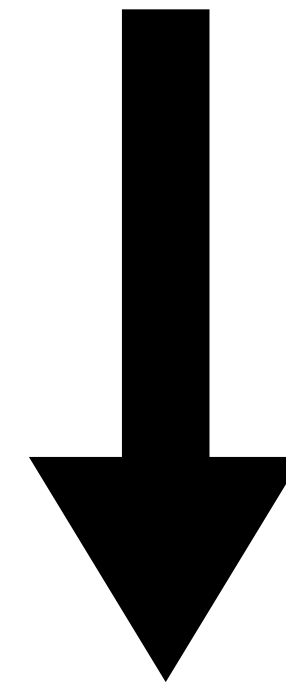


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Physics-motivated framework

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Fixed DOS approximation

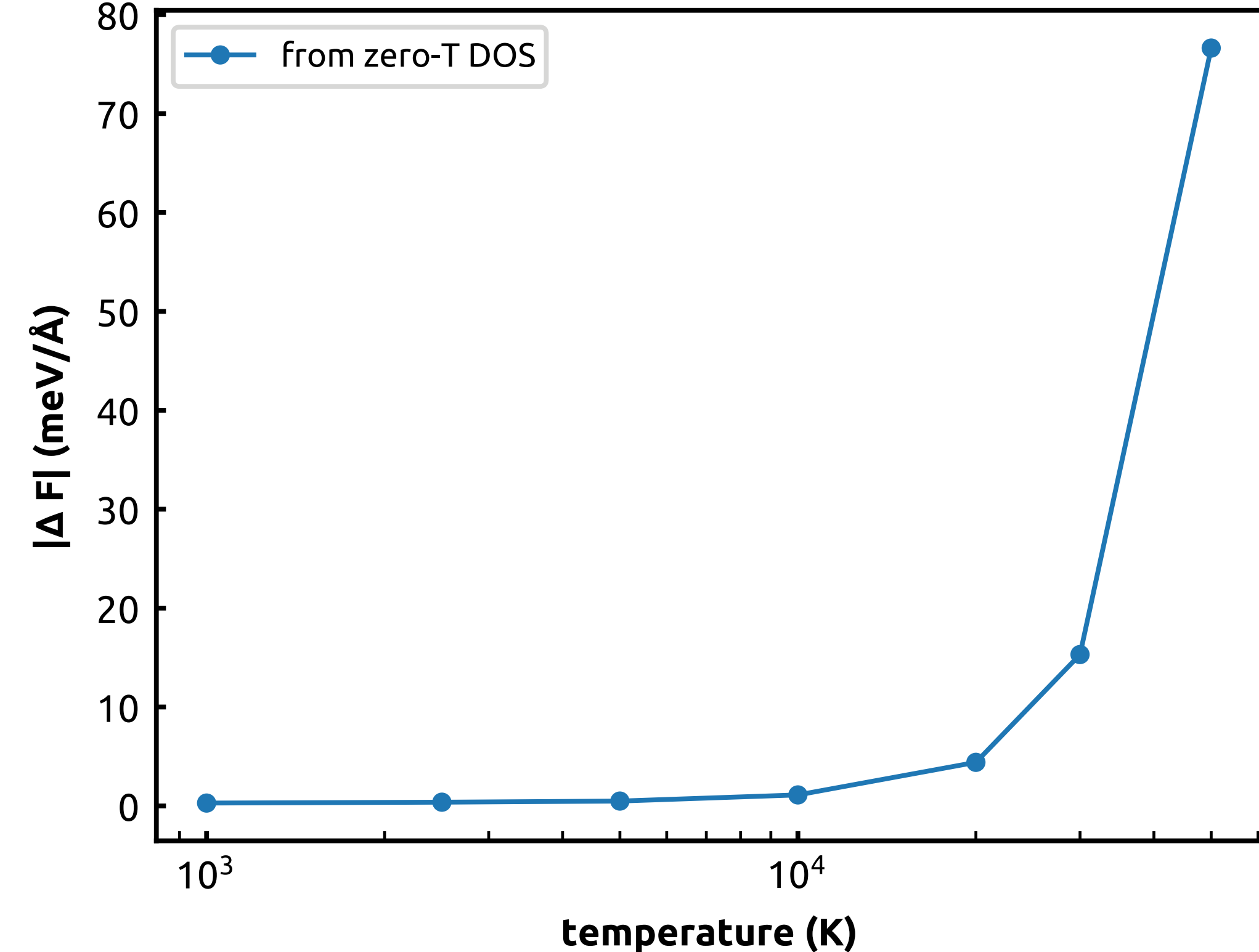
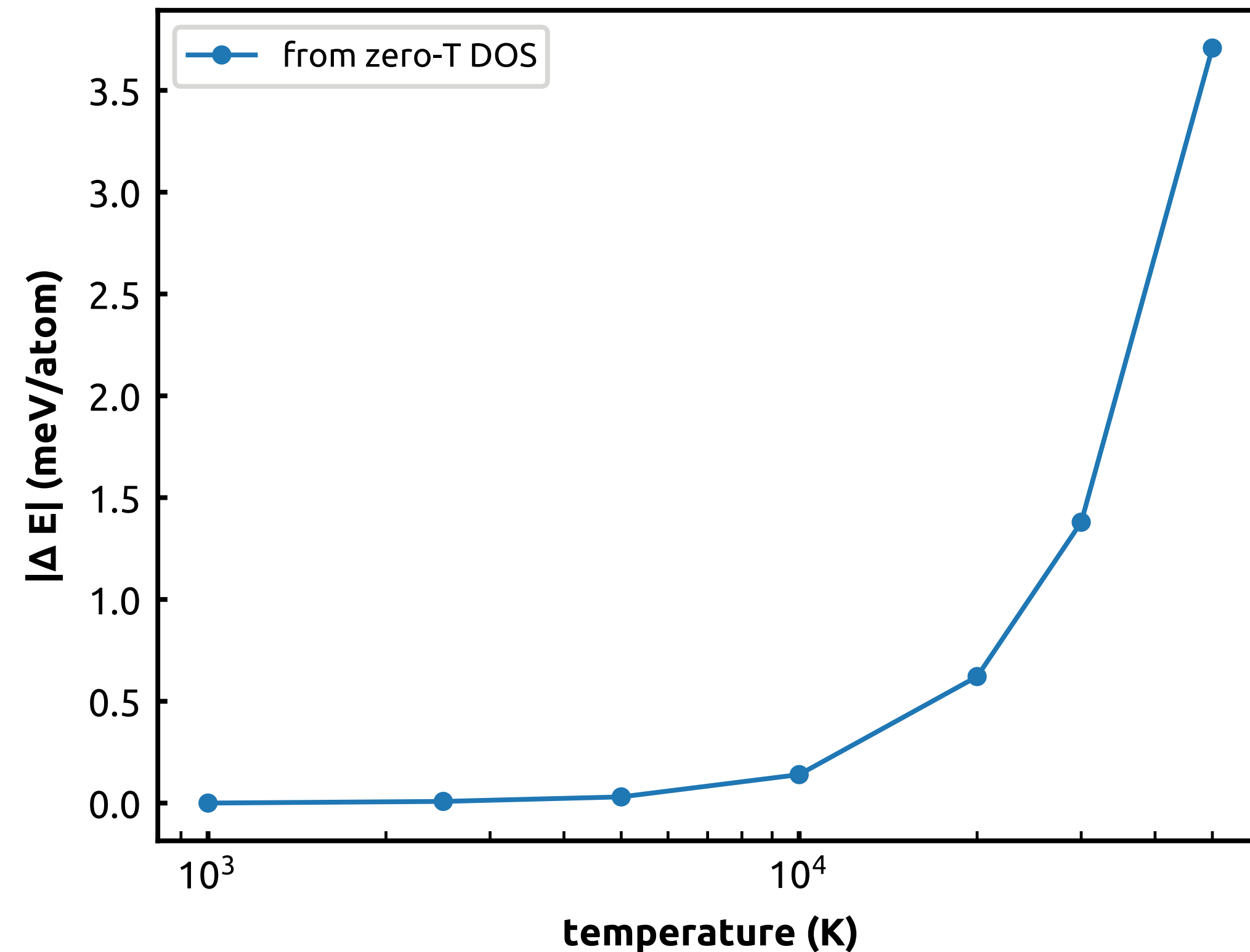
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The temperature T^{el} is an external parameter

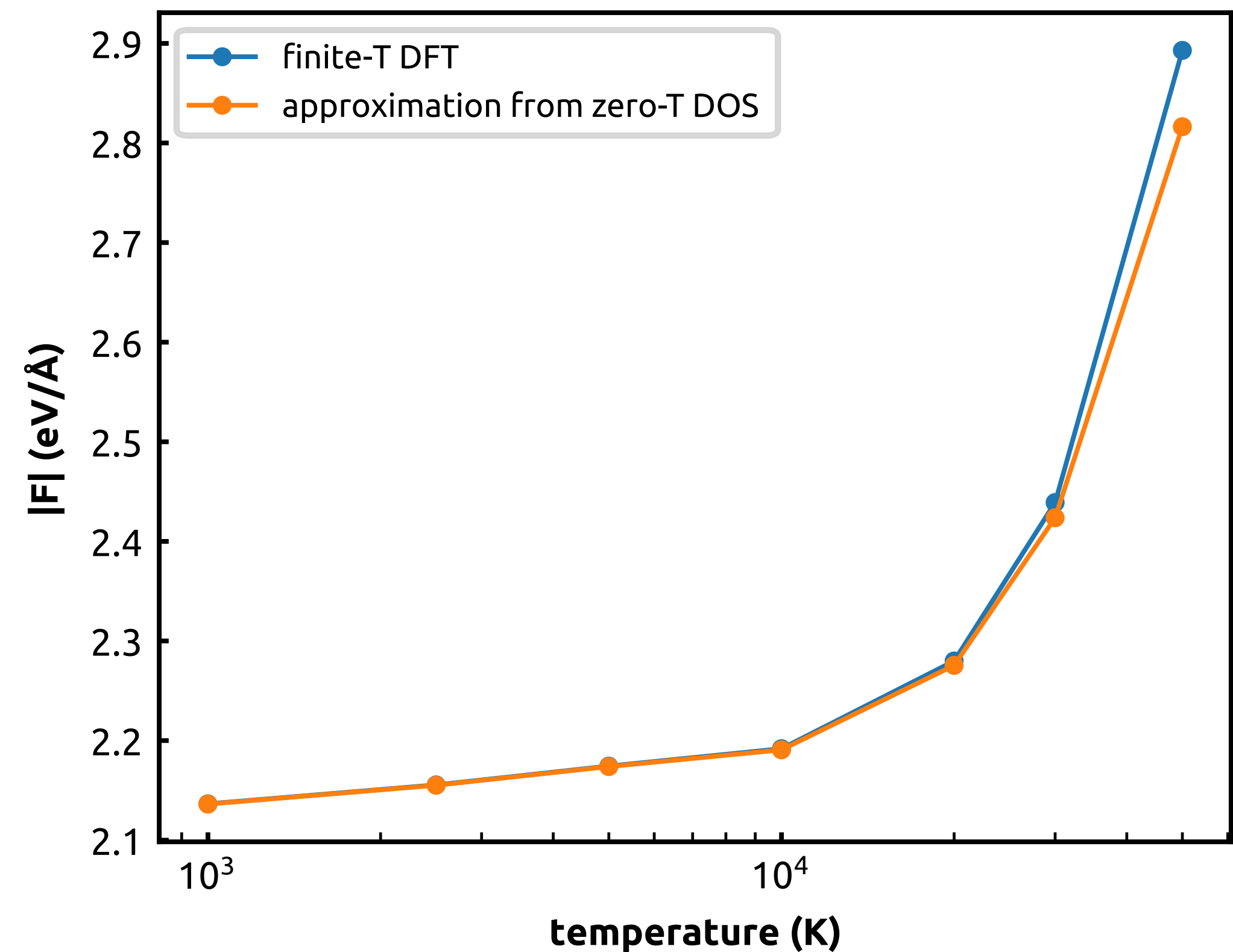
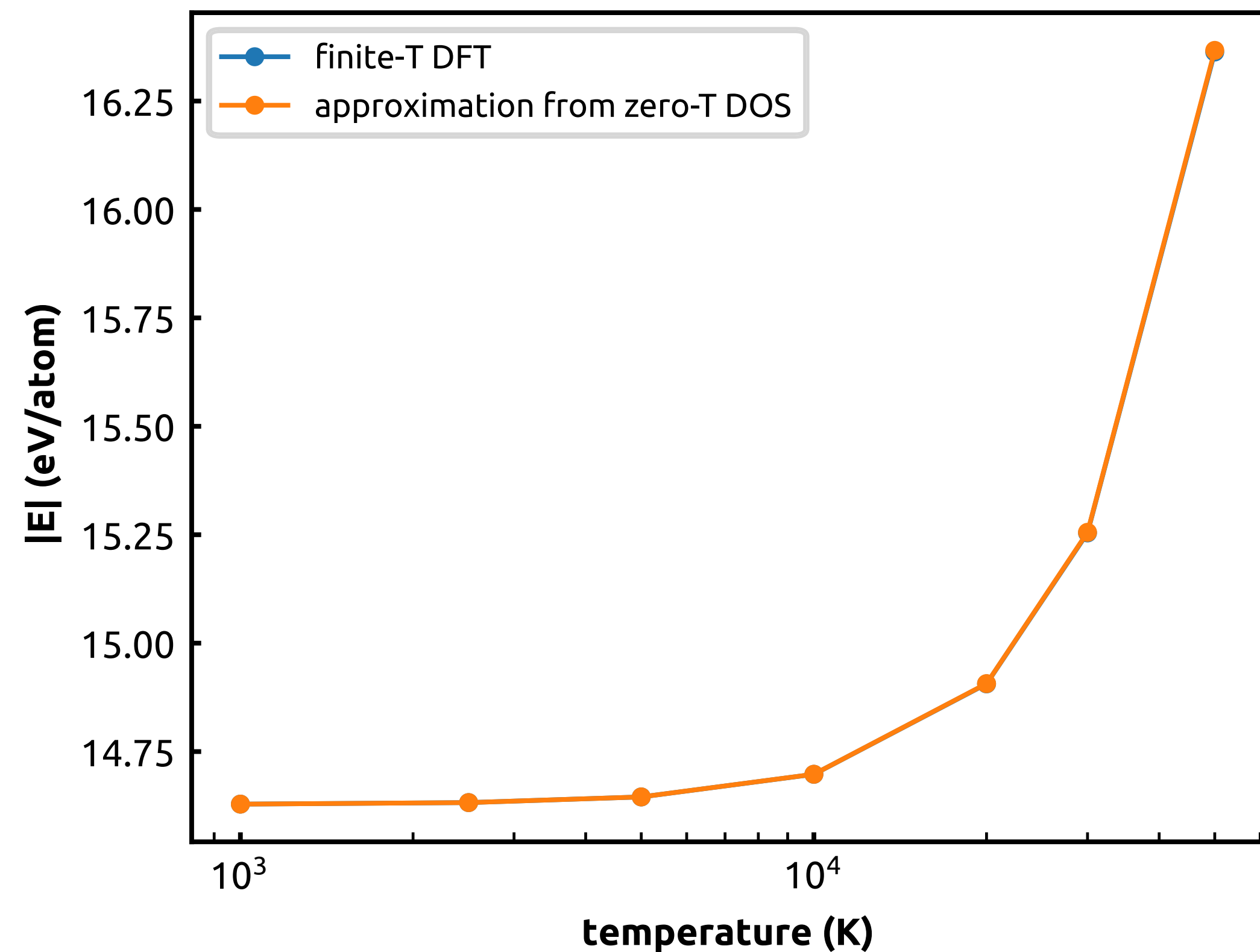
Systematic errors of the approximation

Approximation error in the total free energy and a single force component for a liquid hydrogen structure



Systematic errors of the approximation

Approximation error in the total free energy and a single force component for a liquid hydrogen structure



Application in ML workflows

energy

$$A(T^{\text{el}}) \approx E(0) + \Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}})$$

Application in ML workflows

energy $A(T^{\text{el}}) \approx E(0) + \Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}})$

force $\mathbf{F}_I(T^{\text{el}}) \approx \mathbf{F}_I(0) - \nabla_I (\Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}}))$

stress $\nabla_{\eta} A(T^{\text{el}}) \approx \nabla_{\eta} E(0) + \nabla_{\eta} (\Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}}))$

Application in ML workflows

ANY ML interatomic potentials (GAP, DeepMD, BPNN, (M/Gr)ACE,...)

energy	$A(T^{\text{el}}) \approx$	$E(0)$	$+ \Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}})$
force	$\mathbf{F}_I(T^{\text{el}}) \approx$	$\mathbf{F}_I(0)$	$- \nabla_I (\Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}}))$
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Application in ML workflows

ANY ML interatomic potentials (GAP, DeepMD, BPNN, (M/Gr)ACE,...)

ML DOS models

energy	$A(T^{\text{el}}) \approx$	$E(0)$	+	$\Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}})$
force	$\mathbf{F}_I(T^{\text{el}}) \approx$	$\mathbf{F}_I(0)$	-	$\nabla_I (\Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}}))$
stress	$\nabla_{\eta} A(T^{\text{el}}) \approx$	$\nabla_{\eta} E(0)$	+	$\nabla_{\eta} (\Delta E_{\text{band}}^0(T^{\text{el}}) - T^{\text{el}} S^0(T^{\text{el}}))$

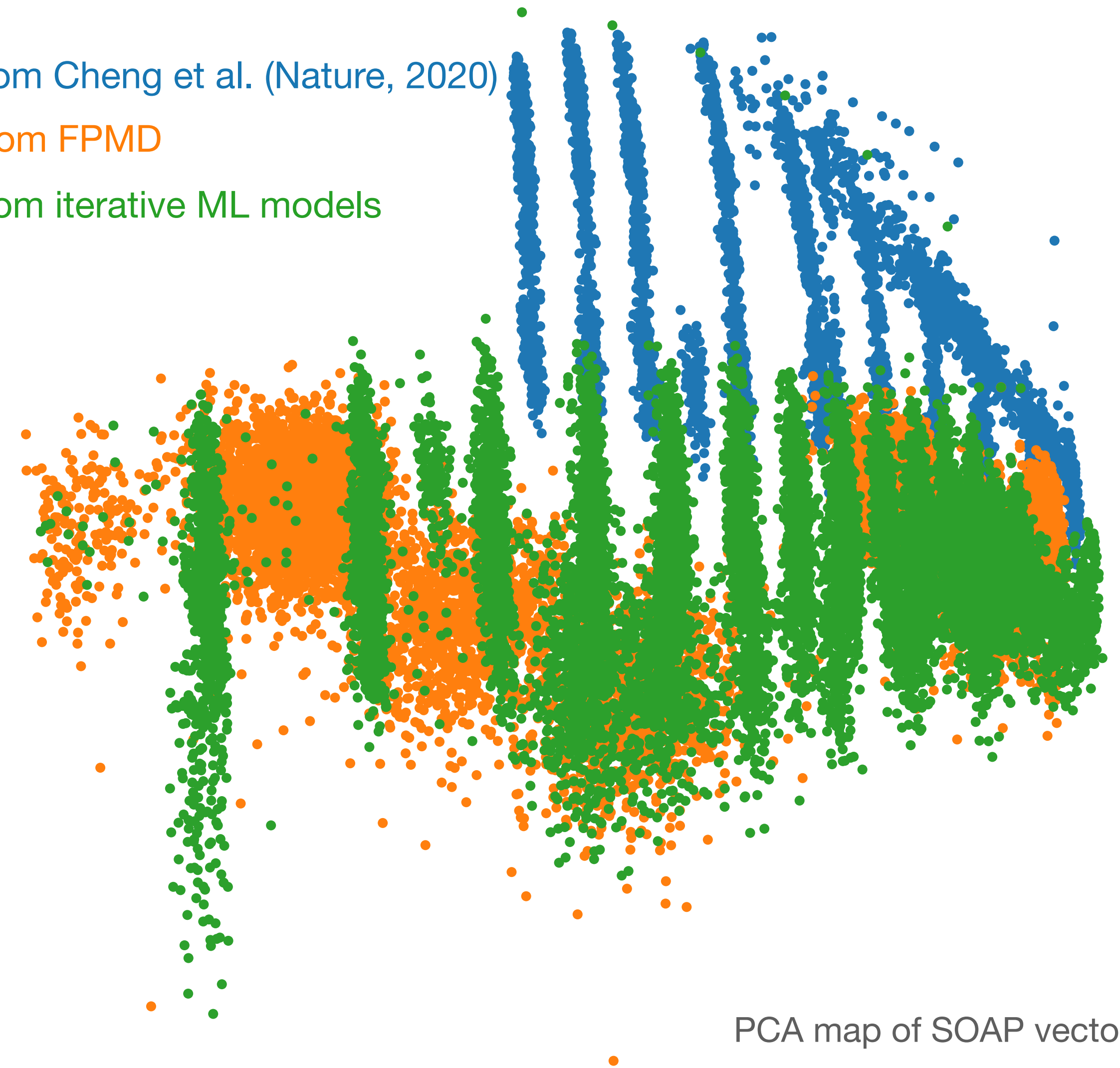
Application to phase diagram of hydrogen

- training set size: ~28,500 structures
- 128 atoms per structure
- DFT calculations on the training set targeting the ground state **ONLY**

from Cheng et al. (Nature, 2020)

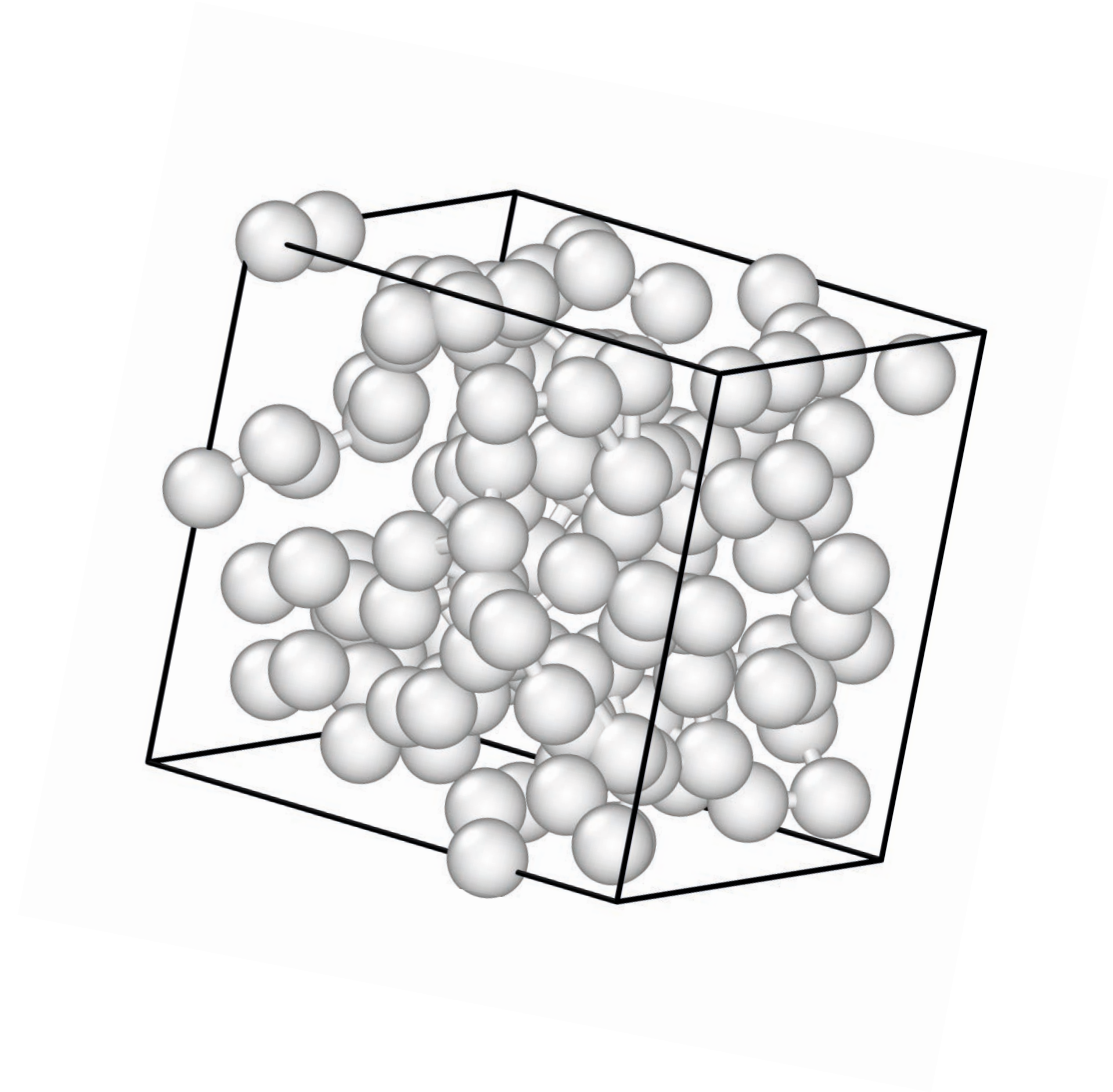
from FPMD

from iterative ML models



PCA map of SOAP vectors

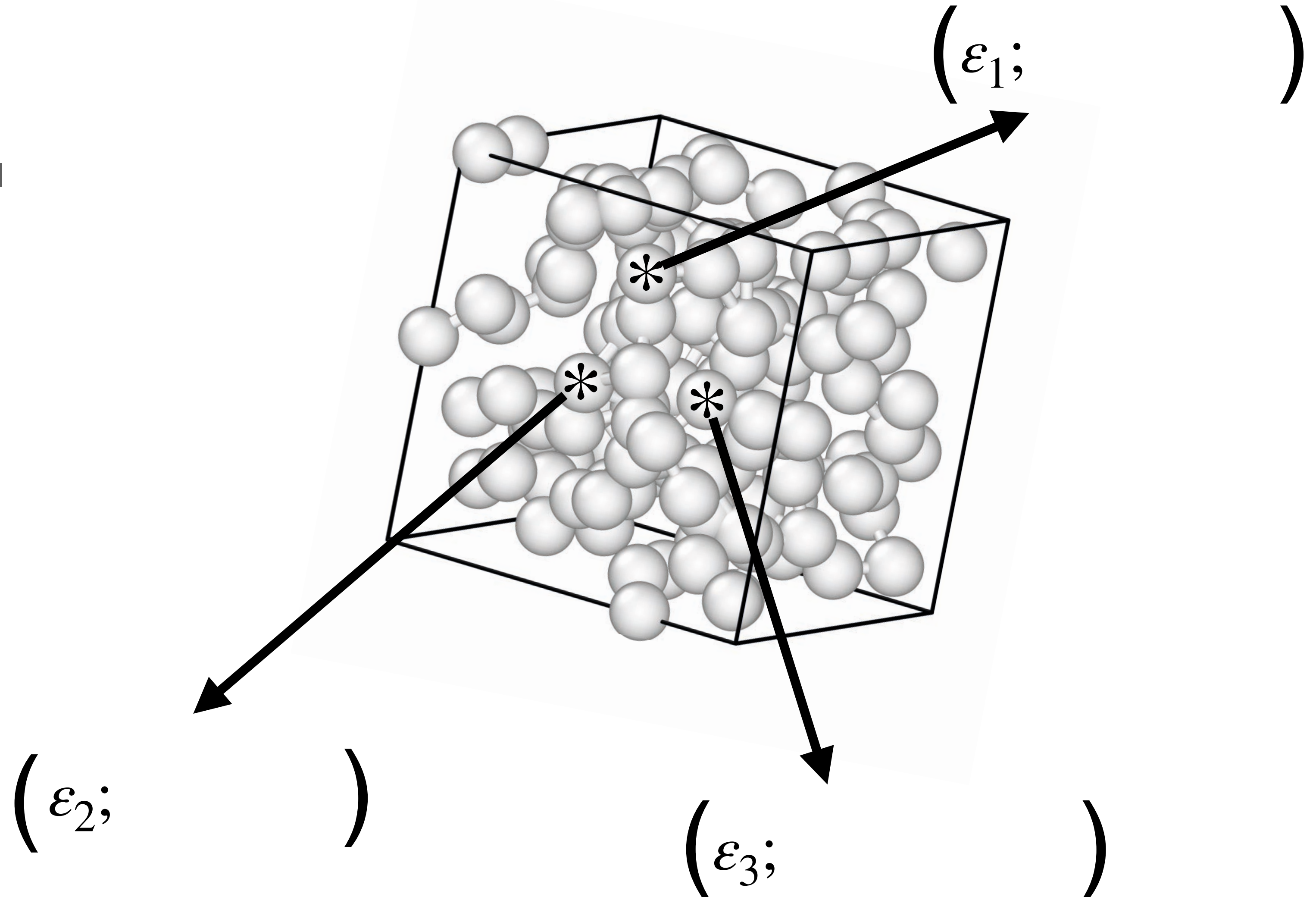
ML models for the energy contributions



ML models for the energy contributions

SOAP based model

- $E(0) = E_{2B} + E_{MB}$

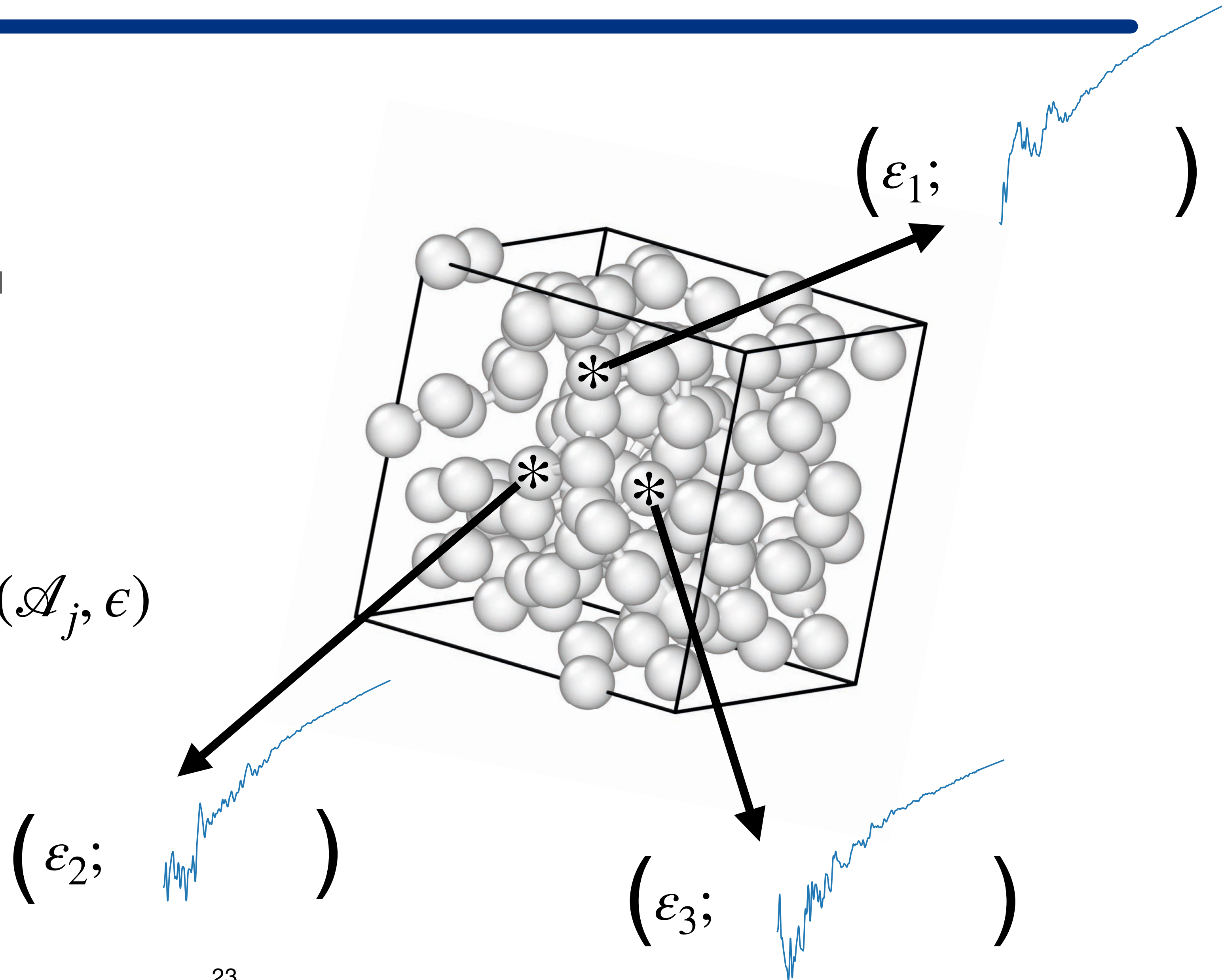


ML models for the energy contributions

SOAP based model

- $E(0) = E_{2B} + E_{MB}$

- $\text{DOS}(\mathcal{A}, \epsilon) = \sum_{\mathcal{A}_j \in \mathcal{A}} \text{LDOS}(\mathcal{A}_j, \epsilon)$

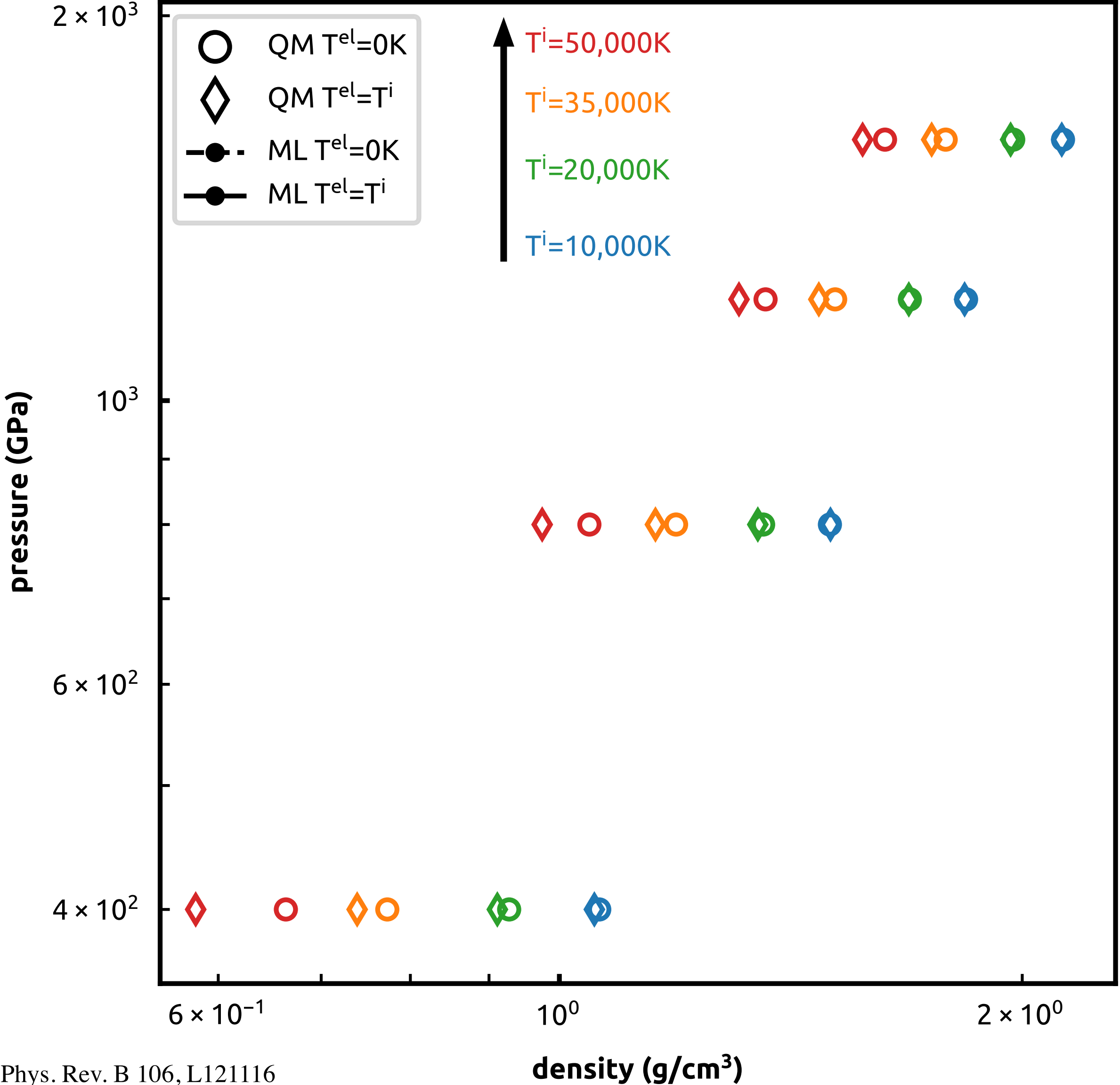


Bartók et al., 2010

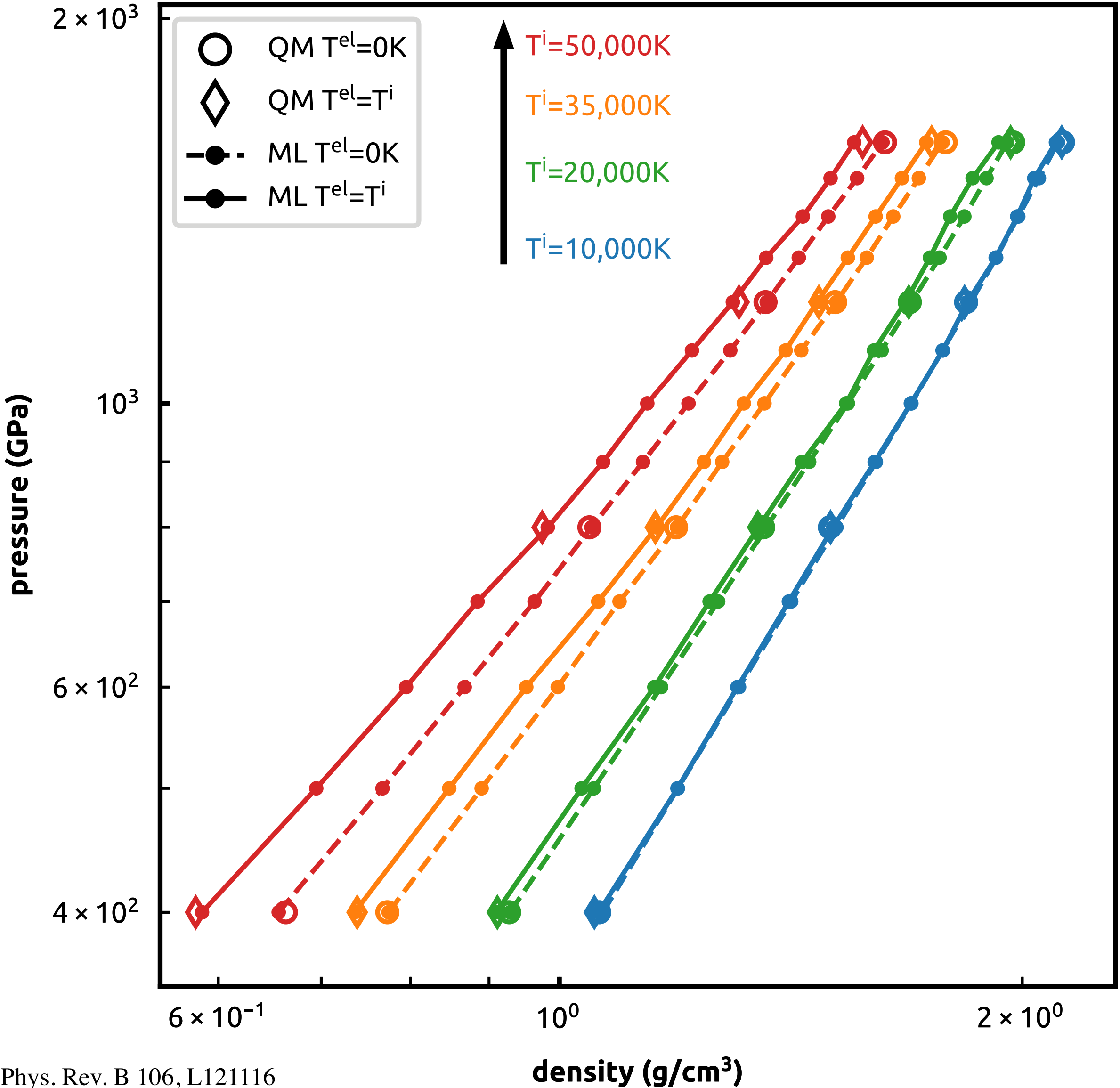
CBM, Anelli, A., Csányi, G., Ceriotti, M., 2020.

Learning the electronic density of states in condensed matter. Phys. Rev. B 102, 235130.

Hydrogen EOS



Hydrogen EOS



CBM, Grasselli, F., Ceriotti, M., 2022.
Predicting hot-electron free energies from ground-state data. Phys. Rev. B 106, L121116

Heat capacity of high pressure hydrogen

- Using finite differences:

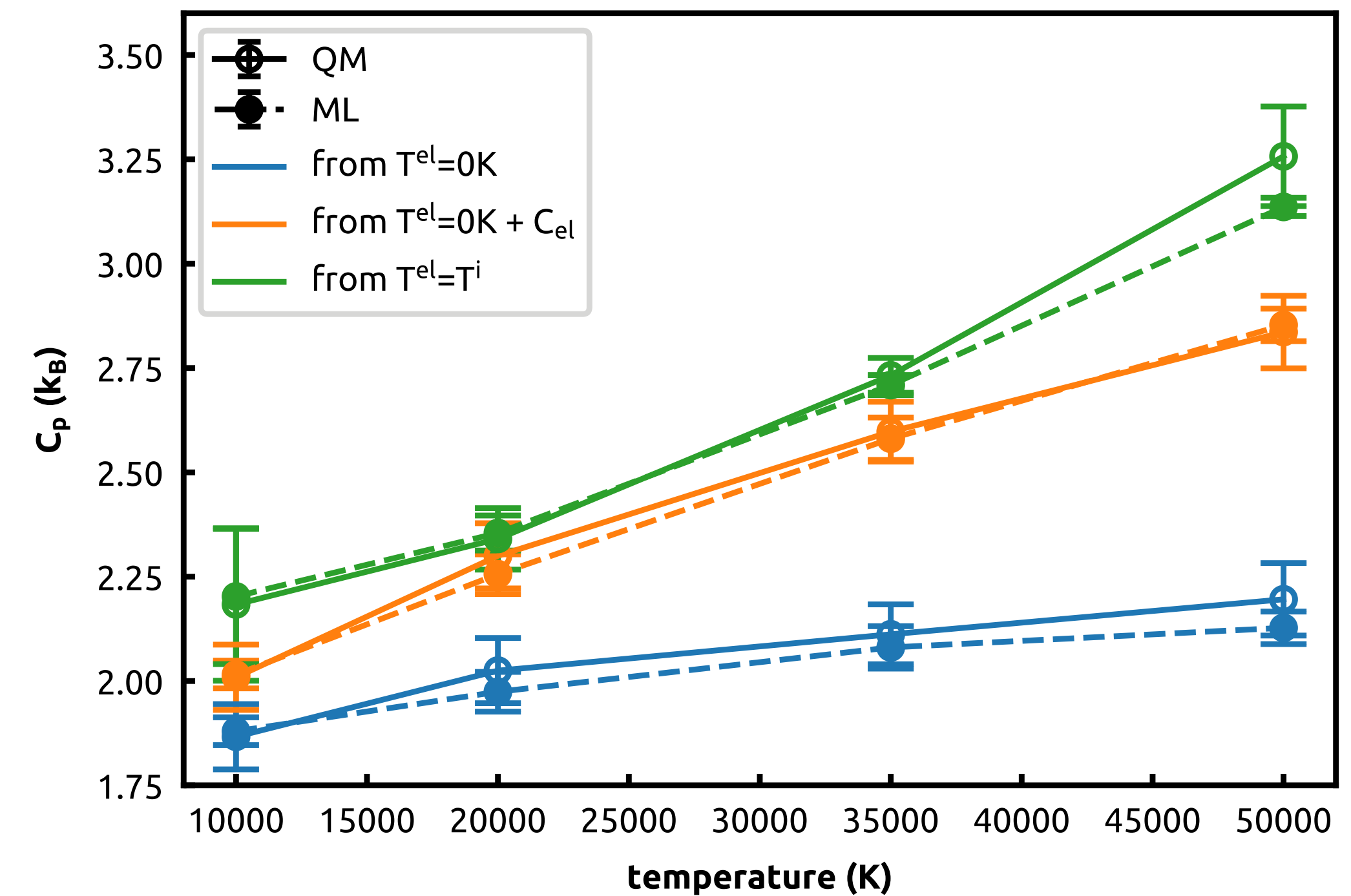
$$C_p = \frac{\langle H(T + \Delta T) \rangle_{Np, T+\Delta T} - \langle H(T) \rangle_{Np, T}}{\Delta T}$$

Heat capacity of high pressure hydrogen

- Using finite differences:

$$C_p = \frac{\langle H(T + \Delta T) \rangle_{Np} - \langle H(T) \rangle_{Np}}{\Delta T}$$

Constant pressure heat capacity of hydrogen at 400GPa



Takehome message

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- Atom-centred DOS models are versatile and transferable

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- ML DOS models can be used in physics-inspired modelling of materials:

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- Atom-centred DOS models are versatile and transferable
- ML DOS models can be used in physics-inspired modelling of materials:
 - characterise local electronic fingerprints

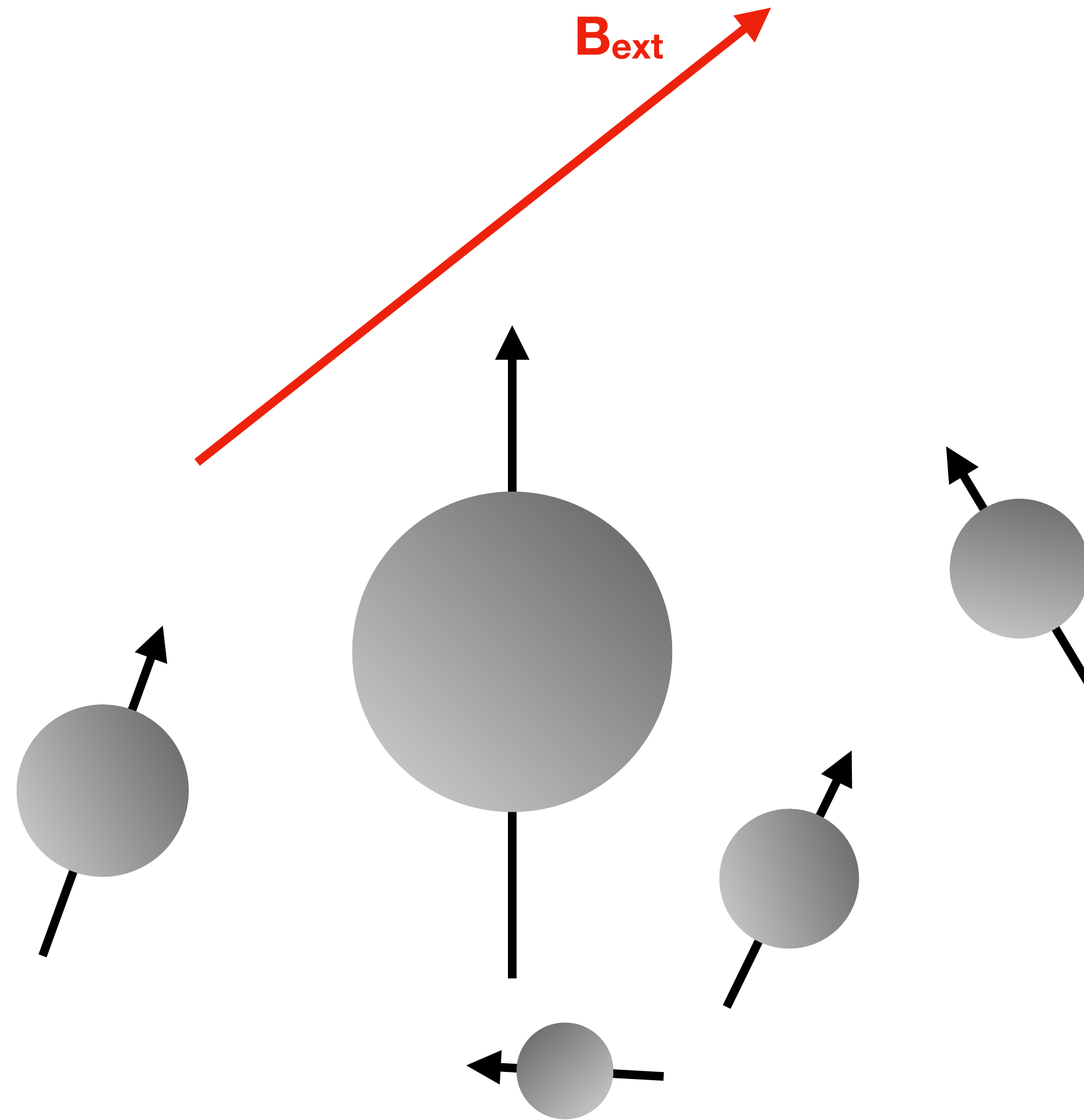
Takehome message

- Atom-centred DOS models are versatile and transferable
- ML DOS models can be used in physics-inspired modelling of materials:
 - characterise local electronic fingerprints
 - incorporate electron-finite-temperature effects

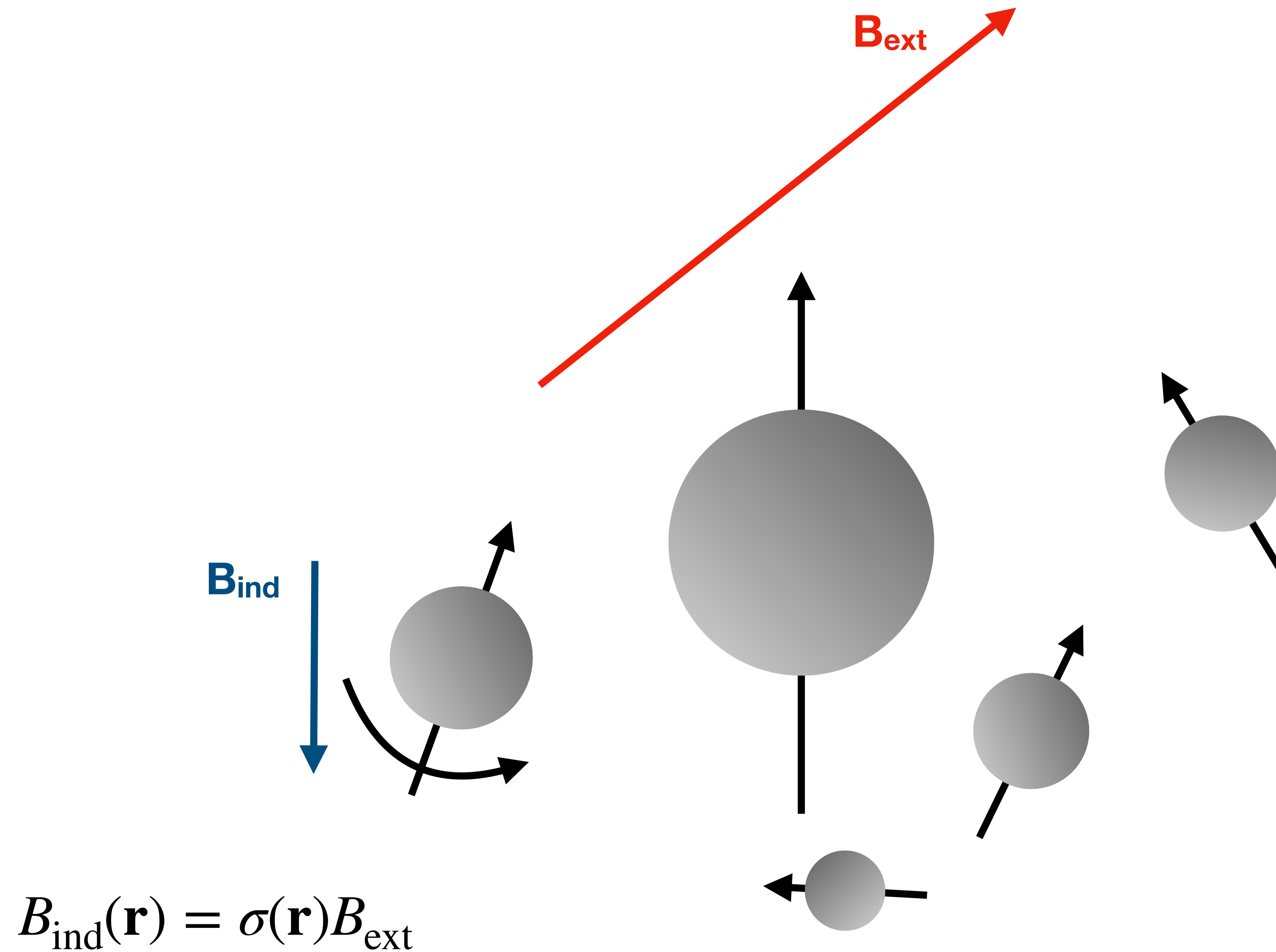
Property 2:

solid-state nuclear magnetic resonance parameters

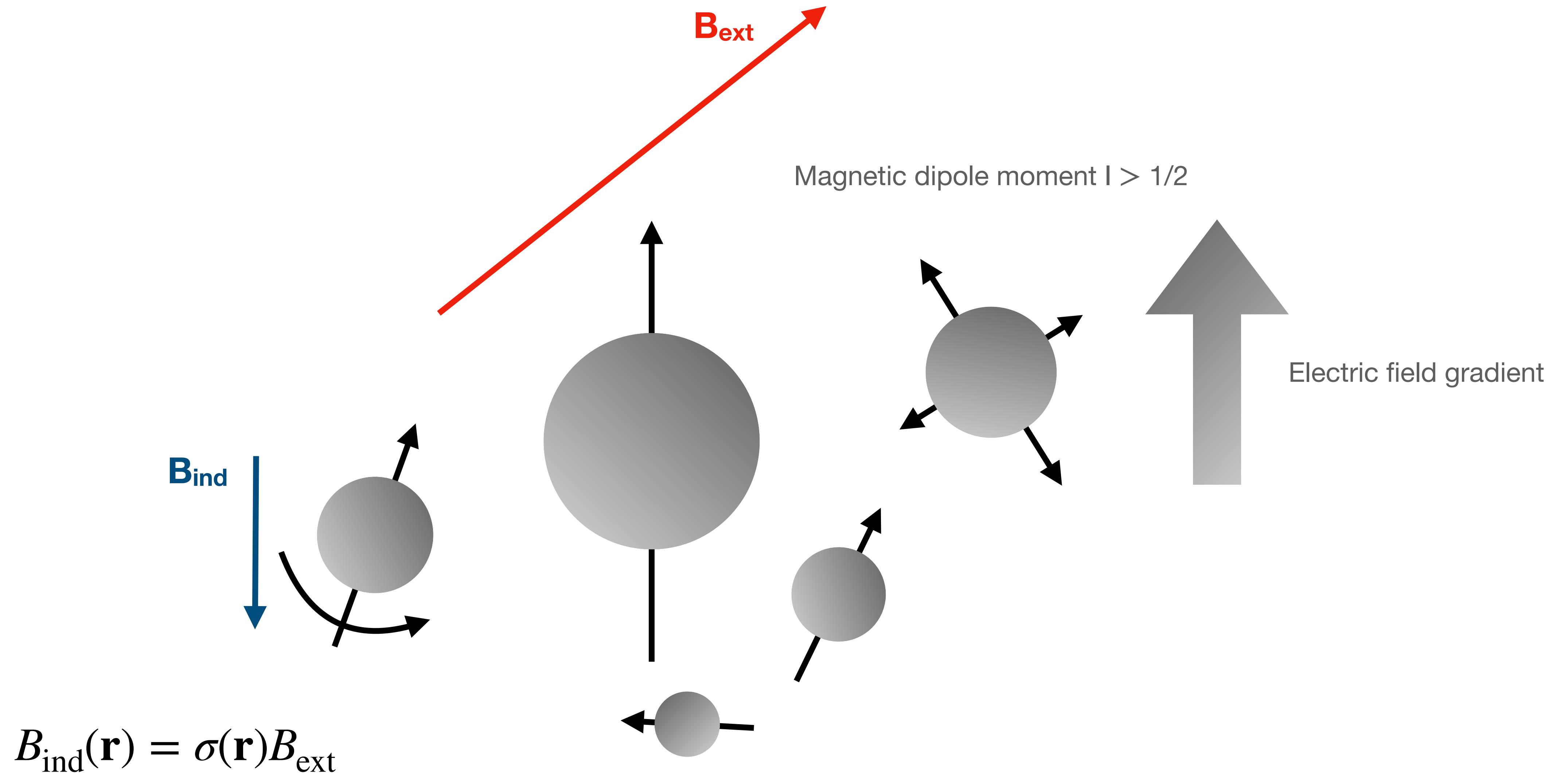
(Quick) Introduction to NMR



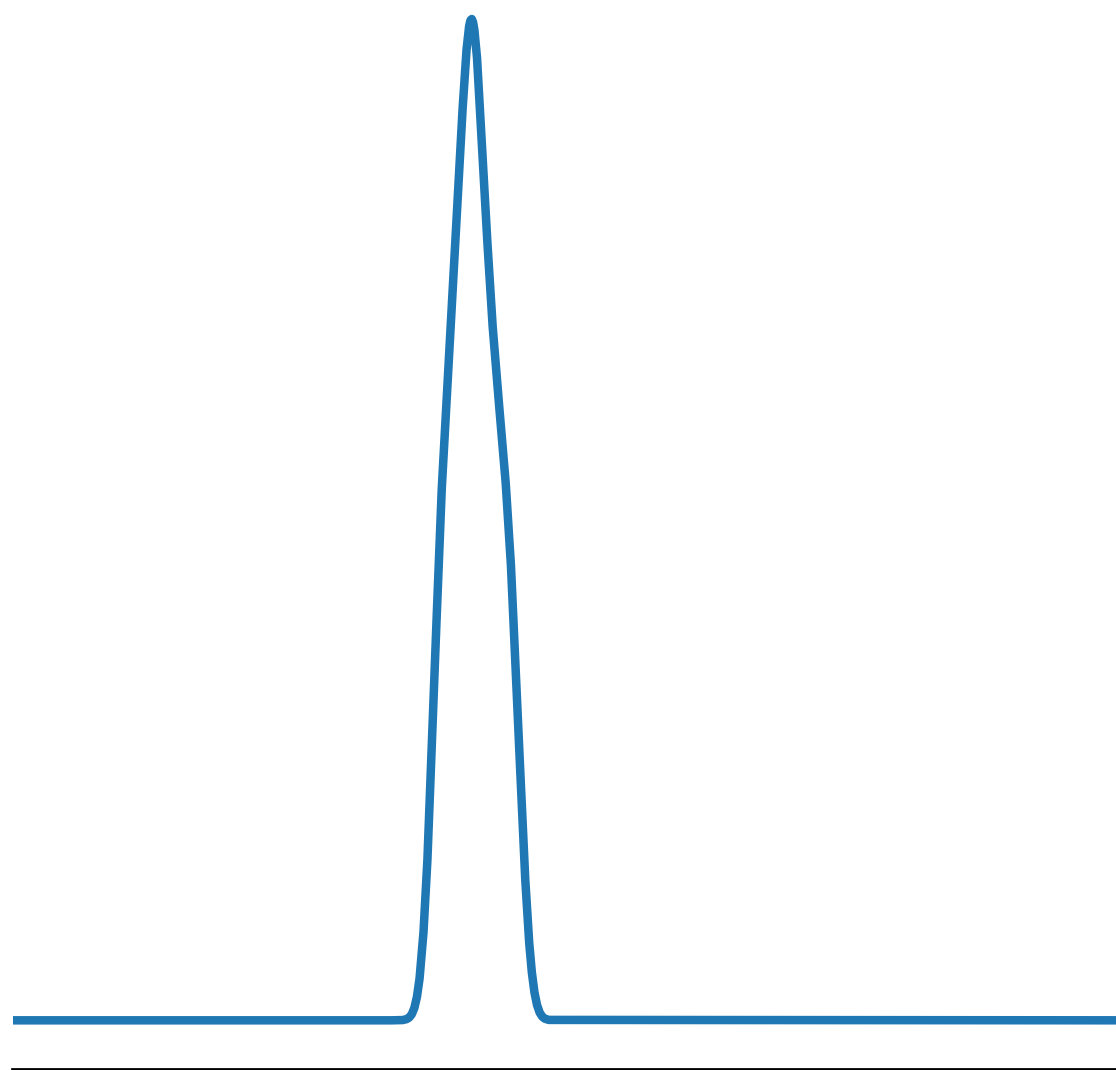
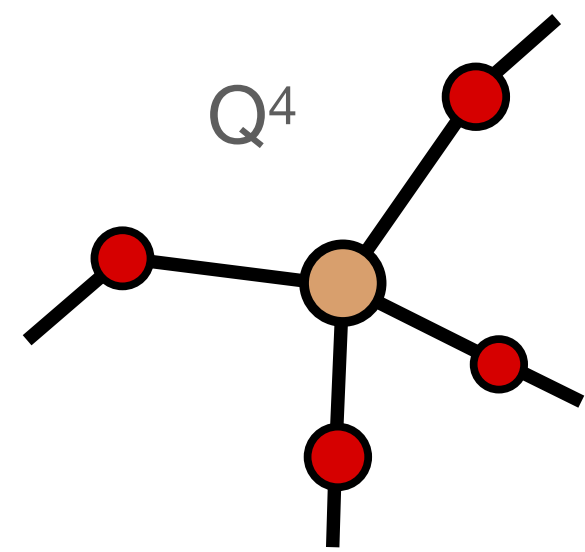
(Quick) Introduction to NMR



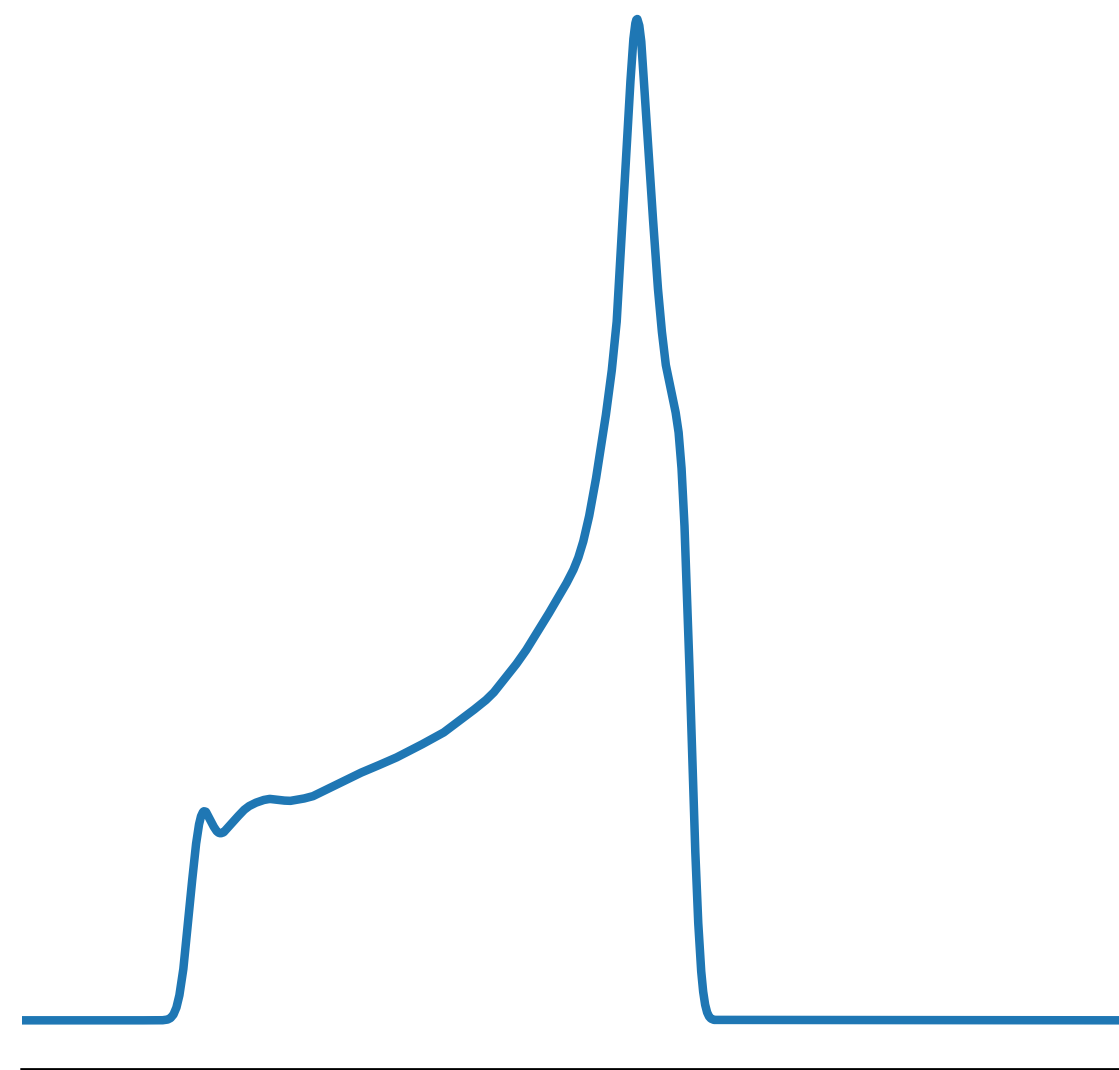
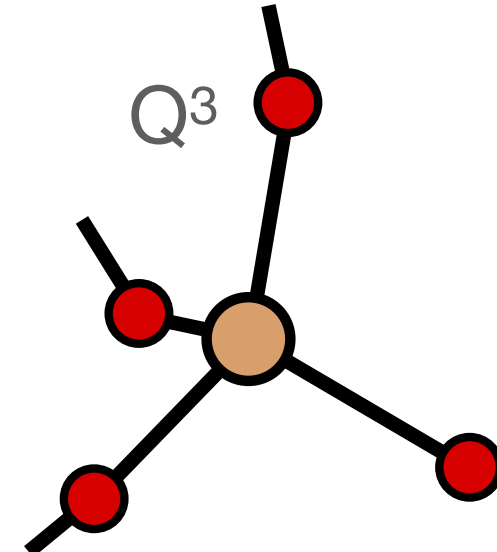
(Quick) Introduction to NMR



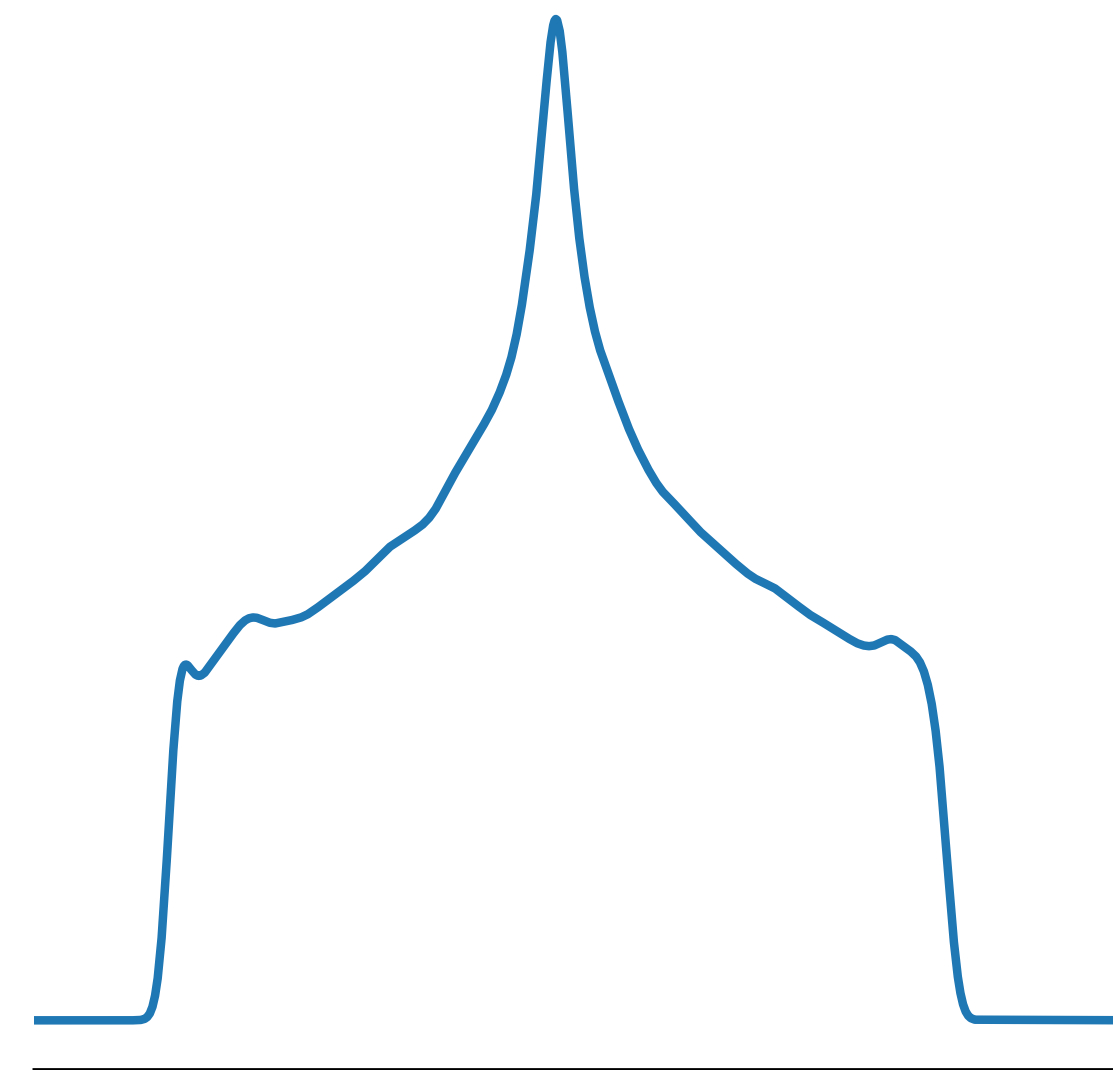
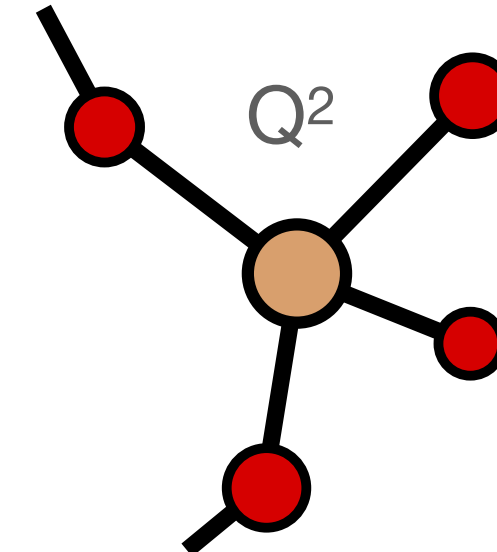
NMR parameters as local fingerprints



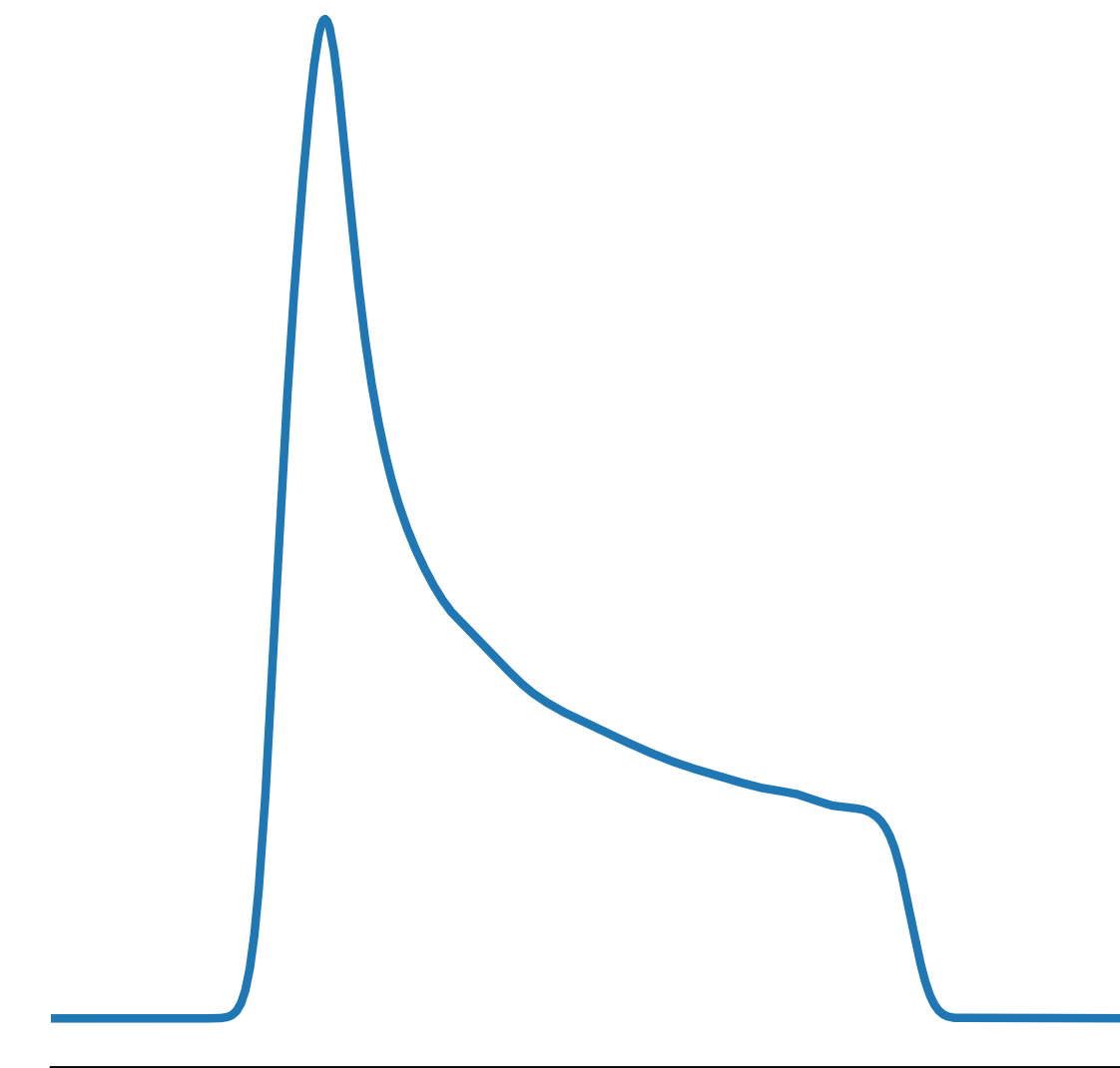
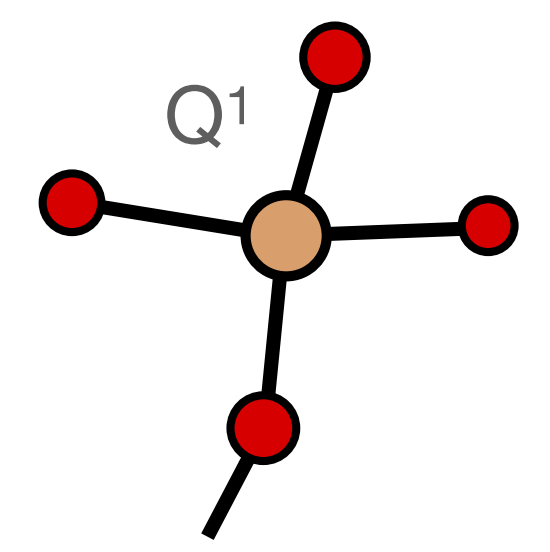
^{29}Si NMR frequency (ppm)



^{29}Si NMR frequency (ppm)



^{29}Si NMR frequency (ppm)

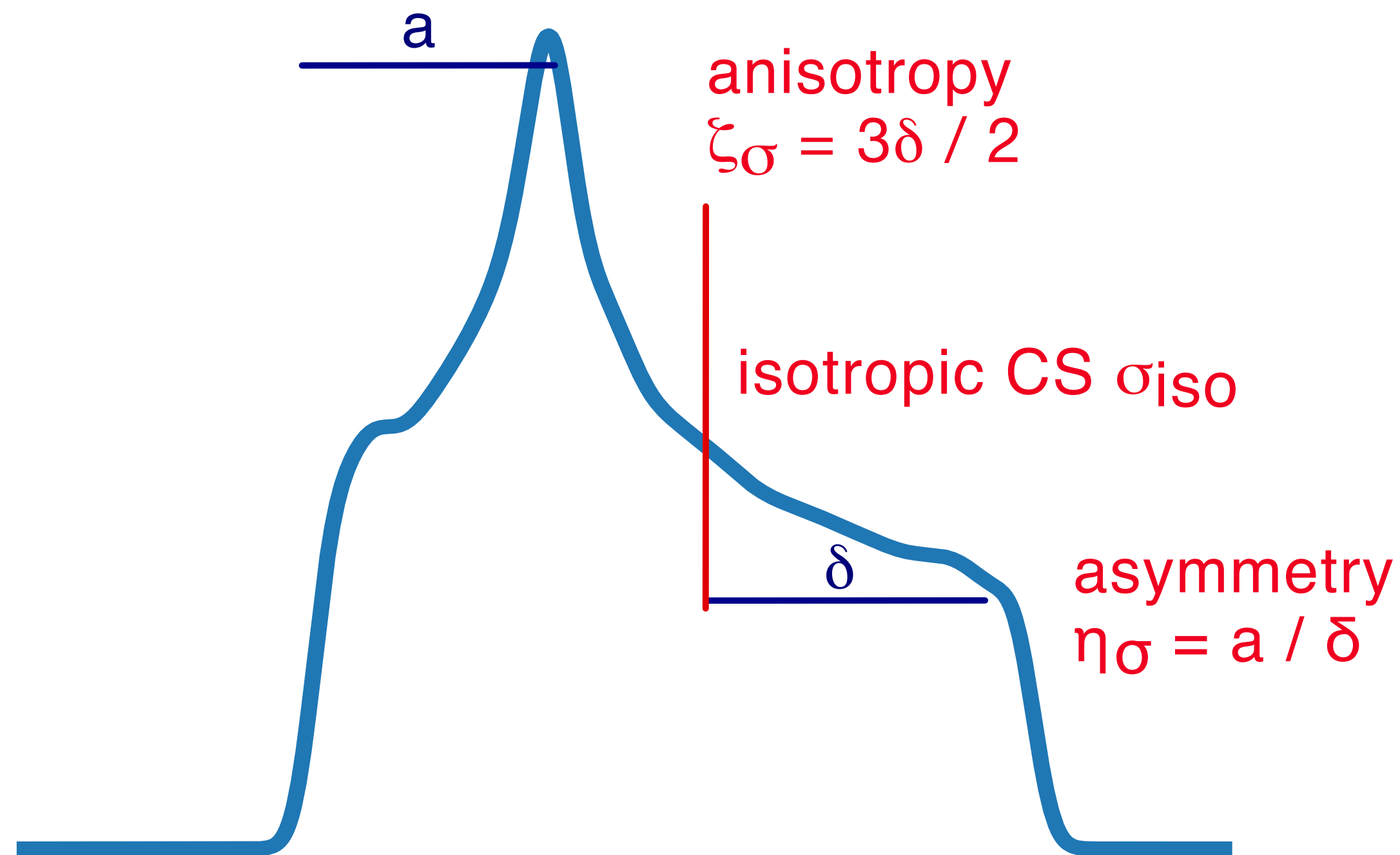


^{29}Si NMR frequency (ppm)

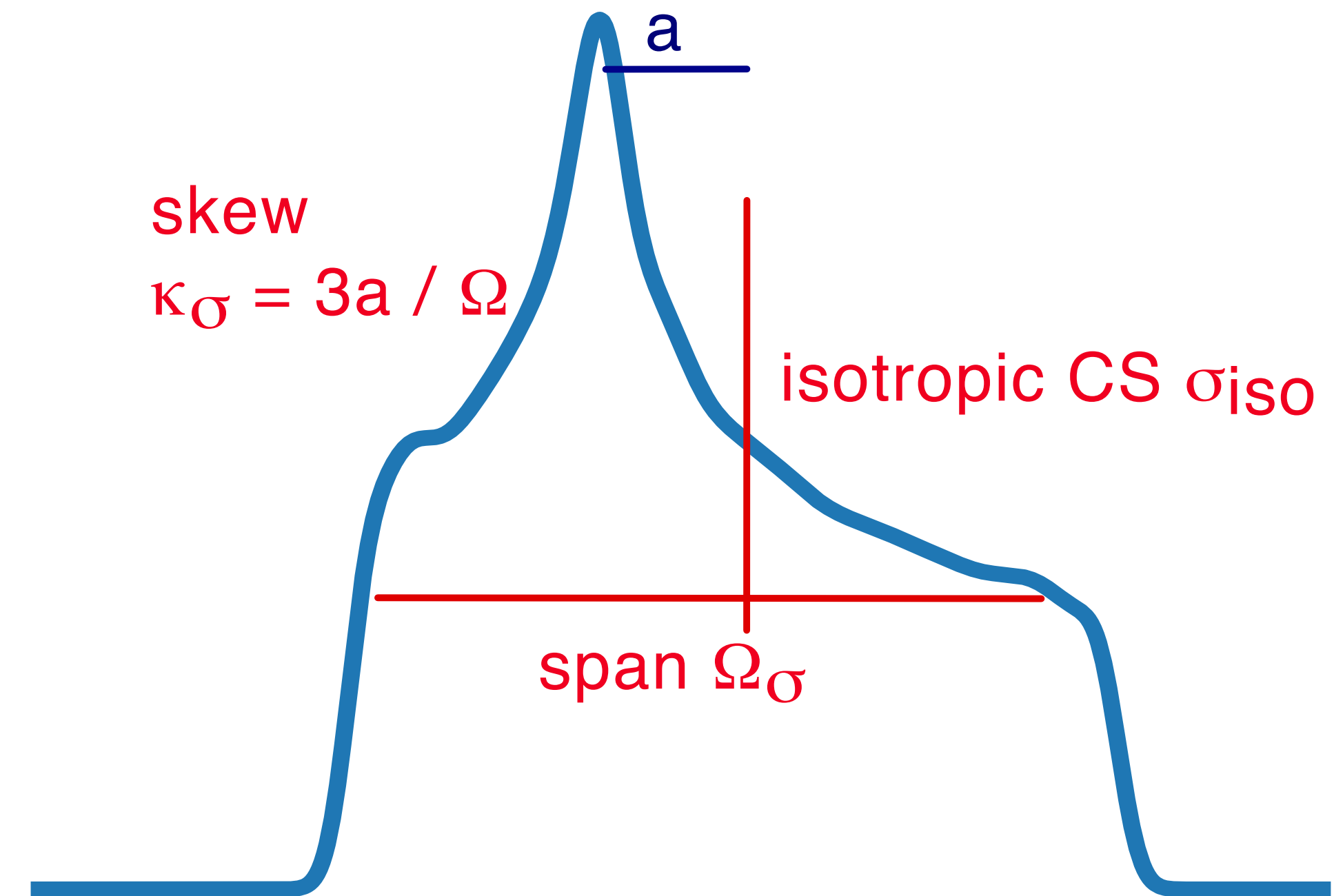
Objectives

- Build a **transferable ML model** for the NMR parameters
- Identify the **challenges** in modelling NMR parameters as tensors
- Apply the ML NMR models to study :
 - the **feature/property** relations
 - dynamics of solids

Conventions for NMR tensors



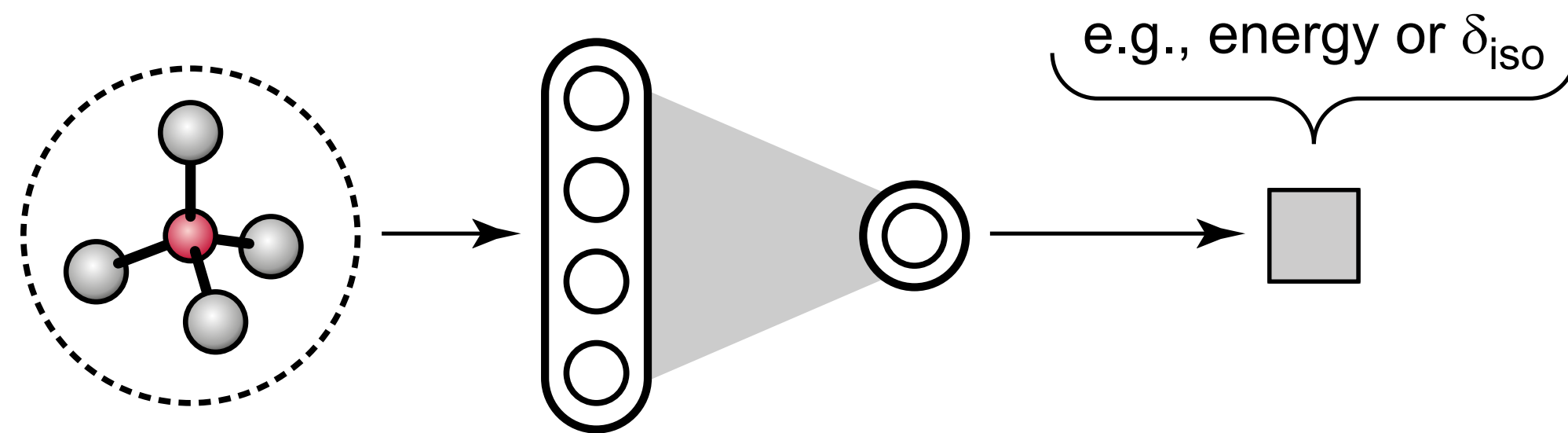
Haeberlen convention



Maryland convention

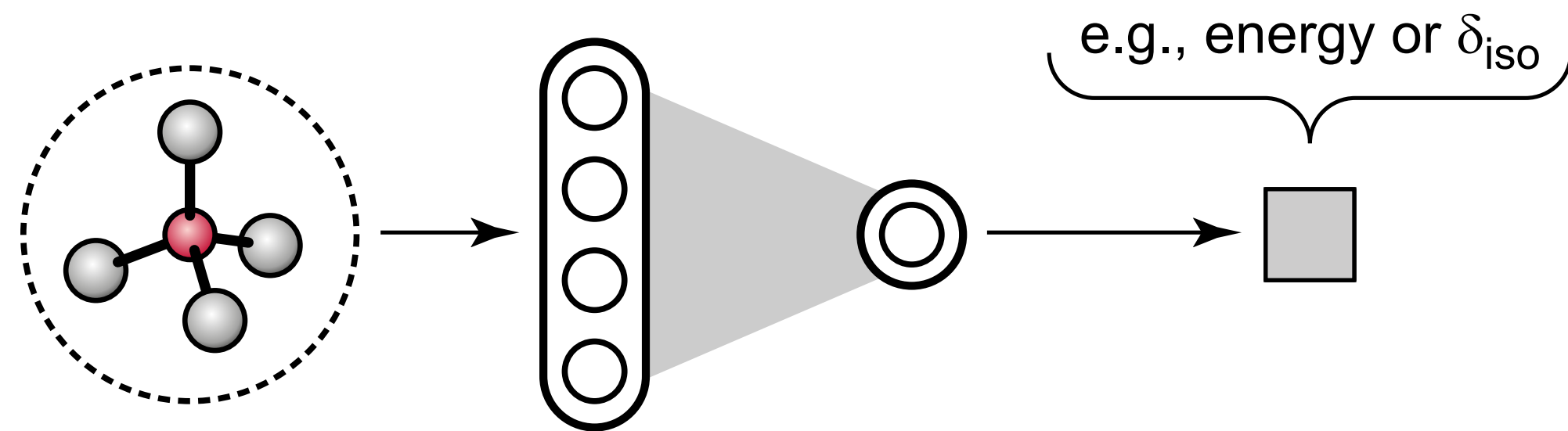
ML architecture

a Scalar atomistic ML

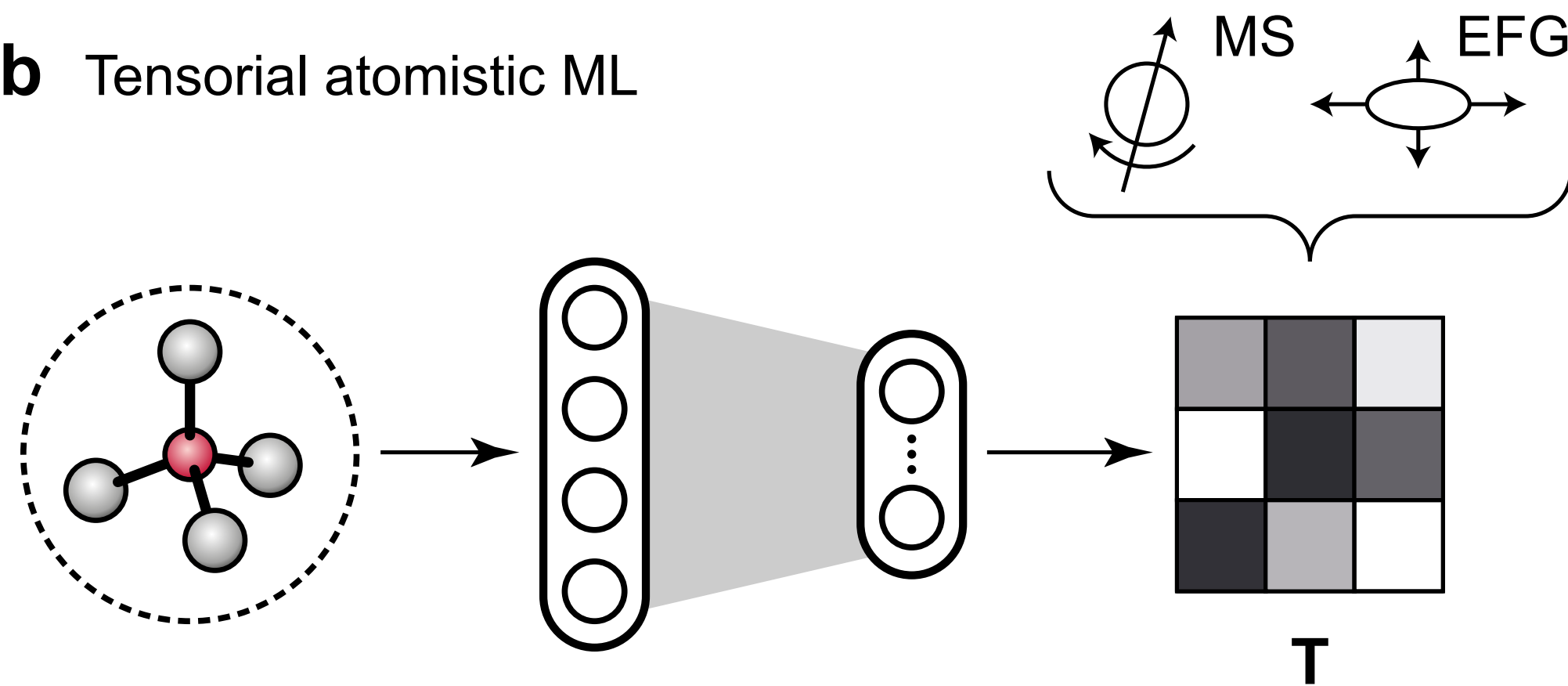


ML architecture

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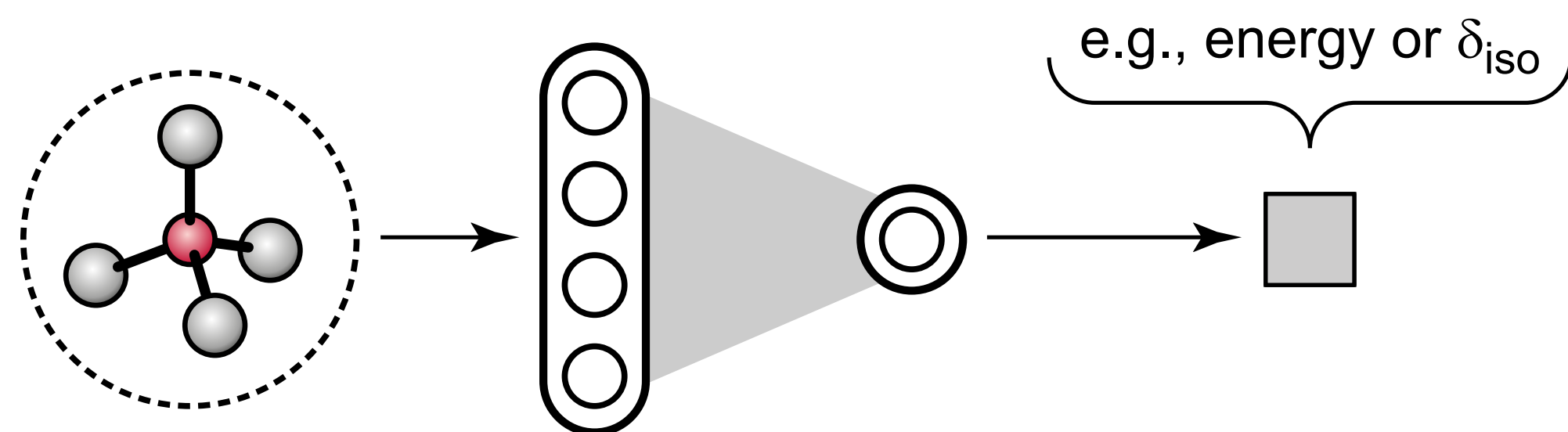


b Tensorial atomistic ML

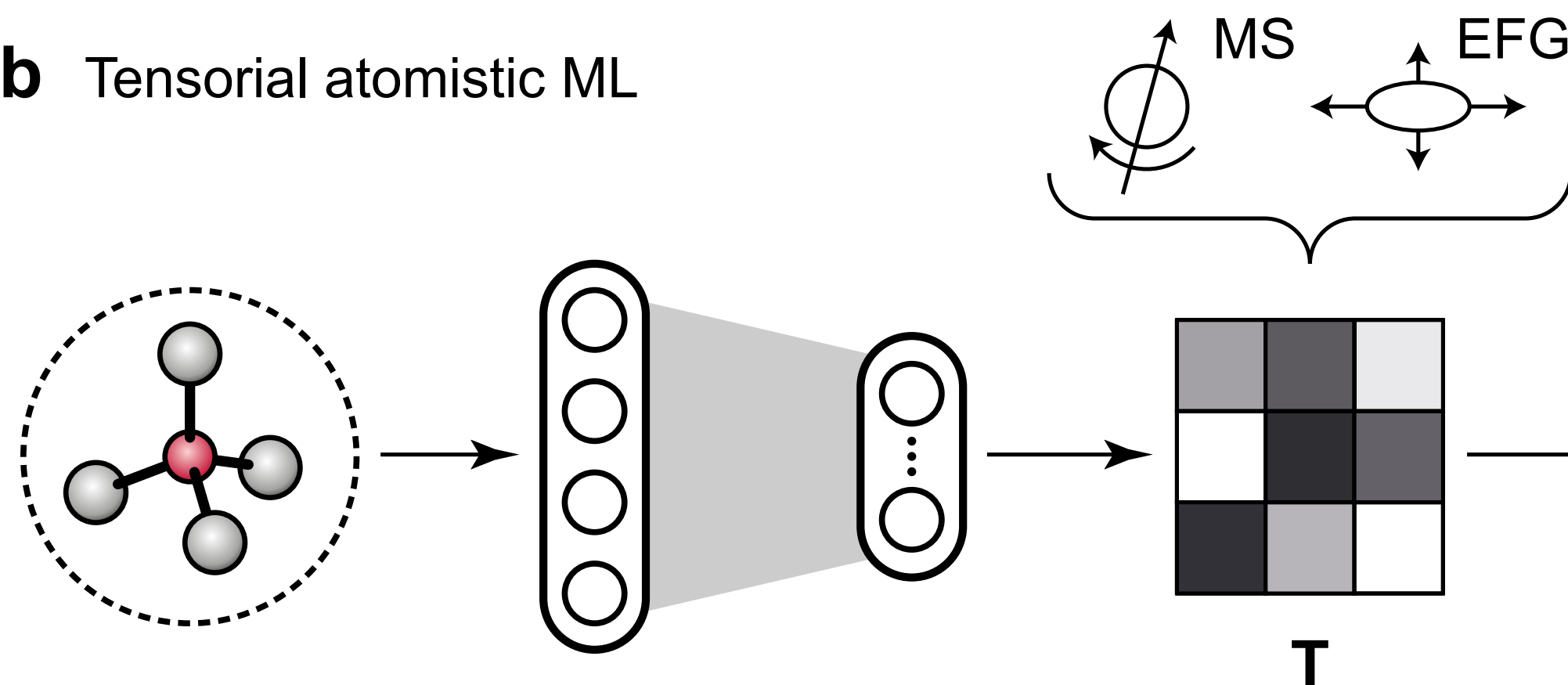


ML architecture

a Scalar atomistic ML



b Tensorial atomistic ML

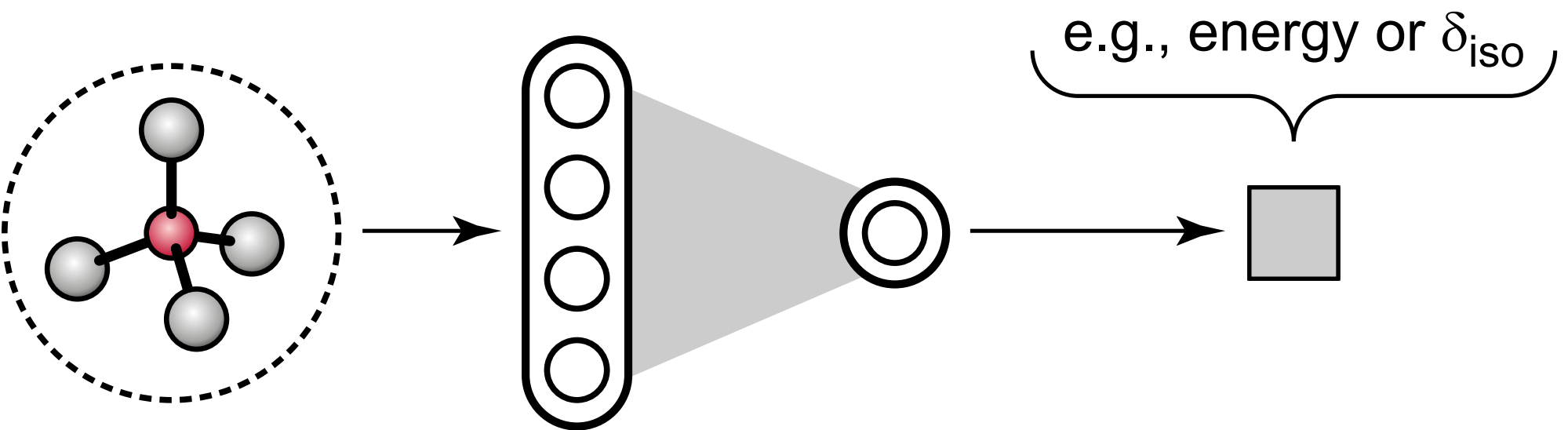


c **Option 1:** Irreducible spherical decomposition (ISD)

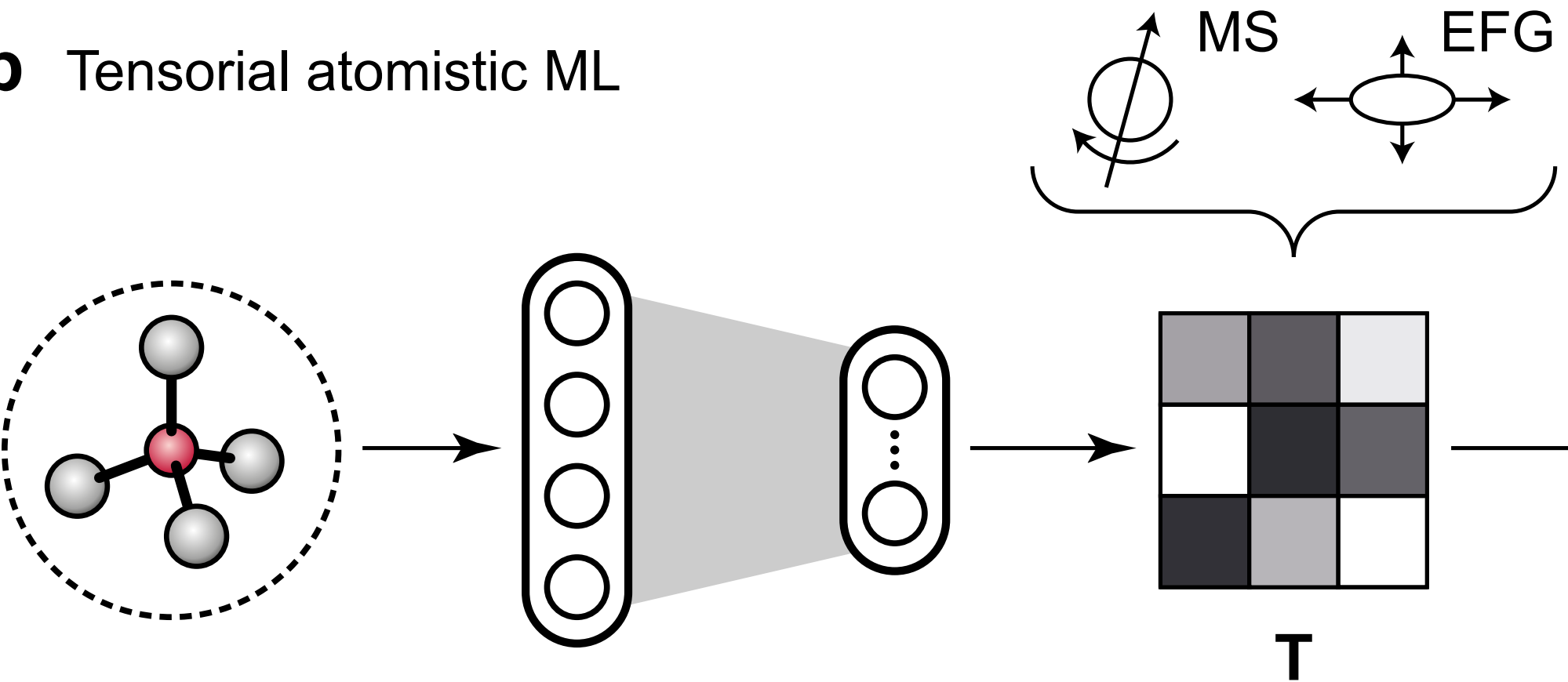
$$\begin{array}{ccc}
 T^{(0)} & \oplus & T^{(1)} \\
 \text{[Orange Box]} & \oplus & \text{[Three Orange Boxes]} \\
 \ell = 0; p = +1 & & \ell = 1; p = +1
 \end{array}
 \quad
 \begin{array}{ccc}
 & \oplus & T^{(2)} \\
 & \oplus & \text{[Five Orange Boxes]} \\
 & & \ell = 2; p = +1
 \end{array}$$

ML architecture

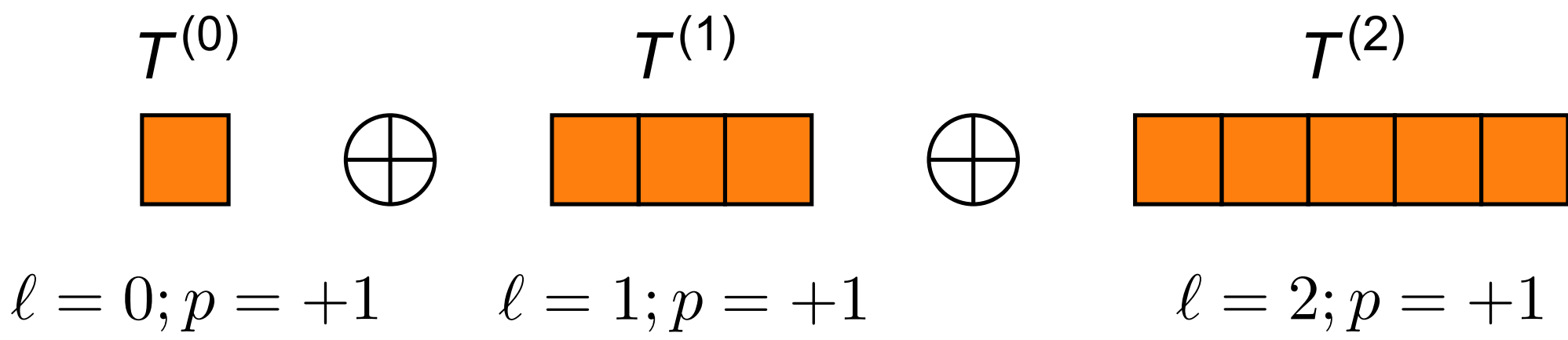
a Scalar atomistic ML



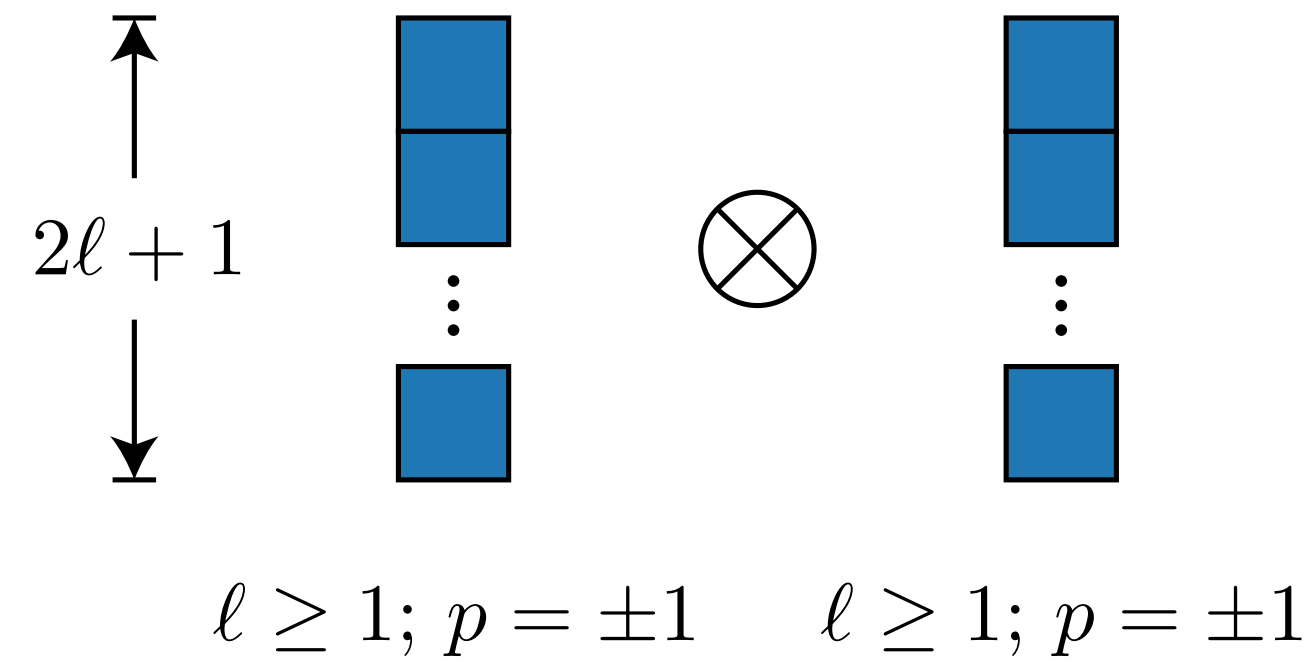
b Tensorial atomistic ML



c **Option 1:** Irreducible spherical decomposition (ISD)

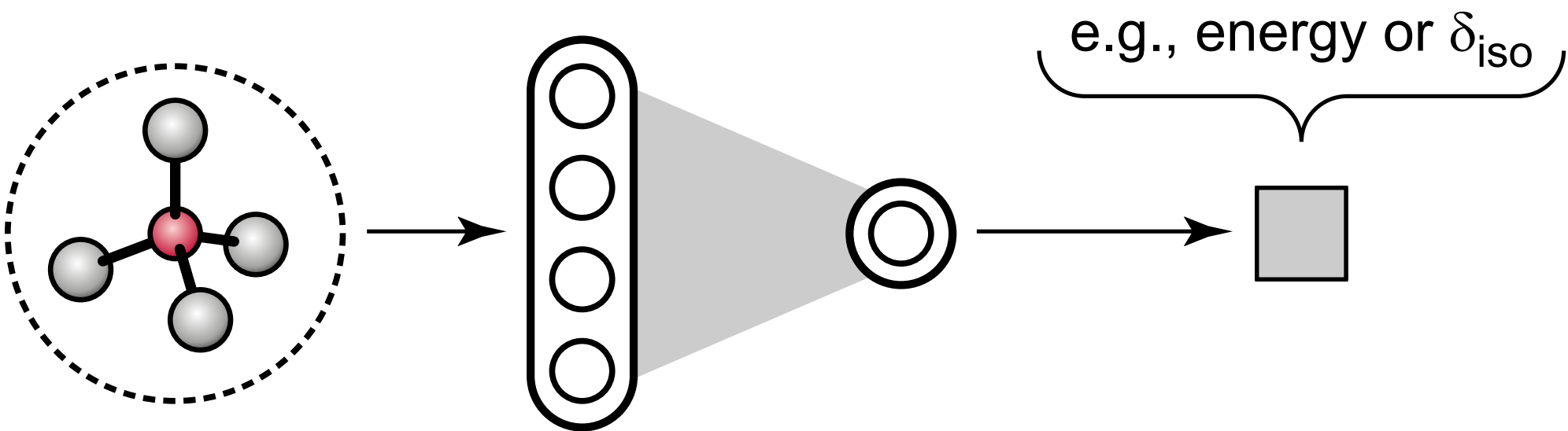


d **Option 2:** Tensor product (TP)

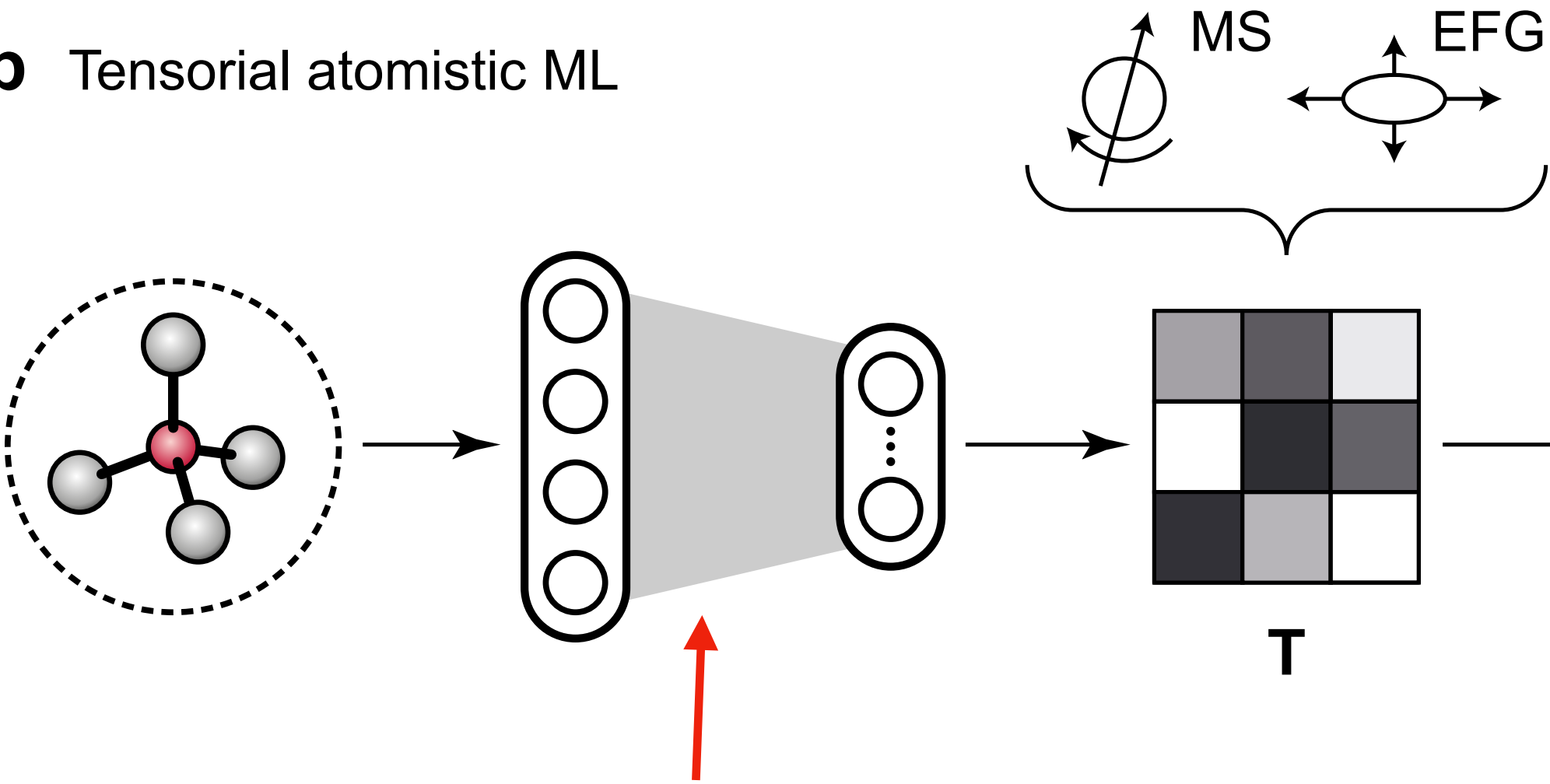


ML architecture

a Scalar atomistic ML

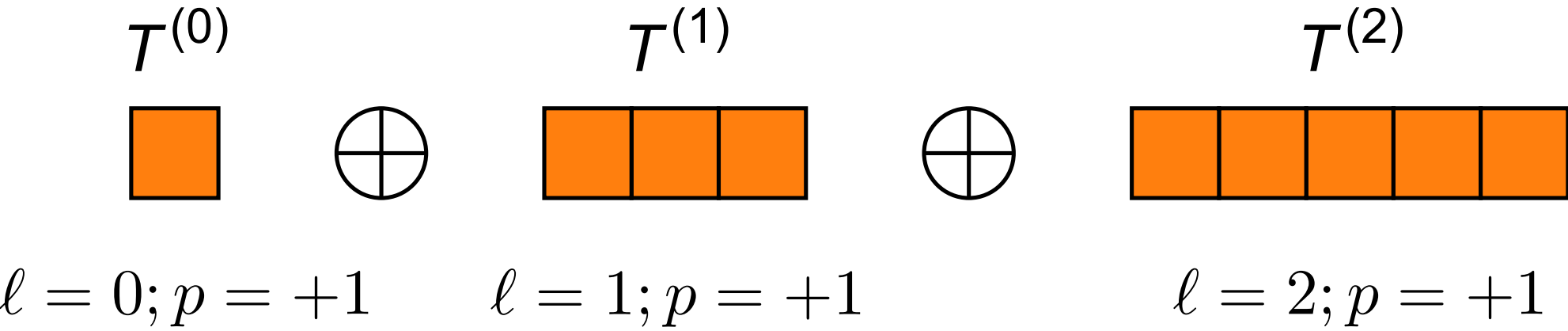


b Tensorial atomistic ML

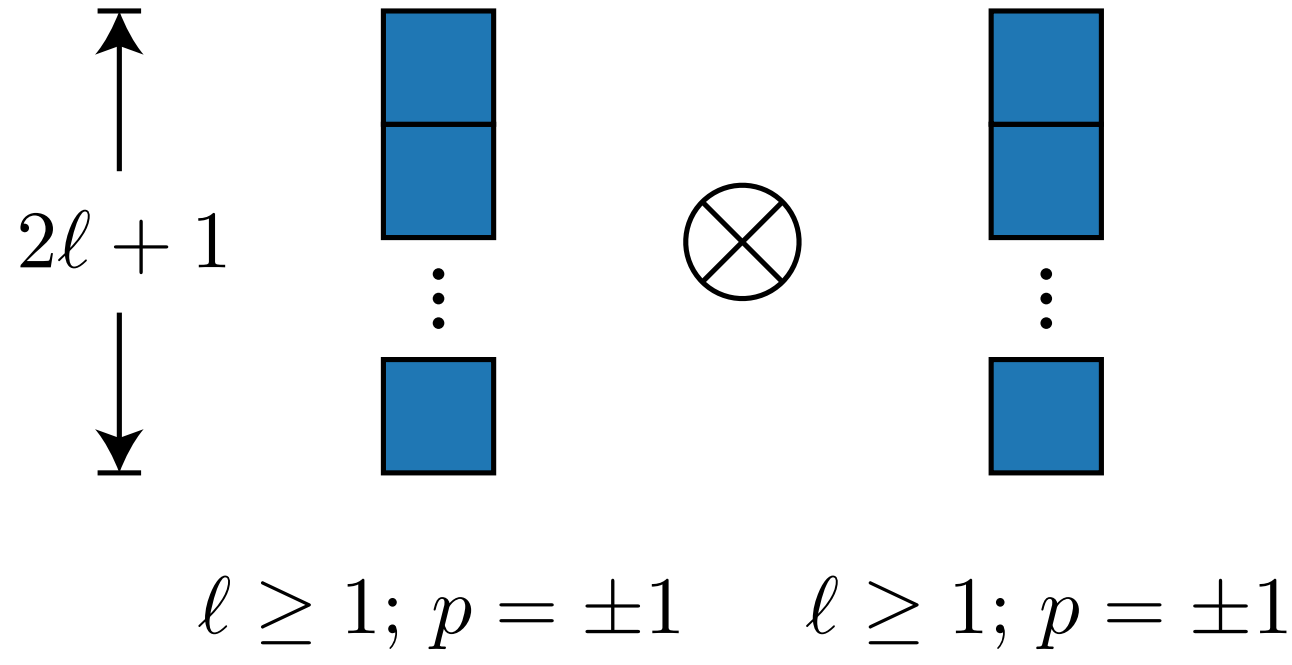


NequIP[1] architecture

c Option 1: Irreducible spherical decomposition (ISD)



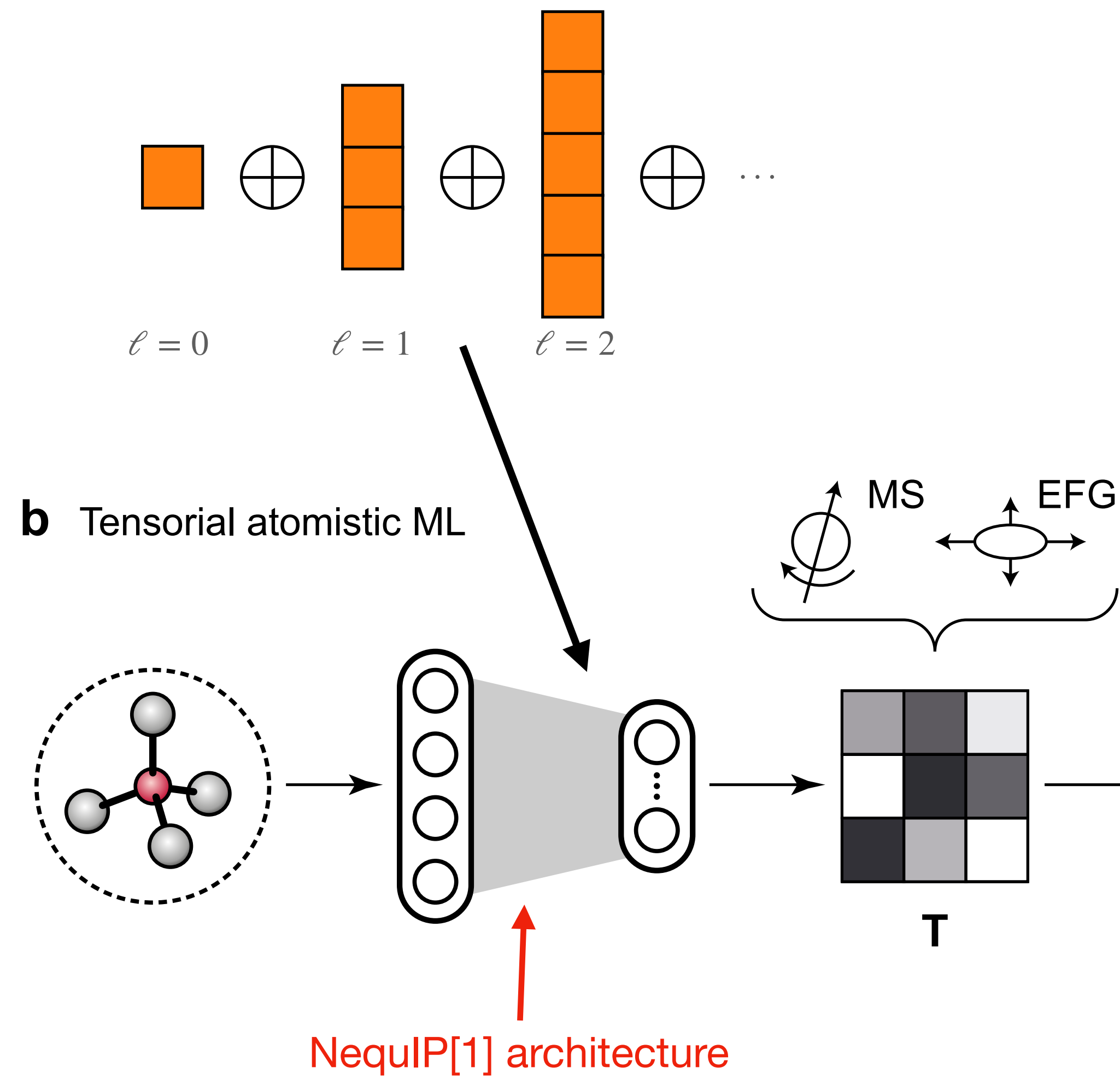
d Option 2: Tensor product (TP)



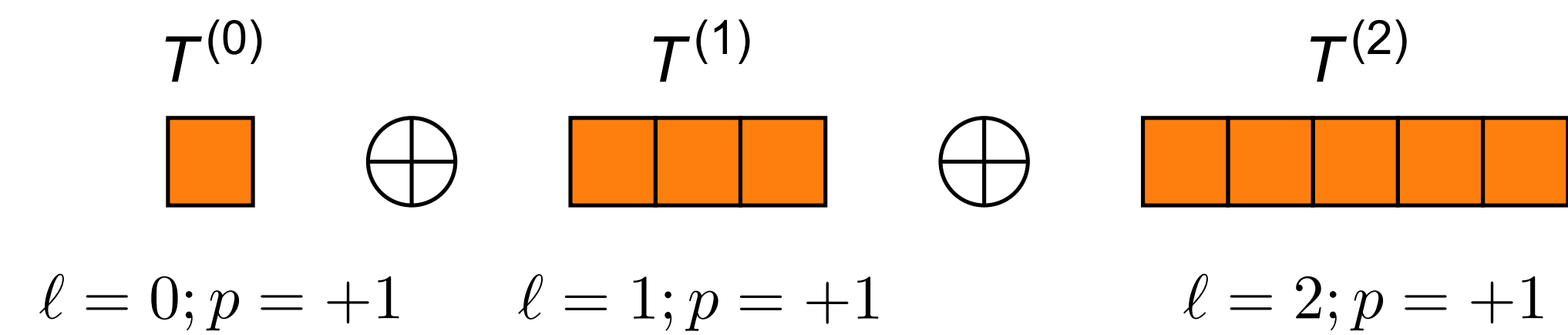
CBM, Rosset, L.A.M., Yates, J.R., Deringer, V.L., 2024.
Graph-Neural-Network predictions of solid-state NMR parameters from spherical tensor decomposition.
arXiv: 2412.15063
[1] Batzner et al., 2022

ML architecture

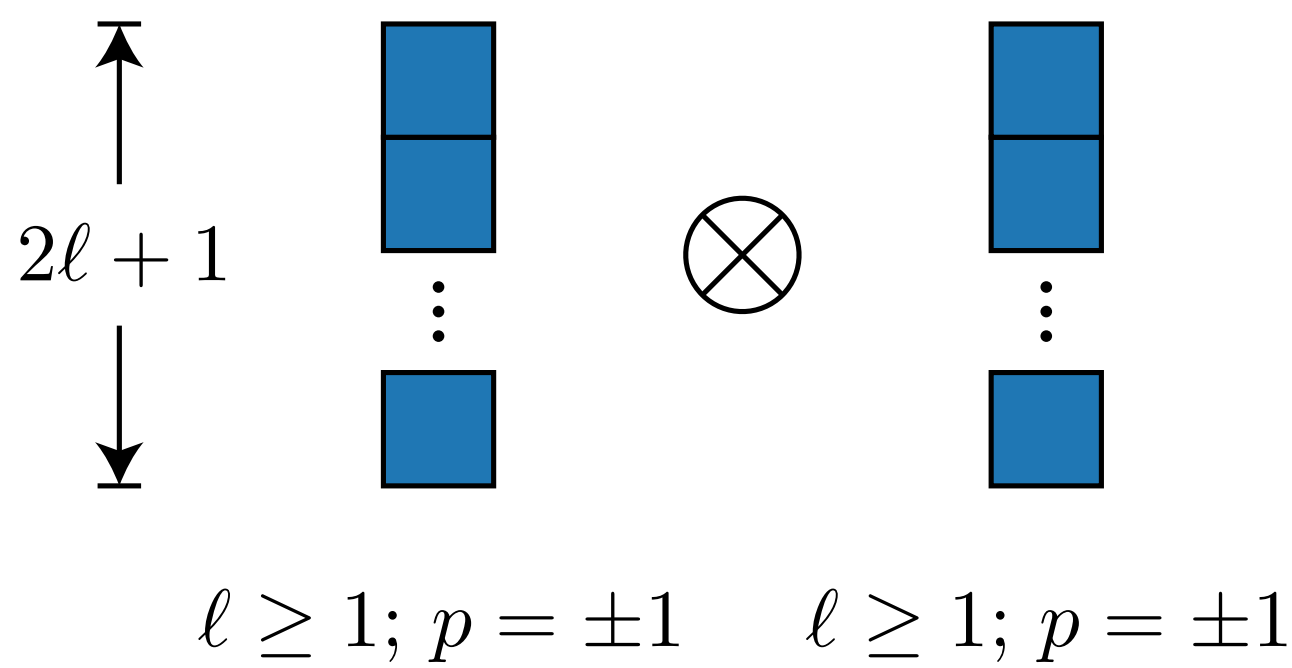
Equivariant features (do change upon structure rotation for example)



c Option 1: Irreducible spherical decomposition (ISD)

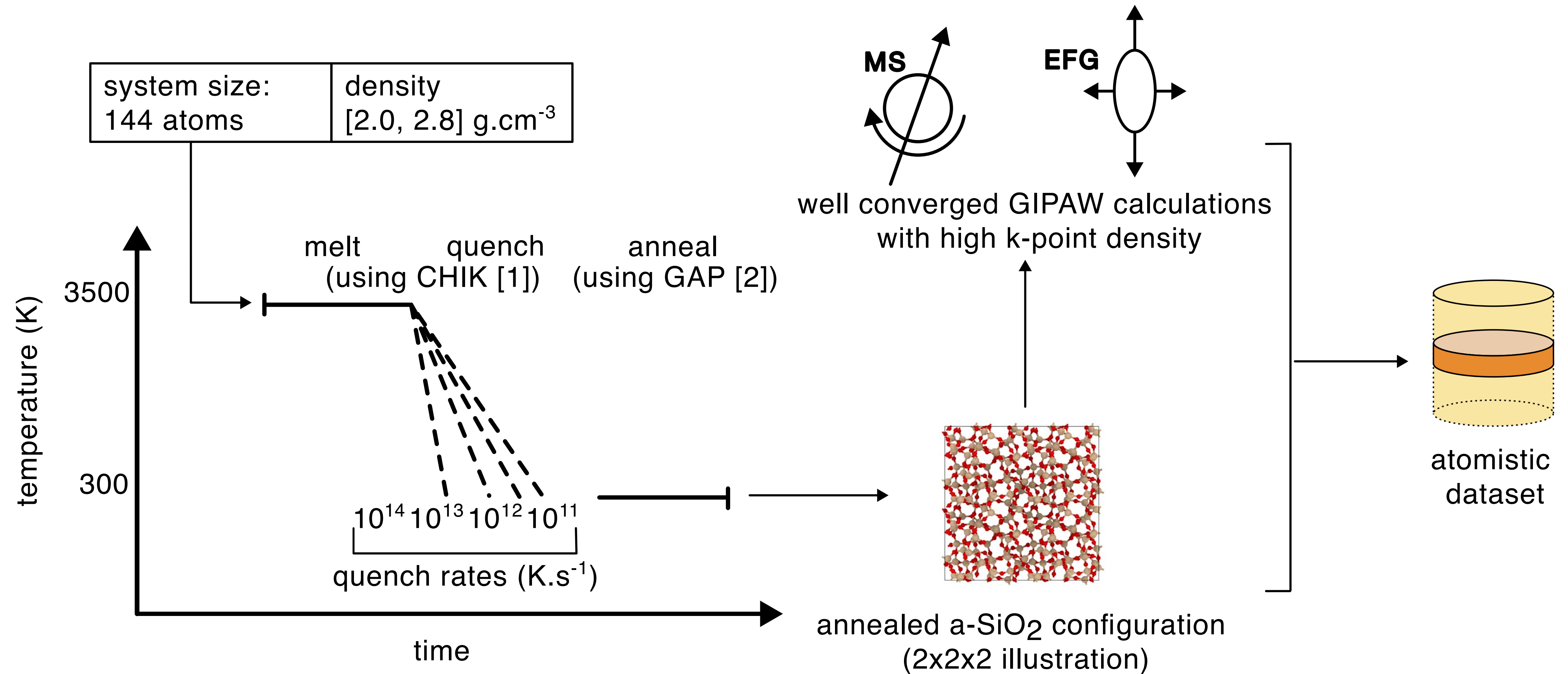


d Option 2: Tensor product (TP)

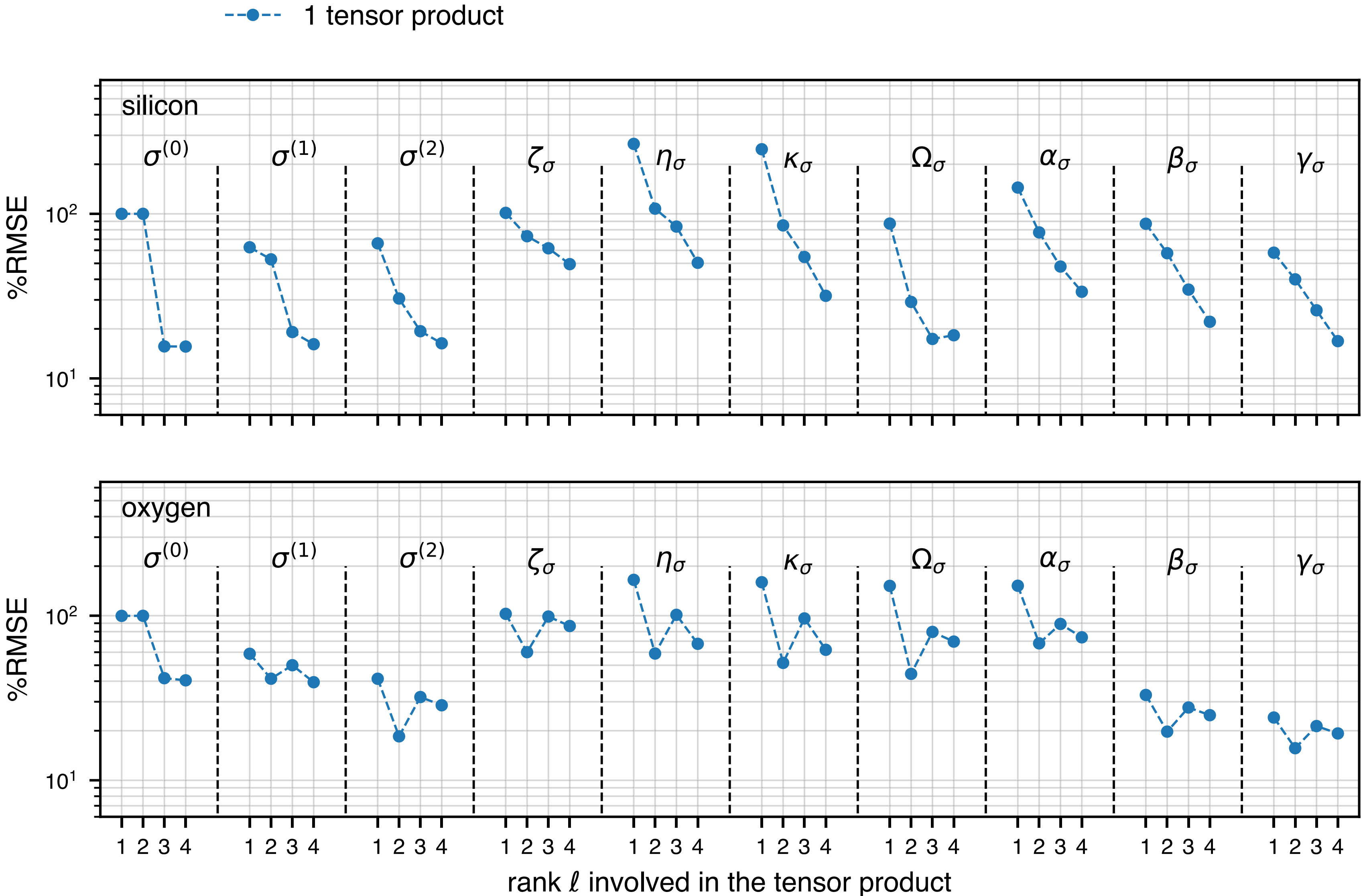


CBM, Rosset, L.A.M., Yates, J.R., Deringer, V.L., 2024.
Graph-Neural-Network predictions of solid-state NMR parameters from spherical tensor decomposition.
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a-SiO₂ dataset generation protocol

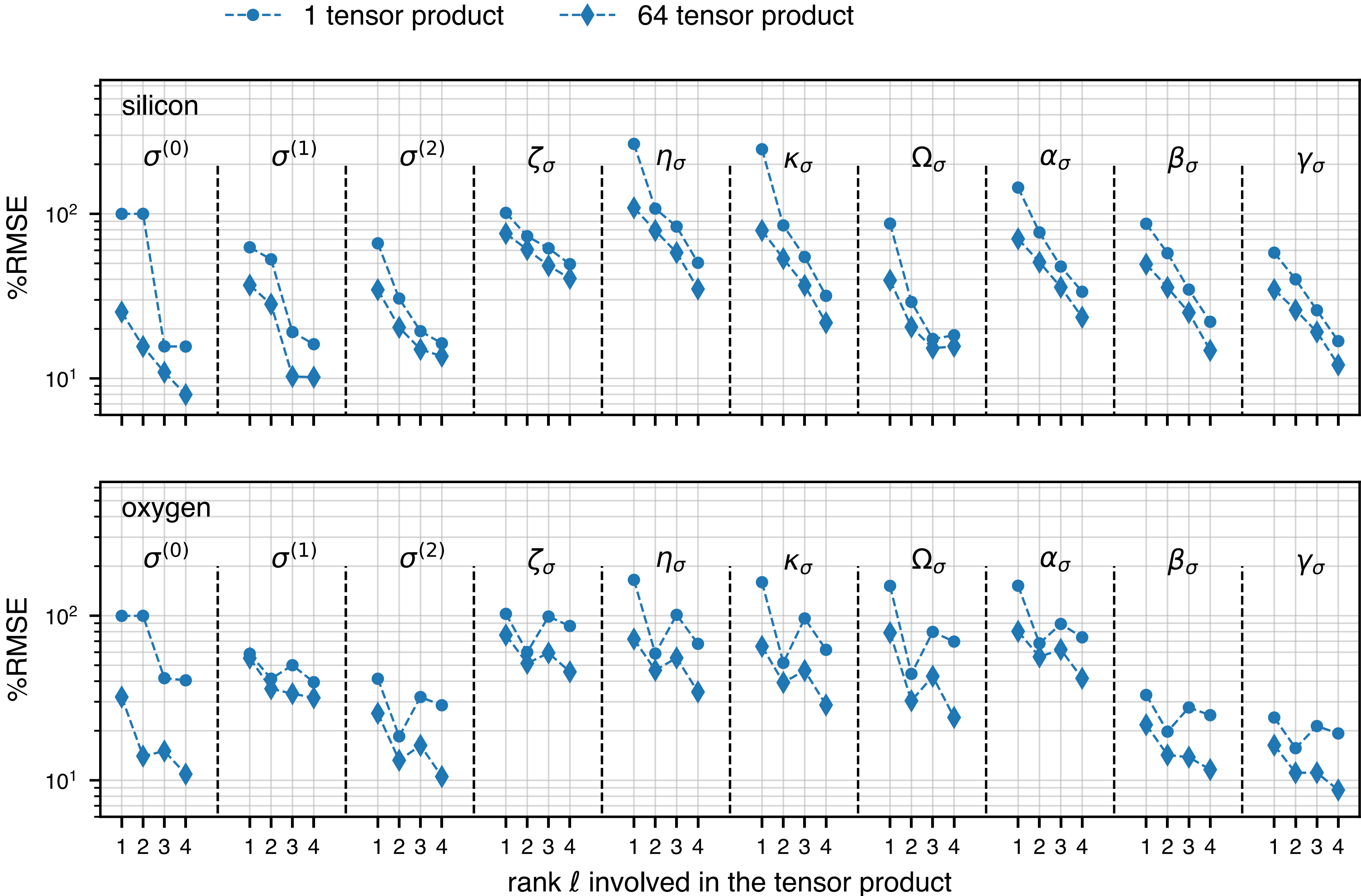


NMR parameters as SH tensor products



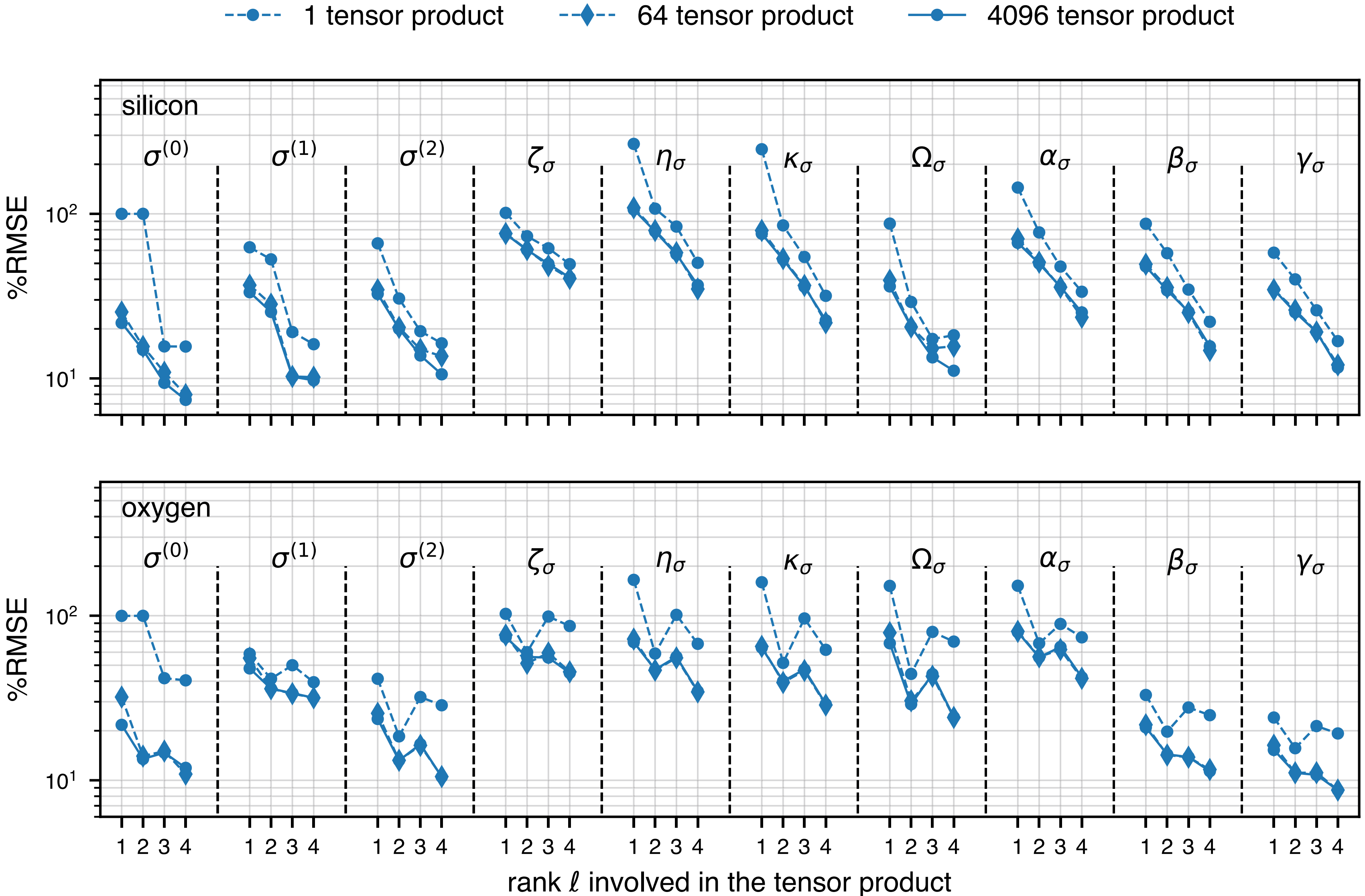
CBM, Rosset, L.A.M.,
Yates, J.R., Deringer, V.L.,
2024.
Graph-Neural-Network
predictions of solid-state
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NMR parameters as SH tensor products



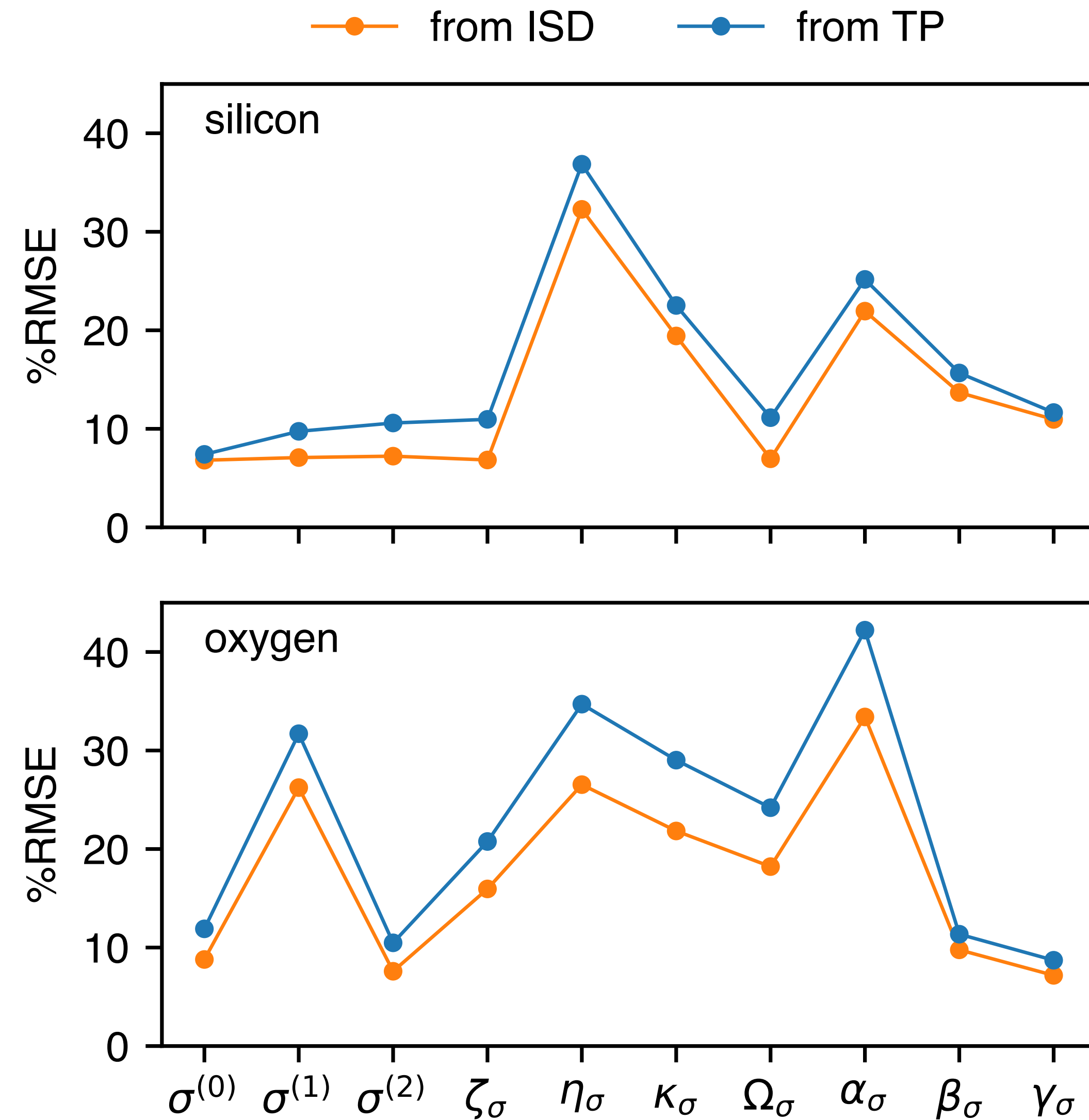
CBM, Rosset, L.A.M.,
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NMR parameters from
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NMR parameters as SH tensor products

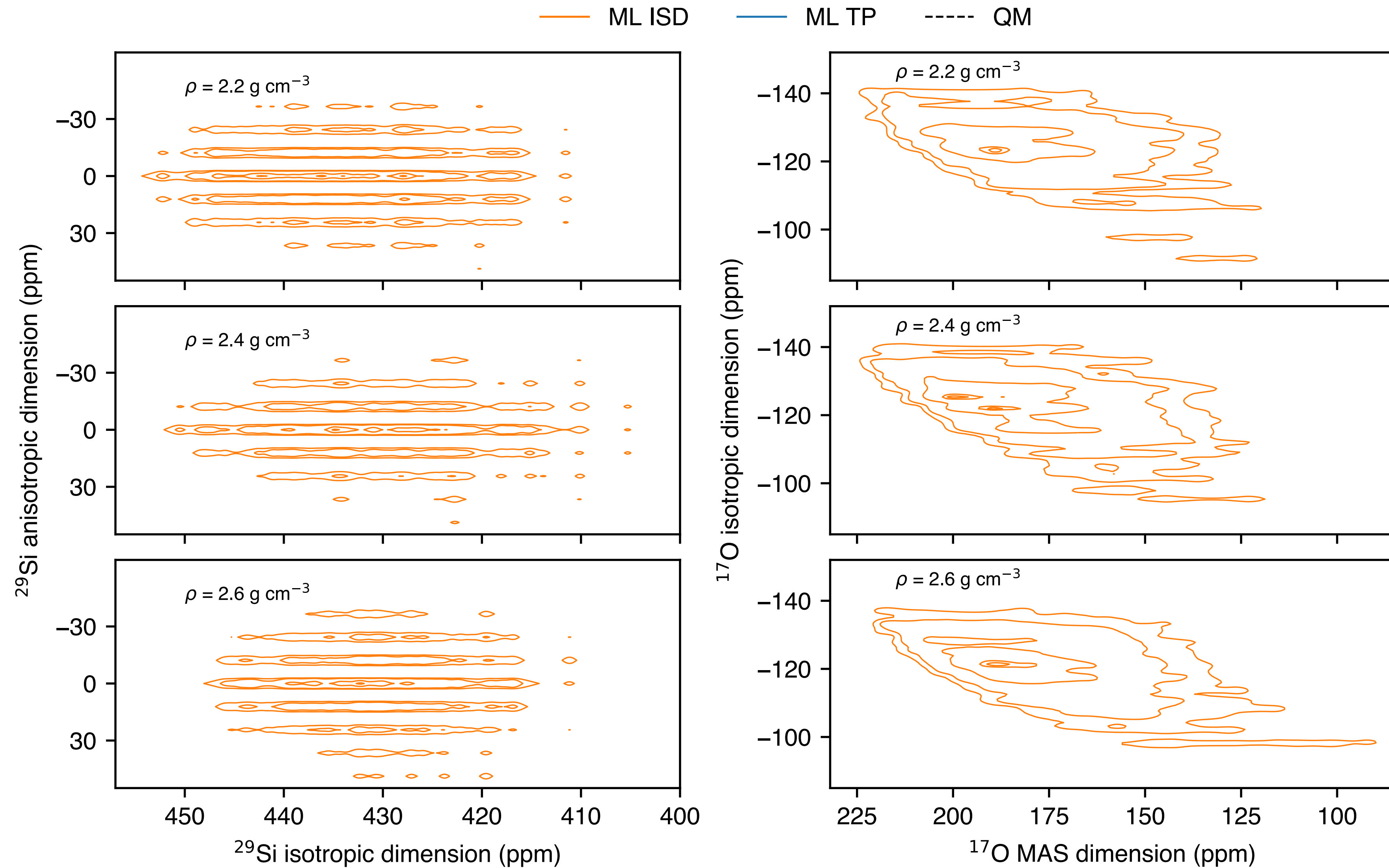


CBM, Rosset, L.A.M.,
Yates, J.R., Deringer, V.L.,
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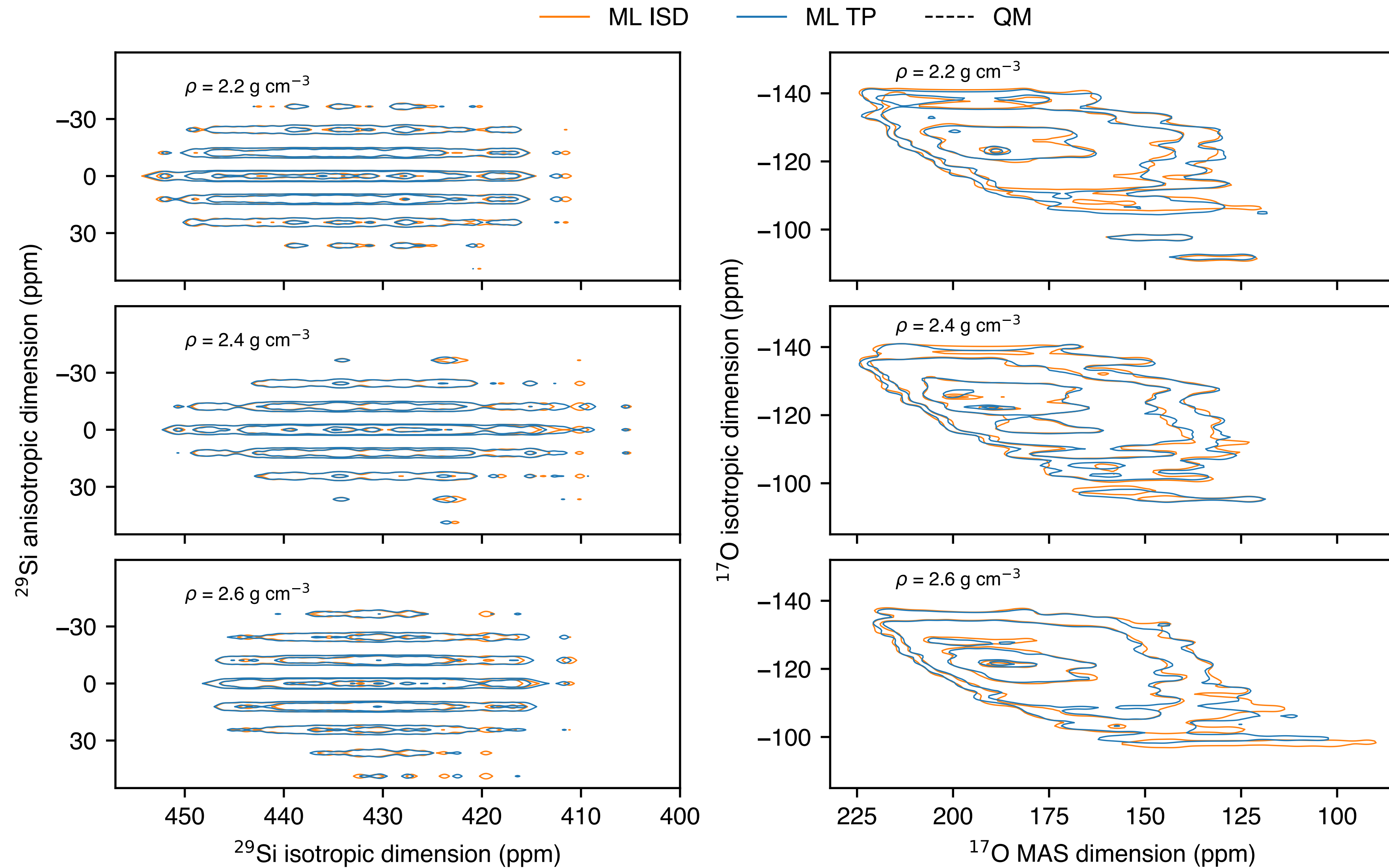
Two (nearly) equivalent representations



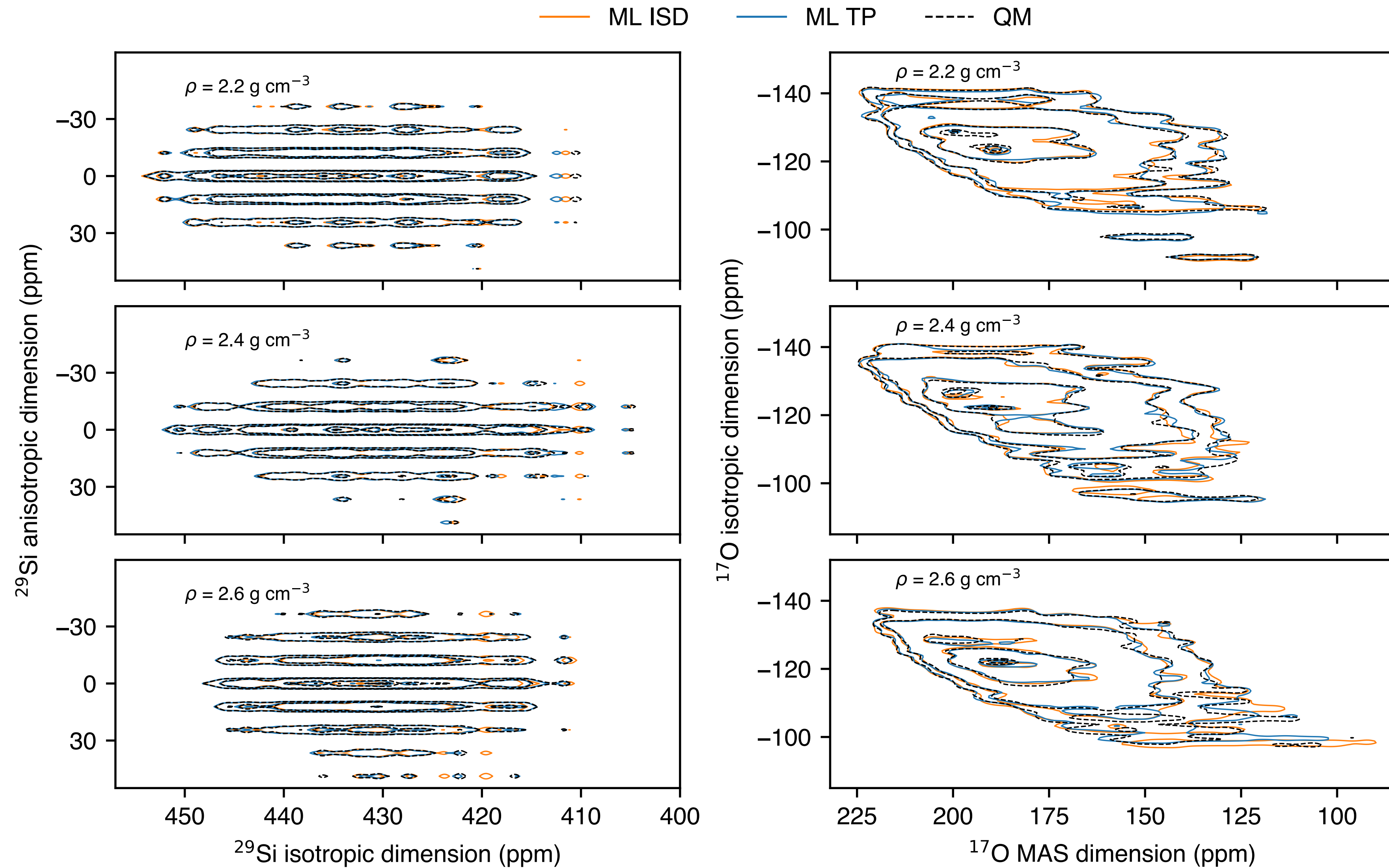
2D NMR spectra of a-SiO₂ models



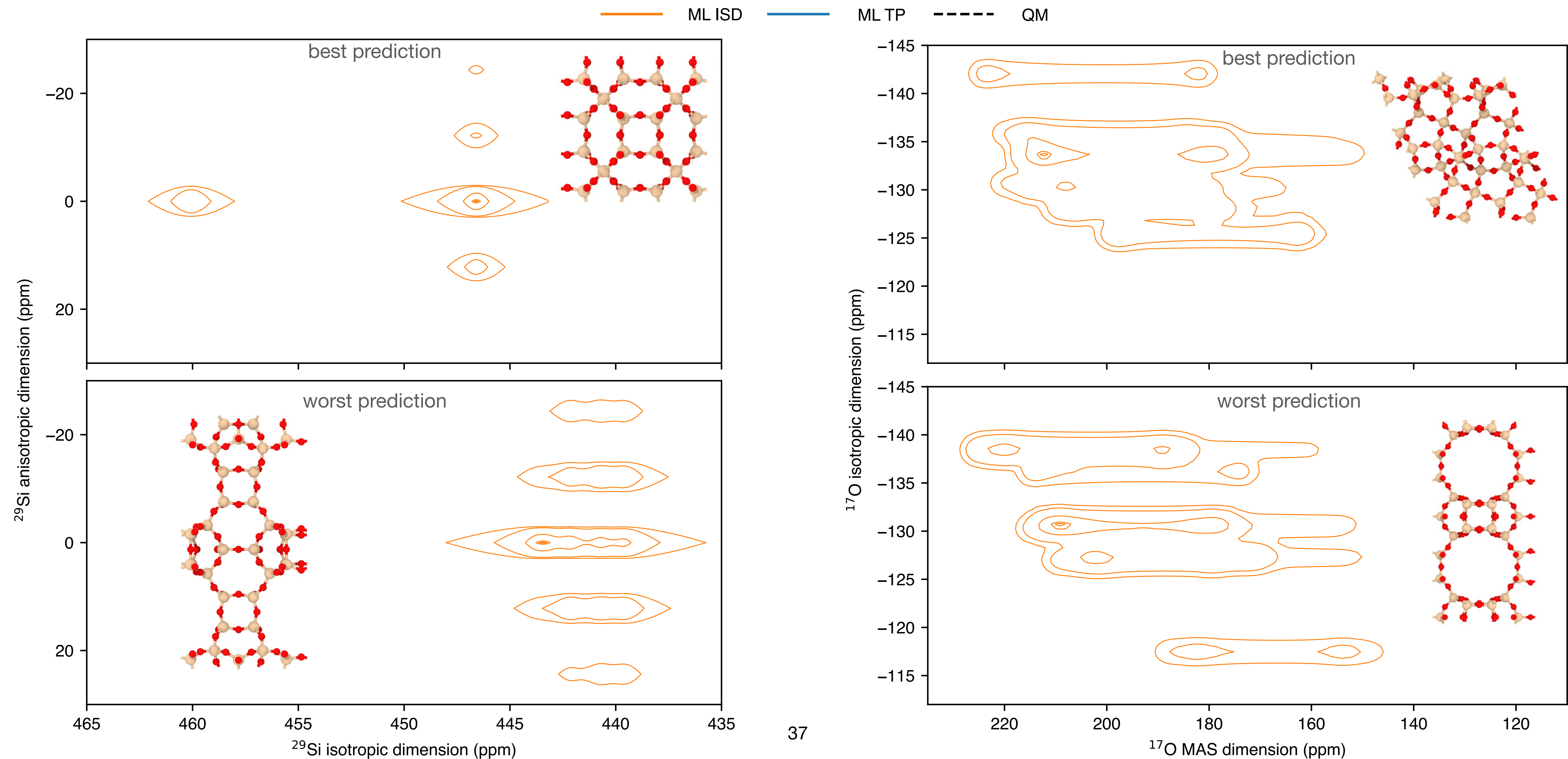
2D NMR spectra of a-SiO₂ models



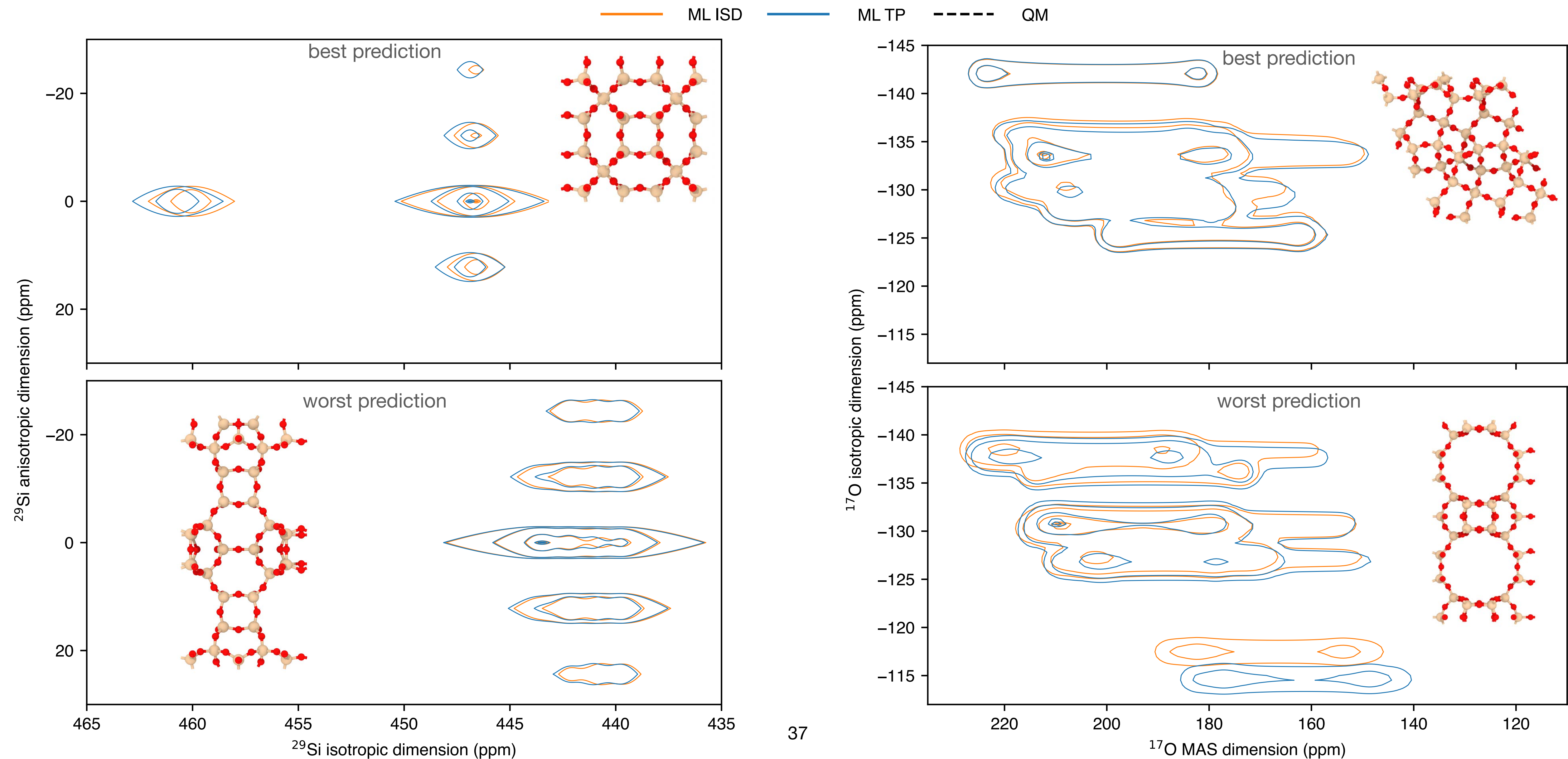
2D NMR spectra of a-SiO₂ models



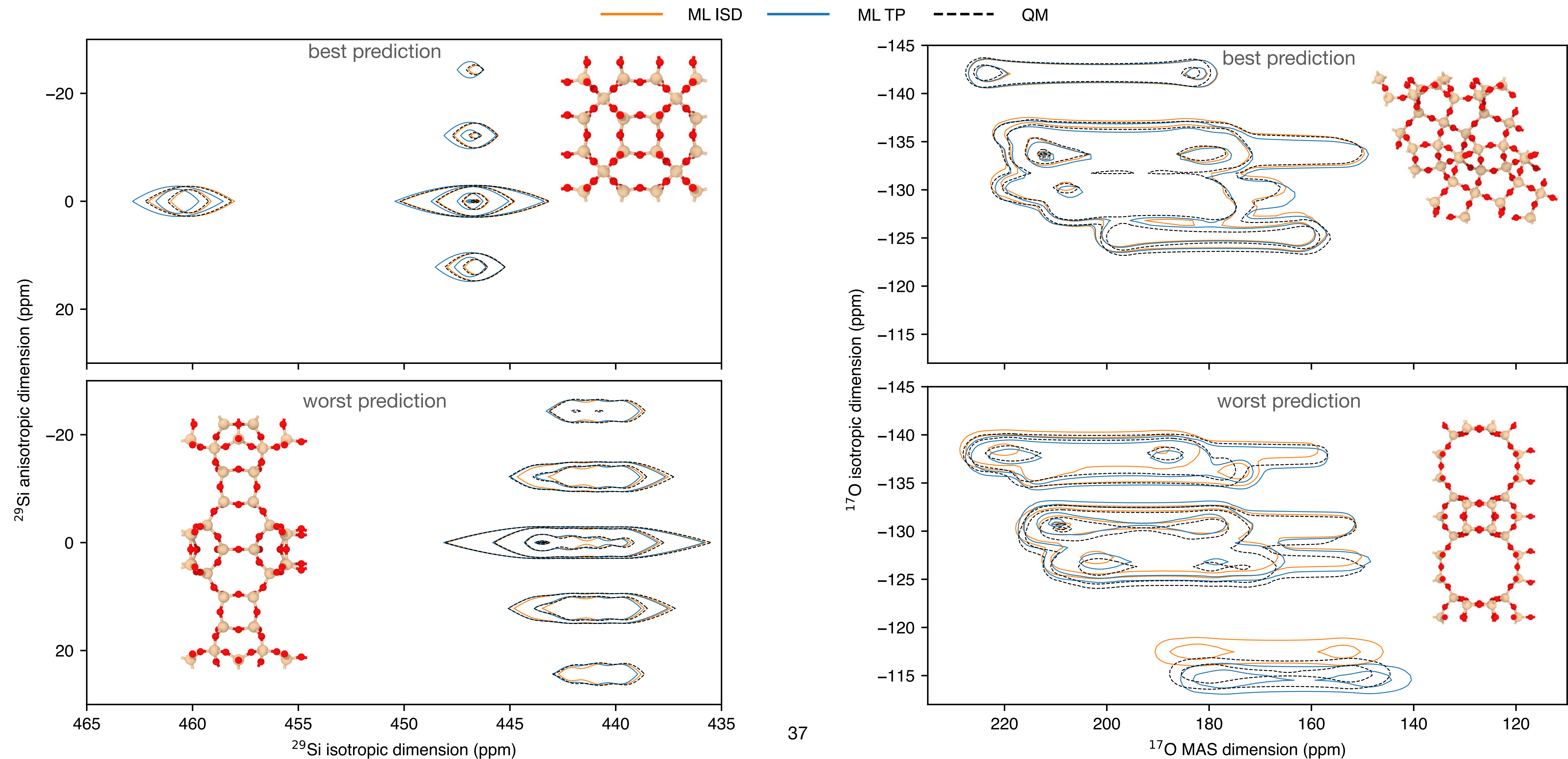
2D NMR spectra for hypothetical zeolites



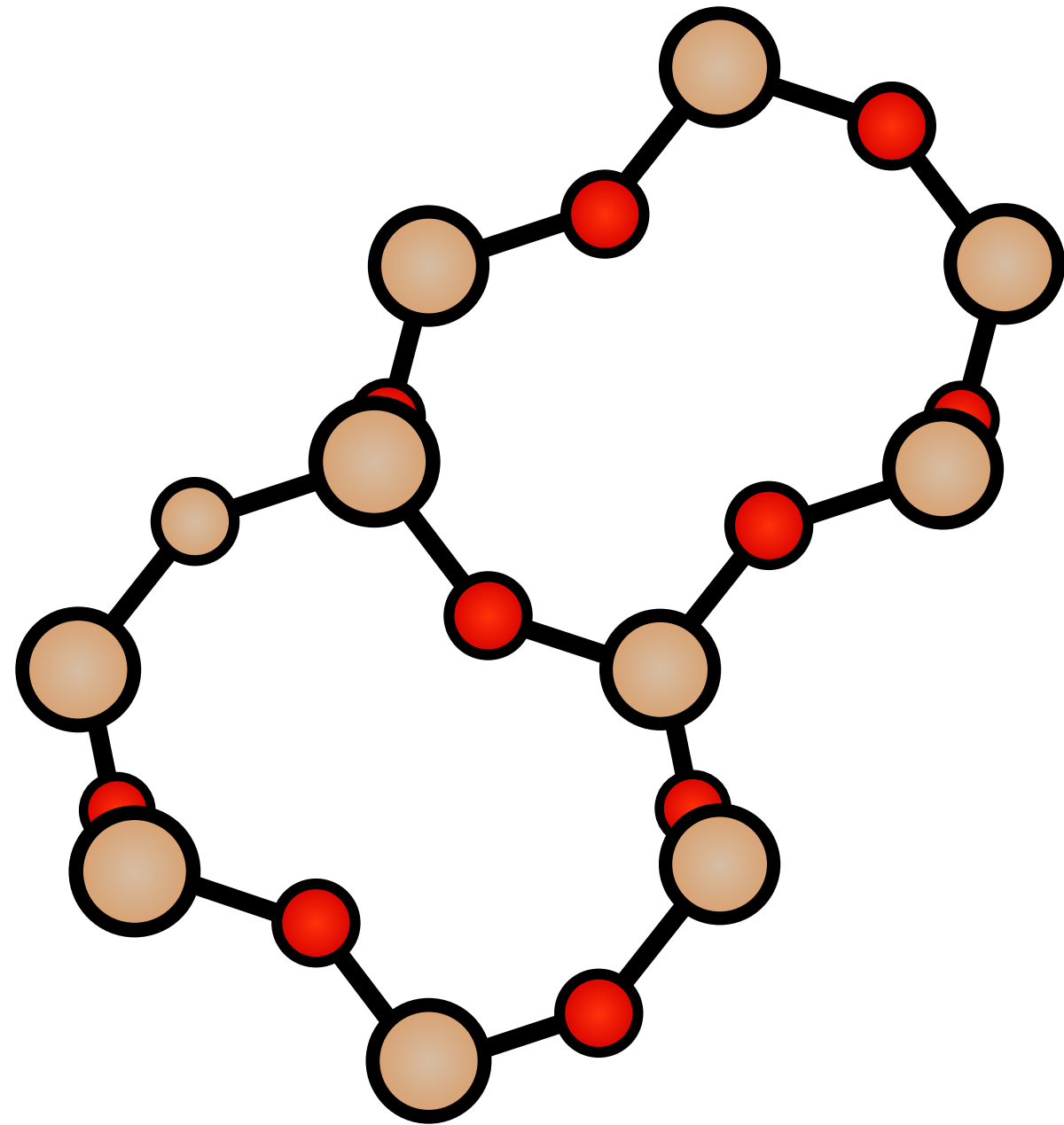
2D NMR spectra for hypothetical zeolites



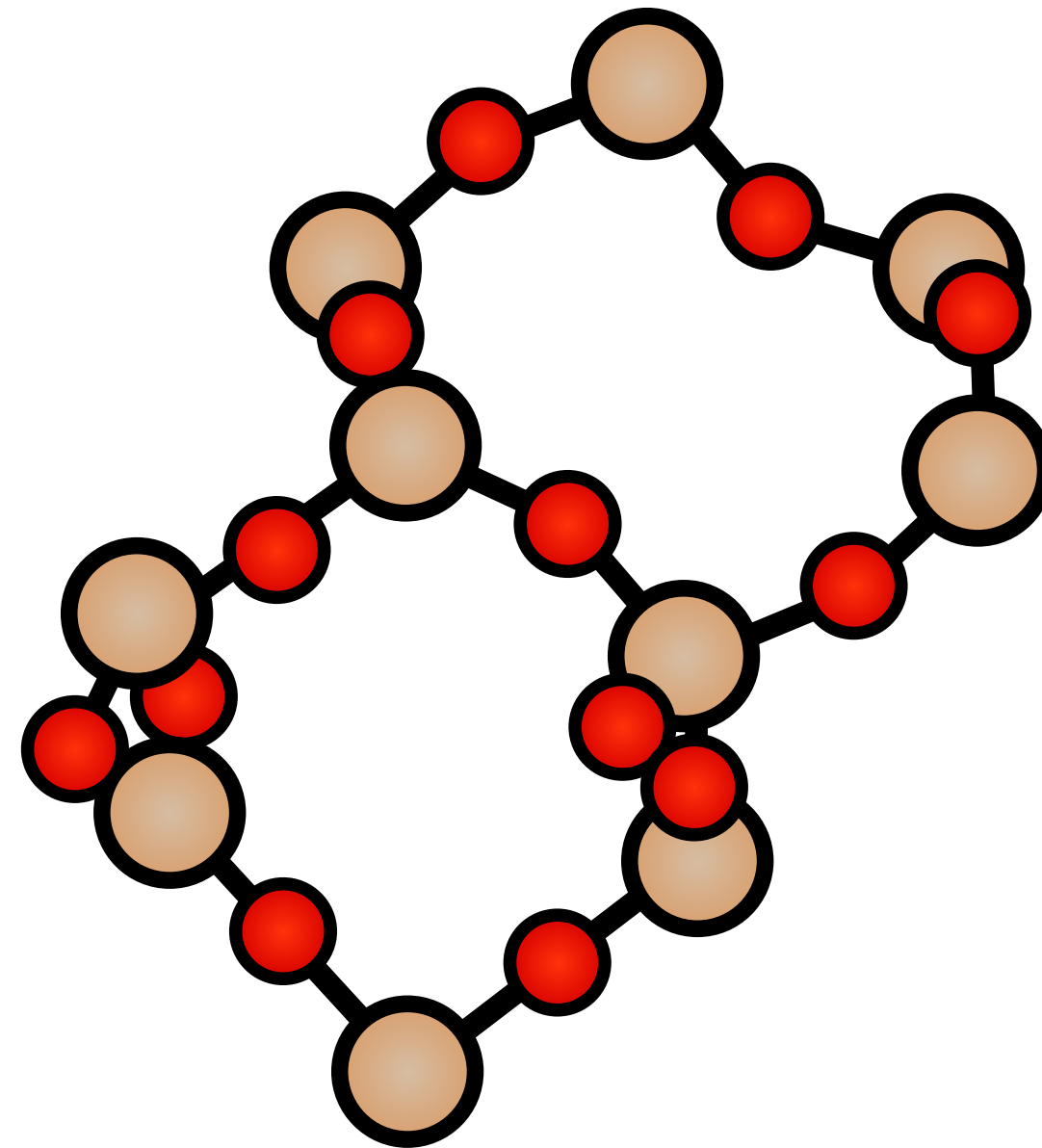
2D NMR spectra for hypothetical zeolites



Dynamics of α — β inversion in cristobalite

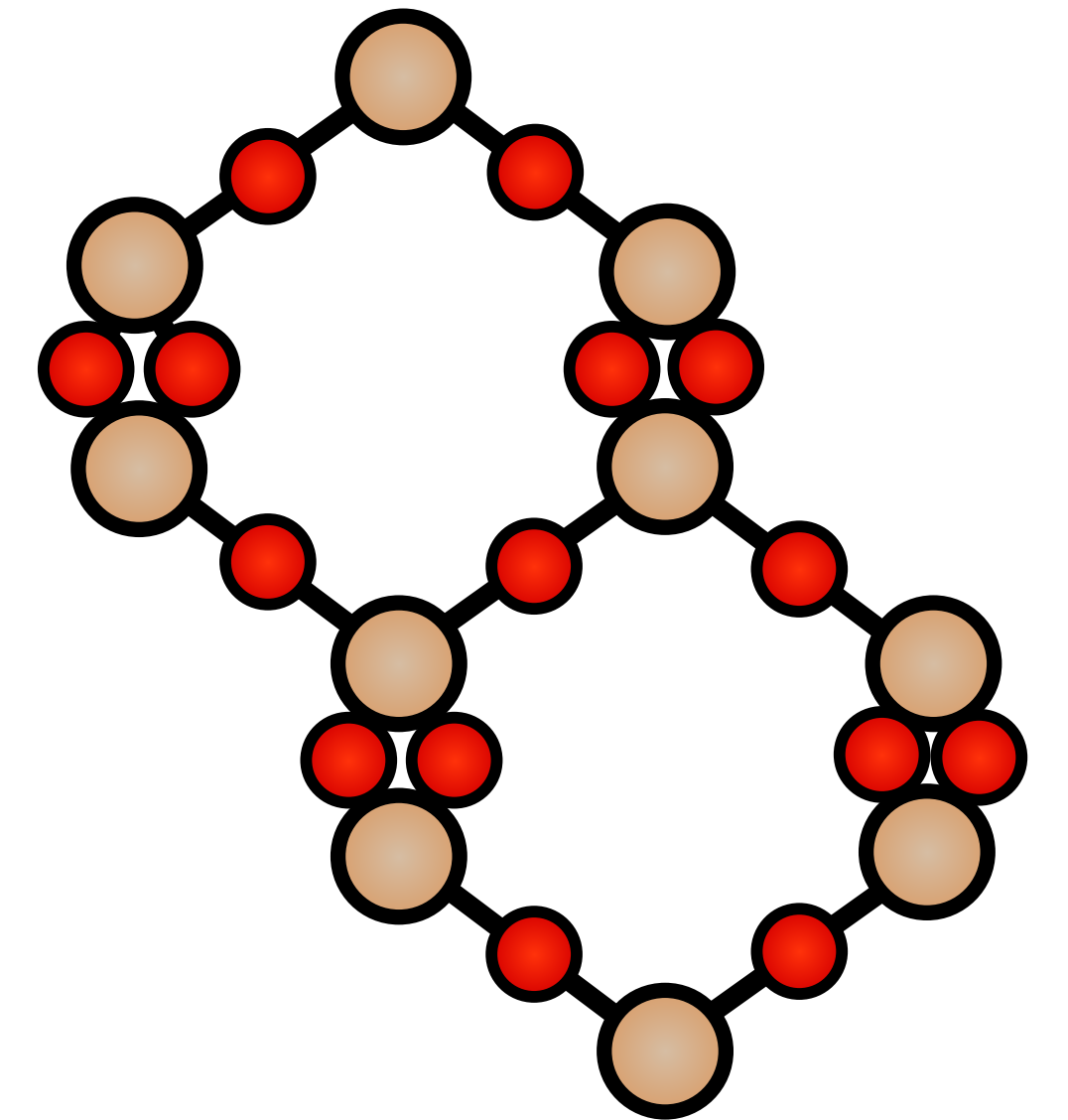


α environments



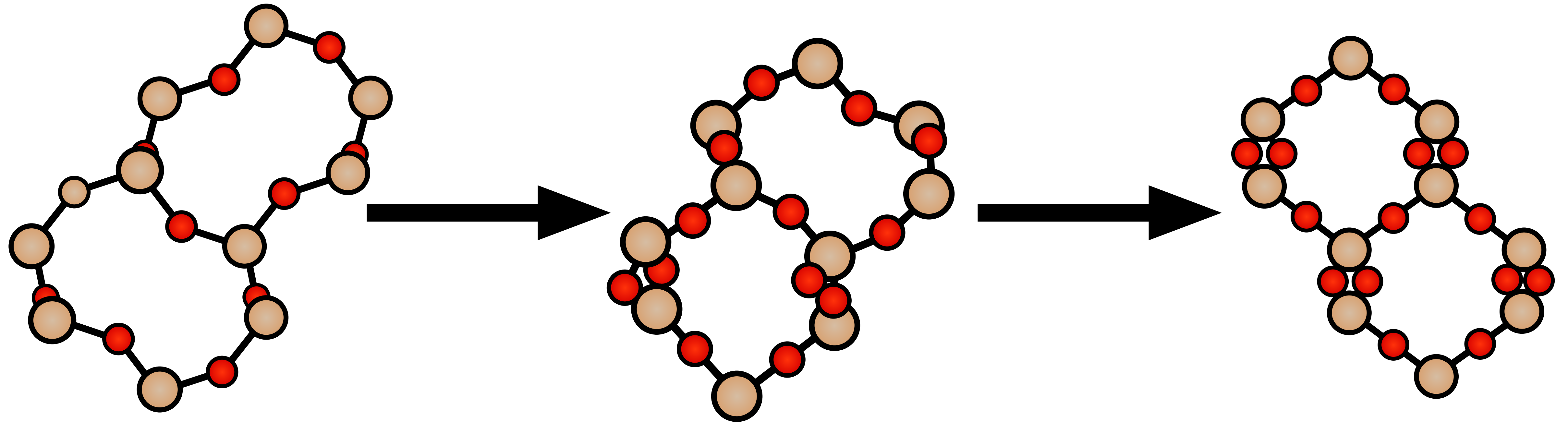
50% β environments

50% α environments



β environments

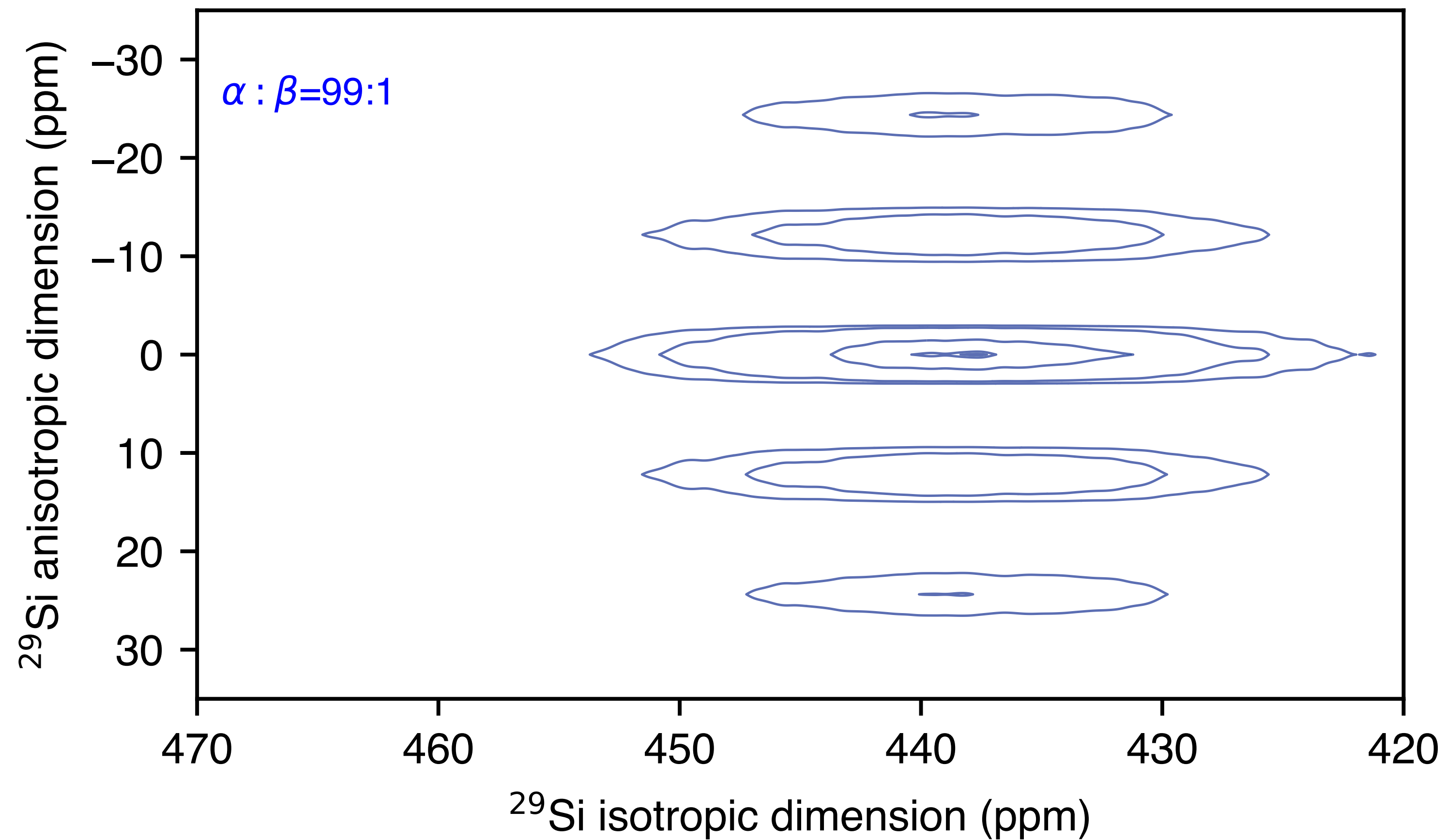
Dynamics of α — β inversion in cristobalite



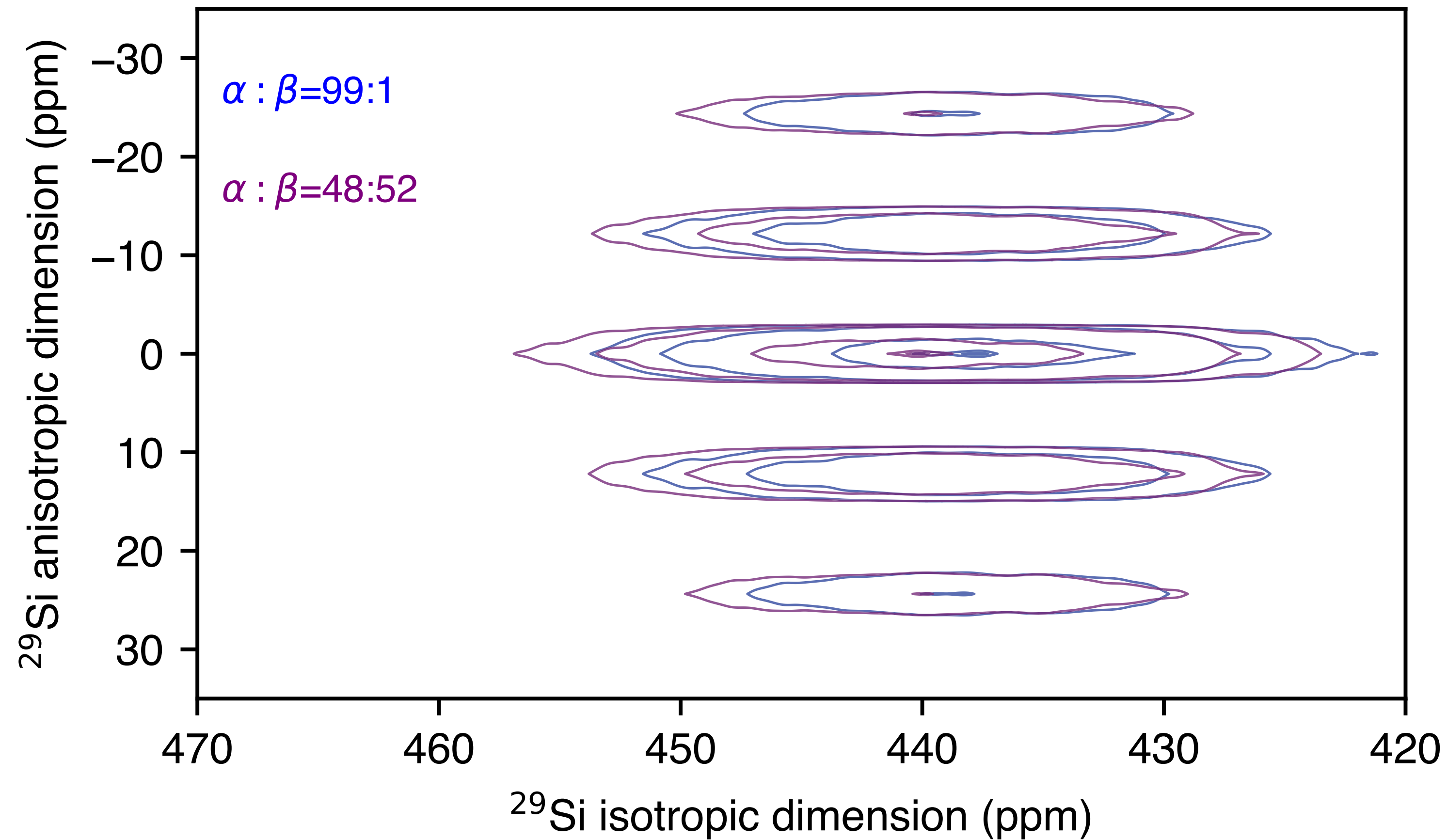
50% β environments
50% α environments

β environments

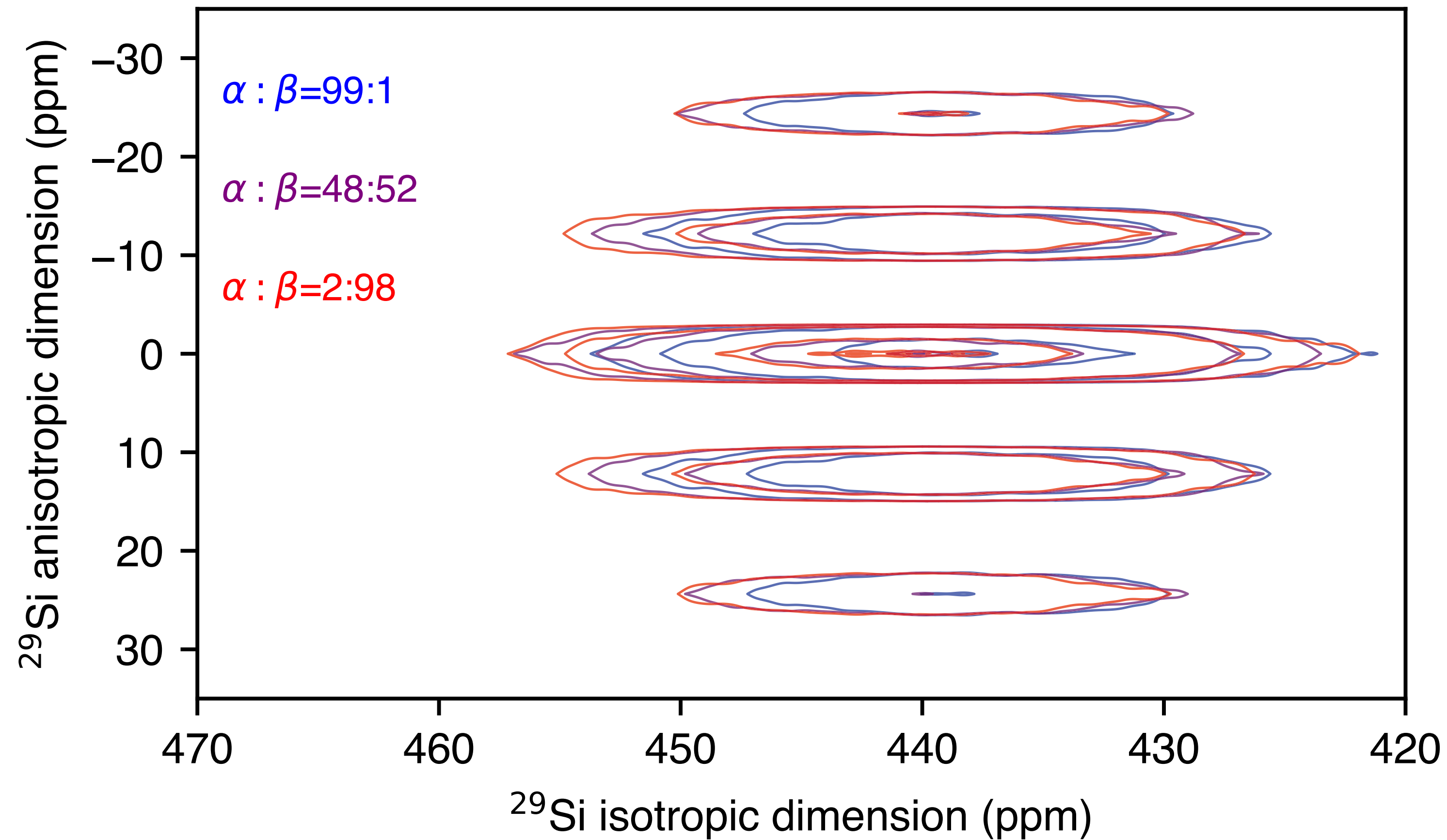
Dynamics of α — β inversion in cristobalite



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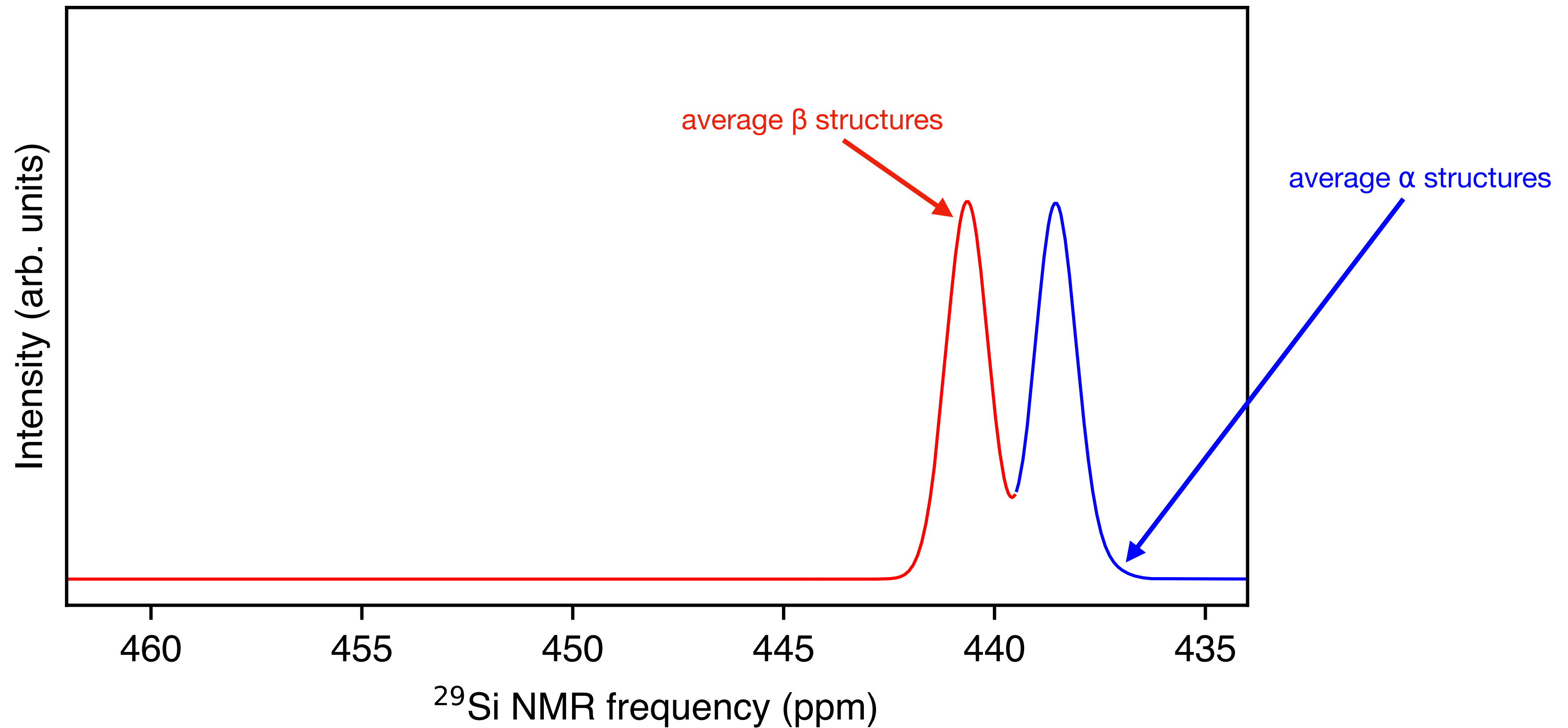


Dynamics of α — β inversion in cristobalite



Dynamics of α — β inversion in cristobalite

MAS spectra: no anisotropy



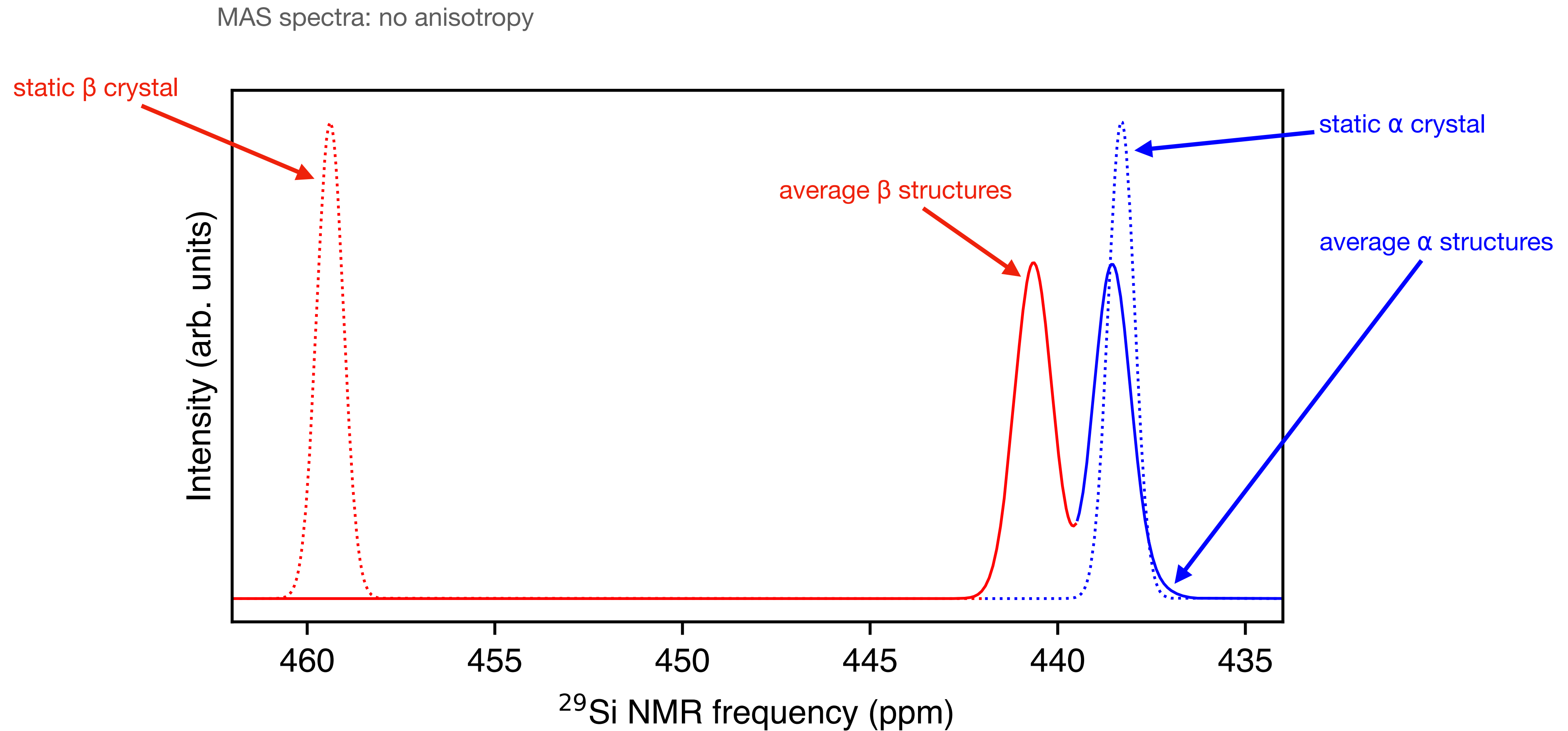
CBM, Rosset, L.A.M., Yates, J.R., Deringer, V.L., 2024.

Graph-Neural-Network predictions of solid-state NMR parameters from spherical tensor decomposition.

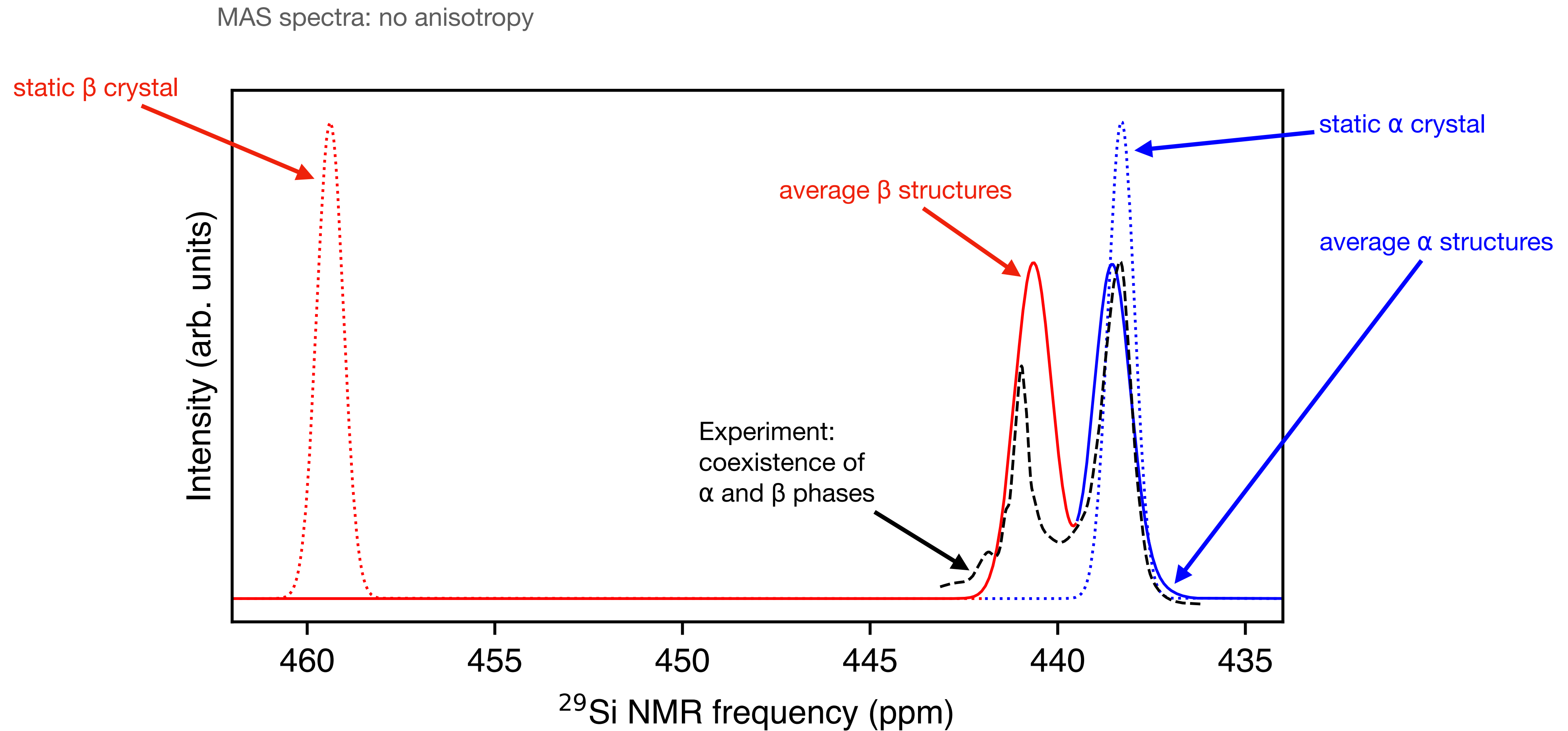
arXiv: 2412.15063

Spearing, Farnan, and Stebbins, 1992

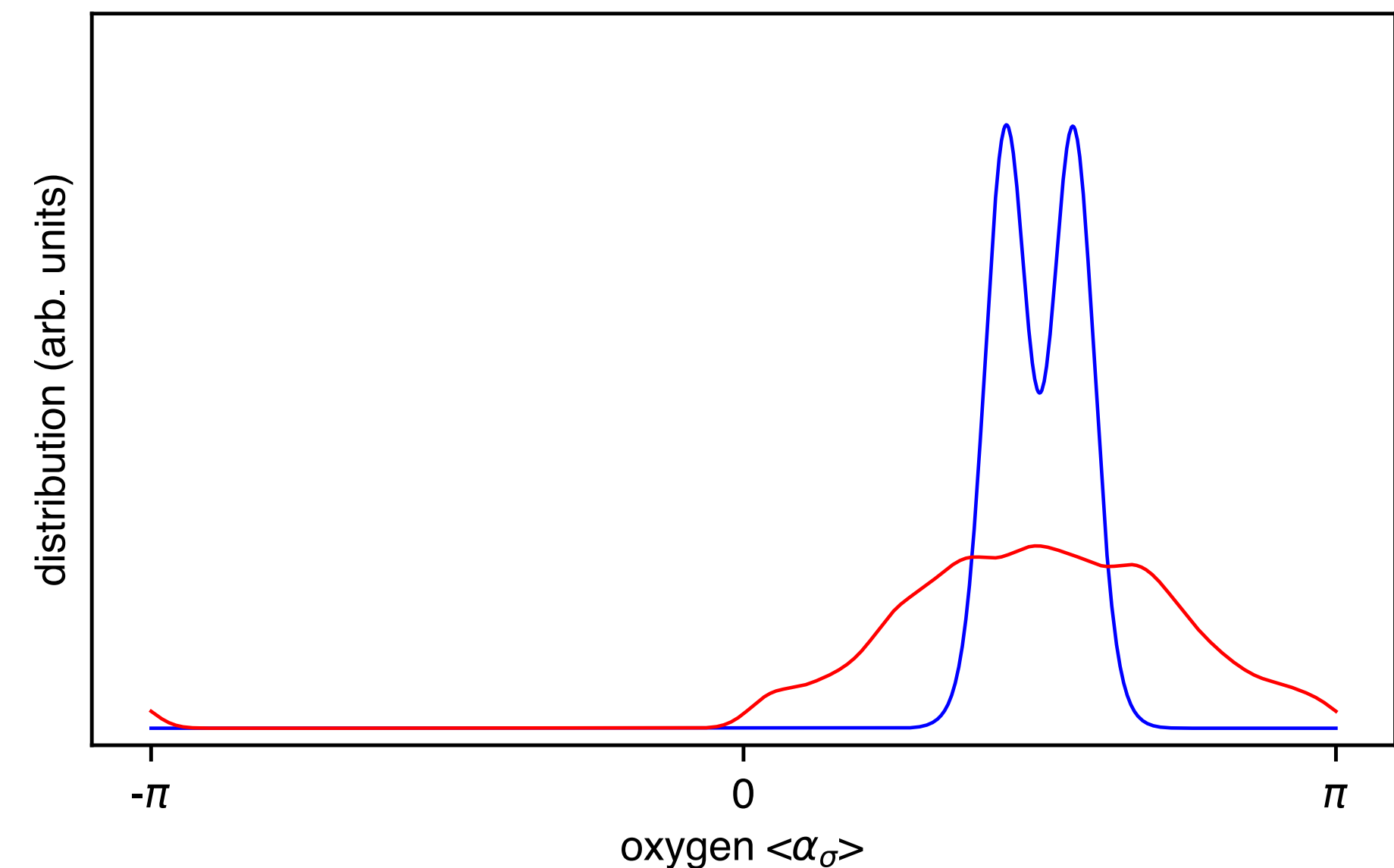
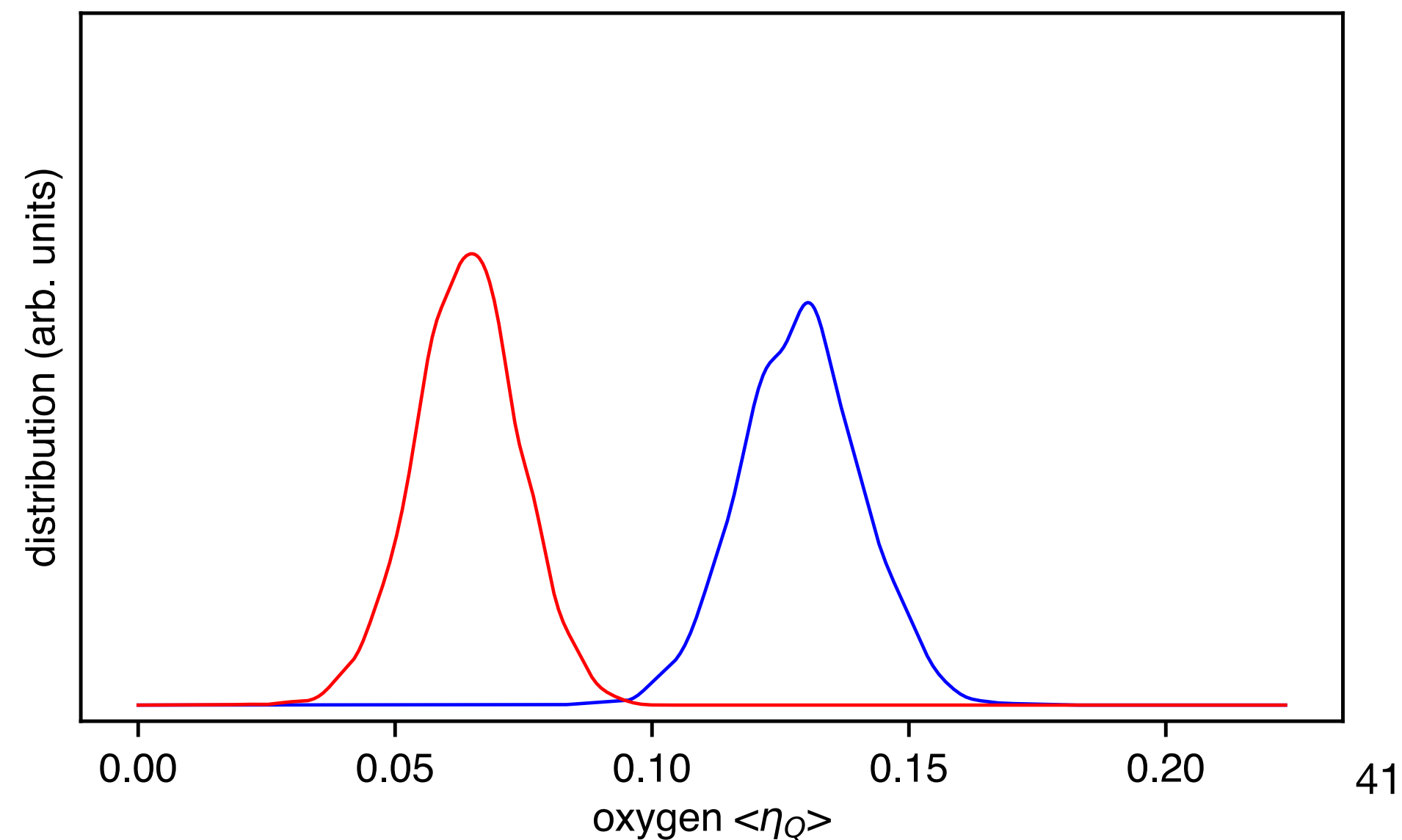
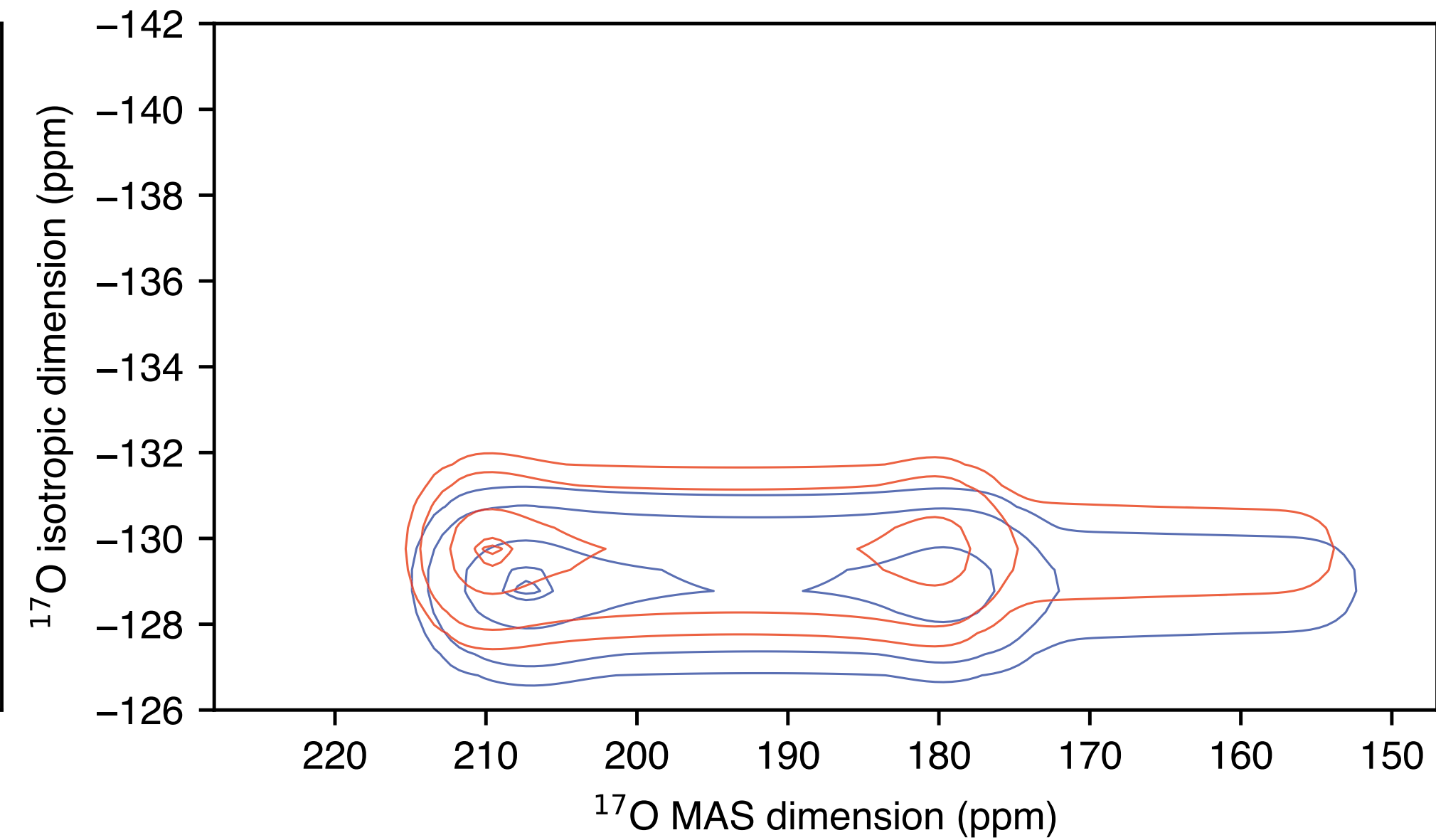
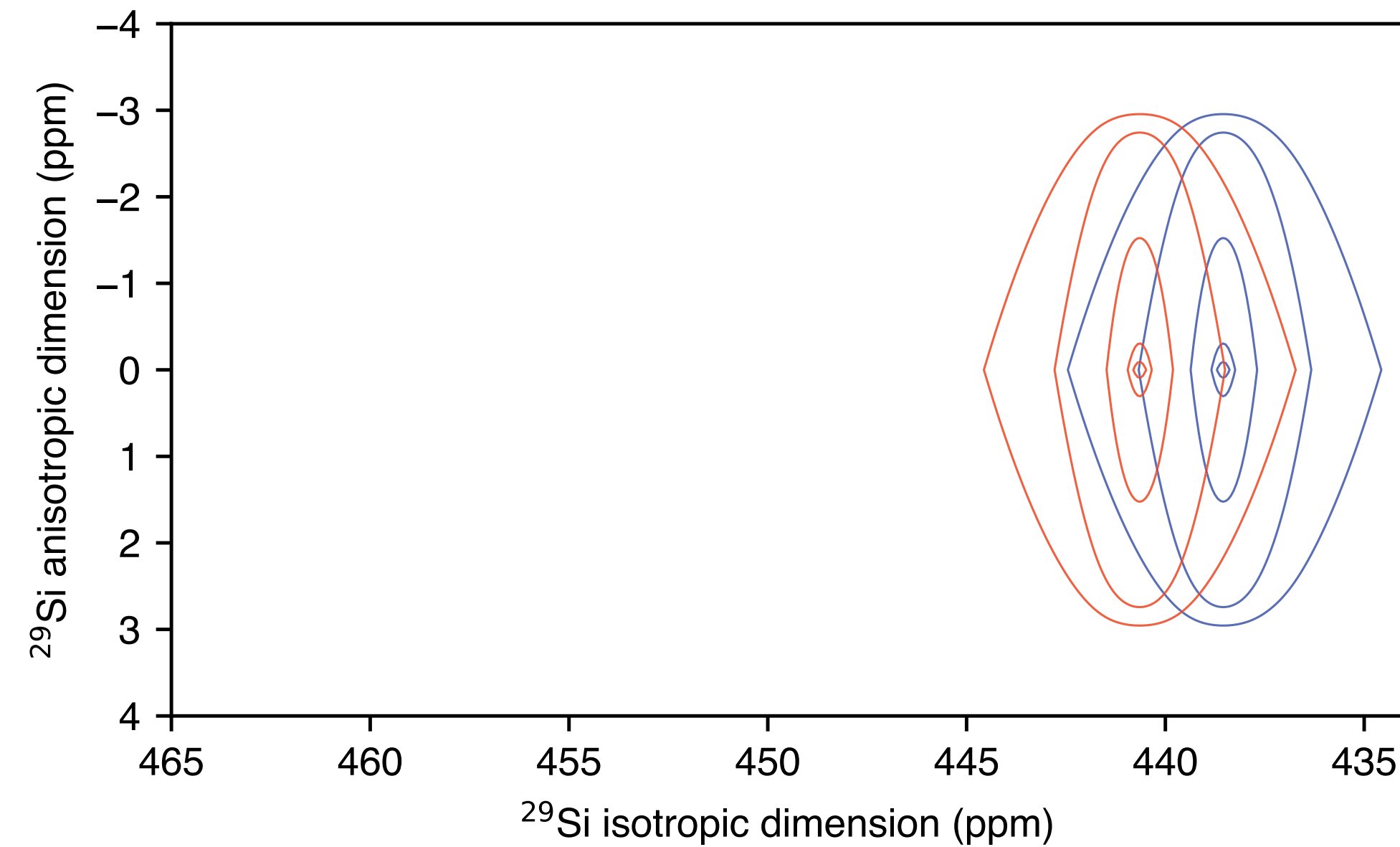
Dynamics of α — β inversion in cristobalite



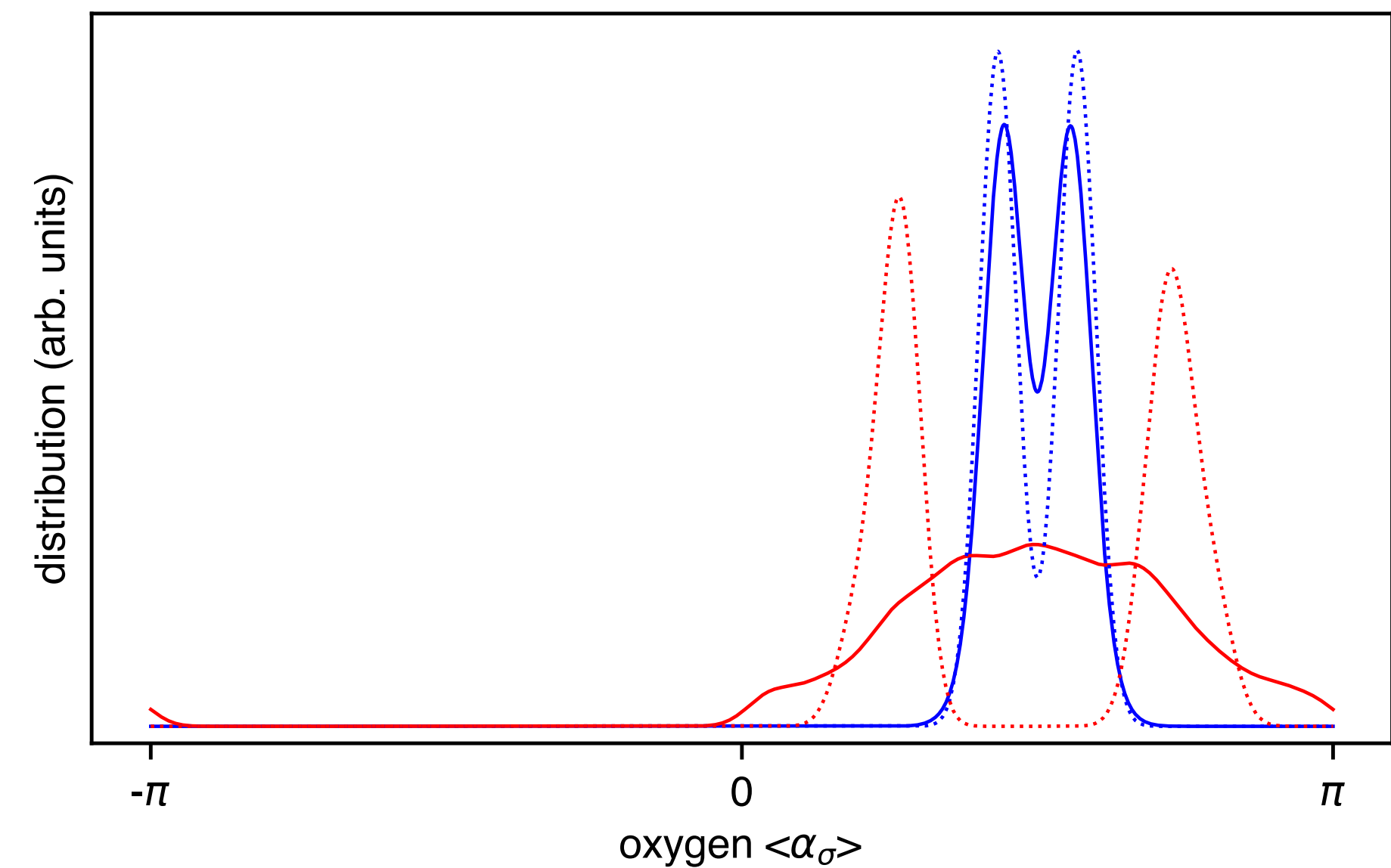
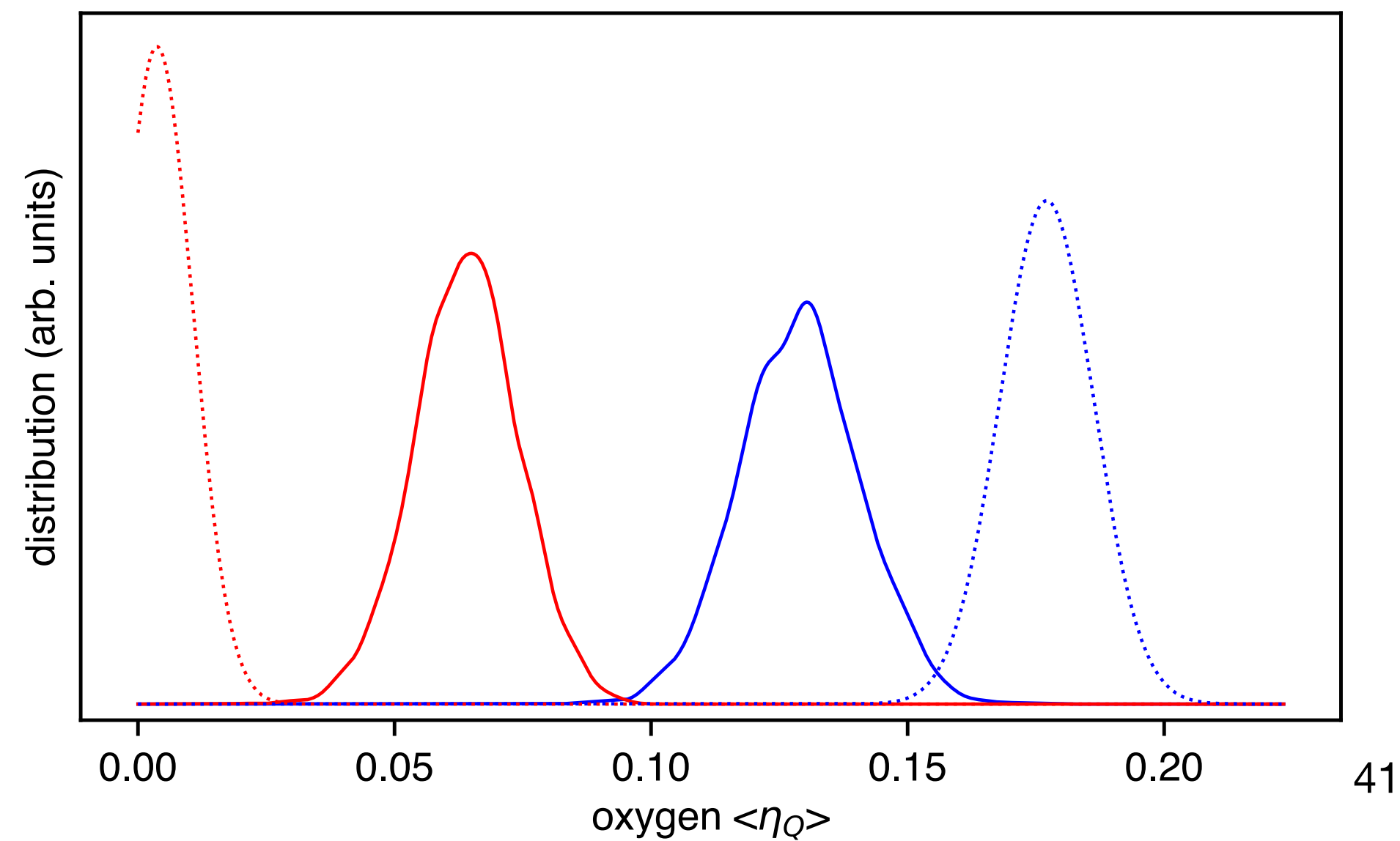
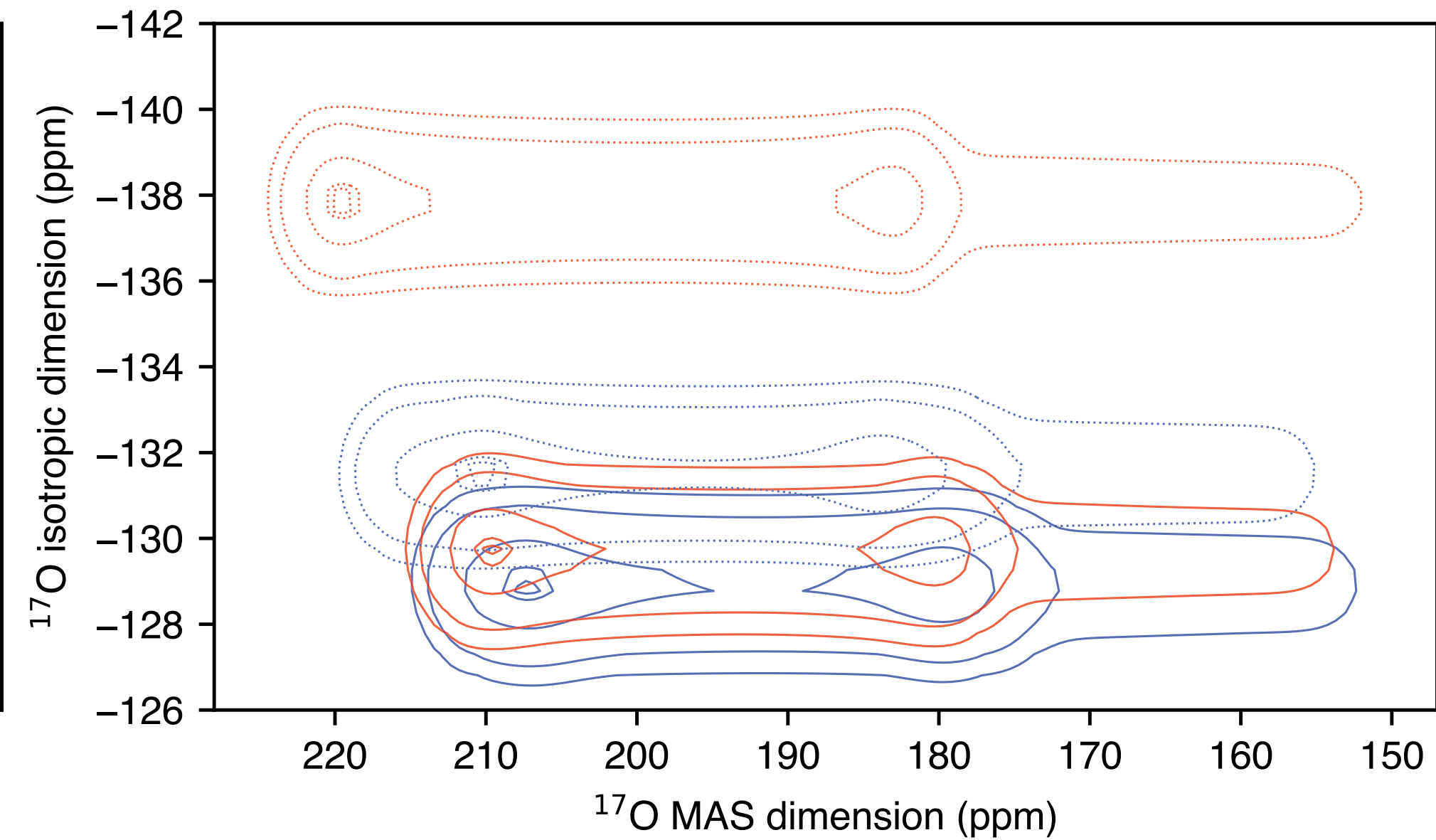
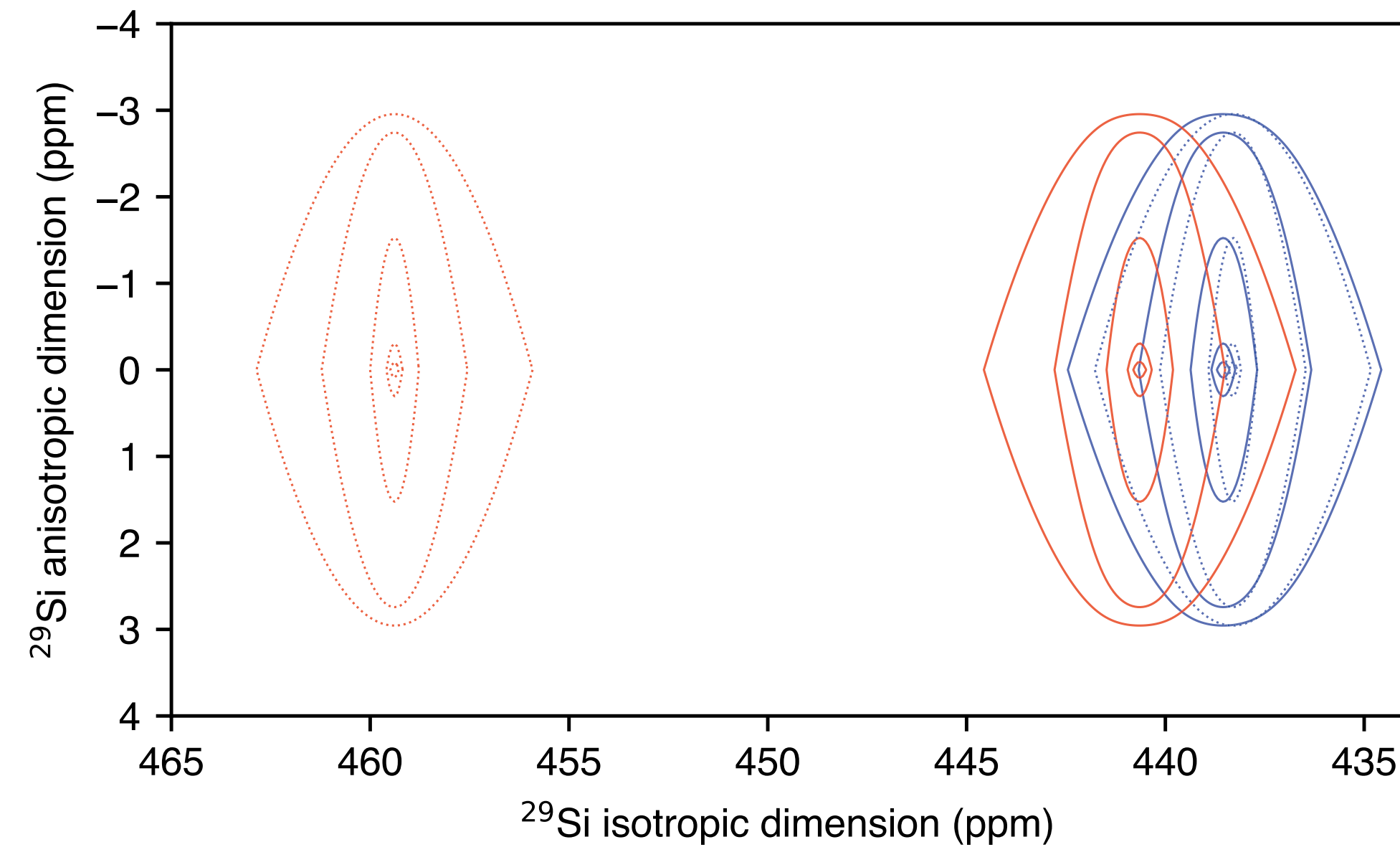
Dynamics of α — β inversion in cristobalite



Dynamics of α — β inversion in cristobalite



Dynamics of α — β inversion in cristobalite



Takehome message

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- Atom-centered atomistic ML formulation extends naturally to NMR parameters

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- NMR models provide an atomistic interpretation of experimental observables for solids and amorphous phases

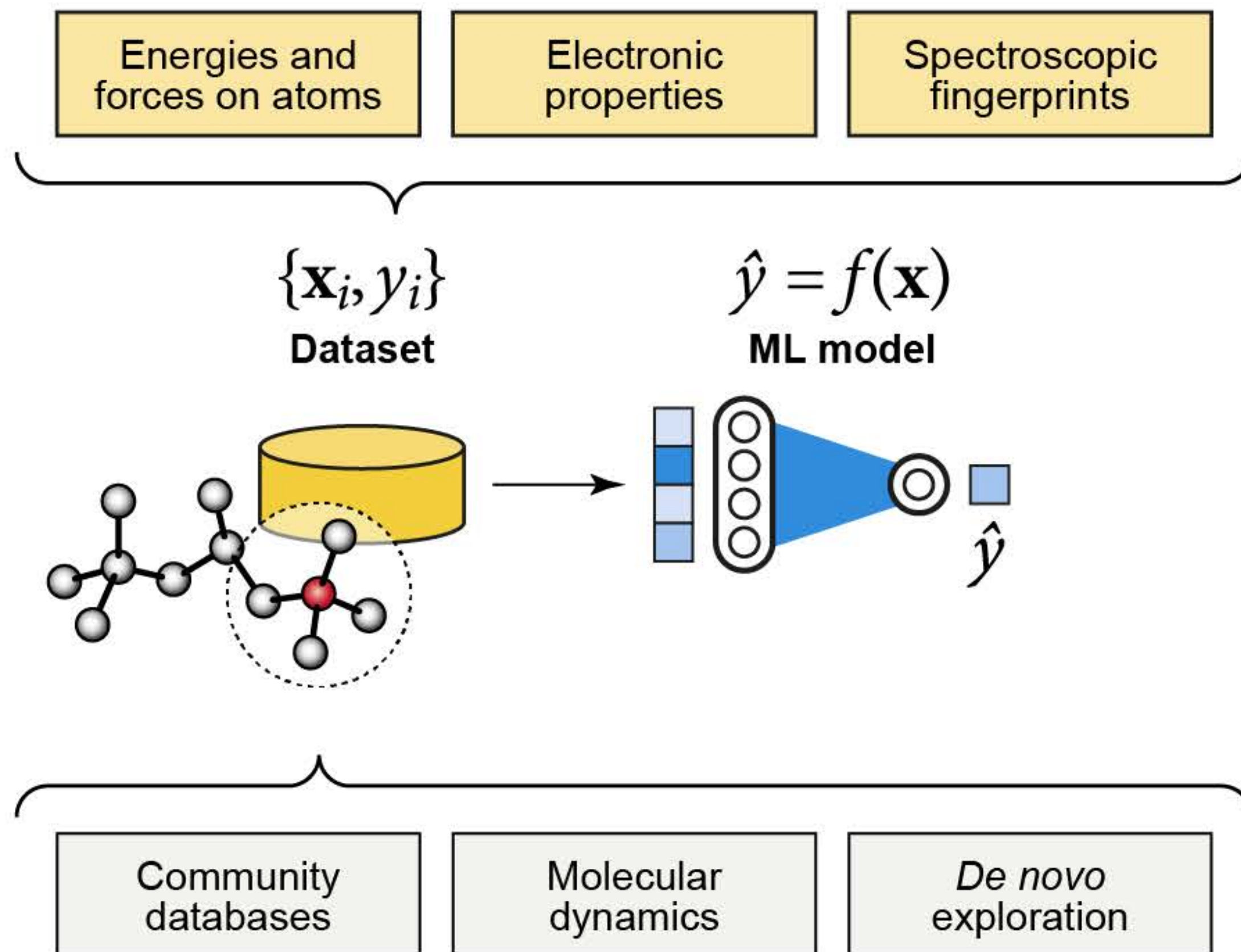
Takehome message

- Atom-centered atomistic ML formulation extends naturally to NMR parameters
- NMR models provide an atomistic interpretation of experimental observables for solids and amorphous phases
- Flexible architectures open the door for multi-property learning

Property 3:
potential energy surface

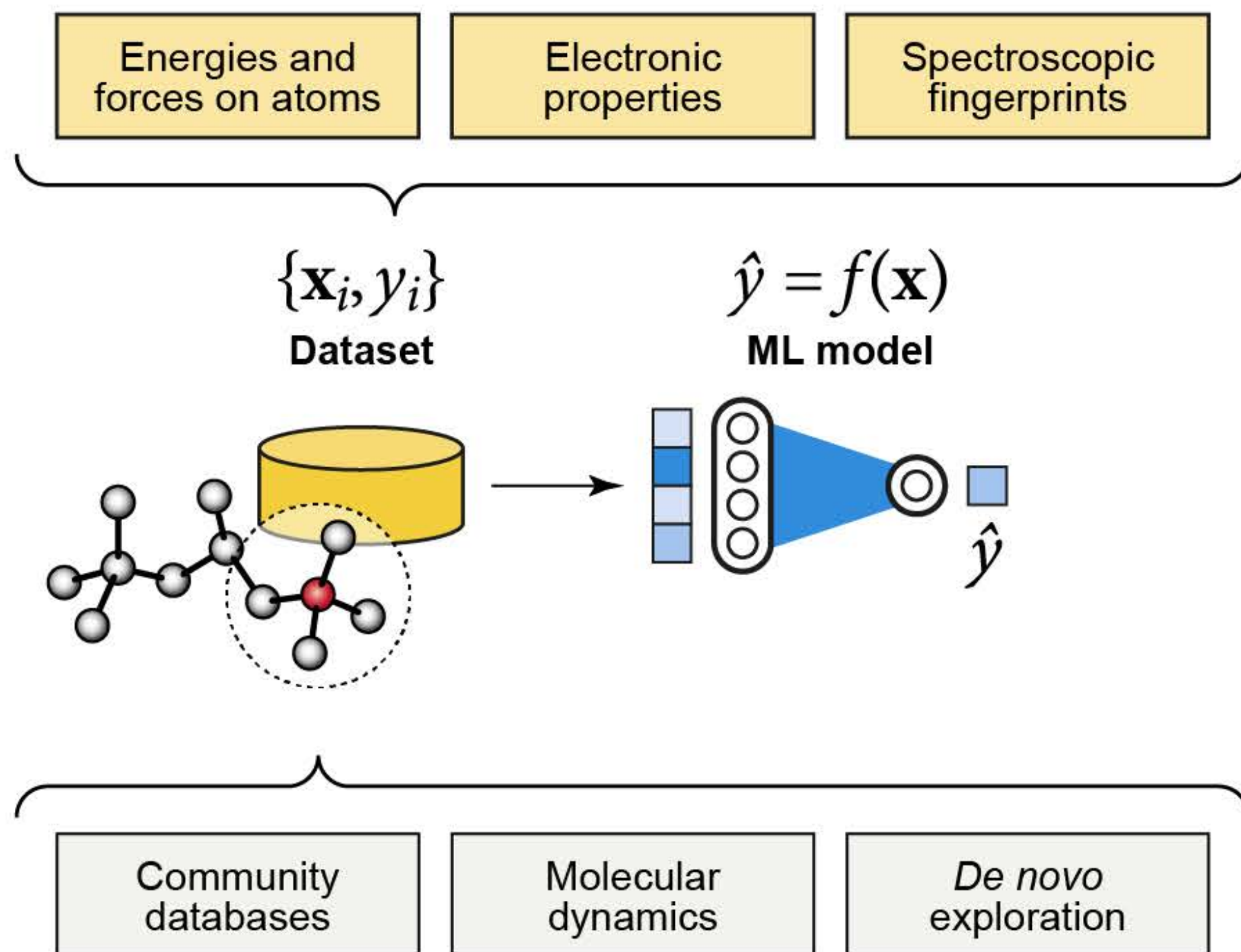
Towards general-purpose models/datasets

a Data enable atomistic machine learning

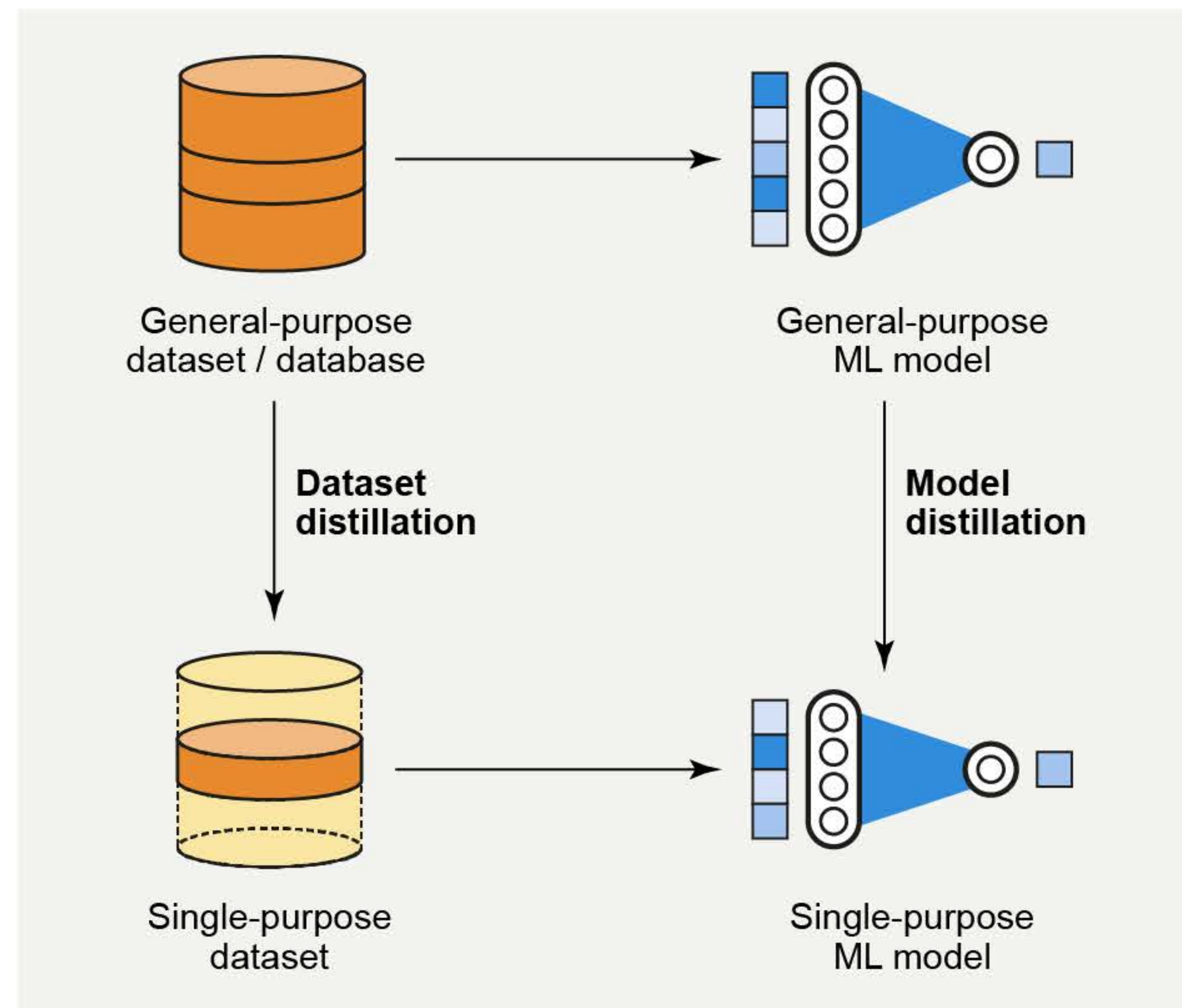


Towards general-purpose models/datasets

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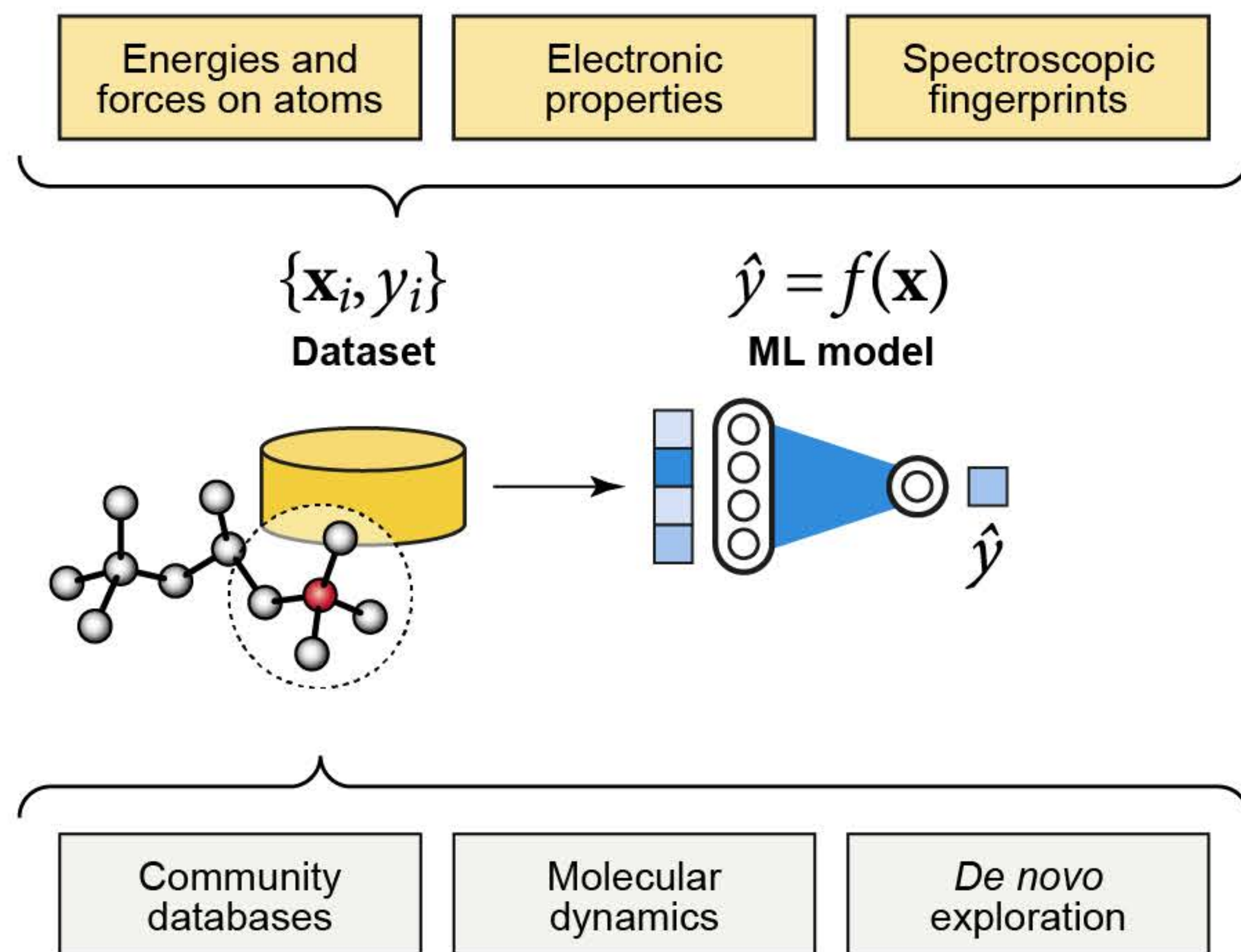


b Towards mainstream foundation models

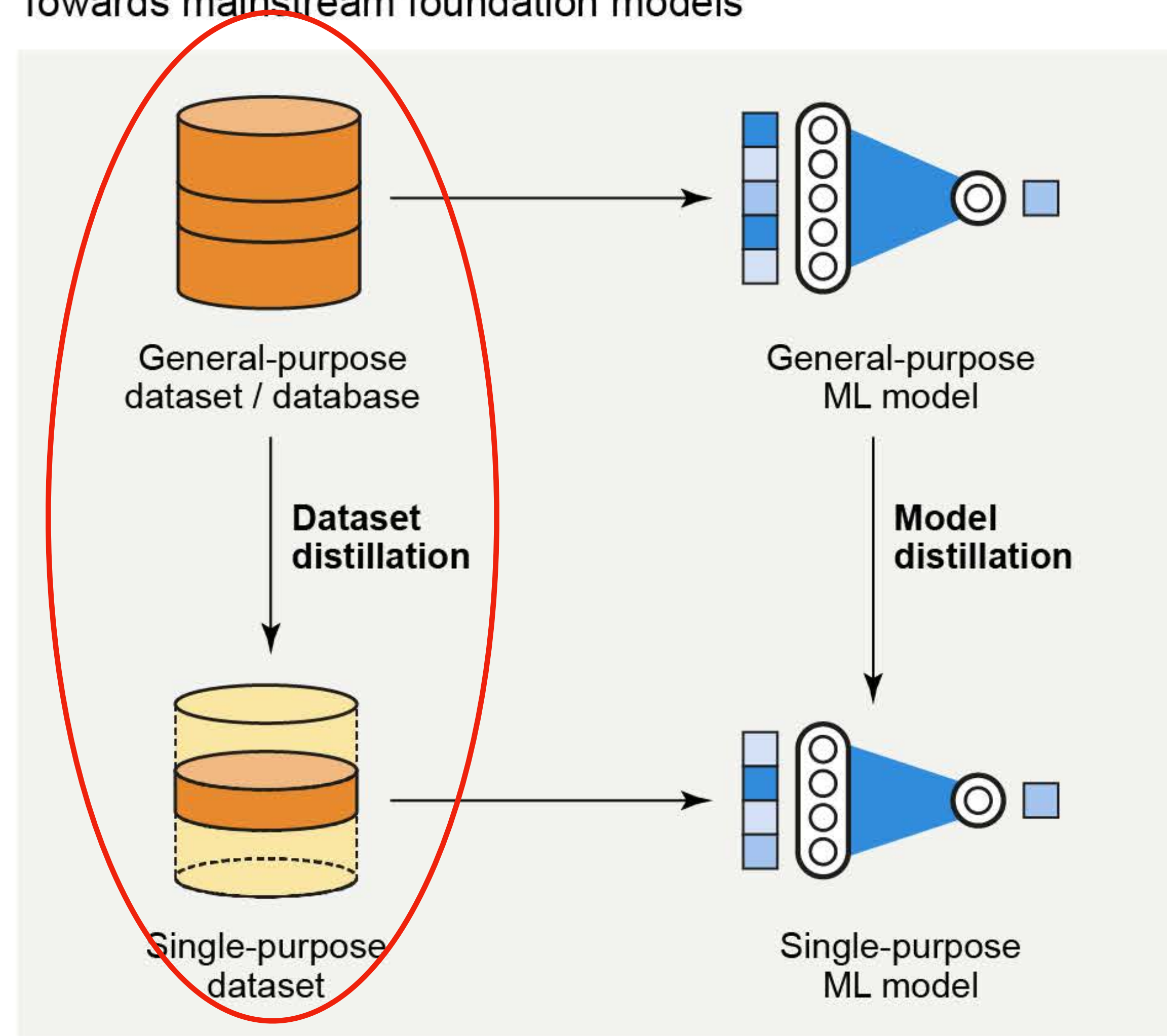


Towards general-purpose models/datasets

a Data enable atomistic machine learning

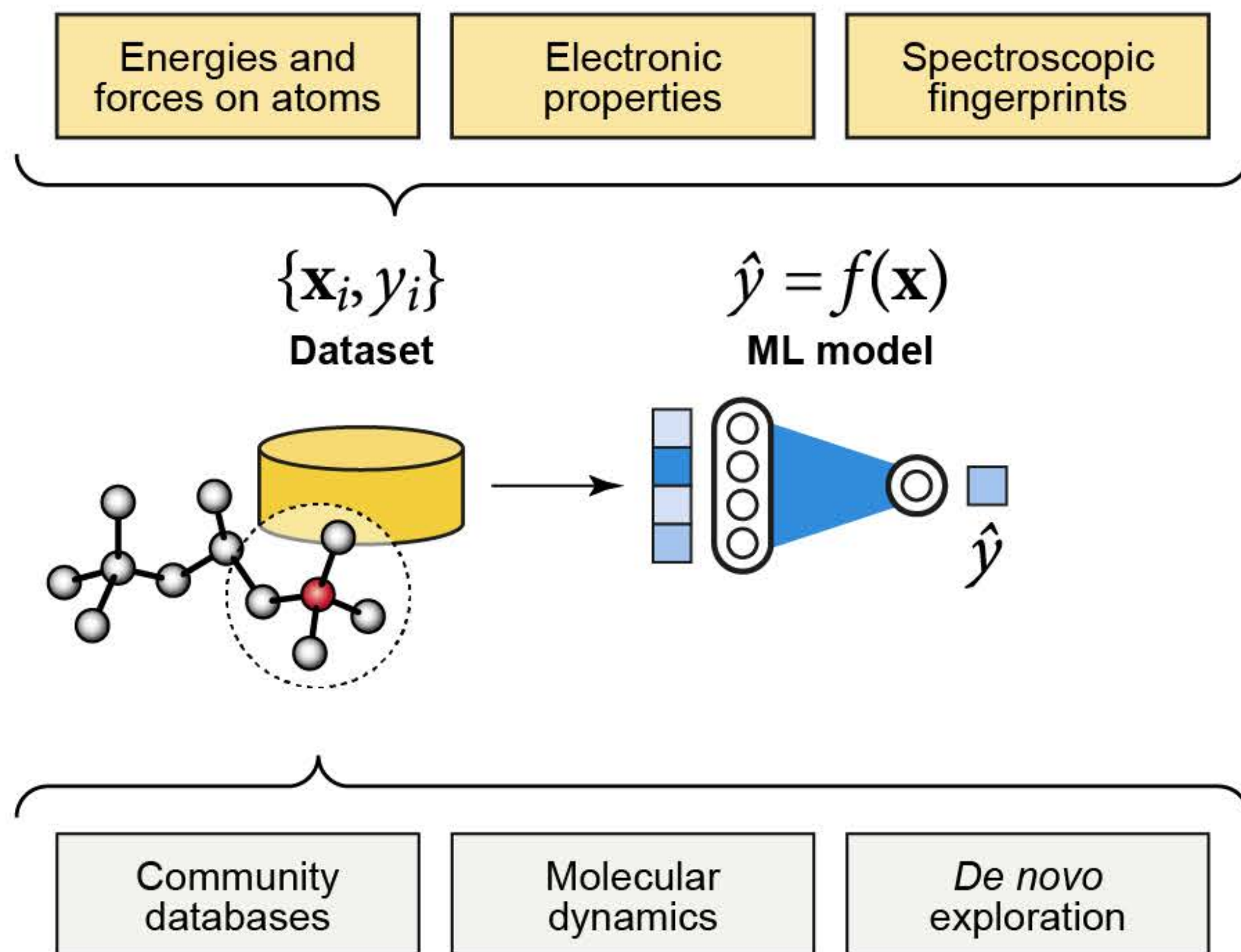


b Towards mainstream foundation models

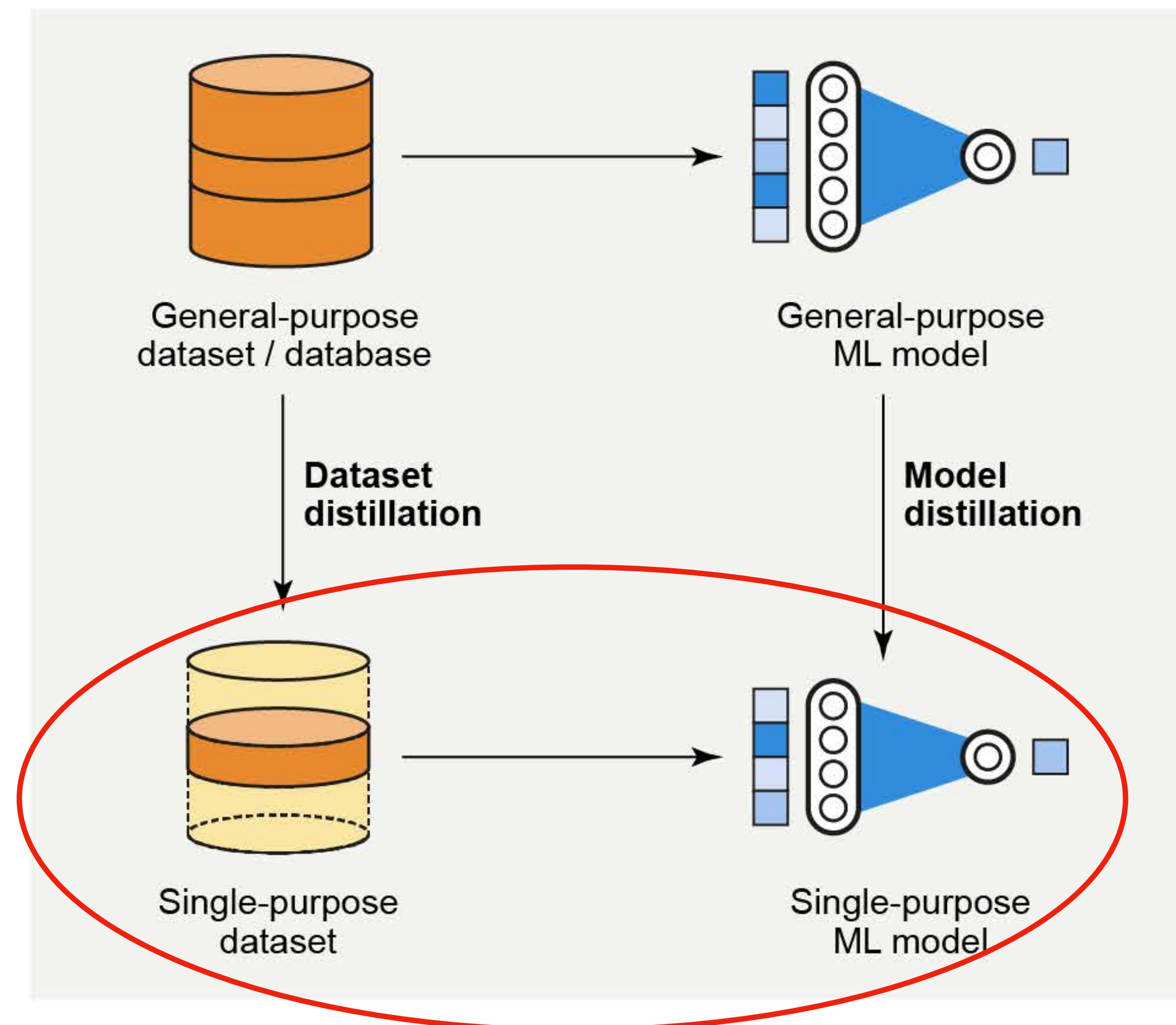


Towards general-purpose models/datasets

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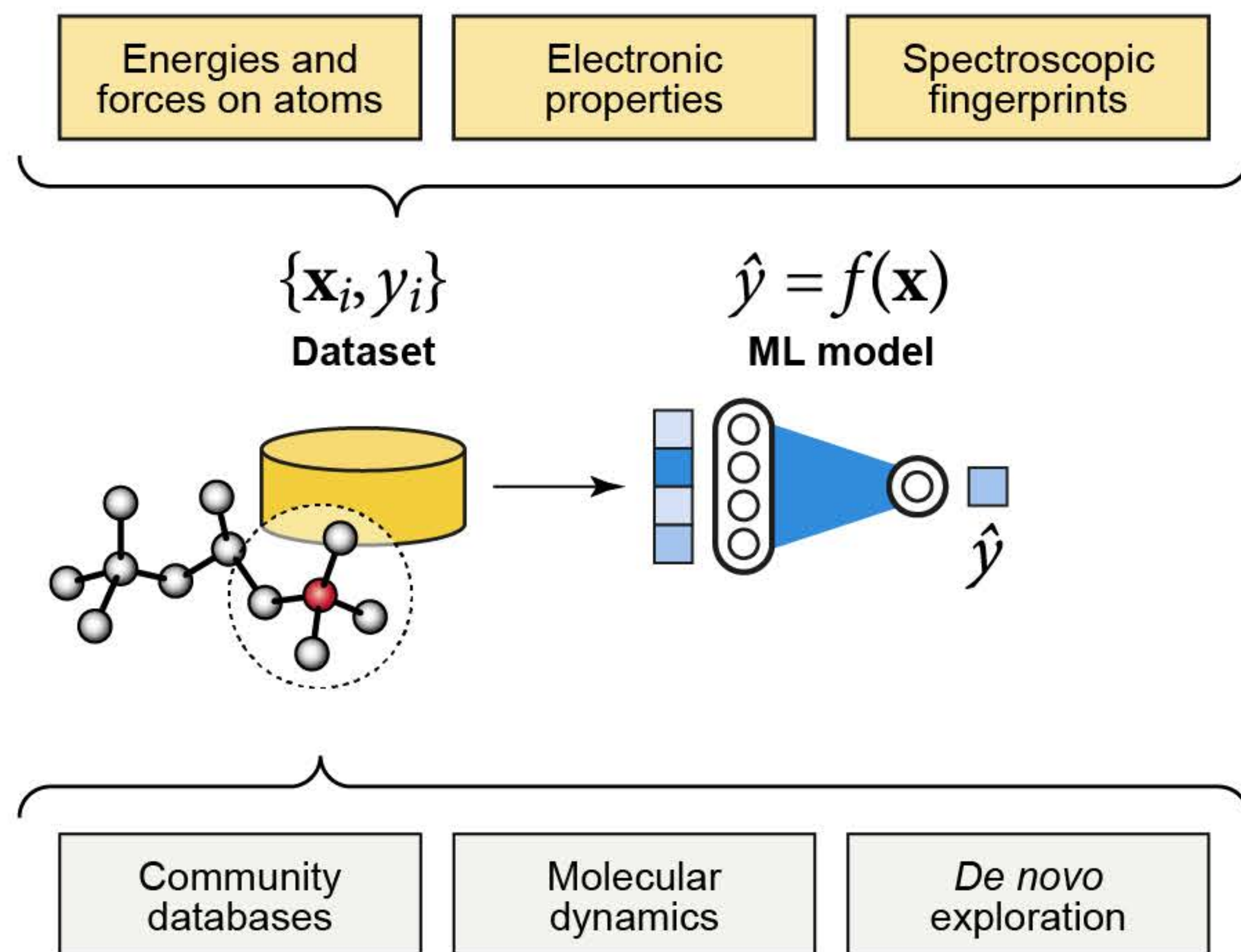


b Towards mainstream foundation models

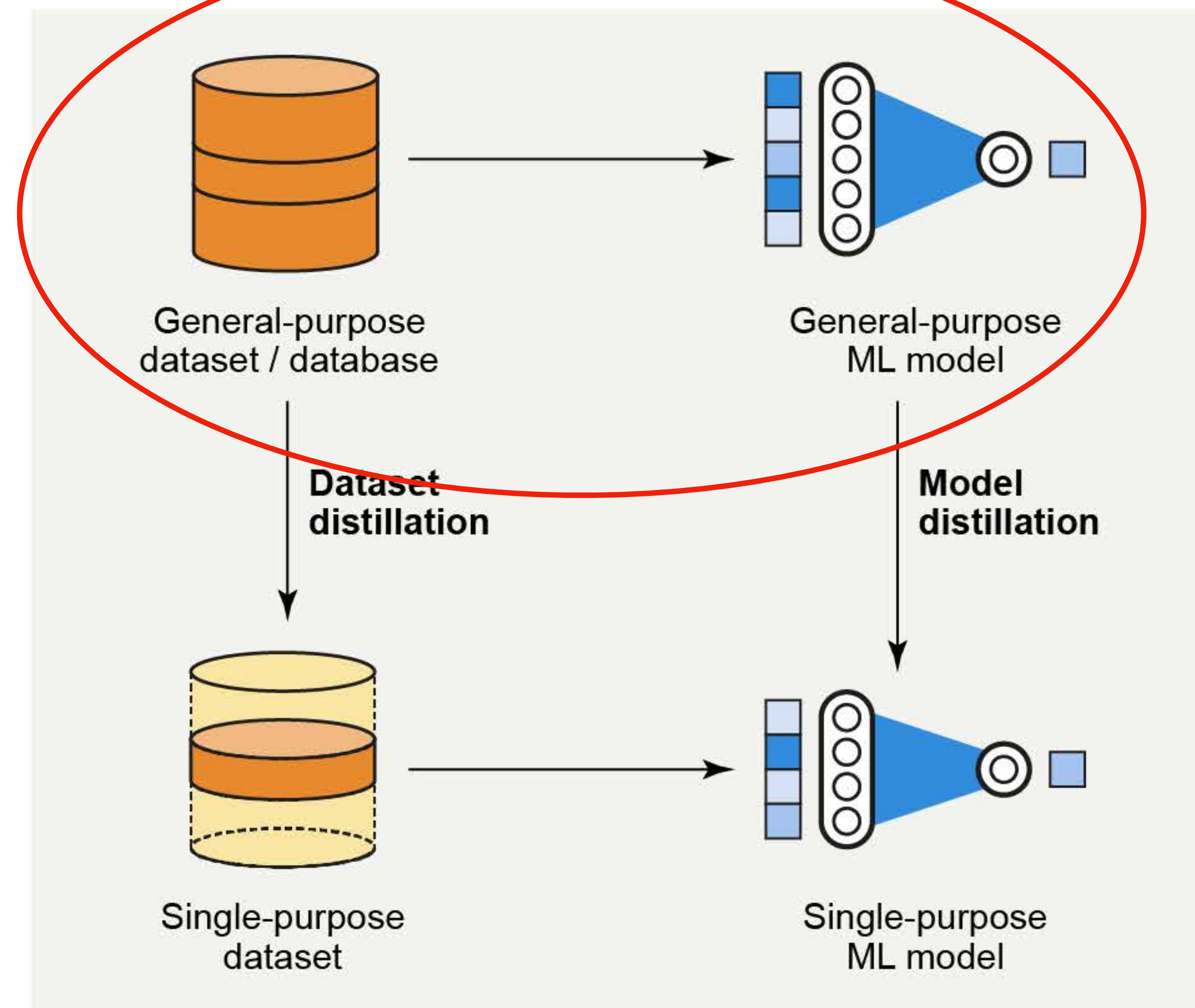


Towards general-purpose models/datasets

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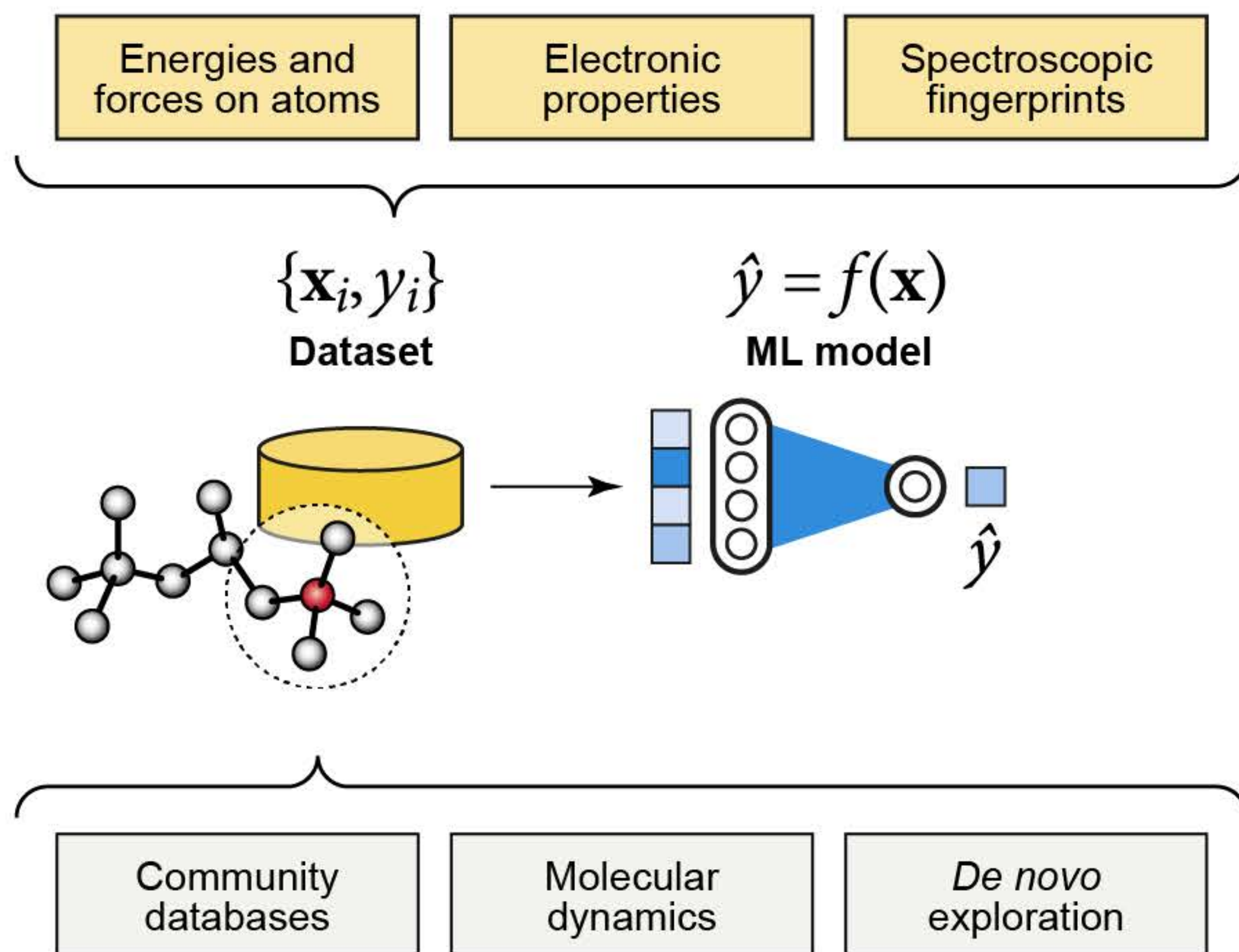


b Towards mainstream foundation models



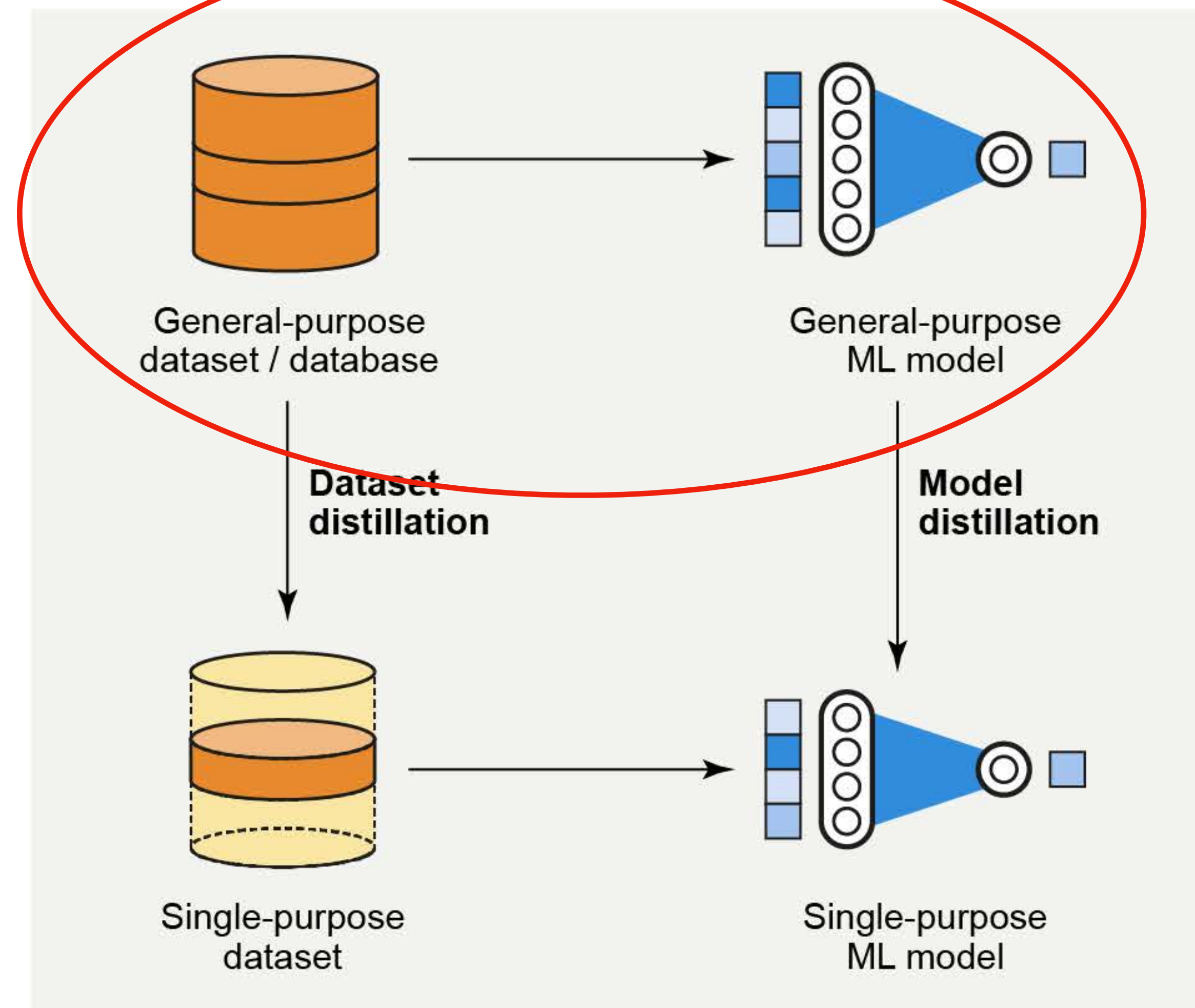
Towards general-purpose models/datasets

a Data enable atomistic machine learning



b Towards mainstream foundation models

Foundation / pretrained models



Example of a pretrained graph neural networks

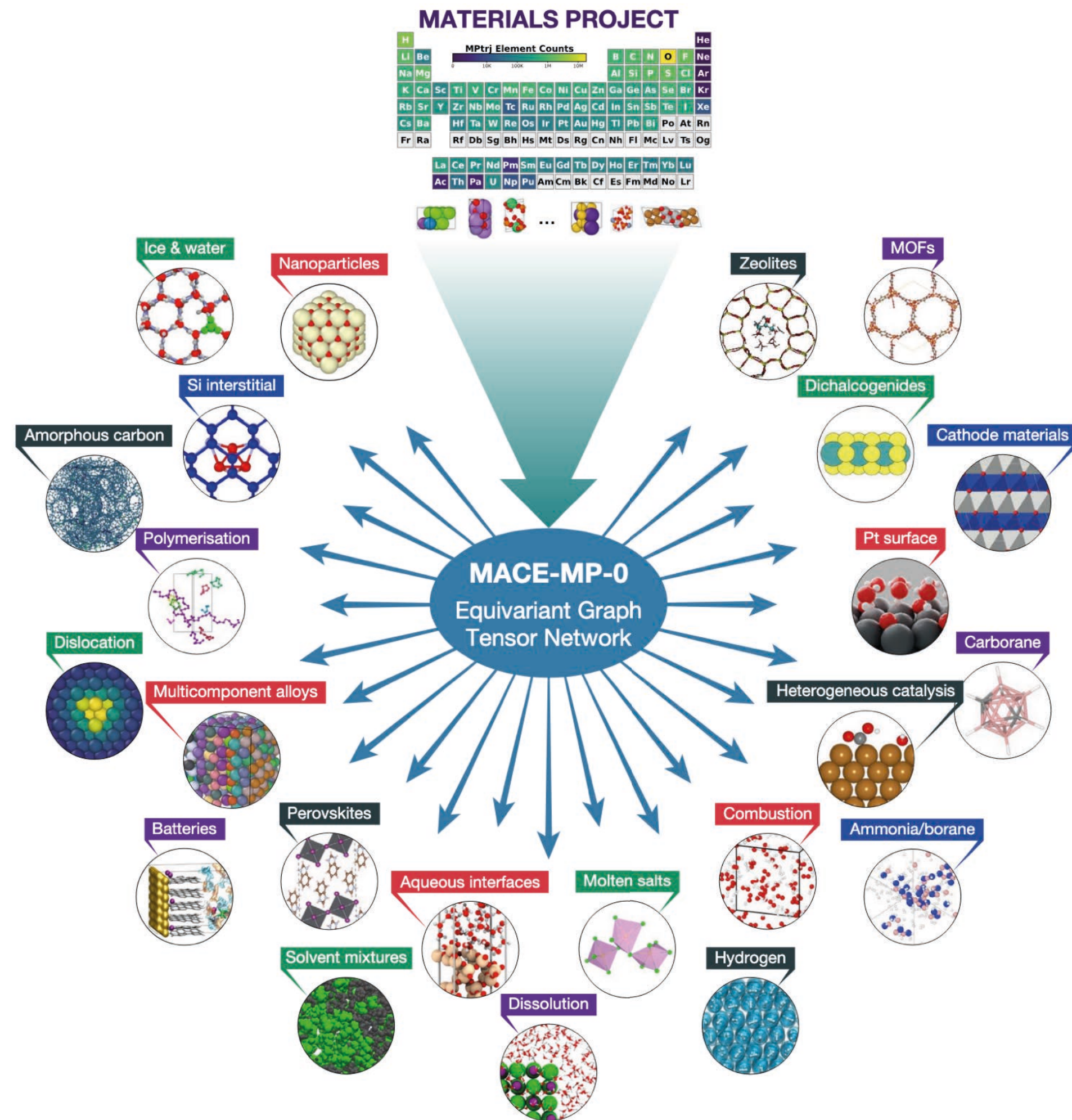


Figure 1: **A foundation model for materials modelling.** Trained only on Materials Project data (19) which consists primarily of inorganic crystals and is skewed heavily towards oxides, MACE-MP-0 is capable of molecular dynamics simulation across a wide variety of chemistries in the solid, liquid and gaseous phases.

Example of a pretrained graph neural networks

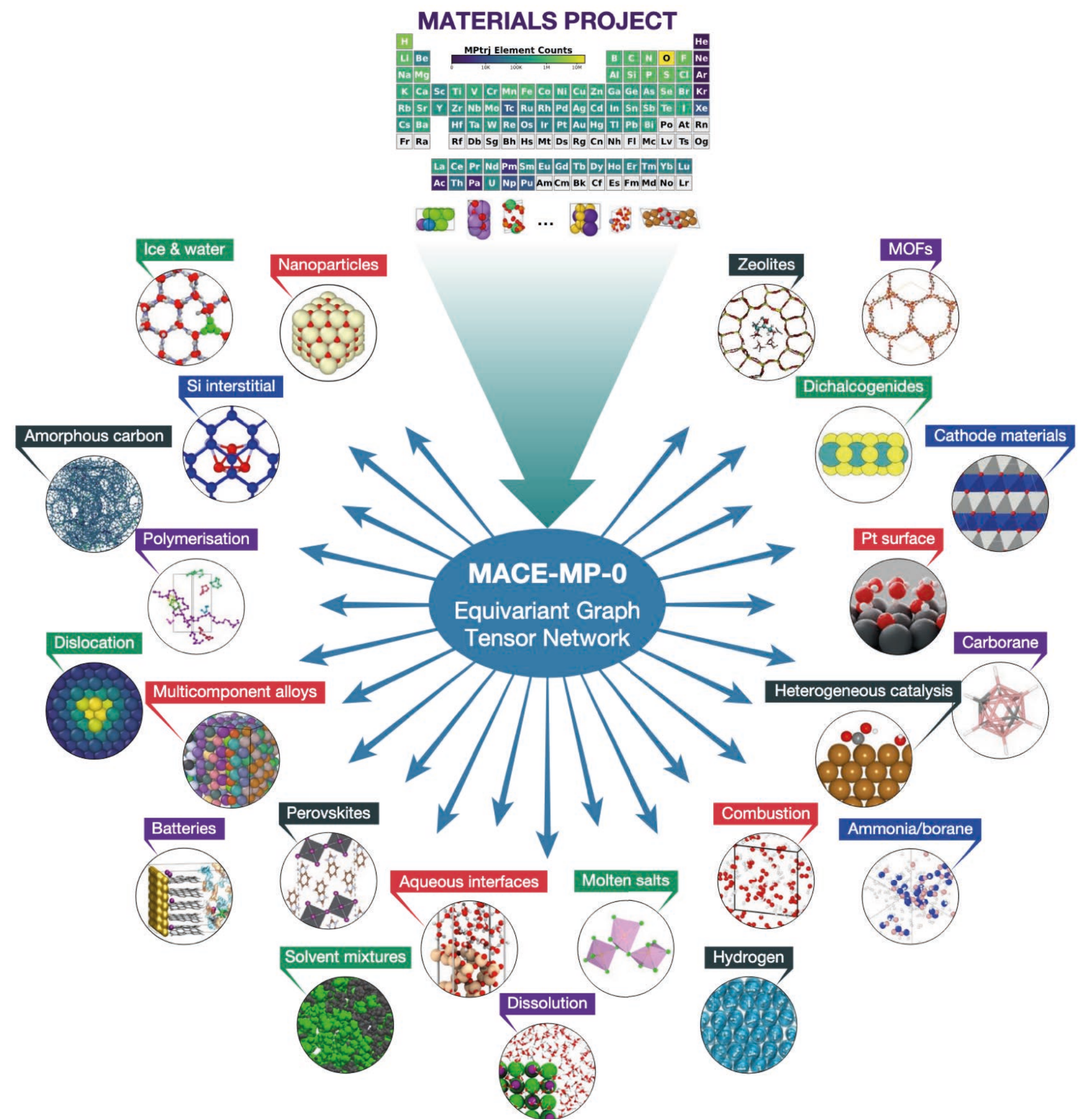


Figure 1: **A foundation model for materials modelling.** Trained only on Materials Project data (19) which consists primarily of inorganic crystals and is skewed heavily towards oxides, MACE-MP-0 is capable of molecular dynamics simulation across a wide variety of chemistries in the solid, liquid and gaseous phases.

Batatia et al., 2023

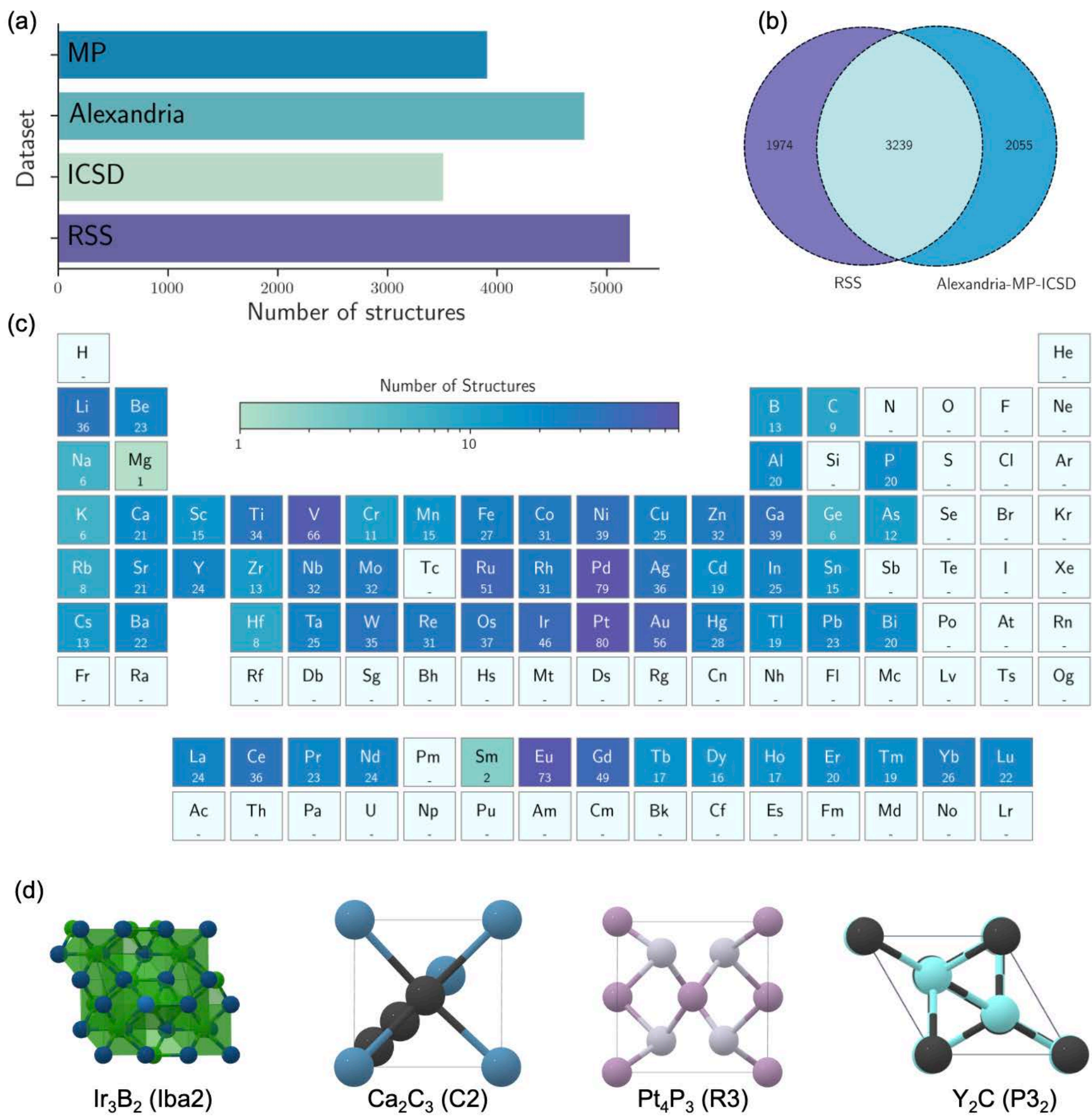


Fig. 3: MatterSim as a zero-shot emulator empowering materials discovery. (a) and (b) are the contribution of each dataset to the combined convex hull formed by Alexandria-MP-ICSD dataset (see text) and RSS-generated materials; (c) Elementwise appearance distribution[37] of the 852 RSS-generated materials found to be lower than the combined convex hull formed by the Alexandria-MP-ICSD and RSS-generated materials. The materials containing H, Si, N, Sb, O, S, Se, Te, F, Cl, Br, I are removed due to potential issue with how anion corrections are implemented in Materials Project when applied to hypothetical materials[38]. (d) exhibits examples of materials found to be lower than the Alexandria-MP-ICSD hull, with the corresponding space group in the parentheses.

Yang et al., 2024

Example of a pretrained graph neural networks

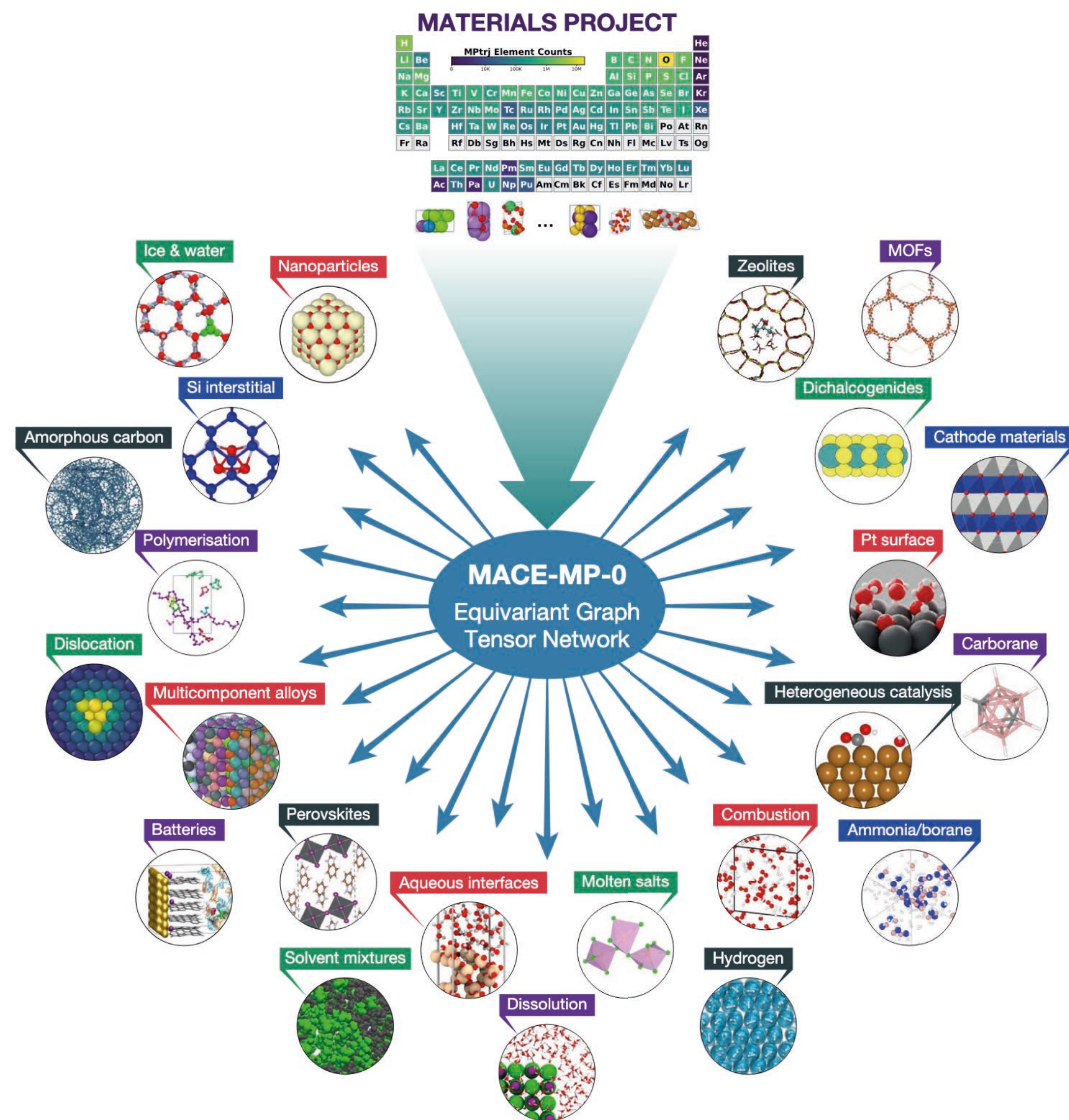


Figure 1: **A foundation model for materials modelling.** Trained only on Materials Project data (19) which consists primarily of inorganic crystals and is skewed heavily towards oxides, MACE-MP-0 is capable of molecular dynamics simulation across a wide variety of chemistries in the solid, liquid and gaseous phases.

Batatia et al., 2023

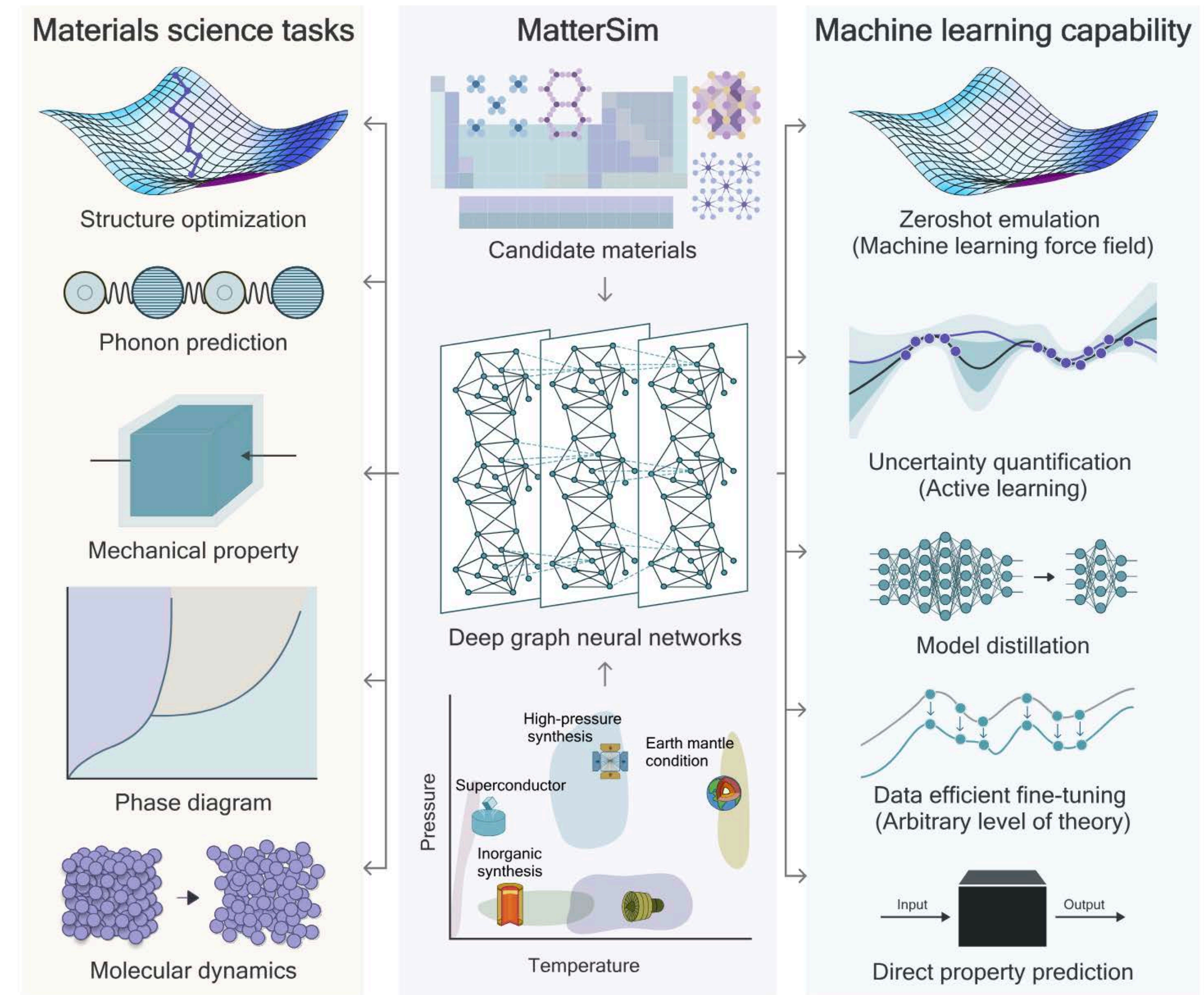
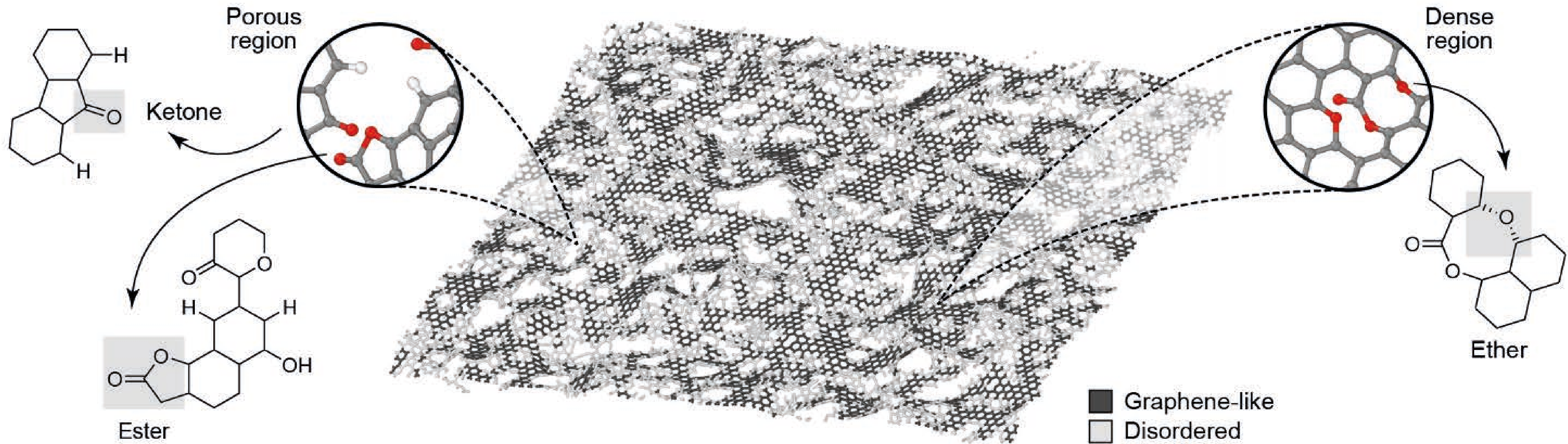


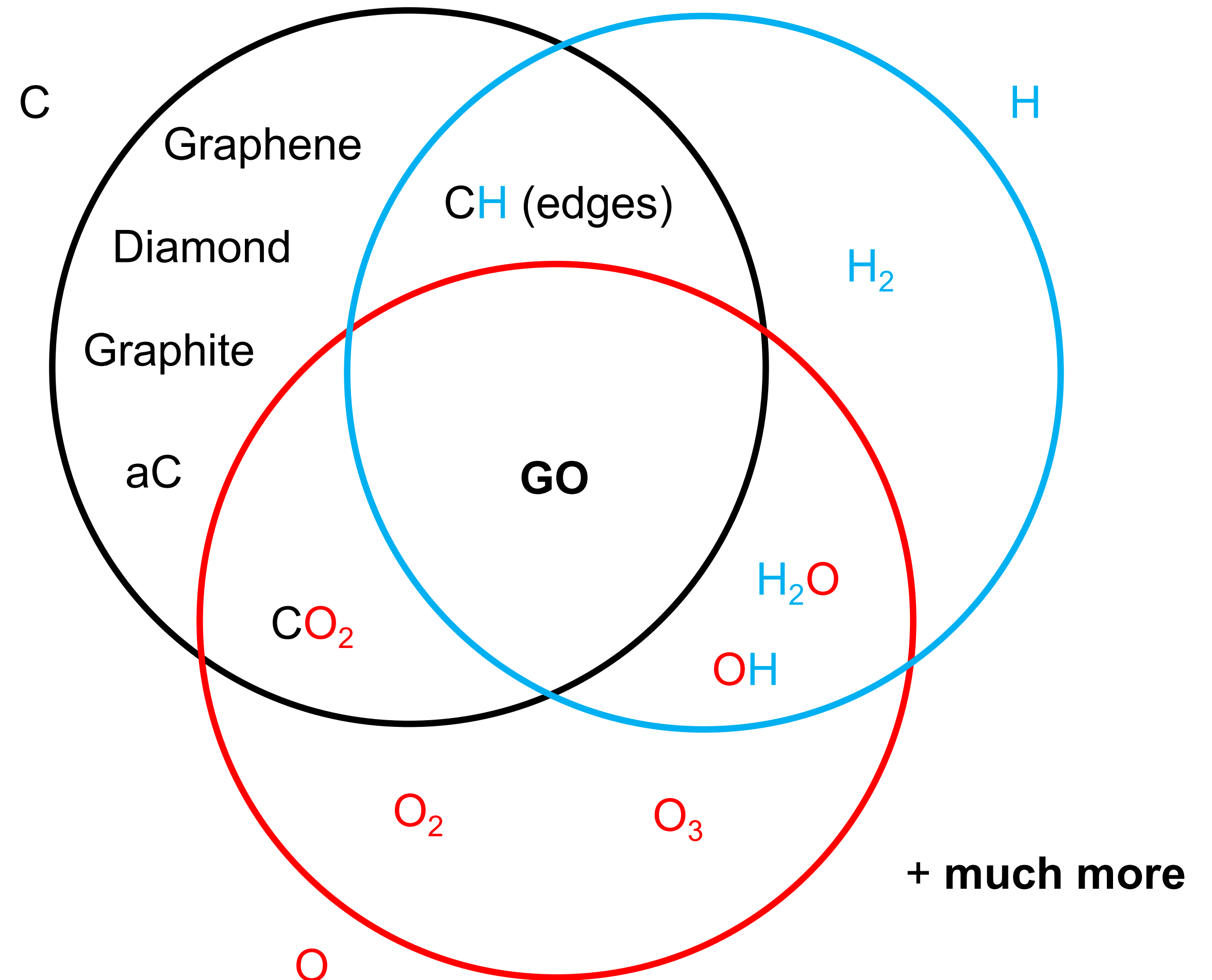
Fig. 1: MatterSim is a deep learning atomistic model for predicting materials properties with high predictive accuracy across chemical elements, temperatures and pressures, enabling a wide range of applicability and functionality.

Graphene oxide as a basis for transferable MLIPs



The structure of reduced GO

OXYGEN GROUPS			
ALCOHOL	ETHER	ALDEHYDE	KETONE
$\text{R}-\ddot{\text{O}}\text{H}$	$\text{R}-\ddot{\text{O}}-\text{R}'$	$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\text{H}$	$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\text{R}'$
(-ol)	(ether)	(-al)	(-one)
EPOXIDE	CARBOXYLIC ACID	ESTER	ACYL HALIDE
$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\text{C}-\text{R}''$	$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\ddot{\text{O}}\text{H}$	$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\ddot{\text{O}}-\text{R}'$	$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\text{X}$
(epoxide)	(-oic acid)	(-oate)	(-oyl halide)
AMIDE		ACID ANHYDRIDE	
$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\text{N}(\text{H})-\text{R}'$		$\text{R}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\ddot{\text{O}}-\overset{\text{:}\ddot{\text{O}}\text{:}}{\text{C}}-\text{R}'$	
(-amide)		(-anhydride)	

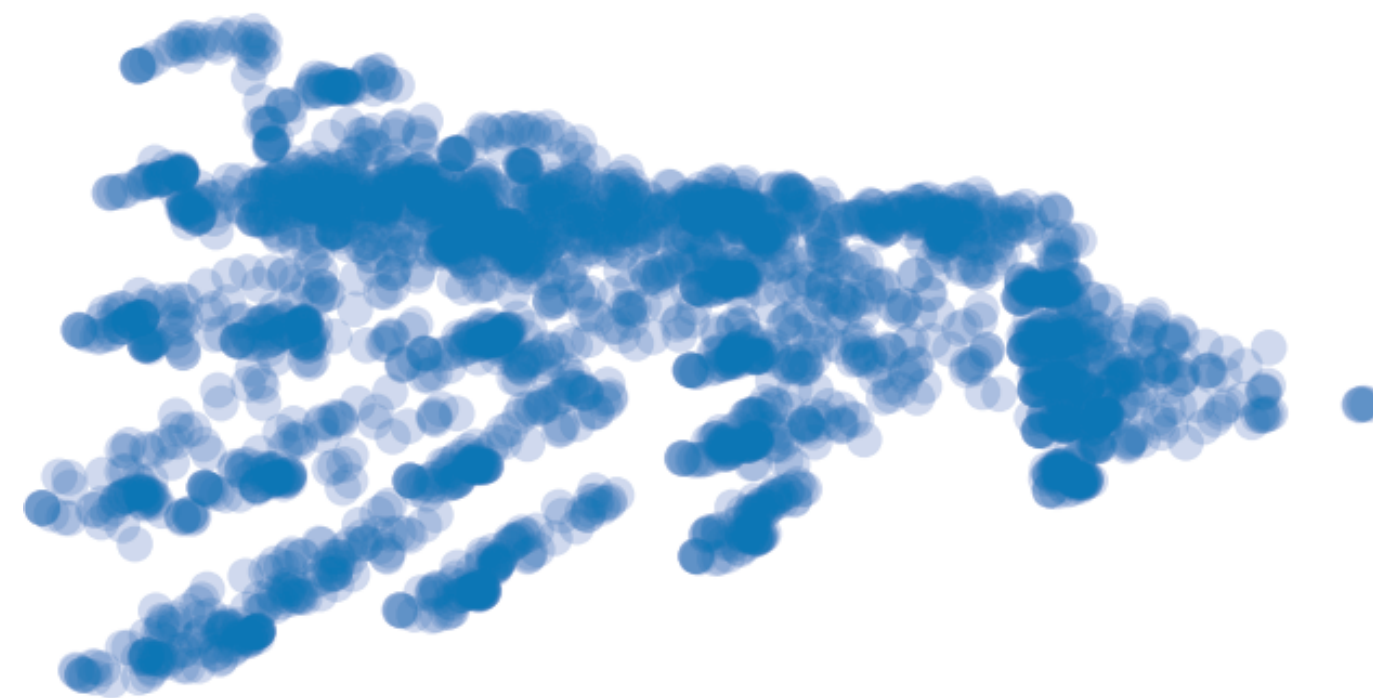


Objectives

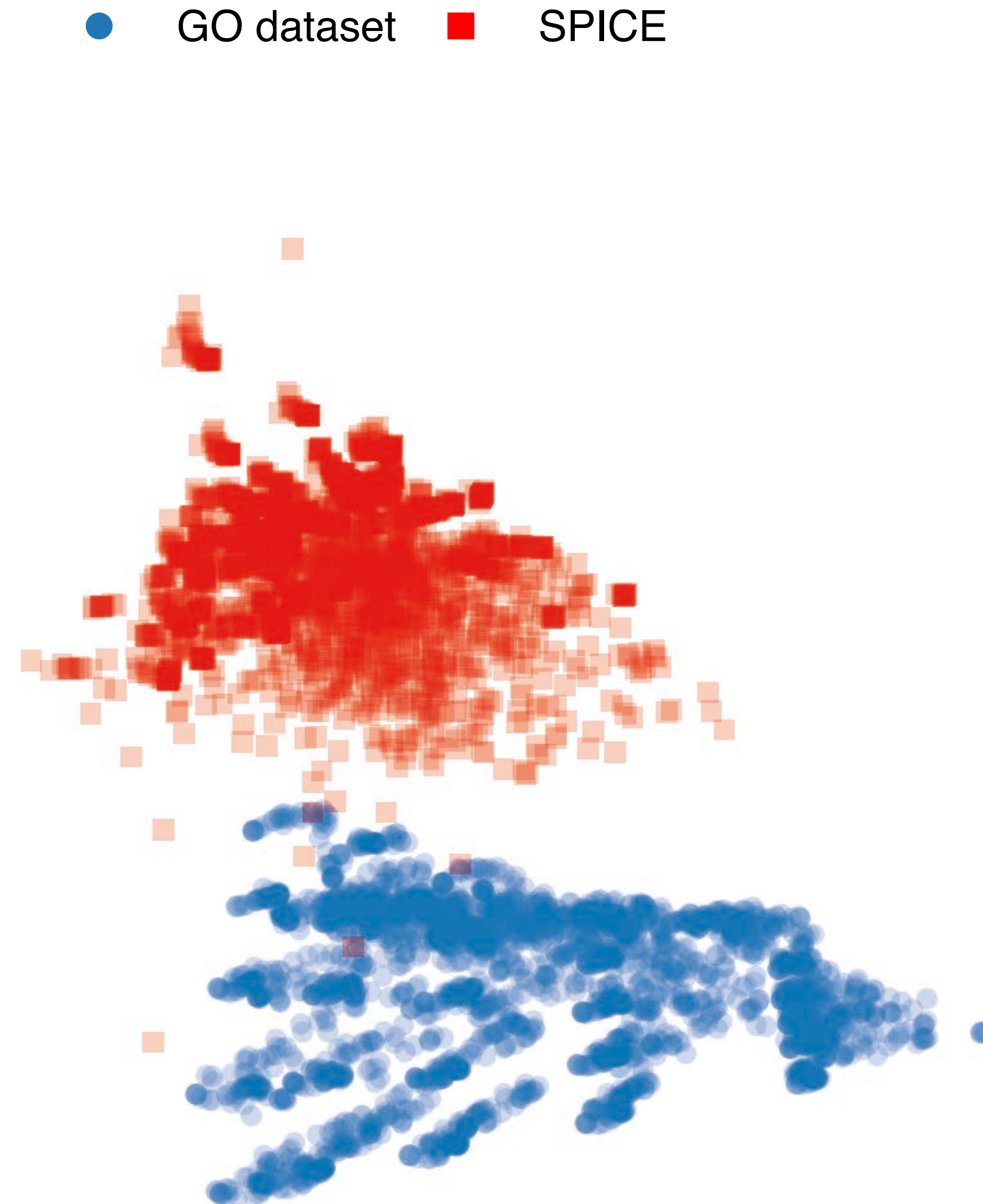
- Not to train an MLIP from scratch (yes, it is painful!)
- Assess the out-of-domain performance of an MLIP trained on (some) chemistry: **GO-MACE-23**
- Quantify the role of data and architecture in out-of-domain performance
- Identify requirements for general-purpose MLIP targeting (organic) chemistry

GO vs molecules from GO-MACE-23 perspective

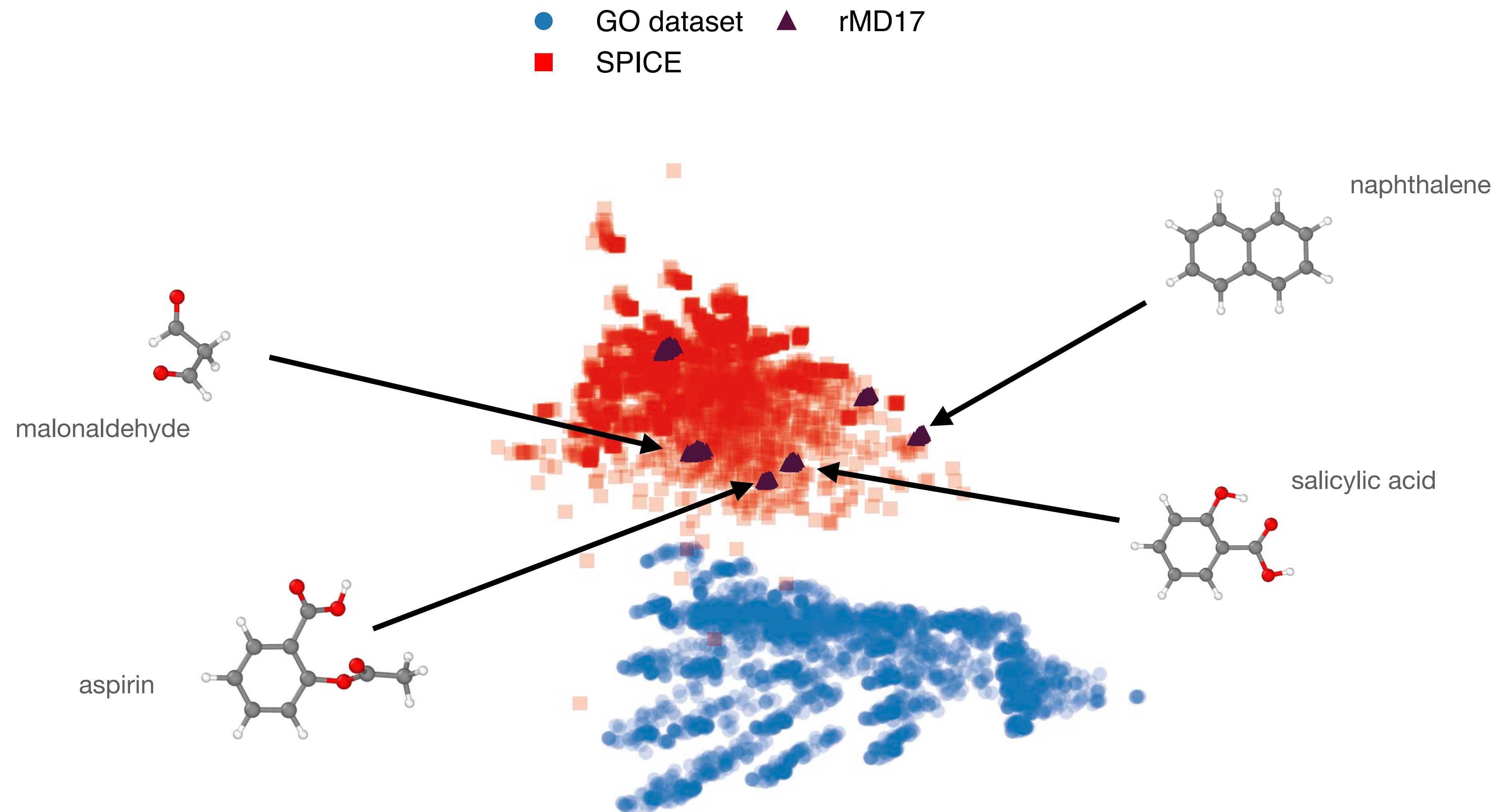
- GO dataset



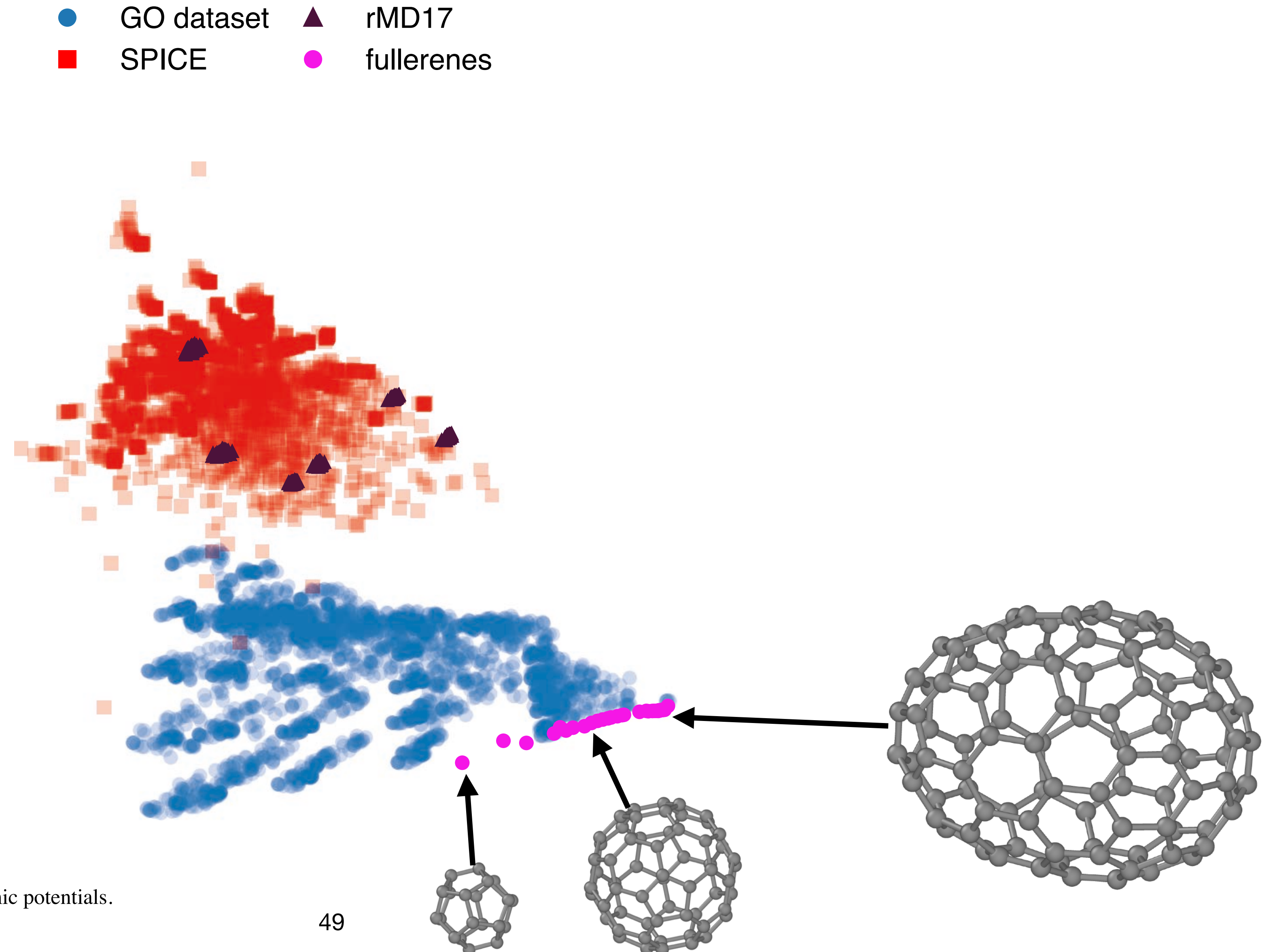
GO vs molecules from GO-MACE-23 perspective



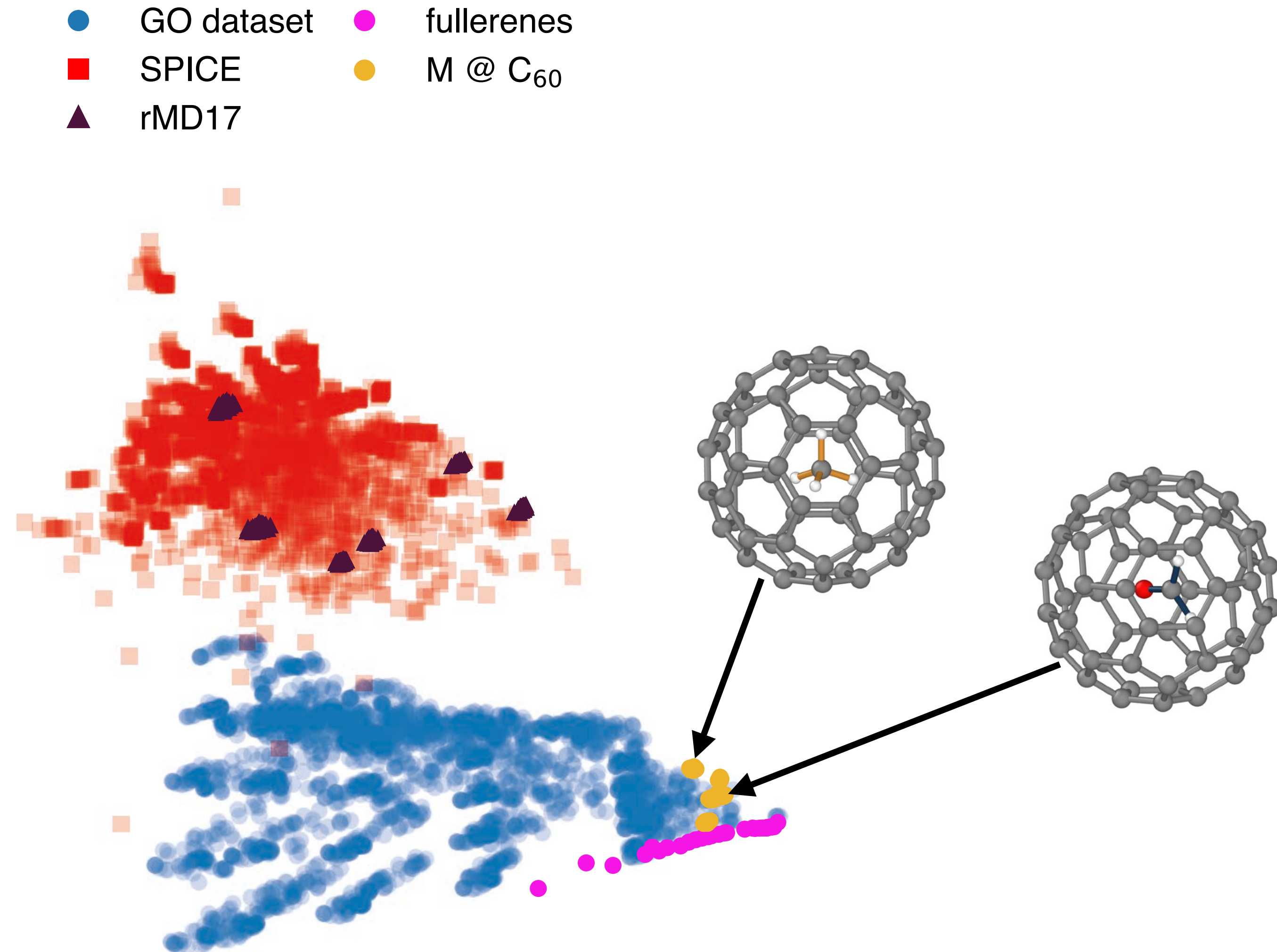
GO vs molecules from GO-MACE-23 perspective



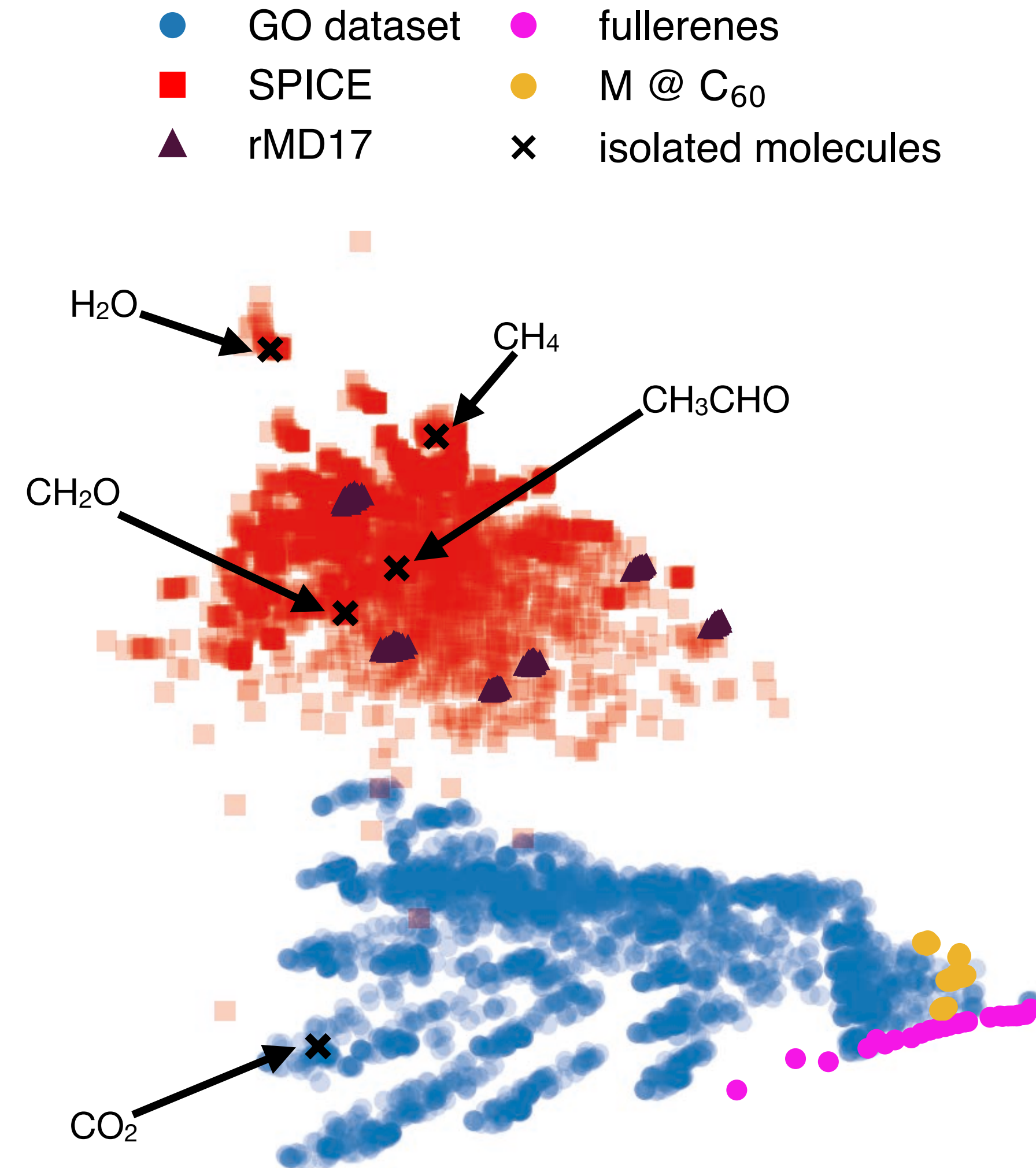
GO vs molecules from GO-MACE-23 perspective



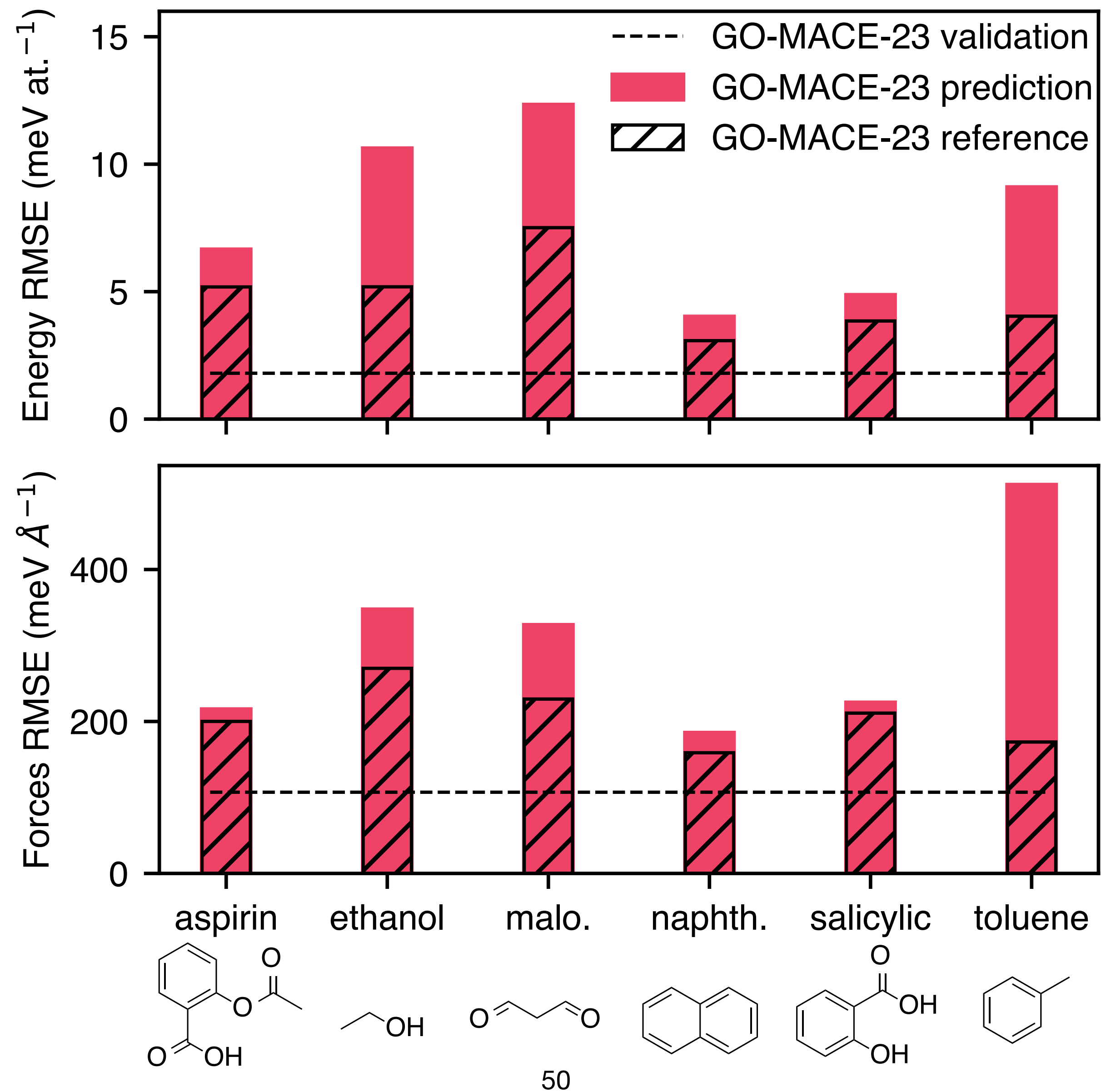
GO vs molecules from GO-MACE-23 perspective



GO vs molecules from GO-MACE-23 perspective

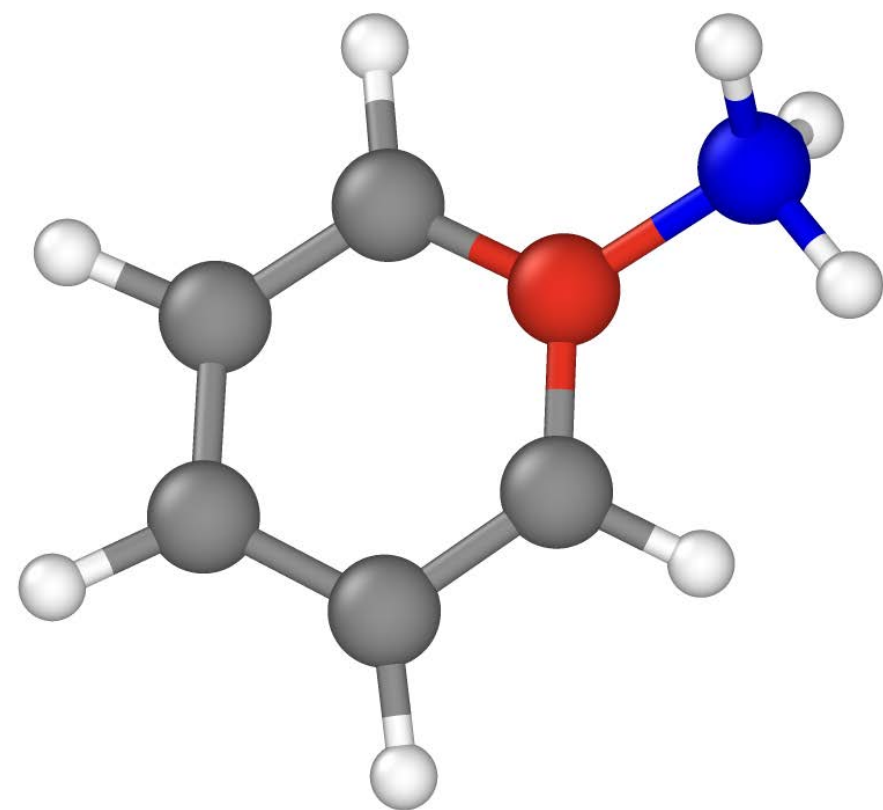


Performance on the rMD17 dataset

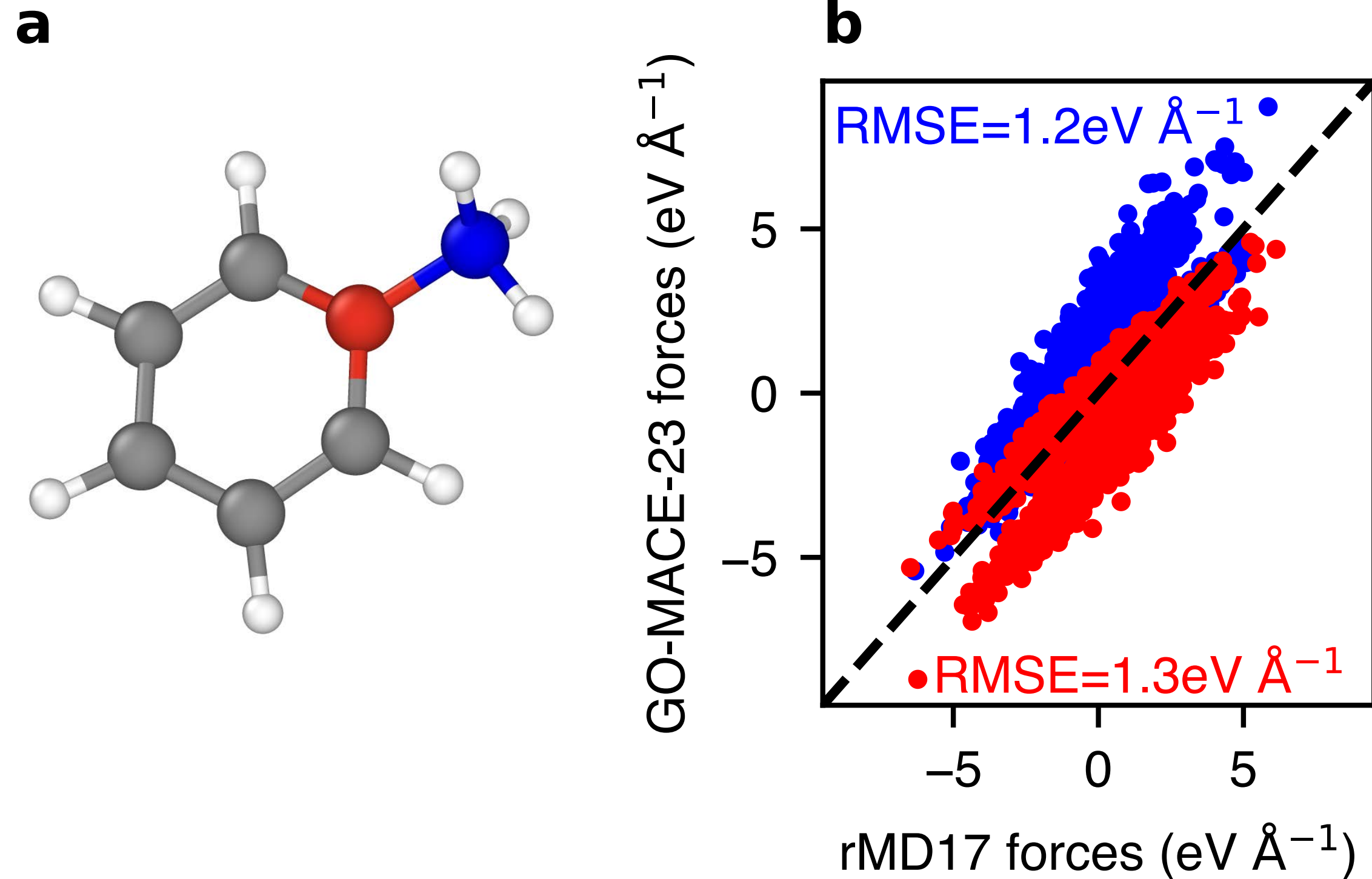


Why the toluene is unlearnable?

a

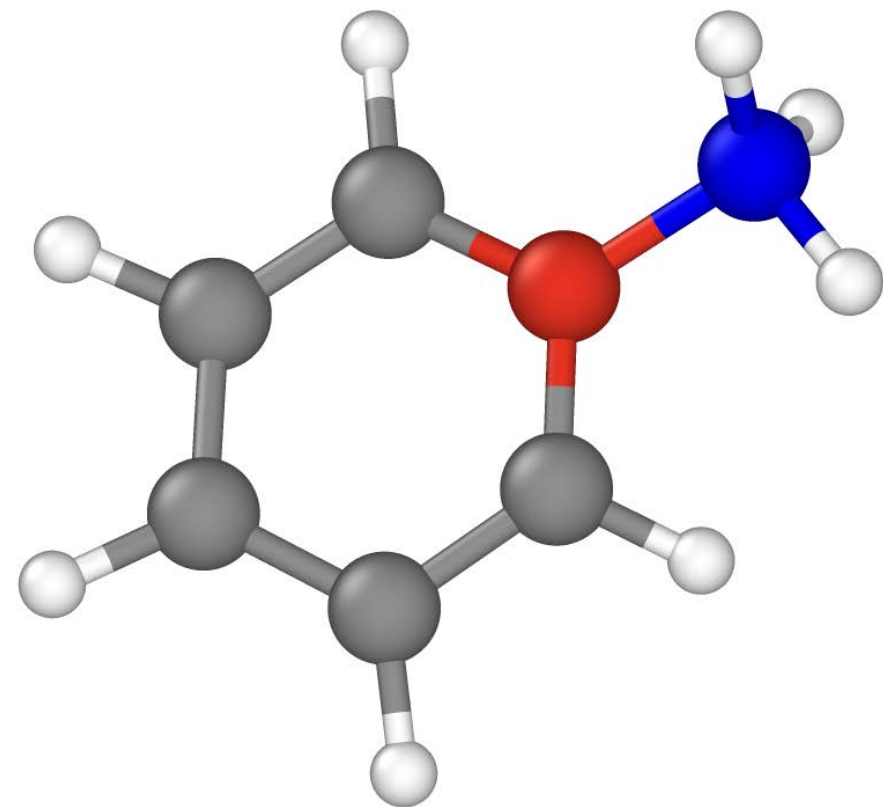


Why the toluene is unlearnable?

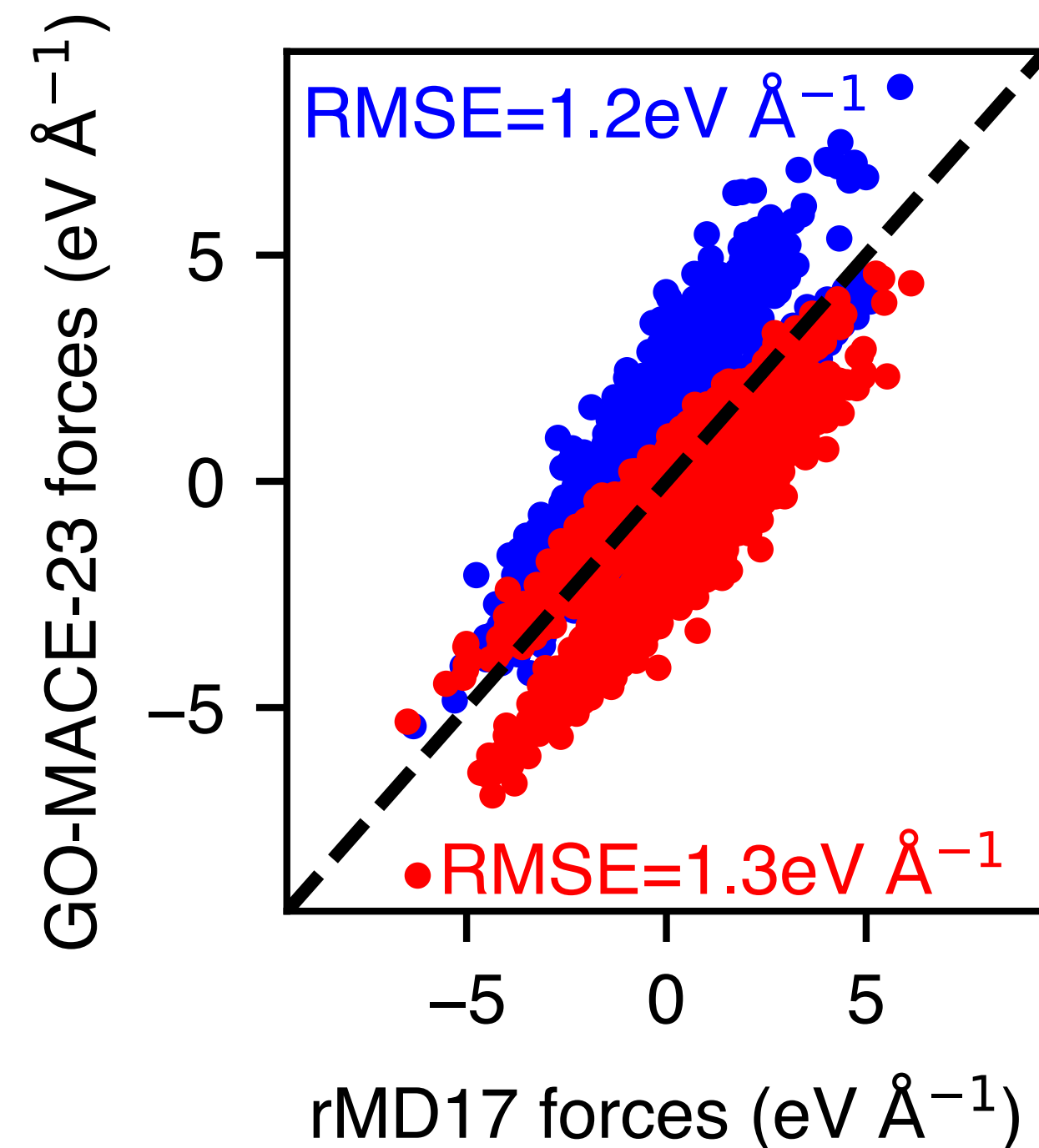


Why the toluene is unlearnable?

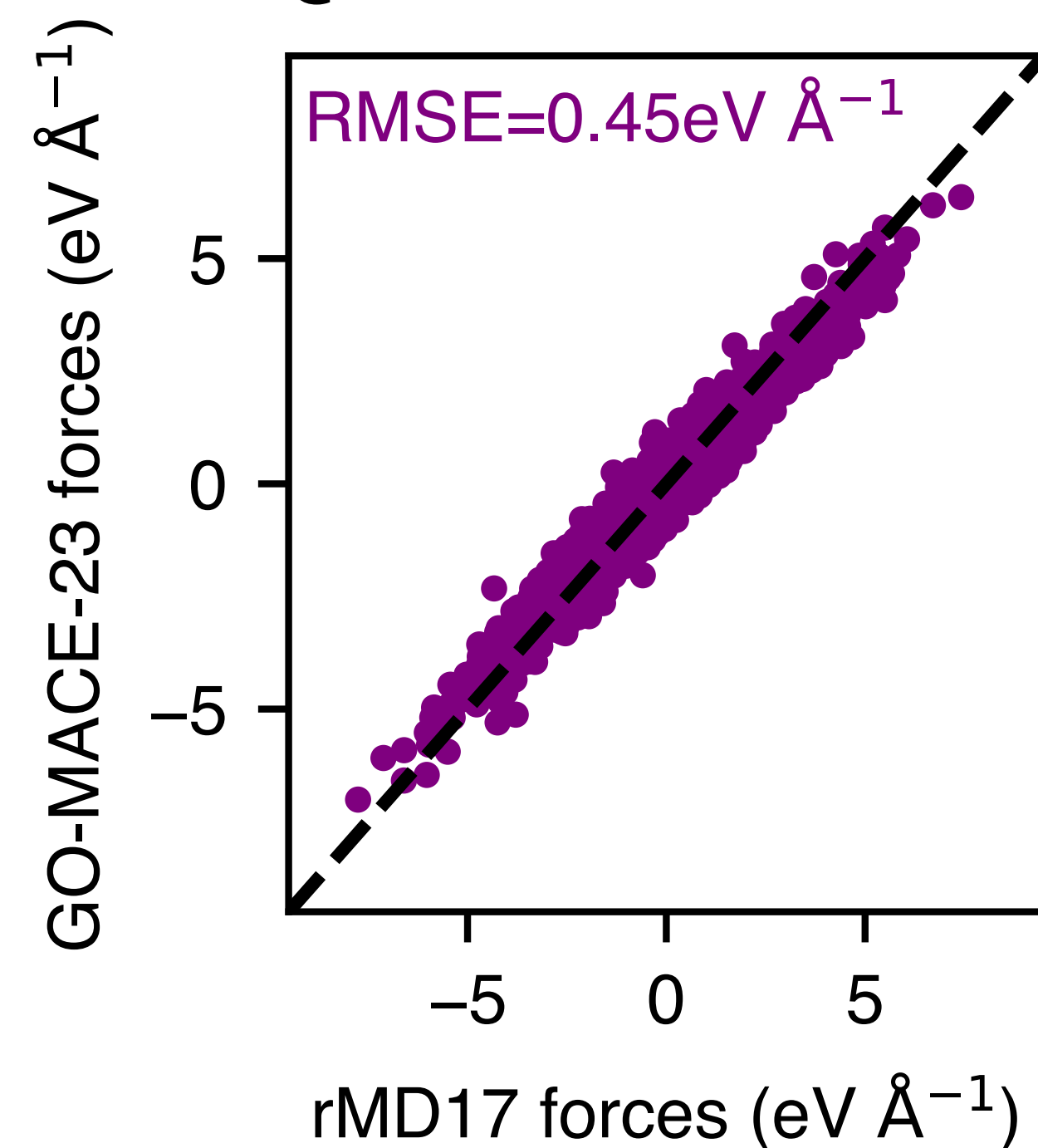
a



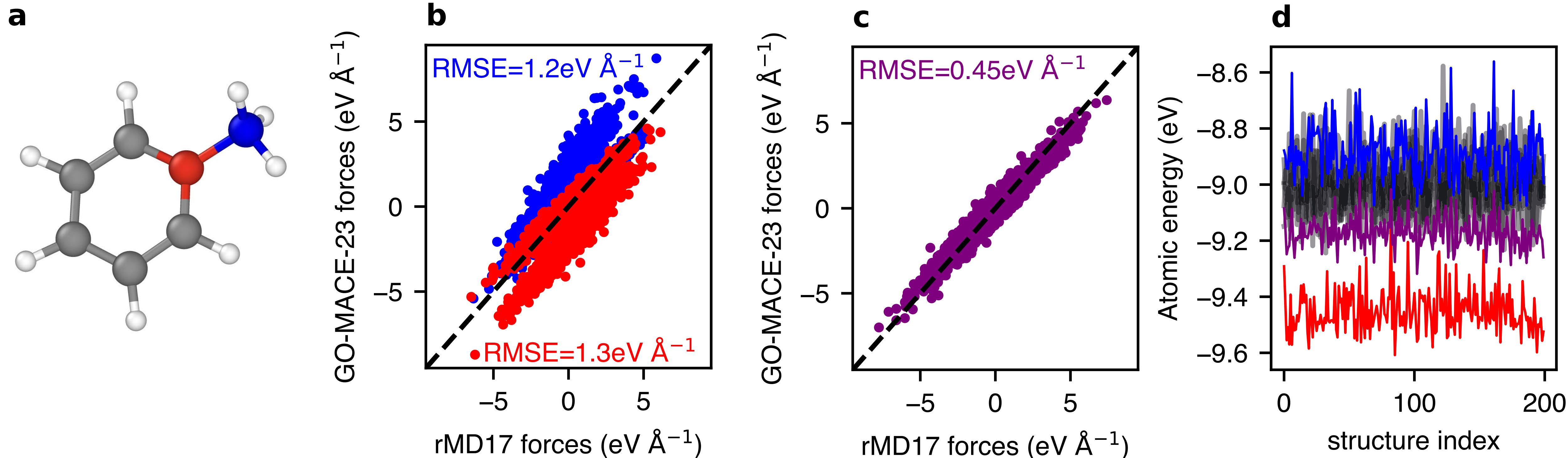
b



c



Why the toluene is unlearnable?

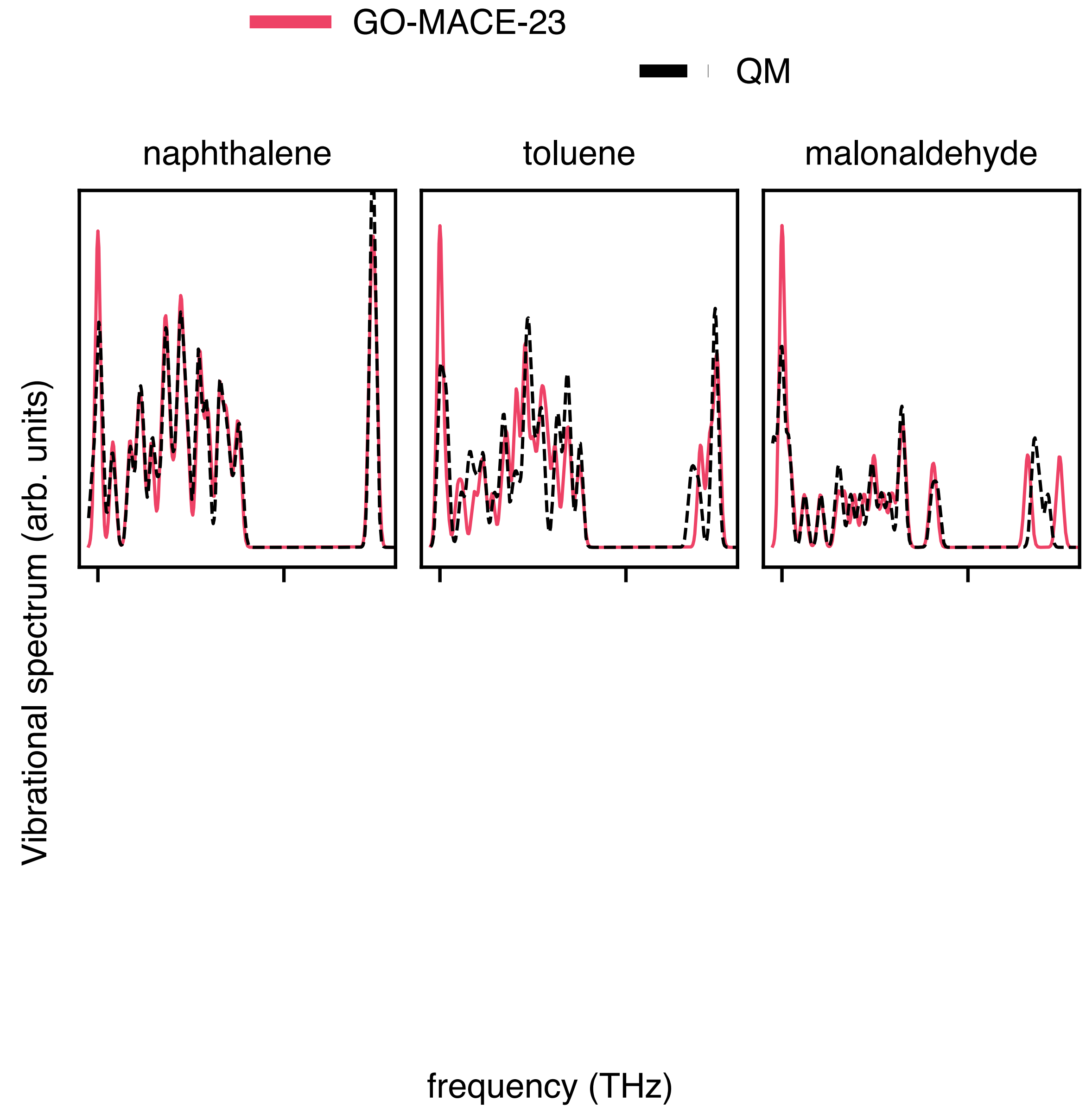


CBM, El-Machachi, Z., Gierczak, K.A., Gardner, J.L.A., Deringer, V.L., 2025.

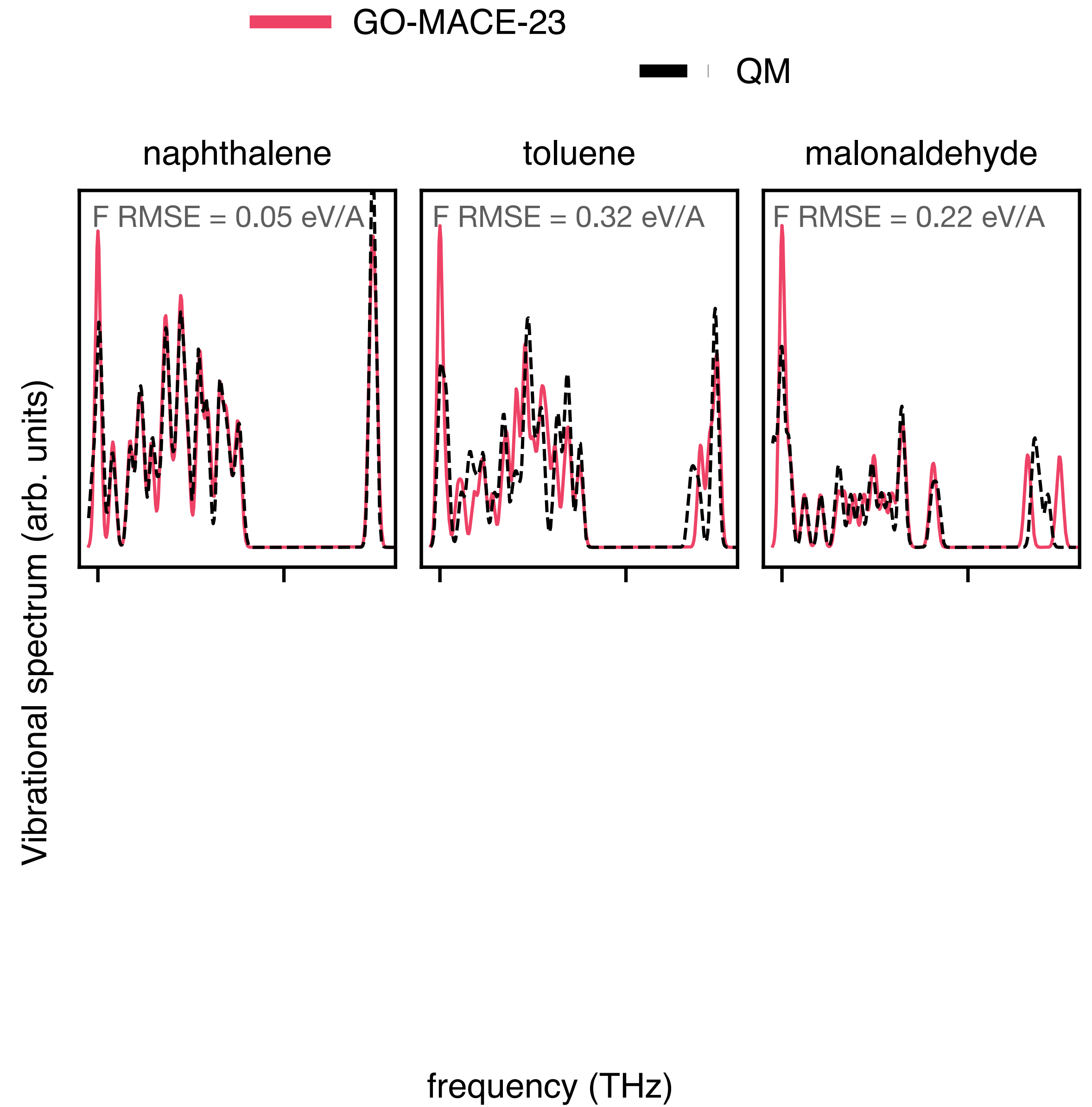
Assessing zero-shot generalisation behaviour in graph-neural-network interatomic potentials.

arXiv: 2502.21317

Vibrational spectra of molecules

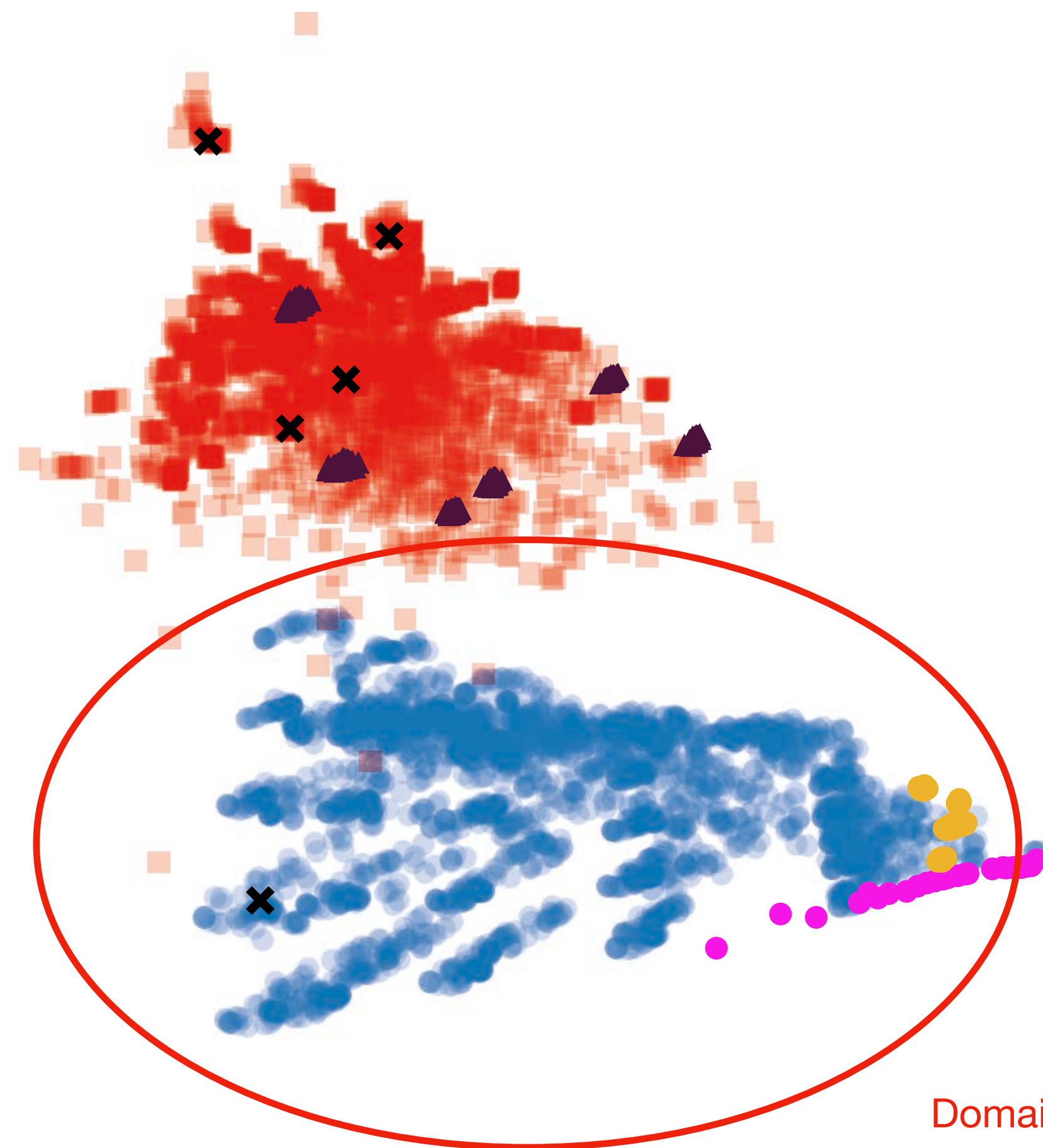


Vibrational spectra of molecules



Vibrational spectra of molecules

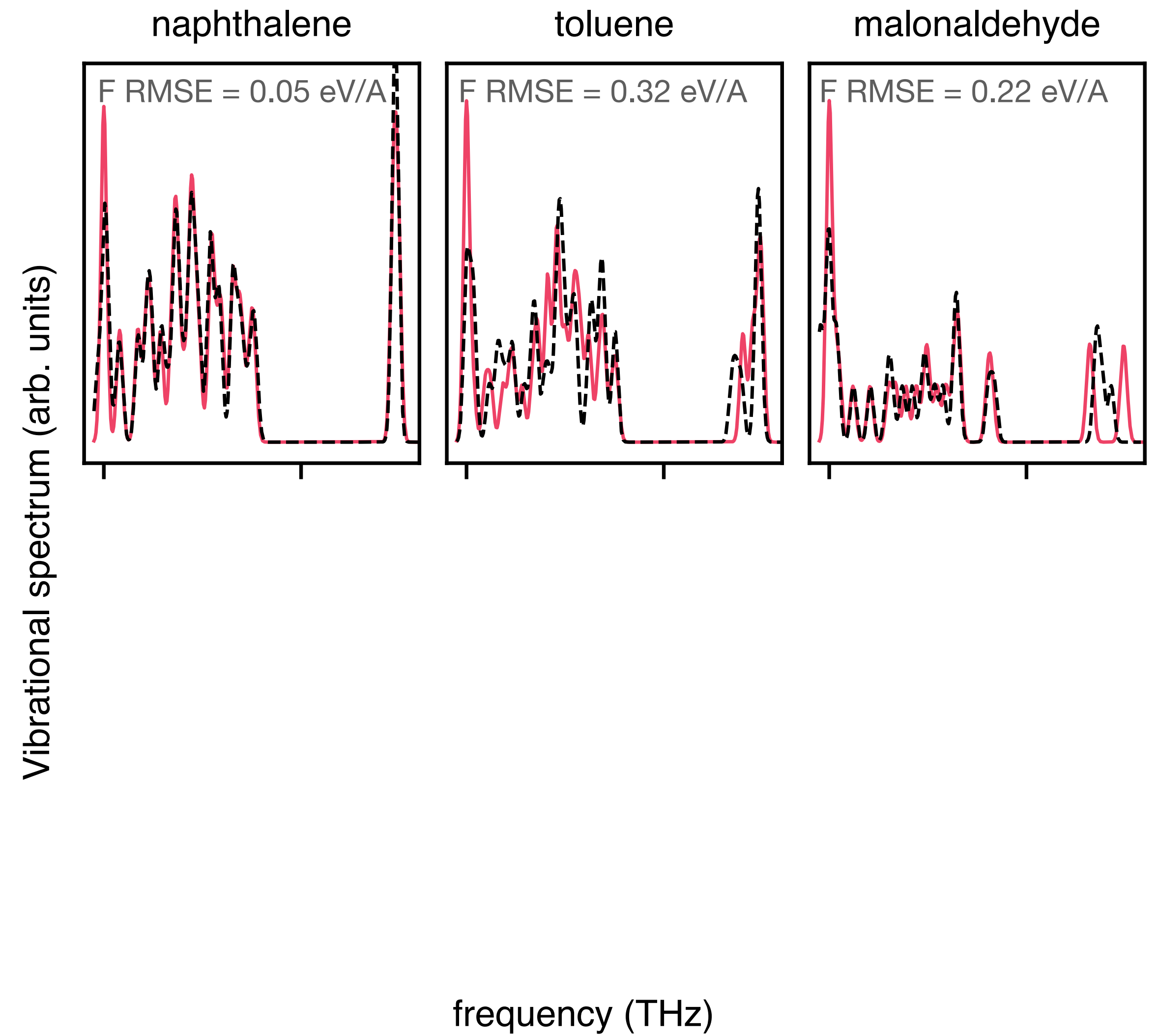
- GO dataset
- SPICE
- ▲ rMD17
- fullerenes
- M @ C₆₀
- × isolated molecules



Domain of GO-MACE-23

— GO-MACE-23

— QM



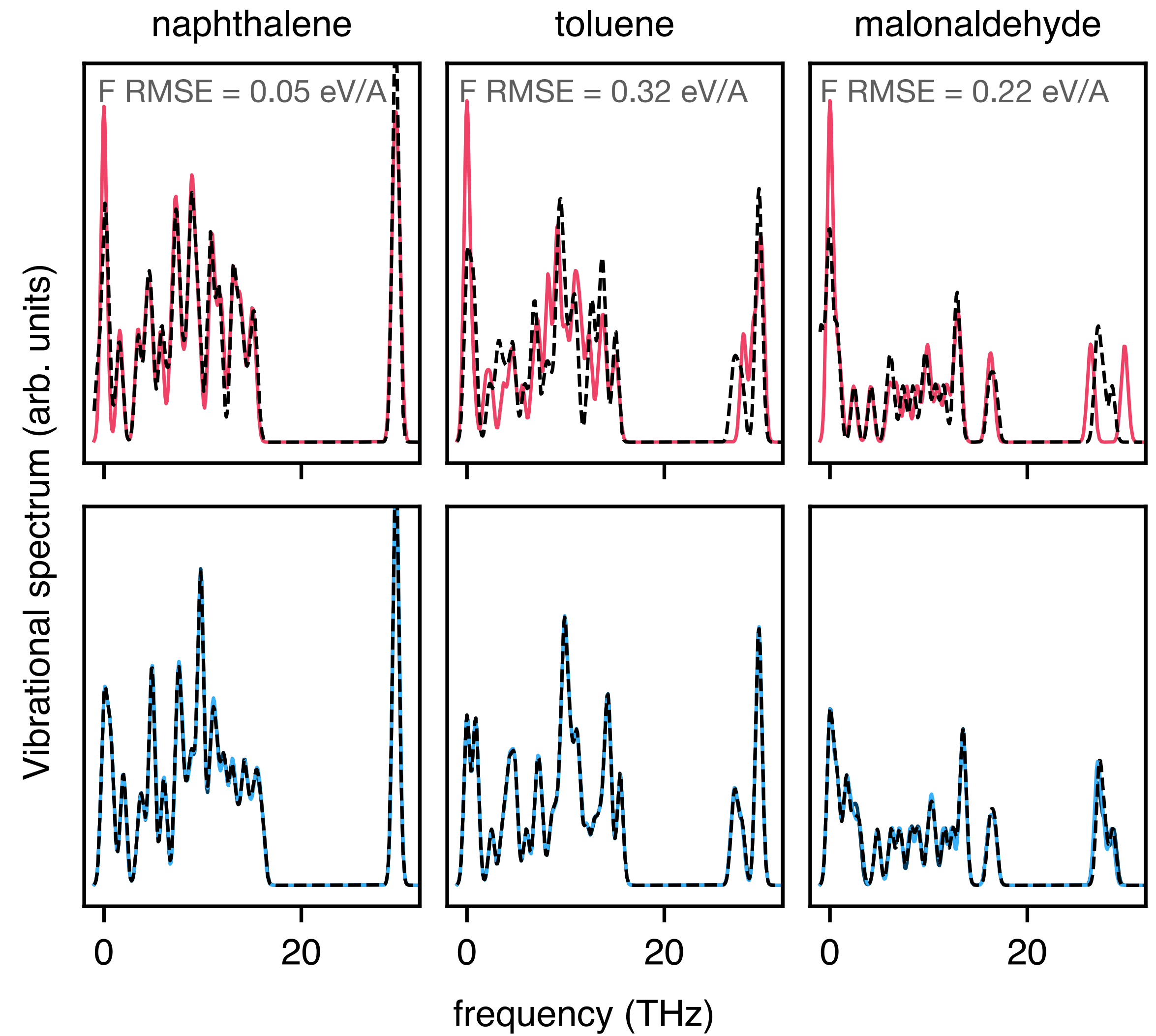
Vibrational spectra of molecules

- GO dataset
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- ▲ rMD17
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Domain of MACE-OFF family

Domain of GO-MACE-23

- GO-MACE-23
- MACE-OFF23
- MACE-OFF24
- QM



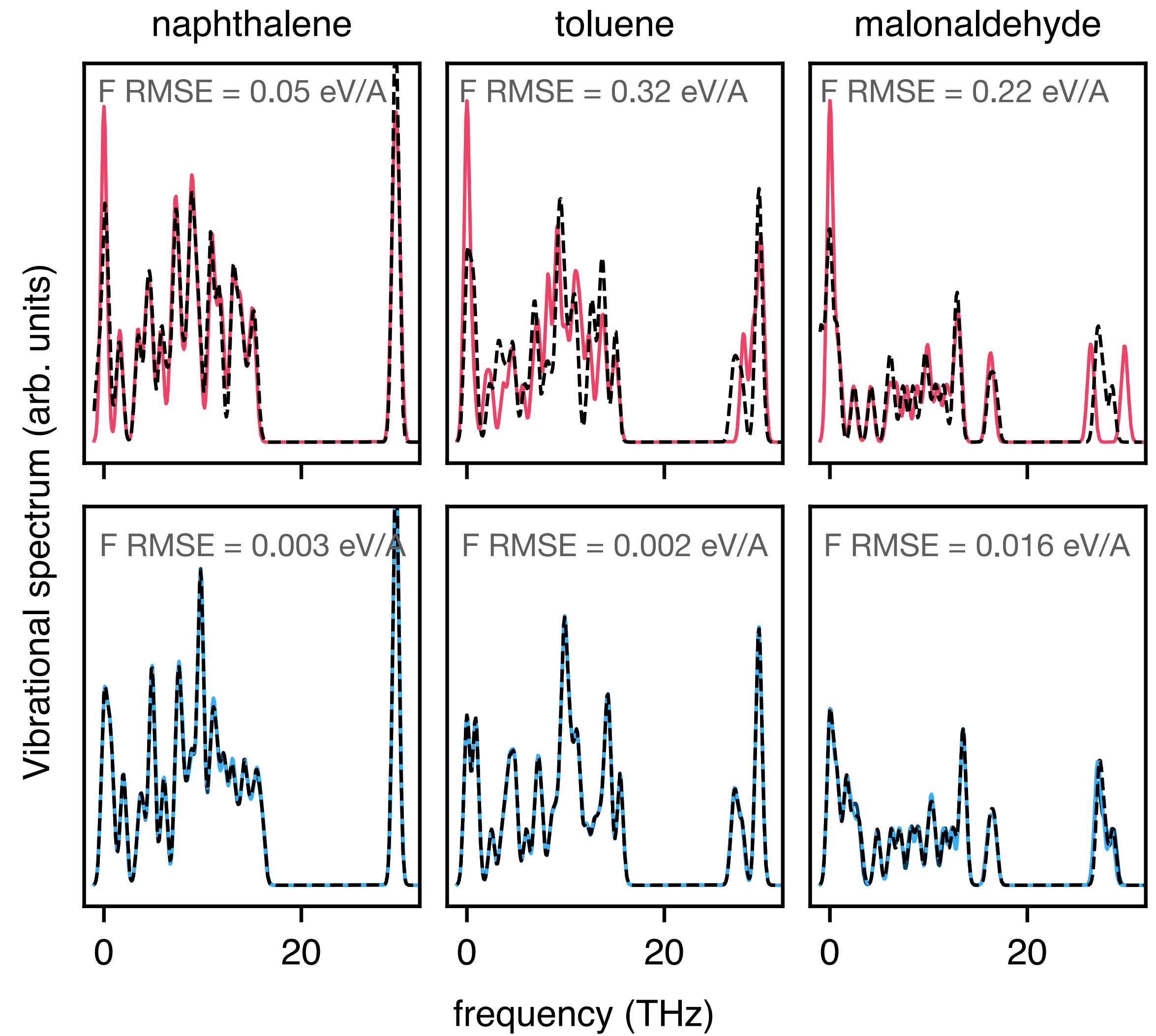
Vibrational spectra of molecules

- GO dataset
- SPICE
- ▲ rMD17
- fullerenes
- M @ C₆₀
- × isolated molecules

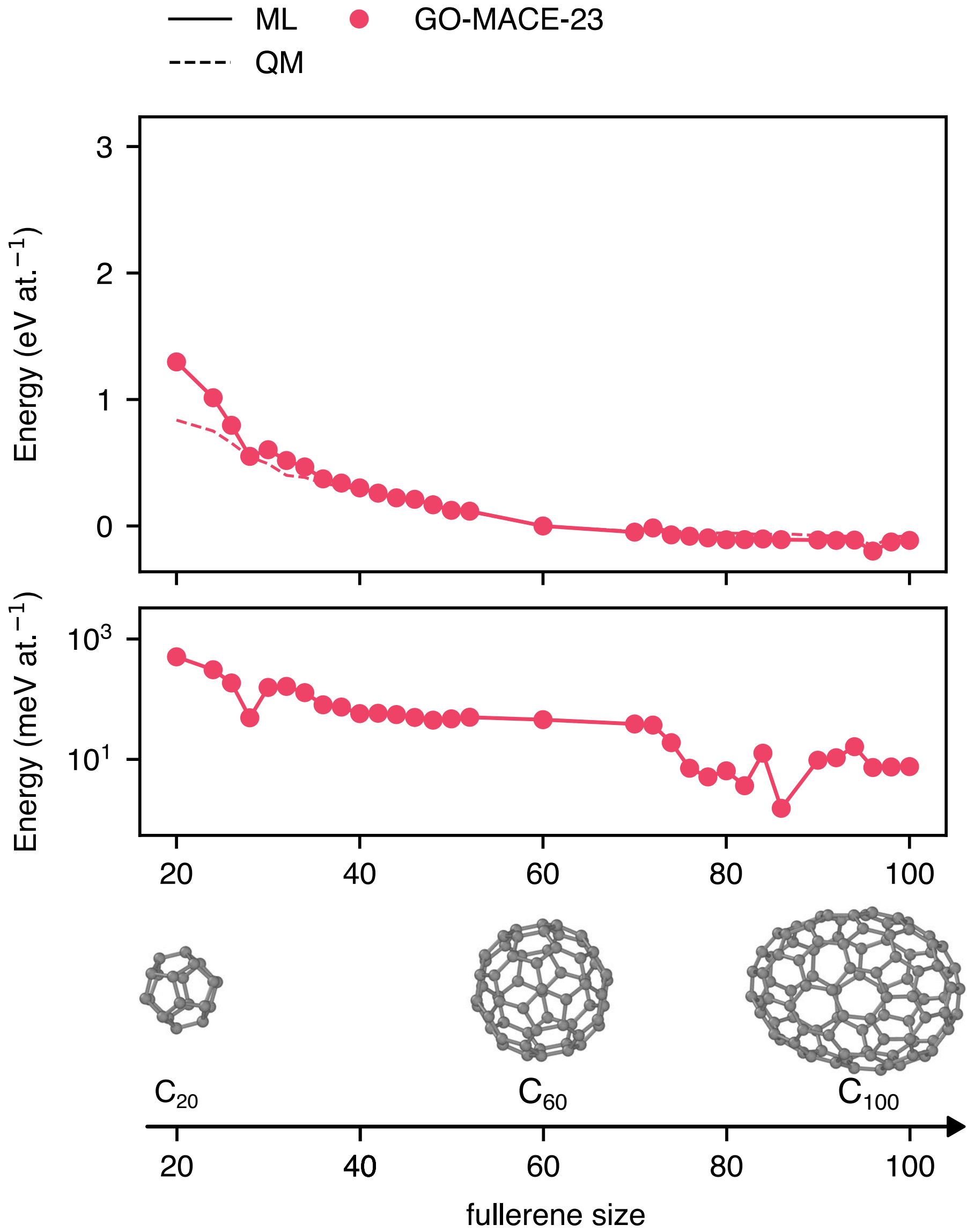
Domain of MACE-OFF family

Domain of GO-MACE-23

- GO-MACE-23
- MACE-OFF23
- MACE-OFF24
- QM

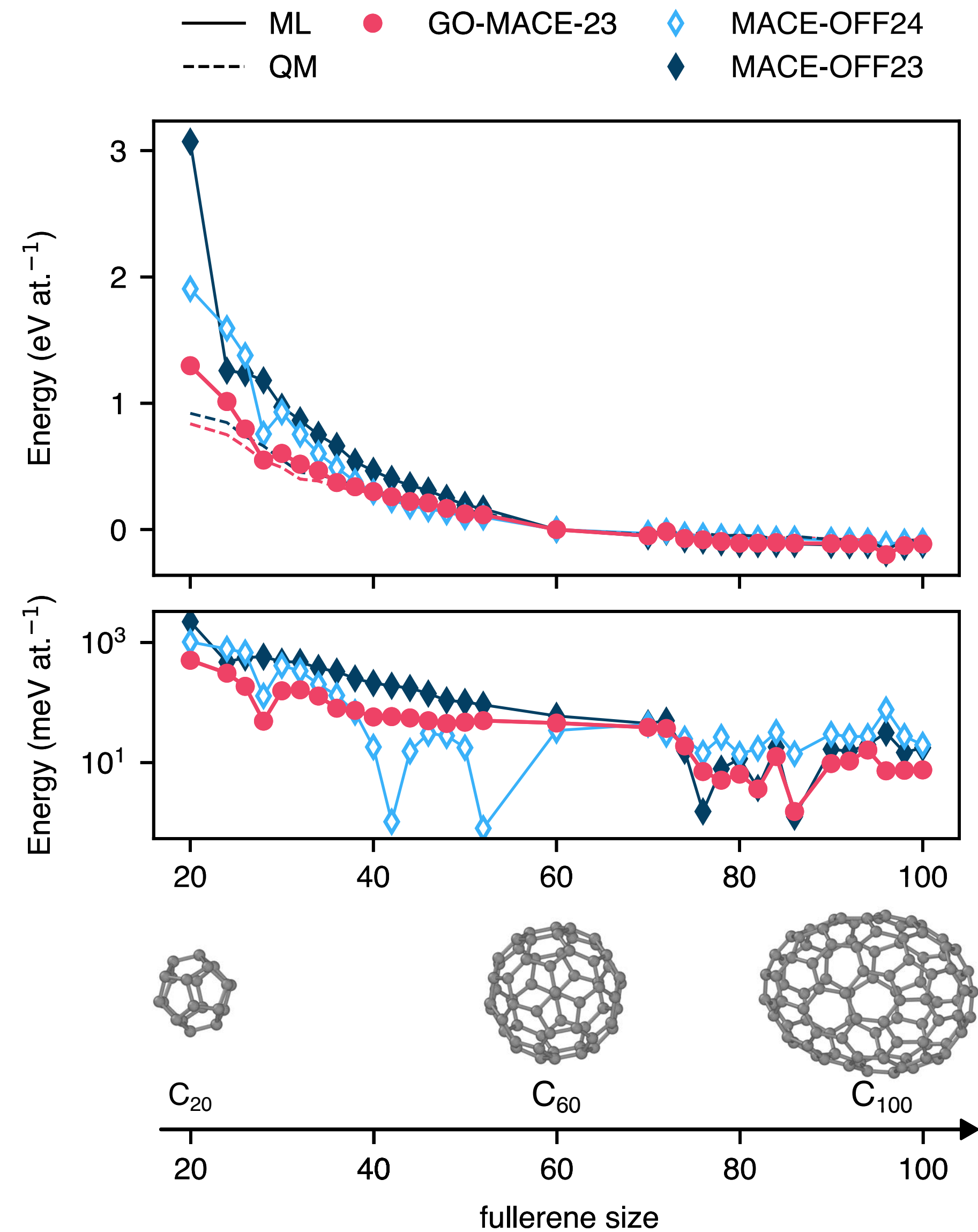


Fullerenes



CBM, El-Machachi, Z., Gierczak, K.A., Gardner, J.L.A., Deringer, V.L., 2025. Assessing zero-shot generalisation behaviour in graph-neural-network interatomic potentials. arXiv: 2502.21317

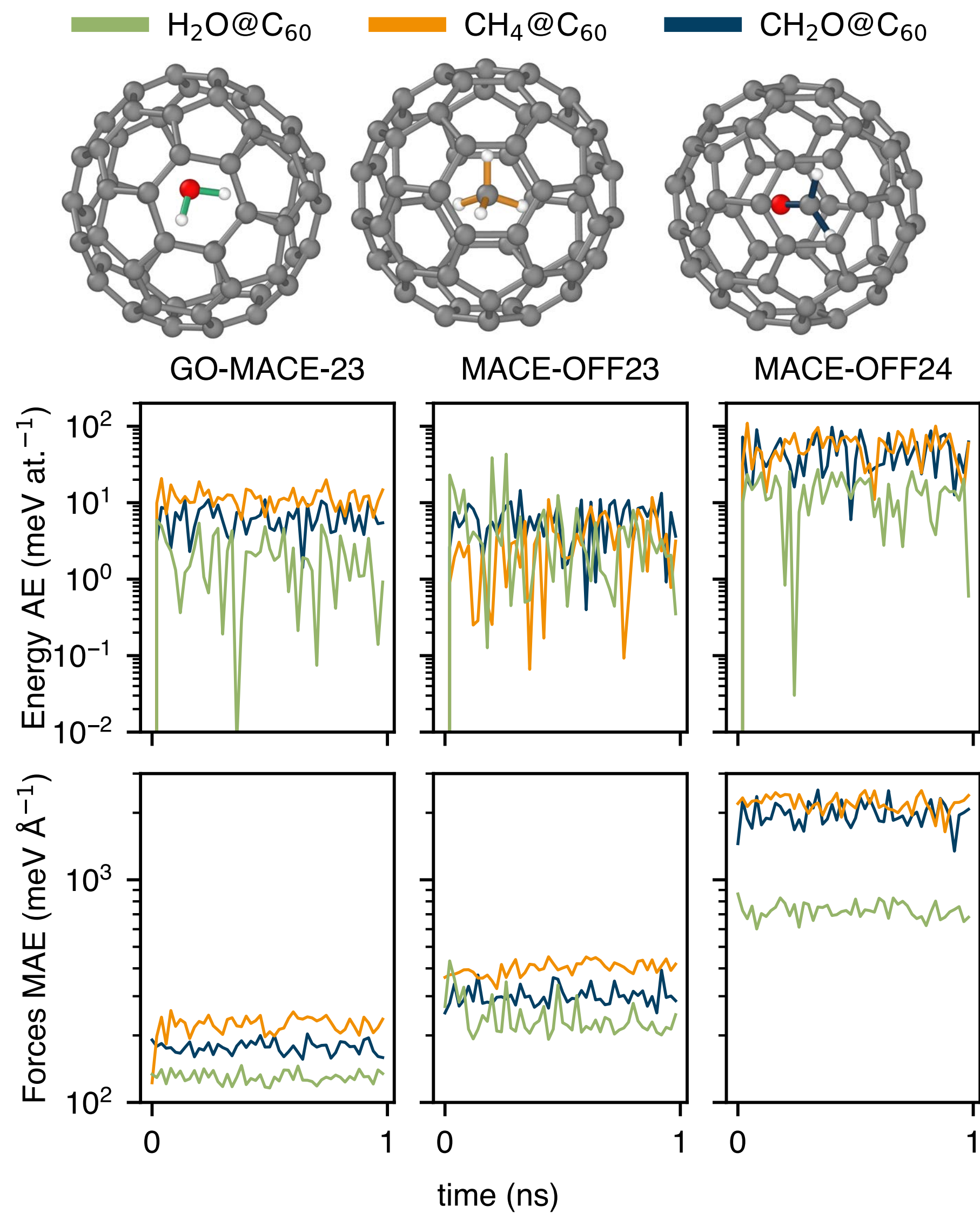
Fullerenes



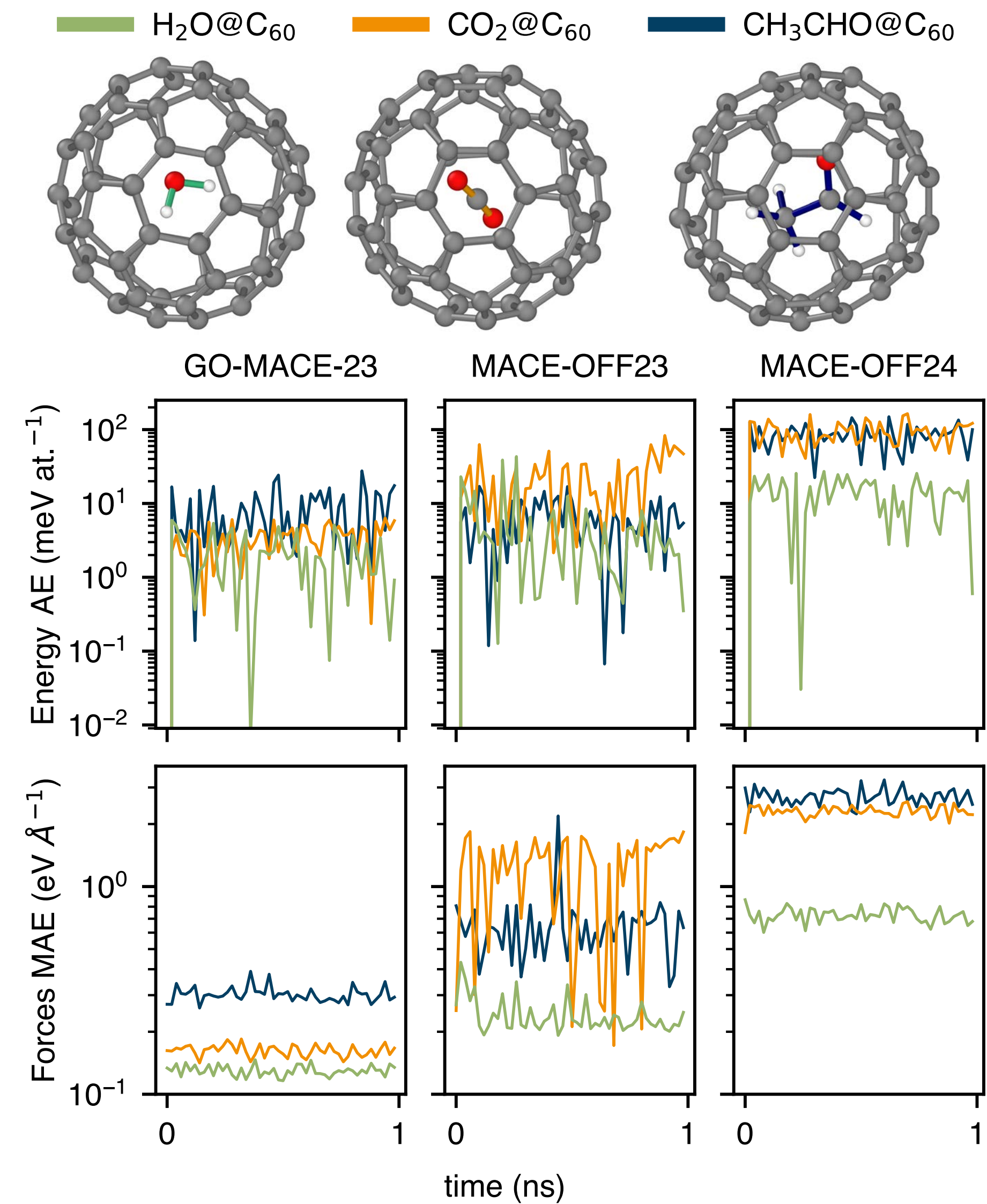
CBM, El-Machachi, Z., Gierczak, K.A., Gardner, J.L.A., Deringer, V.L., 2025. Assessing zero-shot generalisation behaviour in graph-neural-network interatomic potentials. arXiv: 2502.21317

Encapsulated reactions

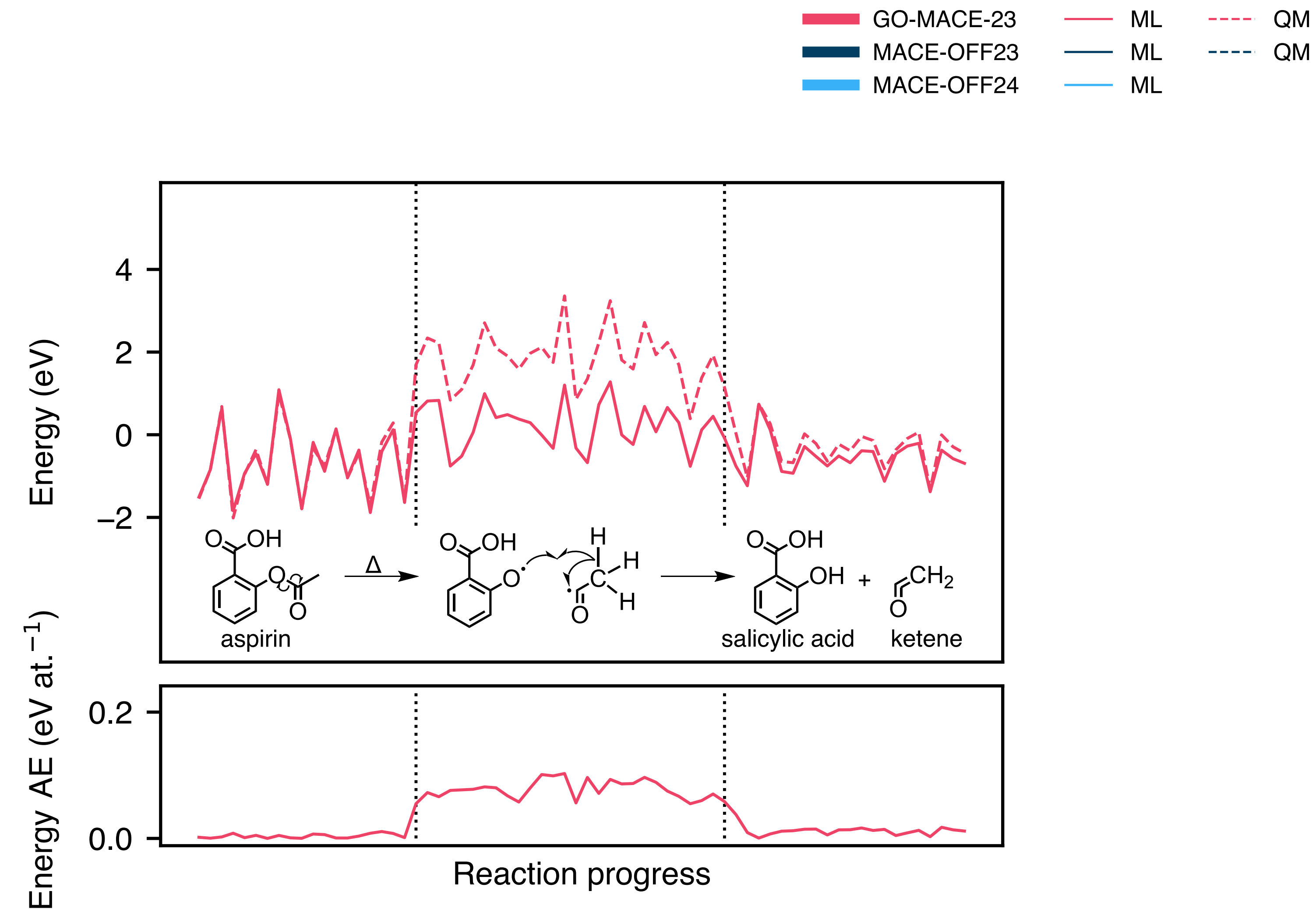
Reported in the literature



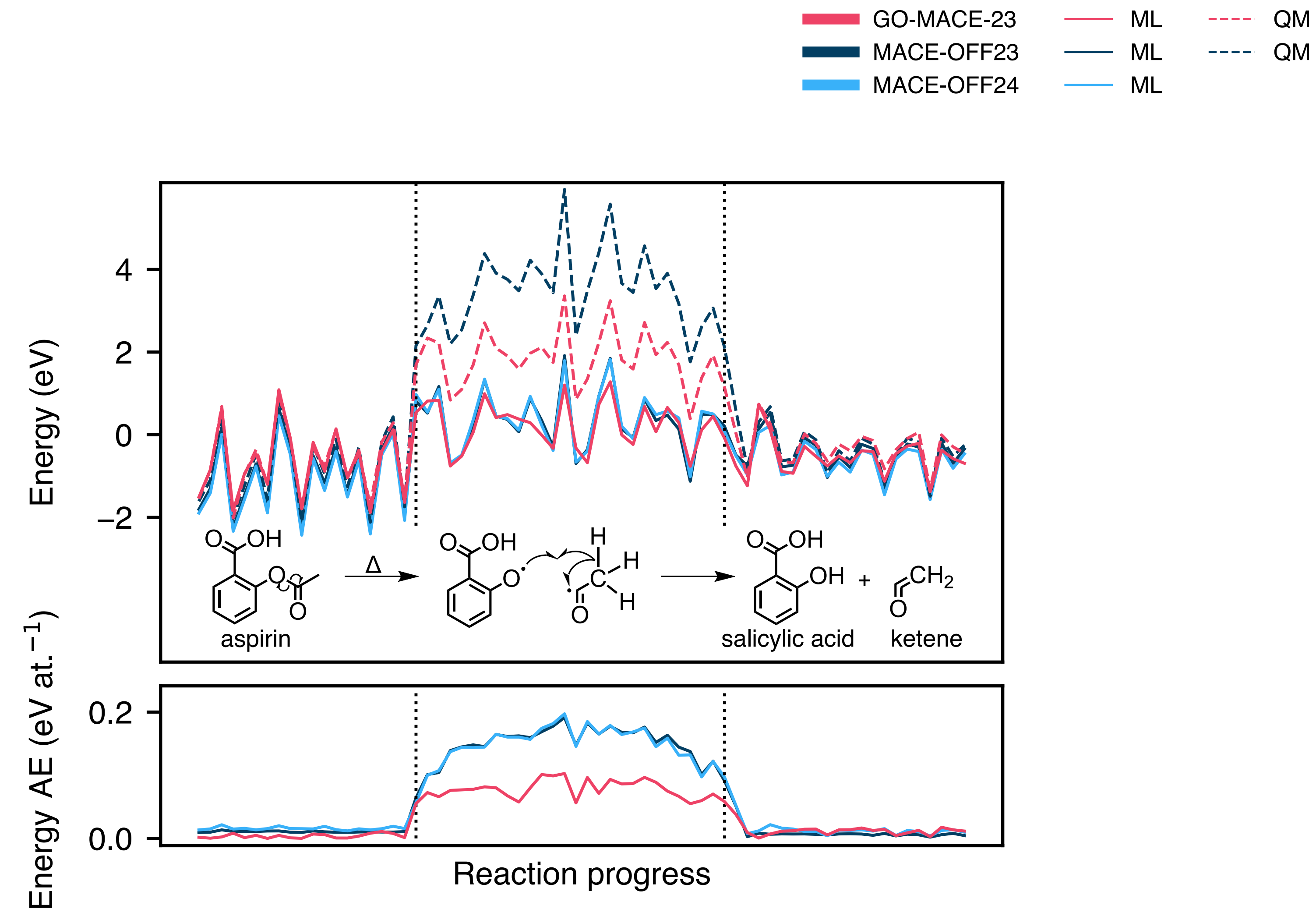
Yet to be reported in the literature



Transferability to reaction pathways

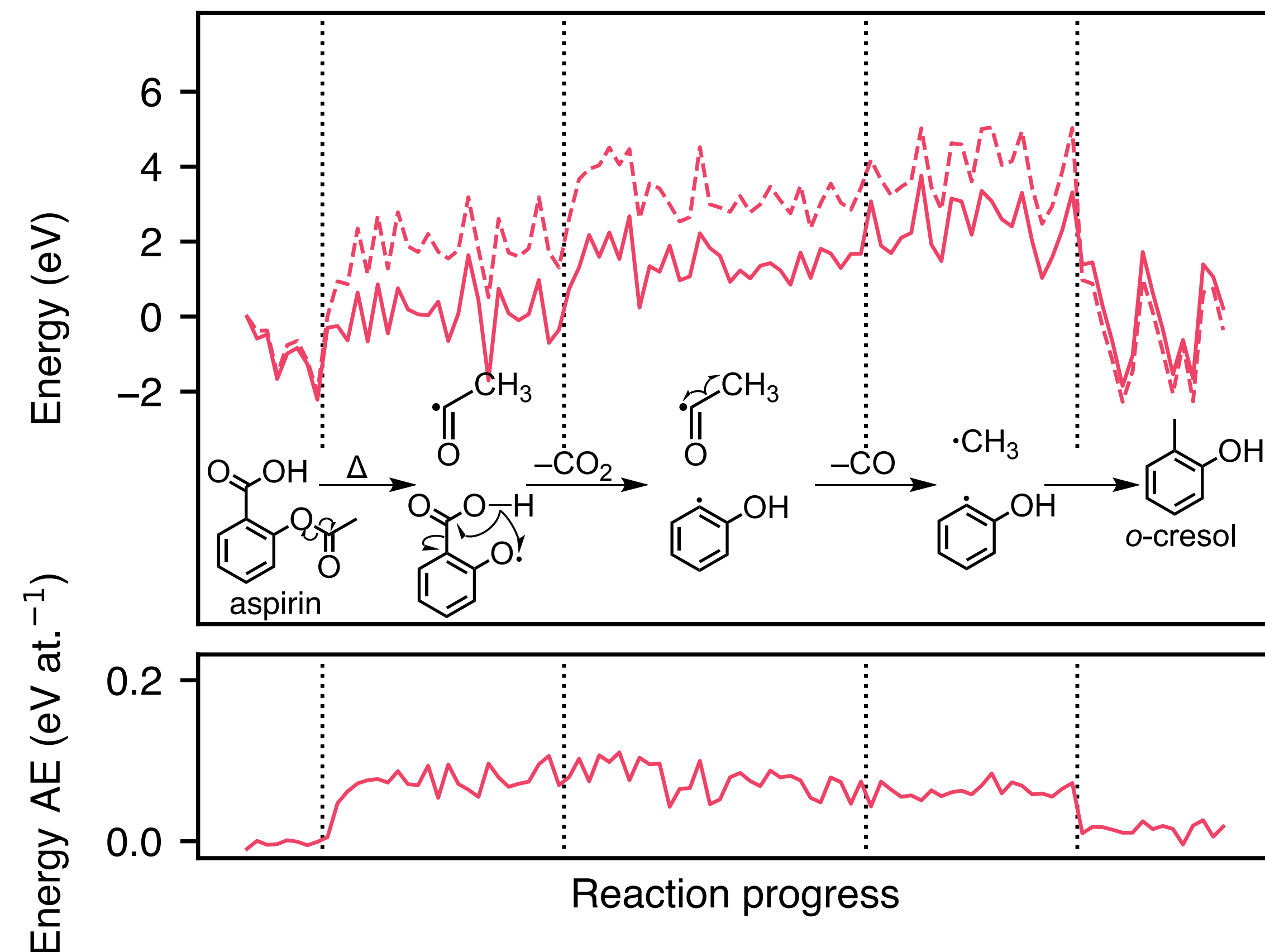
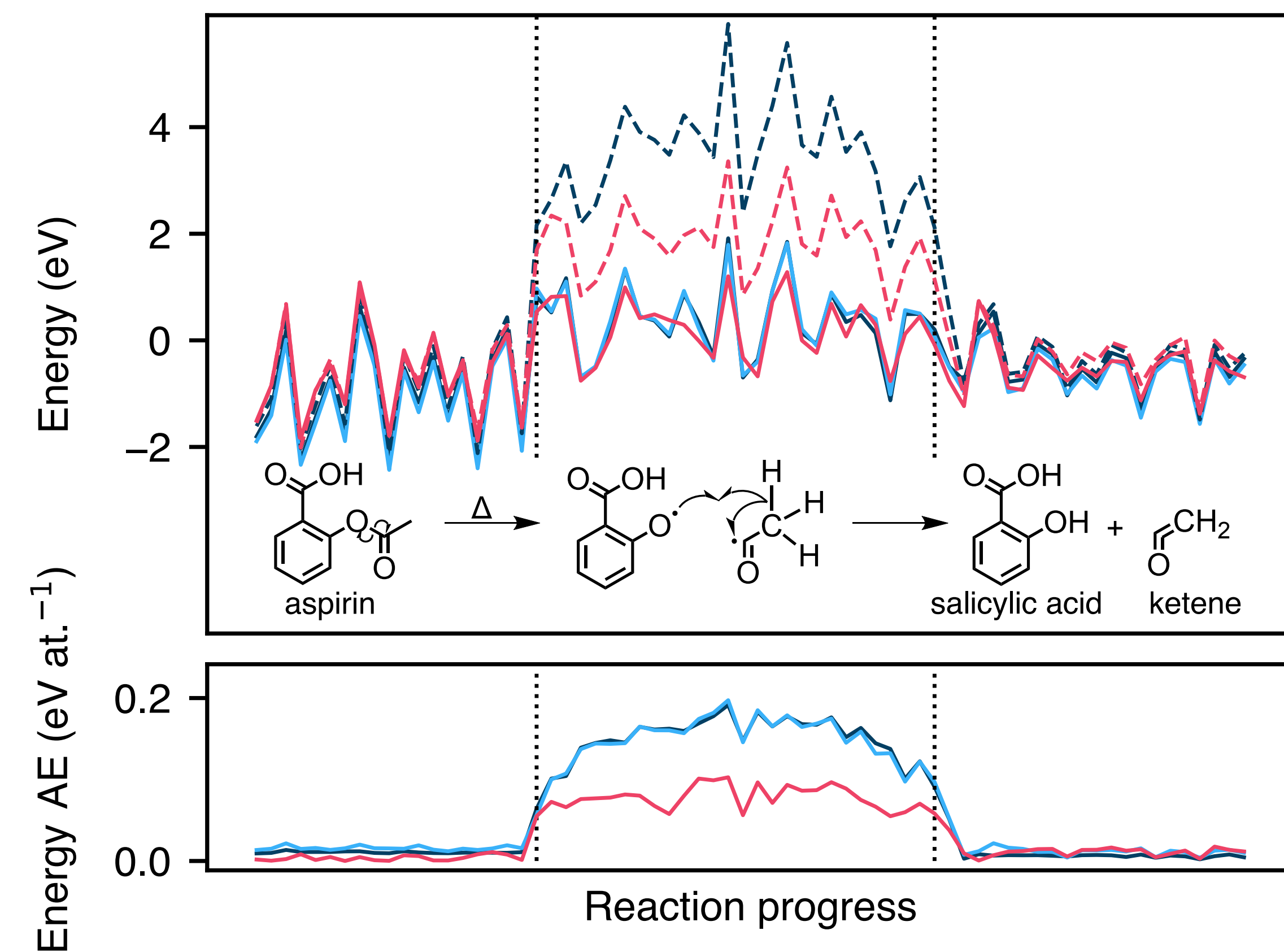


Transferability to reaction pathways



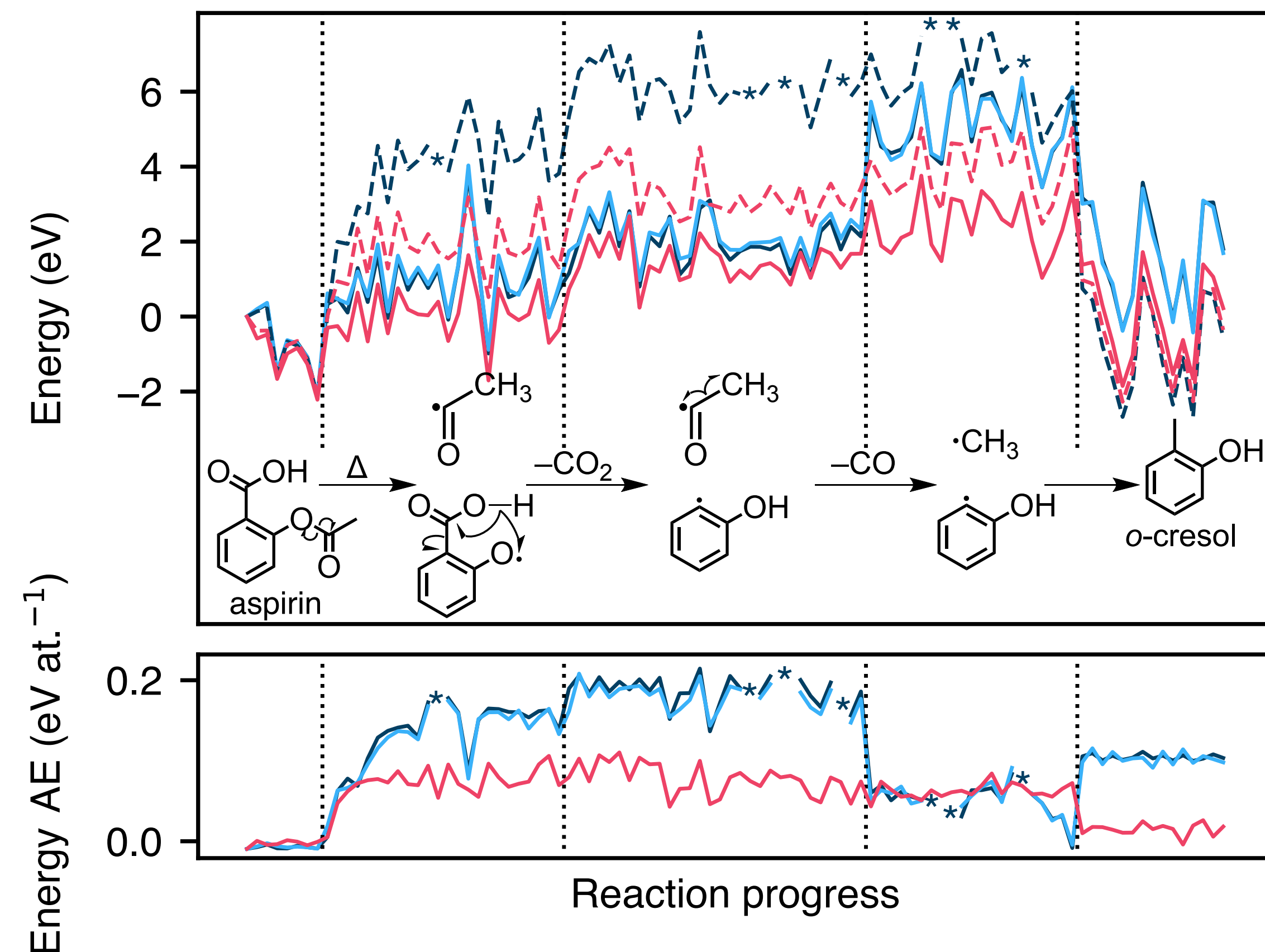
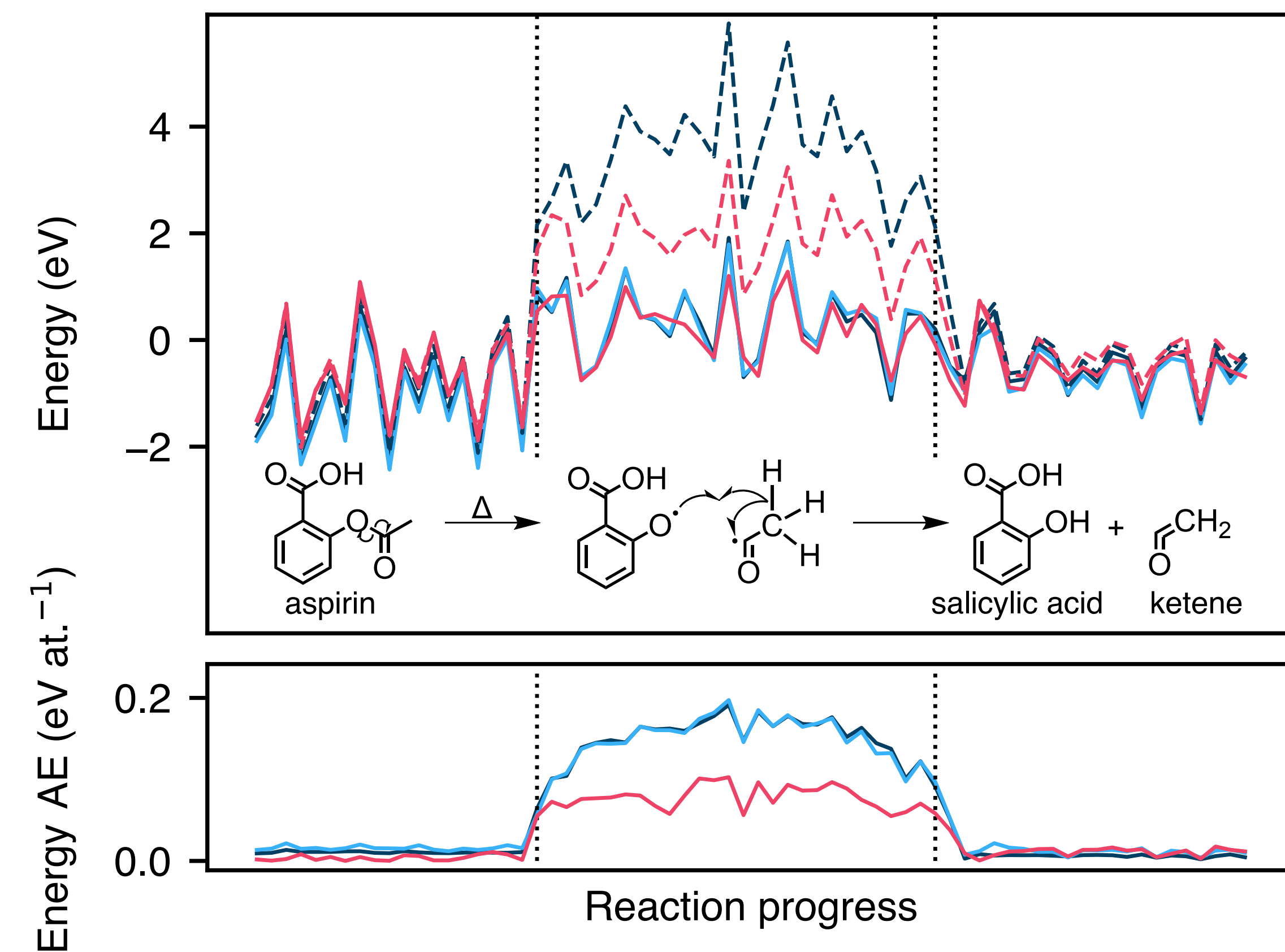
Transferability to reaction pathways

GO-MACE-23 ML QM
 MACE-OFF23 ML QM
 MACE-OFF24 ML

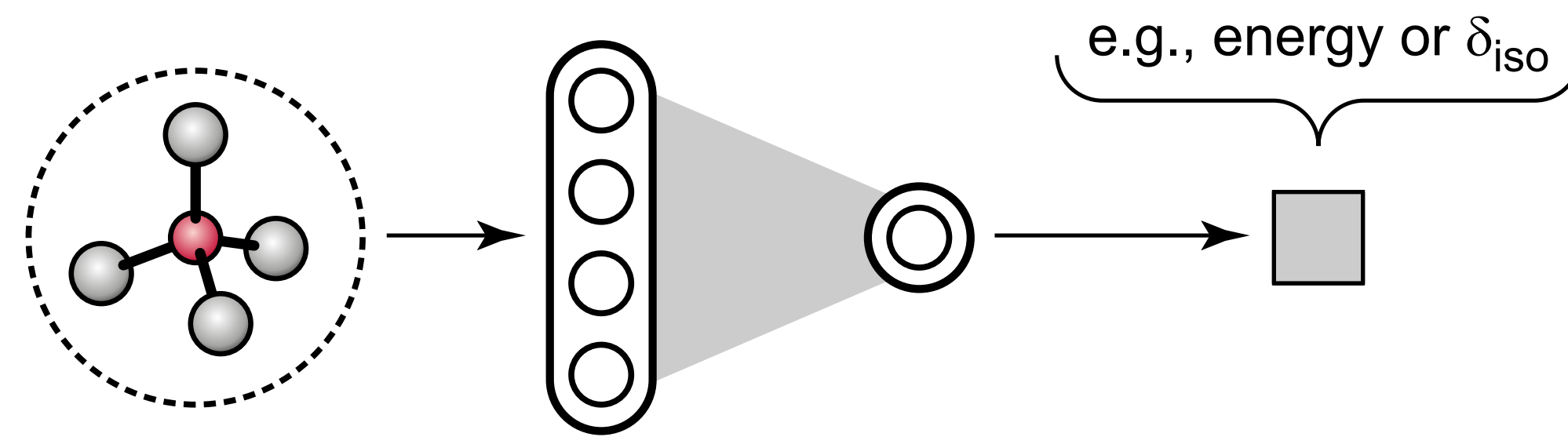


Transferability to reaction pathways

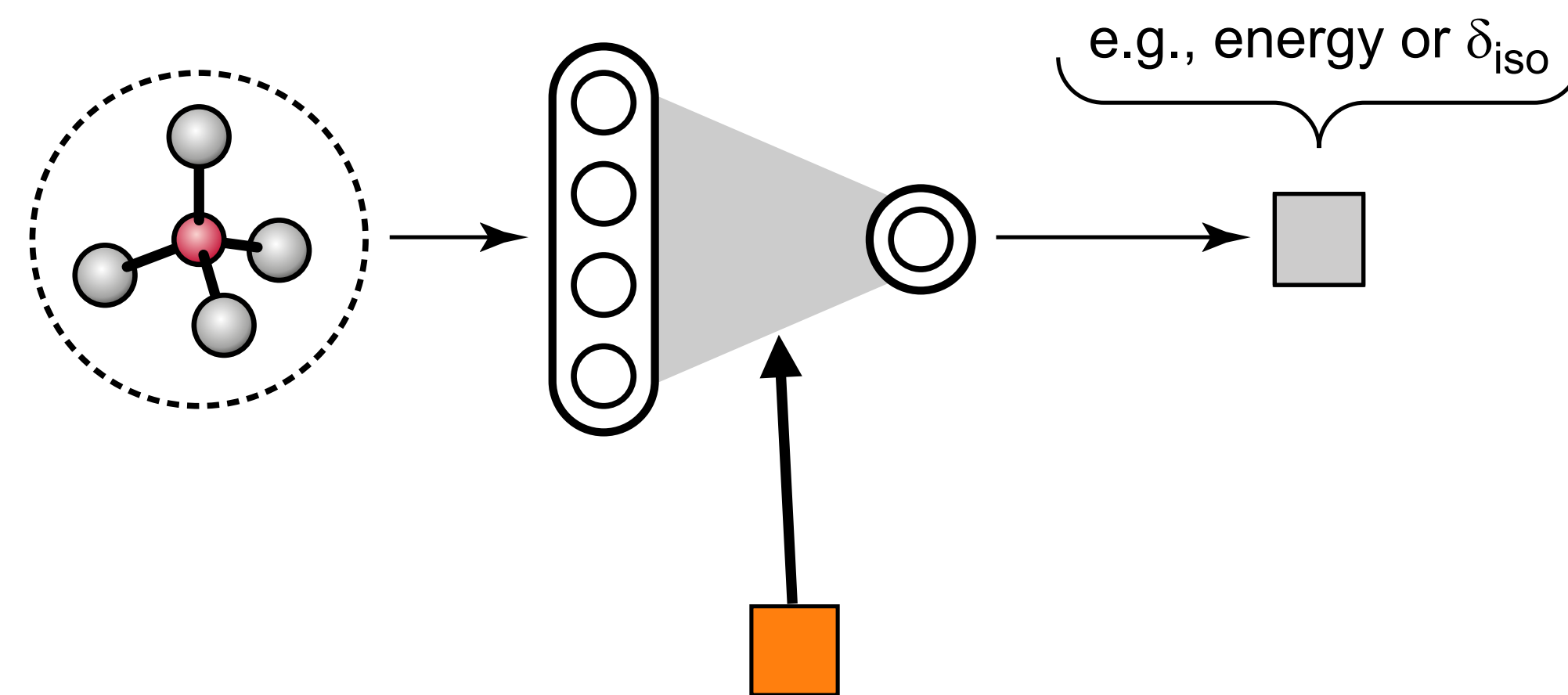
GO-MACE-23 ML QM
 MACE-OFF23 ML QM
 MACE-OFF24 ML QM



Does model architecture matter?

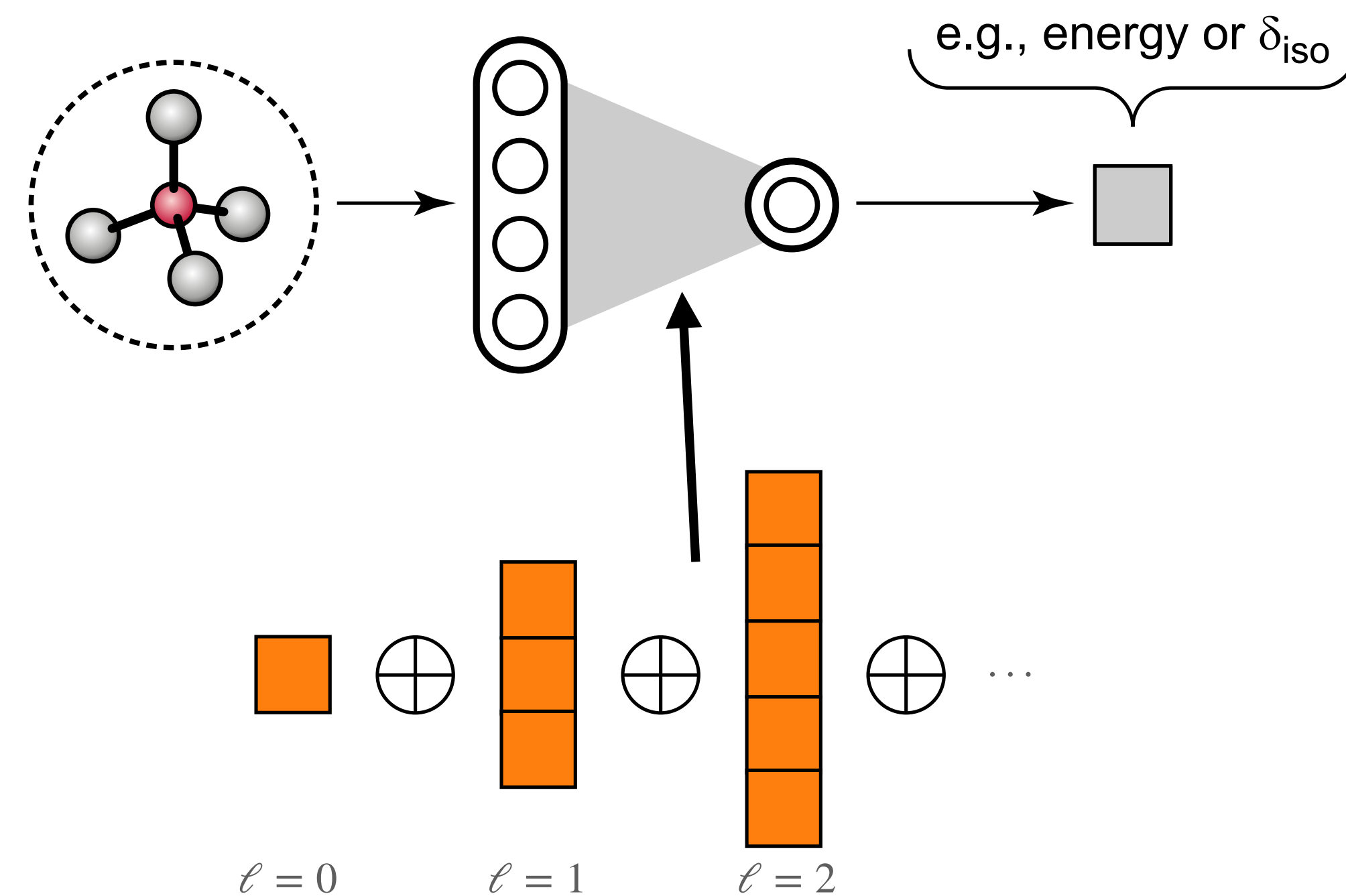


Does model architecture matter?



Invariant features (do not change upon structure rotation for example)

Does model architecture matter?



Equivariant features (do change upon structure rotation for example)

Does model architecture matter?

	Energy RMSE (meV at. ⁻¹)			Force RMSE (eV Å ⁻¹)		
max L	0	1	2	0	1	2
aspirin	6.2	6.6	4.9	0.25	0.22	0.28
ethanol	12.3	10.6	12.2	0.49	0.35	0.48
malonaldehyde	7.7	12.3	9.2	0.28	0.33	0.25
naphthalene	3.3	4.0	3.6	0.18	0.18	0.17
salicylic acid	5.3	4.9	6.8	0.22	0.22	0.26
toluene	5.6	9.1	6.9	0.32	0.51	0.25

CBM, El-Machachi, Z., Gierczak, K.A., Gardner, J.L.A., Deringer, V.L., 2025.
Assessing zero-shot generalisation behaviour in graph-neural-network interatomic potentials.
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Other MLIPs trained on GO

- MLIPs trained with **GraphPES** : a universal interface to train and fine-tune graph-neural network potentials
- GraphPES accessible on: <https://github.com/jla-gardner/graph-pes>
- **pip install graph-pes**

	Energy RMSE (meV at. ⁻¹)				
	GO-MACE-23	SchNet	TensorNet	NequIP	PaiNN
aspirin	6.6	22.4	6.6	5.7	11.3
ethanol	10.6	33.4	17.4	17.2	27.6
malonaldehyde	12.3	38.5	10.8	8.8	17.2
naphthalene	4.0	9.9	5.0	3.9	5.8
salicylic	4.9	19.7	5.6	3.9	7.4
toluene	9.1	16.8	8.7	24.0	14.0

	Force RMSE (eV Å ⁻¹)				
	GO-MACE-23	SchNet	TensorNet	NequIP	PaiNN
aspirin	0.22	0.86	0.38	0.31	0.57
ethanol	0.35	1.13	0.61	0.47	1.01
malonaldehyde	0.33	0.98	0.34	0.33	0.38
naphthalene	0.18	0.54	0.21	0.21	0.30
salicylic	0.22	0.42	0.24	0.19	0.25
toluene	0.51	0.59	0.28	0.38	0.32

CBM, El-Machachi, Z., Gierczak, K.A., Gardner, J.L.A., Deringer, V.L., 2025.
Assessing zero-shot generalisation behaviour in graph-neural-network interatomic potentials.
arXiv: 2502.21317

Takehome message

- Systematic assessment of out-of-domain performance is essential for MLIPs
- Structural variability is as important as chemical diversity when training general-purpose MLIPs for chemical reactions

Conclusions

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- ML tools are essential in atomistic modelling

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- Data is central to atomistic ML

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- The emergence of pretrained/foundation models will close the gap between material and molecular modelling

Conclusions

- ML tools are essential in atomistic modelling
- Data is central to atomistic ML
- The emergence of pretrained/foundation models will close the gap between material and molecular modelling
- Universal models describing structural, electronic, and spectroscopic fingerprints within physics-based approximations can enable more accurate and predictive materials modelling and design

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- Krystian Gierczak

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